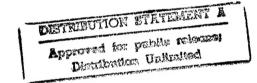


Peroxone Groundwater Treatment Demonstration

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Prepared by:

Montgomery Watson 4525 Wasatch Blvd., Suite 200 Salt Lake City, Utah 94124

TRW, Inc. Space & Technology Division One Space Park Redondo Beach, CA 90278

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TECHNICAL REPORT

PEROXONE GROUNDWATER TREATMENT DEMONSTRATION PROGRAM CORNHUSKER ARMY AMMUNITION PLANT GRAND ISLAND, NEBRASKA

February, 1998

TRW Space & Technology Division One Space Park Redondo Beach, CA 90278

Project No.: 1166031.01091017

Montgomery Watson 4525 South Wasatch Blvd., Suite 200 Salt Lake City, Utah 94124 (801) 272-1900

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ABBREVIATIONS & ACRONYMS

BGS Below Ground Surface

CAAP Cornhusker Army Ammunition Plant

COE U.S. Army Corps of Engineers

CP Control Panel

CSTR Continuously Stirred Tank Reactor
DESA Defense Evaluation Support Activity

EBCT Empty Bed Contact Time

EPA U.S. Environmental Protection Agency

FPM Feet Per Minute

GAC Granular Activated Carbon

GPM Gallons Per Minute

HDPE High density polyethylene

HOA Hand-Off-Auto HP Horse Power

HRT Hydraulic Retention Time

lb Pound

LOX Liquid Oxygen mA Milli Amp

mg/l Milligram per liter

NDEQ Nebraska Department of Environmental Quality

O&M Operations and Maintenance ORP Oxidation-Reduction Potential

PFD Process Flow Diagram
PSA Pressure Swing Absorption

RDX Hexahydro-1,3,5-trinitro-1,3,5-triazine

RO Reverse Osmosis

SCFH Standard Cubic Feet Per Hour SCFM Standard Cubic Feet Per Minute SOC Synthetic Organic Chemical

SOW Statement of Work SS Stainless Steel

TDH Total Dynamic Head
TNT 2,4,6-trinitrotoluene
TNB 1,3,5-trinitrobenzene
TOC Total Organic Carbon

USAEC U.S. Army Environmental Center VOC Volatile organic compounds VSA Vacuum Swing Absorption

WES Waterways Experiment Station

μg/l Microgram per liter

1.0 INTRODUCTION

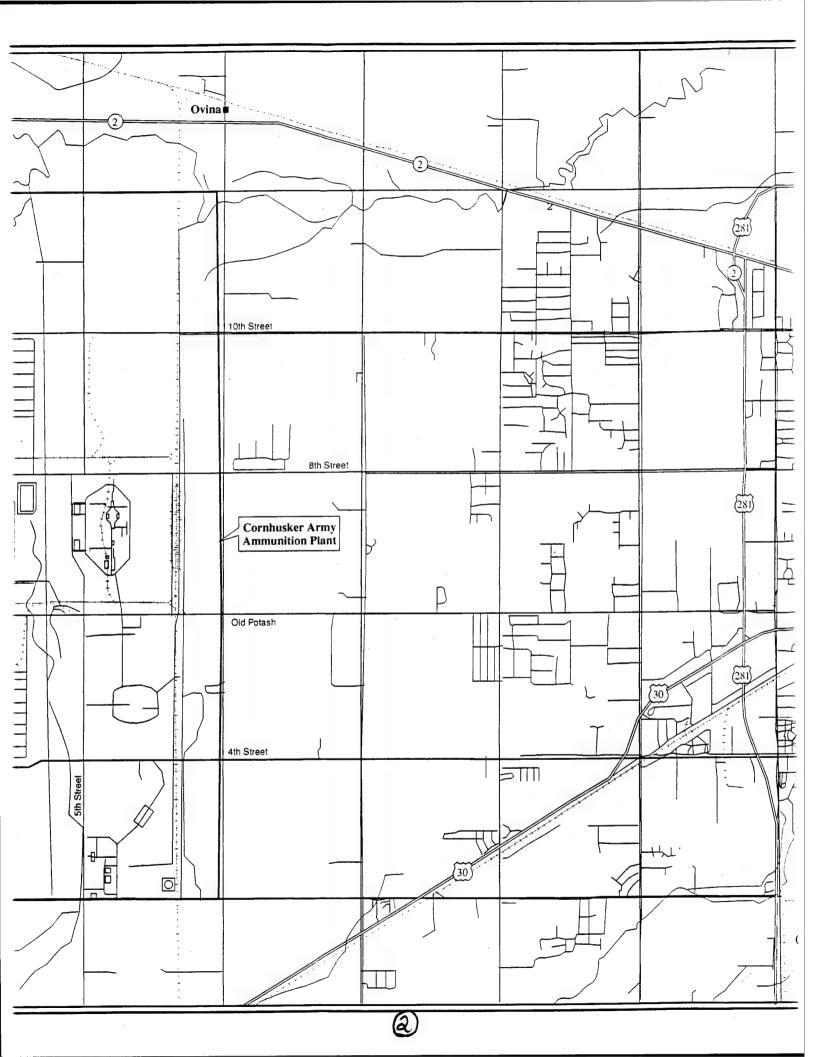
- 1.0.0.1. This document presents the objectives, design details, and results of the Peroxone Groundwater Treatment Demonstration Program (Program) that was conducted at the Cornhusker Army Ammunition Plant (CAAP) in Grand Island, Nebraska (Figure 1-1). The Program was carried out under the auspices of the US Army Environmental Center (USAEC) with technical assistance from the US Army Corps of Engineers, Omaha District (COE) and the Defense Evaluation Support Activity (DESA). A Project Advisory Board was formed from representatives of the above organizations, as well as two project technical advisors: Professor William Glaze from the University of North Carolina, and Mr. Kerwin Rakness of Process Applications, Inc. Dr. Glaze is an international expert on advanced oxidation processes, and Mr. Rakness has extensive experience in the design and optimization of ozonation systems. The Project Advisory Board reviewed the project progress and provided guidance to the project team throughout the project duration. All major project decisions were made with consultation and approval from the Advisory Board.
- 1.0.0.2. The Program, which was implemented by TRW and Montgomery Watson was intended to demonstrate the effectiveness of Peroxide/Ozone (Peroxone) oxidation treatment for groundwater impacted with explosive compounds. Explosives-contaminated groundwater exists at CAAP as a result of load, assembly, and packing (LAP) of explosives into munitions for World War II, the Korean conflict, and the Vietnam conflict. The contaminants of concern include 2,4,6-trinitrotoluene (TNT); 1,3,5-trinitrobenzene (TNB); hexahydro-1,3,5-trinitro-1,3,5-triazine (RDX); and other nitrobodies.

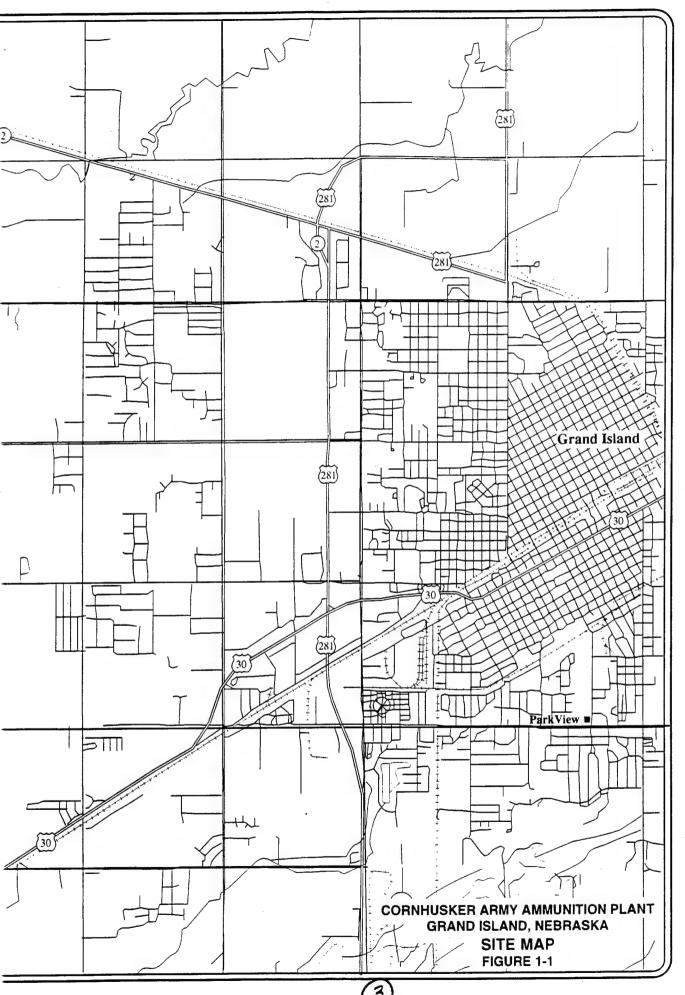
1.1 PURPOSE AND OBJECTIVES

- **1.1.0.1.** The purpose of the Program was to demonstrate the technical and economic feasibility of the Peroxone system to remediate explosives-contaminated groundwater at the CAAP.
- **1.1.0.2.** The following objectives were established for the demonstration program:
 - Further define the Peroxone system treatment requirements for nitrobodies

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- Design and construct a field-scale Peroxone system based on the requirements developed by the Technical Advisory Board and included in the Statement of Work (SOW) and results of the WES pilot-scale testing.
- Conduct demonstration testing of the Peroxone system and gather the necessary data to perform a technical and economic evaluation of the Peroxone system for treatment of explosives-contaminated groundwater.
- Develop recommendations on the feasibility of using Peroxone technology for a full-scale treatment system.

1.1.0.3. This document presents a summary of the activities undertaken to complete the above objectives and the results obtained during the demonstration testing. This document further evaluates the demonstration testing results to provide recommendations for a full-scale Peroxone system.

1.2 PROJECT BACKGROUND

1.2.0.1. Numerous US Army installations have sites that contain groundwater that has been contaminated with explosives. The use of granular activated carbon (GAC) is listed as the best available technology by the United States Environmental Protection Agency (U.S. EPA) for removal of such organic compounds from water. The disadvantage of using GAC is that it accumulates organic compounds on the carbon medium instead of actually destroying the contaminants. There are also problems associated with disposal of explosives-laden GAC. Processes which result in the immediate destruction of the contaminants and are more cost effective than GAC are being sought for the restoration of Army sites.

1.2.0.2. The effectiveness of chemical oxidation is highly dependent on the nature of the organic compounds, the oxidant used, and other contaminants in the water. Among the most promising oxidation processes is the ozone decomposition initiated by hydrogen peroxide. Hydrogen peroxide alone is a moderately powerful oxidizer, but in combination with ozone it is even more powerful because hydroxyl radicals are generated. The hydroxyl radicals that form in a Peroxone system are more effective than ozone alone for oxidation of natural and synthetic organics.

AWWARF & CGE. "Ozone in Water Treatment: Applications and Engineering," Cooperative Research Report, Lewis Publishers, Chelsea, MI, (1991).

1.2.0.3. The Corps of Engineers Waterways Experiment Station (WES) has developed a laboratory scale Peroxone system for the treatment of explosives-contaminated groundwater. Preliminary laboratory results have shown that TNT and RDX are oxidized by this system. In August 1995, a 2-gpm laboratory scale pilot system was field-tested by WES at the CAAP.

1.3 SCOPE OF THE DEMONSTRATION PROGRAM

1.3.0.1. The scope of the demonstration program was limited to the following:

- Design, construct, and operate a 25-gpm Peroxone groundwater treatment system at the CAAP in accordance with the requirements of the Technical Advisory Board.
- Conduct a 12-week demonstration test in accordance with the approved experimental plan.
- Analyze data from demonstration testing to evaluate effectiveness of the Peroxone system in treating explosives-contaminated groundwater.
- Develop recommendations for a 1,000 gpm Peroxone system based on the demonstration testing results.

1.4 ORGANIZATION OF THE REPORT

1.4.0.1. Section 1.0 presents the Program goals and objectives and provides the background for the Peroxone technology and the CAAP Program. The Peroxone system design details are presented in Section 2.0. Section 3.0 describes the activities undertaken during the Peroxone system construction and installation. Details about the demonstration testing and results obtained during the testing period are presented in Section 4.0. The system demobilization and the site restoration activities performed at the conclusion of the demonstration testing are summarized in Section 5.0. Results obtained during the demonstration testing were evaluated to develop recommendations for a full-scale Peroxone system. Section 6.0 presents the evaluation process and provides recommendations for a full-scale Peroxone system for treatment of explosives-contaminated groundwater.

Fleming, E.C., M.E. Zappi, J. Miller, R. Hernandez, and E. Toro (1997). "Evaluation of Peroxone Oxidation Techniques for Removal of Explosives From Cornhusker Army Ammunition Plant Waters", Technical Report SERDP-97-2, U.S. Army Engineer Waterways Experiment Station, Vicksburg, MS.

1.4.0.2. The sequence of construction for the Peroxone system is presented using a series of photographs which are included in Appendix A. The results obtained during the optimization period are listed in Appendix B. The results of the demonstration testing are summarized in Appendix C. The as-built drawings for the Peroxone system are included in Appendix D. The project Experimental Plan is included in Appendix E. (The actual experimental approach differed slightly from the experimental plan due to the ongoing analysis of the results during the course of the project. The deviations are explained in Section 4.0 of this report). The project team contact list is included in Appendix F.

2.0 PEROXONE SYSTEM DESIGN

2.1 INTRODUCTION

2.1.0.1. This section presents the details of the Peroxone system used to conduct the CAAP demonstration testing program. It is noted that the process selection and configuration was specified by the Technical Advisory Board with the concurrence of USAEC. The system design criteria is presented in Table 2-1 (A schematic of the treatment system is depicted in Appendix E, Figure 1). It is noted that photographs of the treatment system during construction are shown in Appendix A of this report.

Table 2-1

Demonstration Program Peroxone System Design Criteria

Equipment	Description	Criteria
Extraction Well	Number of existing wells	2
	Existing well casing	4" or larger
	Groundwater level	11 feet bgs (appr.)
	Well head finishing	Above ground with no vaults
Extraction Pump	Number of pumps	2 (one for each well)
-	Type	Submersible, electrical
	Capacity	25 gpm each
	Total dynamic head	One at 75 feet TDH; One at 90 feet TDH
	Pump horsepower	3/4 Hp (each)
	Control	Local control panel
	Manufacturer	Grundfos - Clovis, CA
Conveyance Line	Total length	1,130 feet (600 ft and 530 ft from
Conveyance Eme	Total length	the wells, respectively)
	Size and material	2" PVC, Sch 80
	Type	Single wall, above ground
Influent Flow Meter	Danca	0.20 cm
influent Flow Meter	Range	0-30 gpm Paddle wheel
	Type Indicators	Instantaneous flow/totalizer
		4-20 mA
	Signal type Manufacturer	Signet Scientific - El Monte, CA
	ivialiulactulel	Signet Scientific - El Wome, CA

Table 2-1

Demonstration Program Peroxone System Design Criteria (Continued)

Equipment	Description	Criteria
Ozone Contactor	Number of contactors Type	6 Unpacked column with co-current and counter-current flow; saddles for packing ring
	Capacity Size	500 gal (each) 3 feet diameter, 10 feet above diffuser; 13 feet total height
	Material Diffuser type Level indicator Fabricator	304 SS Ceramic Dome diffuser Sight glass Denver Mineral Corporation - Denver, CO
Effluent Tank	Capacity Type Control	500 gal HDPE High-level alarm (system shut off)
Sump Pump	Number Type Capacity	Low-level stop switch 1 Submersible, electrical 25 gpm
	Total dynamic head Control Manufacturer	25 feet (max) Internal float switch Little Giant - Ryan Herco (rep)
GAC	Number of vessels EBCT Carbon Quantity Manufacturer	3 in series 30 min. total at 25 gpm flow 1,000 lb./unit Calgon Corp Pittsburgh, PA
Effluent Flow Meter	Range Type Indicator Signal type Manufacturer	0-30 gpm Paddle wheel Instantaneous/totalizer 4-20 mA signal Signet Scientific - El Monte, CA

Table 2-1

Demonstration Program Peroxone System Design Criteria (Continued)

Equipment	Description	Criteria
Ozone Generator	Capacity Ozone Dosage (each vessel) Flow measurement Control Panel Dosage control Manufacturer (rented)	100 lb./day 55 mg/l at 10% O ₃ at 25 gpm Rotameter (internal) Local control panel Flow paced Ozonia - Lodi, NJ
Hydrogen Peroxide System	Capacity Storage Applied dosage Feed pump type Number of pumps Pump flow rate Control panel Dosage control	16 lb./day of 35% solution 55-gallon drums of 35% solution 18 mg/l (total) at 1.5% solution at 25 gpm Pulsafeeder 6 (one for each reactor) 0.75 gph (max) Local control panel Flow paced
Sodium Thiosulfate System	Storage Applied dosage Feed pump type Number of pumps Pump flow rate Control panel Dosage control	50 lb. bags; photo grade 7 mg/L per mg/L residual ozone Pulsafeeder 1 3 gph (max) Local control panel Flow paced
Oxygen Storage Tank	Capacity Type Owner (rented)	3,000 gal Liquid oxygen mixed with 3% nitrogen Linweld - Grand Island, NE
Oxygen Vaporizer	Capacity Controls Owner (rented)	500 scfh at 15 psi discharge Local/Manual Linweld - Grend Island, NE

Table 2-1 .

Demonstration Program Peroxone System Design Criteria (Continued)

Equipment	Description	Criteria
		8 scfm
Ozone Destructor	Capacity Controls	Local/Manual
	Manufacturer (rented)	Ozonia - Lodi, NJ
	Manufacturer (rented)	Ozoma - Loui, 143
Ozone Monitor	High concentration	1
	Ambient concentration	1
Off-gas Stack	Height	20 ft
	Velocity	100 fpm
	Size	2 inch
	Material	Carbon steel
		TT' 1 1 - 1 - 1 - 1
Alarm System	System shut-down mode	High-level alarm in effluent tank Major equipment failure
	Control logic	PLC control with auto dialer for
	Common 10810	off-hour operation
Process Piping	Type	Single wall
1100035 1 Iping	Size and Material	3" PVC, Sch 80
		·
Containment Pad	Size	30-feet x 40-feet x 12-in (wall)
Utilities (provided by	owner)	
Water	Capacity	150 gpm tap water
Electricity	Capacity	480V, 3 Phase, 200 Amp
Sewer	Capacity	150 gpm
Control/Equipment	Number	2
Room (provided by	Size	20-feet x 20-feet (each)
owner)		
Operation Mode	Normal mode	Automatic with PLC control and monitoring
	Number of operators	2
	Operator hours	40 hr/wk/operator
		•

2.1.0.2. The basic design of the process components was defined by the Technical Advisory Board with concurrence of USAEC based on the results of the previous testing conducted by WES. Minor modifications were made during the construction and demonstration testing phases, however, and these changes are discussed in the following text. The information presented in the SOW was used to develop the design criteria and to prepare the Peroxone system design.

2.1.1. Sources of Contaminated Water

2.1.1.1. Two (2) existing groundwater wells (Well No. 66 and New TRW Well, also referred to as Wells A and B, respectively, in Appendix E) were used to provide contaminated water to the Peroxone system. Each well was assumed to be capable of producing a continuous flow of 25 gallons per minute (gpm). However, this assumption proved to be false for one of the wells as testing progressed.

2.1.2. Design Hydraulic Capacity

- **2.1.2.1.** The Peroxone system was sized for a maximum hydraulic capacity of 25 gpm. This assumed that either one well is operated at a given time, or that the combined flow from the two wells would not exceed 25 gpm.
- **2.1.2.2.** The components designed for a 25 gpm hydraulic capacity include the groundwater extraction well pumps and the conveyance pipe, ozone contactors, activated carbon vessels, and appurtenance. The system support facilities were also sized to handle a maximum flow of 25 gpm at the design influent concentrations discussed below. Flexibility was provided to turn-down the Peroxone system to handle a lower flow; however, no provisions were provided for an effective treatment at a higher flow rate.

2.1.3. Influent Concentrations and Treatment Goals

- 2.1.3.1. The anticipated influent concentrations in the groundwater are listed in Table 2-
- 2. During the design period, no information was available on expected influent concentrations from individual wells. After that time, data were collected for actual influent concentrations from individual wells and are included in Section 4.0. The treatment goals listed in Table 2-2 represent the effluent limits established by the Nebraska Department of Environmental Quality (NDEQ) for discharge of treated water to

a swale or to a storm drain. The GAC effluent stream met these limits without exception during the demonstration period.

Table 2-2

Design Influent Concentrations and Treatment Goals

Contaminant	Design Influent Concentration (µg/L)	Target Treatment Goals (µg/L)
TNT	500	2
RDX	200	2
TNB	100	2
Total Nitrobodies	1,000	30

2.2 EXTRACTION SYSTEM

2.2.1. Extraction Wells

- **2.2.1.1.** The existing groundwater wells were 4-inch in diameter with a capped riser on top. The boring logs indicated that the groundwater table was at approximately 11 feet below ground surface (bgs) for both wells. This proved to be true in the field.
- 2.2.1.2. Each well was equipped with an electric submersible pump rated for 25 gpm maximum flow. The electric pump assembly was to include a power supply and a local disconnect switch for isolation of individual wells. A junction box was installed instead of the disconnect switch. This still allowed removal of the pump, but ensured that the operator turned off and locked out power at the main control panel.
- **2.2.1.3.** A 2-inch Schedule 80 PVC pipe was used for connecting the wellhead to the conveyance pipe. A ball valve was installed to control the actual flow from each well. Due to the short duration of the demonstration project, an above-ground well vault was not provided; however, temporary barricades were located at the wellhead to protect the wellhead equipment from accidental damage.

2.2.2. Conveyance Piping

2.2.2.1. An above-ground, single-walled, 2-inch Schedule 80 PVC pipe was used to convey extracted groundwater to the Peroxone system. The Schedule 80 pipe offered

additional strength to minimize accidental damage. Individual pipe runs from extraction wells were manifolded to provide a single run to the Peroxone system.

2.2.2.2. A flow meter with a range of 0-30 gpm and equipped with a local indicator and totalizer was installed on the conveyance pipe to record flow rates from individual wells and to provide information needed for system operation. The flow meter was located at the treatment pad for ease of readout and maintenance. The flow meter was calibrated using a 50-gallon barrel and a stop watch for three flow rates.

2.3 PEROXONE SYSTEM DESCRIPTION

2.3.1. Ozone Contactors

- **2.3.1.1.** Six (6) conventional, bubble-diffuser type contactors were used to accomplish the chemical oxidation. Each contactor was 3 feet in diameter with a 10-foot side wall depth above the diffuser base. Each contactor provided a retention time of approximately 20 minutes at a flow rate of 25 gpm. A 2-foot head space was provided in each contactor above the water column for off-gas collection.
- 2.3.1.2. The contactors were 1/8-inch thick, 304 stainless steel (SS 304) shells with 3/16-inch SS 304 top and bottom plates. Each contactor included a 20-inch manway integral to the contactor shell. The manway was located near the bottom of the contactor and was used to position the dome diffuser. It was noted during construction and debugging that the manway was critical to making alignments and repairs. A clear sight glass was included with each contactor for visual observation of the water level inside the contactors. However, visual inspection of the interior of the contactor through the sight glass proved to be difficult due to the inavailability of sufficient lighting to the inside of the contactor.
- **2.3.1.3.** The first contactor was provided with additional features to allow further studies in the future. An additional 20-inch manway opening was located near the top of the contactor. The top manway may be used in the future to fill the contactor with packing material. A saddle ring to facilitate packing of the contactor in the future and two, 2-inch clear acrylic windows were also included in the first contactor to allow observation of the bubble pattern and size.

- 2.3.1.4. Two dome diffusers were located at the bottom of each contactor to facilitate even distribution of ozone inside the contactors. The number of diffusers per contactor was selected based on the required gas flow rate and the manufacturer's specifications for those specific diffusers. The dome diffusers were 8-inch in diameter and constructed of ceramic material which offers excellent resistant to ozone corrosion. Each contactor was fitted with the diffusers as planned, although several diffusers were replaced during debugging due to irregular bases that did not allow for an air-tight seal around the diffusers.
- 2.3.1.5. The contactors were designed to operate in both the co-current and the counter-current flow conditions. The piping and valves between the contactors were installed such that the contactors could be manually switched to operate in either co-current or counter-current flow mode. The interconnecting piping between the contactors was Schedule 80 PVC with plastic valves. Although the system was piped for both flow conditions, it was only operated in the counter-current condition during the demonstration testing.

2.3.2. Ozone Generation and Feed System

- 2.3.2.1. Liquid oxygen was used for ozone generation at the demonstration plant, and was selected for ease of operation and maintenance. Liquid oxygen was stored in a supplier-provided bulk storage tank located adjacent to the treatment pad. A local supplier set up and stocked the tank as planned, then emptied the tank and removed it as planned at the end of the demonstration testing.
- 2.3.2.2. The ozone generator was designed to deliver an applied ozone dosage of 55 mg/L to each contactor at 10 percent by weight ozone concentration and at a maximum flow rate of 25 gpm. This equates to 100 pounds of total ozone per day for the Peroxone system. A gas flow meter with a local indicator and an ozone monitor was provided to track the ozone generation rate. At each injection point, a local rotameter with a manual control valve was used to calibrate the ozone dosage to an individual contactor. The ozone generator was delivered with all the features expected and produced over 108 pounds per day of ozone, although the full capacity was not used during the demonstration testing. An air compressor was provided to deliver a nitrogen-containing air stream to the oxygen feed flow. The added nitrogen is believed to result in a catalytic reaction that may increase the efficiency of ozone generation by as much as 15% to 20%.

This was based on Montgomery Watson's experience with the design of ozone systems, and the recommendation of the ozone generator supplier.

2.3.2.3. The supply piping from the liquid oxygen tank to the ozone generator was 1 1/2-inch copper pipe specifically designed for liquid oxygen systems. The ozone feed piping was 1 1/2-inch 304 SS. Flexible polyethylene tubing was to be used to connect the ozone feed pipe to individual contactors. The cooling water pipe for the ozone generator was 1 1/2-inch Schedule 40 PVC capable of providing the 70-gpm cooling water flow required for the operation of the ozone generator (required by the generator manufacturer). The copper piping for the oxygen and the stainless steel piping for the ozone feed worked well. During the System Debugging task, the polyethylene tubing degraded under the ozone concentrations used, and was then replaced with high-grade teflon tubing which proved to be resistant to the operational environment at the demonstration testing.

2.3.3. Chemical Feed System

- 2.3.3.1. The hydrogen peroxide storage system was designed for a thirty-five percent industrial grade solution stocked in 55-gallon drums at the site. The peroxide solution was diluted to 2 percent strength using deionized water from an on-site, ion exchange system. Two, 275-gallon day tanks were used to stock 2 percent peroxide solution which was fed to the Peroxone system. The purpose of diluting the peroxide solution was to increase the volume of the solution actually fed to the contactors, thereby allowing a more precise control over pumping rates and system operation. At times the strength was reduced to 1.5 percent and 1 percent to maintain better control of the peroxide dose.
- 2.3.3.2. The peroxide solution was fed to the contactors through flexible, polyethylene tubing connected through injection points located in the piping between the contactors. Positive displacement pumps were used to feed peroxide into the system. An individual, dedicated pump was used for each contactor. All pumps fed off a single day tank to ensure that the concentration of the peroxide solution fed to each contactor was constant. Back-pressure control valves were added to the positive displacement pumps to prevent loss of prime on the suction side of the pumps.
- 2.3.3.3. Sodium thiosulfate was selected to neutralize the residual ozone in the effluent from the contactors. Fifty-pound bags of photograde thiosulfate crystals were stored at the site for this purpose. Thiosulfate solution was prepared on a daily basis using

deionized water from an on-site, ion exchange system. Thiosulfate was then fed directly into the effluent tank through a 1/2-inch Schedule 40 PVC pipe. A positive displacement pump was used to feed the thiosulfate solution to the effluent tank. The 1/2-inch PVC feed line into the effluent tank was replaced with a 3/8-inch polyethylene tubing to increase the flow velocity and thus allow for a more efficient pumping system.

2.3.4. Effluent Tank and Effluent Pump

- 2.3.4.1. Effluent from the ozone contactors was fed into the effluent tank via gravity flow. The effluent tank was used as a reaction tank to neutralize the residual ozone before discharging the water into the GAC vessels as described in Section 2.3.5. This tank was placed next to a sump built into the containment pad. The tank worked well for equalization of the effluent and addition of the thiosulfate solution. It was also connected to the sump pump to permit transfer of rain water or spills on the pad into the tank allowing treatment through the GAC vessels before discharge.
- **2.3.4.2.** An end suction, centrifugal pump rated for 25 gpm at 15 psi total head was used to transfer treated water from the effluent tank through GAC vessels to the discharge connection. The pump worked as expected, outpacing the gravity flow to the effluent tank and allowing intermittent operation of the pump.

2.3.5. Activated Carbon Polishing System

2.3.5.1. Treated water from the Peroxone system was routed through GAC vessels for additional treatment. The GAC system was a vendor-supplied package consisting of three (3) vessels operated in series. Each vessel contained 1,000 pounds of virgin activated carbon and provided 10-minute retention time at a flow of 25 gpm. The total retention time for the GAC system was 30 minutes. The carbon vessels worked successfully, preventing discharge of any contaminants above the permit requirements.

2.3.6. Ozone Destruction System

2.3.6.1. Off-gas from the ozone contactors was collected and treated through an ozone destruction unit. The ozone destruction unit consisted of a dual catalyst bed with an electric heating coil which converts ozone to innocuous byproducts. Exhaust from the ozone destruction unit was discharged to the atmosphere through a stack. The ozone

destruction unit was delivered with the ozone generator and worked as planned, reducing ozone gas concentrations in the stack to non-detect levels.

- **2.3.6.2.** Polyethylene tubing was installed on top of each of the ozone contactors for offgas collection. The tubing was then manifolded into 2-inch Schedule 80 PVC pipe which ran to the ozone destruction unit. During the testing, the polyethylene tubing proved to be incompatible with high ozone concentrations, therefore, the off-gas tubing was changed to teflon to prevent further failures. This piping setup worked well, with no failures detected during the demonstration testing.
- 2.3.6.3. The exhaust stack from the destruction unit was to be a field-installed PVC vent, but it was changed to 2-inch steel pipe to allow for a more rigid installation without guy wires. Sampling points were located downstream of the destruction unit to collect air samples into a single ozone analyzer. A high ozone condition in the exhaust stack triggered an alarm. The discharge stack was also equipped with a local flow meter to monitor the exhaust flow but the flow rate out of the stack proved to be so low that a manometer and pitot tube had to be used instead to check the flow.

2.4 PROCESS CONTROL NARRATIVE

2.4.1. Groundwater Extraction System

- **2.4.1.1.** As shown in the as-built drawings in Appendix D, each extraction well pump was controlled from a hand switch on the control panel (HS-101 or HS-102). The well pumps shut down automatically from the low flow switch (FSL-200) on the influent pipe, with a time delay to restart the pumps. The well pumps also shut down at a high-level alarm from the first ozone contactor (LSH-301), and the well pump power was interlocked with the main treatment system alarm (Alarm Level I) for an emergency shut off.
- **2.4.1.2.** A flow meter (M-1) was provided with a local indicator/totalizer (FIT-203/FIQ-203) and a pen recorder (FIR-203) to monitor the influent flow rates.
- **2.4.1.3.** The first ozone contactor (OT-1) was equipped with a high water level switch (LSH-301). LSH-301 signaled the system alarm (Alarm Level I).

2.4.2. Peroxide Feed System

- 2.4.2.1. Hydrogen peroxide was fed from one of the two day tanks (DT-1/DT-2). Each day tank had a low level switch (LSL-601/LSL-602). A selector switch (LSS-600) was included to determine which day tank was to be in service and thus which level switch was functional. LSL-601/LSL-602 were used to shut off the peroxide metering pumps and signal the system alarm (Alarm Level I). The day tanks were connected to the feed lines using ball valves. When the operator wanted to draw from a specific tank, the appropriate valve was opened and the other tank valve closed. The tank low-level switches LSL-601 and LSL-602 were installed as described, but instrumented together. Both switches were operational at the same time, allowing the operator to withdraw solution from both tanks simultaneously.
- **2.4.2.2.** Chemical mixers (MX-1/MX-2) in the peroxide day tanks were controlled manually at the local switch (see Facility Plan in as-built drawings).
- **2.4.2.3.** All peroxide metering pumps were turned on by a single local switch (HS-603) that was interlocked with the low flow switch (FSL-200). In the "ON" position, the pumps automatically turned on or off. The dosage from each metering pump was adjusted manually from the speed and stroke controls on the pump. A dedicated peroxide metering pump was used for each ozone contactor. All peroxide metering pumps were shut off by the level switch (LSL-601/LSL-602) in the peroxide day tank, and interlocked with the system alarm (Alarm Level I).

2.4.3. Ozone Feed and Destruction Systems

- **2.4.3.1.** Ozone was fed from the ozone generator OG-1. A flow meter (FI-300) with a local indicator and a central ozone monitor (AI-300) was included to track the ozone generation rate. At each injection point, a local rotameter (FI-301 through FI-306) with a manual control valve was used to calibrate the ozone dosage to an individual contactor. On the off-gas line from each ozone contactor, the residual ozone concentrations were monitored with a central ozone monitor (AI-300) to track the actual ozone transfer efficiency. The ozone monitoring worked as planned, allowing the operator to observe the ozone absorption concentrations and to adjust each rotameter during the demonstration testing.
- **2.4.3.2.** The ozonator was equipped with vendor-supplied control panel (LCP) for ozone generation rate control, liquid oxygen (LOX) usage, and cooling water systems. The LCP

was turned on manually and it was interlocked with the low flow switch (FSL-200). The LCP included an ozonator alarm to shut off the ozonator and to initiate the system alarm (Alarm Level I). The ozonator was also interlocked with the system alarm (Alarm Level I). The LCP alarm was triggered by ozonator malfunction, ozone leak, or an LOX feed problem. The vendor-supplied LCP provided all the controls for the ozone generator, but required a remote connection from the main control panel to ensure a shut down of ozone production if there was a main system alarm or a failure of the ozone destruction unit. This connection worked well and was tripped during actual operation.

2.4.3.3. Off-gas from each ozone contactor was forced through the ozone destruction unit (OD-1) prior to discharge. The destruction unit was equipped with vendor-supplied LCP for controls. The LCP was turned on manually and it was interlocked with a low-flow switch (FSL-200) (see as-built drawing I-1). The LCP was equipped with an alarm to shut off the ozone destruction unit and to initiate the system alarm (Alarm Level I). The ozone destruction unit was also interlocked with the system alarm (Alarm Level I). Discharge from the destruction unit was monitored by the central ozone monitor (AI-300), and was tied to a high ozone concentration alarm (Alarm Level II) at the control panel. The off-gas discharge stack included a flow meter (FI-702) with a local indicator to monitor the discharge flow rate through the stack (OGS-1). These features were installed as specified, although the vendor-supplied LCP did not shut off the ozone destruct unit if a low-flow occurred. Since the unit worked independently of all other components, this was not changed.

2.4.4. Effluent Tank

2.4.4.1. Effluent from the last ozone contactor was designed to gravity flow to the effluent tank (TK-2) which was equipped with three level switches. Switch LSHH-501 (which stands for Level Switch High-High-501) signaled a high-high level alarm (Alarm Level II); switch LSH-501 would turn on the effluent transfer pump (P-2), and switch LSL-501 shut off the effluent transfer pump (P-2) and the thiosulfate metering pump (DF-7). LSLL-501 (which stands for Level Switch low-low-501)signaled an alarm (Alarm Level II).

2.4.5. Thiosulfate Feed System

- **2.4.5.1.** Sodium thiosulfate was fed from a day tank (DT-3). The day tank was equipped with a low-level switch (LSL-611). LSL-611 shut off the thiosulfate metering pump and signaled an alarm (Alarm Level II).
- **2.4.5.2.** Chemical mixer (MX-3) in the thiosulfate day tank was turned on manually at the local switch.
- **2.4.5.3.** The thiosulfate metering pump was turned on by a local switch (HS-610) and it was interlocked with the effluent transfer pump (P-2) fail status. The metering pump was adjusted manually from the speed and stroke controls on the pump. The thiosulfate metering pump was turned off from the low-level switch (LSL-611) in the thiosulfate day tank and it was interlocked with the system alarm (Alarm Level I).

2.4.6. Effluent Transfer Systems

- **2.4.6.1.** Effluent transfer pump (P-2) had a hand-off-auto (H-O-A) switch (HS-502) located on the control panel. When the pump was in AUTO, the effluent transfer pump (P-2) was controlled by the level switches (LSH-501 and LSL-501) in the effluent storage tank (TK-2). The pump was interlocked with the system alarm (Alarm Level I). Instead of locating the H-O-A switch on the control panel, it was located by the pump in the field. Otherwise the pump functioned as designed.
- **2.4.6.2.** Each GAC vessel was to be equipped with a pressure gauge fitting to allow visual observation, but the vendor did not supply the gauges. Since this was not an essential parameter to measure, the pressure gauges were left out.
- **2.4.6.3.** Discharge from the Peroxone system was to be monitored through a flow meter (M-2). The flow meter was equipped with a local indicator/totalizer (FIT-701/FIQ-701) and a pen recorder (FIR-701) located at the control panel. This flowmeter was installed down stream of the carbon contactors.

2.4.7. Support Systems

2.4.7.1. The system alarm (Alarm Level I) was designed to shut off the entire system; the alarm status was displayed on the Control Panel (CP-1). An auto-dialer was to be used to

notify the operator of any alarm condition during off-hour operation. The system alarm was connected as designed and successfully shut down the entire system as required. The auto dialer was not installed since the treatment process was monitored full time during the day and would shut down automatically if a failure occurred at night. However, it is noted that no failures or shut-downs occurred during the testing period.

2.4.7.2. Alarm Level II was displayed on the Control Panel (CP-1); however, Alarm Level II would automatically re-set when the alarm condition disappears. This worked as planned with the annunciation light coming on during each Level II alarm.

2.5 TREATMENT PAD

- **2.5.0.1.** The treatment pad was sized to accommodate all components of the Peroxone system except for the liquid oxygen tank and chemical storage. The treatment pad was designed for a seismic zone 1 and for other local conditions per the Uniform Building Code (UBC).
- **2.5.0.2.** A 12-inch berm was provided on all sides of the pad for secondary containment. The containment pad and the berm were designed to provide adequate capacity to hold the volume of all contactors and GAC vessels plus 10 percent. The berm and the pad were constructed as designed and were poured monolithic allowing for a more water-tight structure.

3.0 PEROXONE SYSTEM CONSTRUCTION

3.0.0.1. This section contains a review of the construction portion of the Peroxone system demonstration program. It is organized chronologically by weekly progress. This discussion contains information about successful components of the construction process as well as lessons learned during assembly of the system.

3.1 ADVANCE PREPARATION

3.1.1. Procurement

- 3.1.1.1. During development of the conceptual design for the system, it was proposed that the individual components be shipped to Montgomery Watson's test facility in California and pre-assembled to ensure proper operation at the CAAP. Montgomery Watson suggested that the schedule could be expedited if the system was assembled and tested on site using a field engineer. After agreement that the schedule was tight and the system should be assembled on site, it became necessary to procure equipment and services immediately as the design was being developed. In order to allow ten weeks of operation before winter set in, it was necessary to construct the system within a window of four to five weeks.
- **3.1.1.2.** Early procurement involved ordering and fabricating the contactor vessels two months ahead of the scheduled construction period. Since the design was in process, the design group was diverted to focus on the long lead items first. The fabrication company was enlisted to help with detailed design issues and the contactors were designed in parallel with the rest of the treatment system. The same approach was used to select the ozone generator.
- 3.1.1.3. The design documents were abbreviated in detail to expedite the schedule. Any details that were not shown on the design drawings were completed by the on-site Montgomery Watson engineer during field fabrication. This expedited approach saved three weeks in additional engineering time and reduced the subsequent cost of engineering.
- **3.1.1.4.** As soon as the size, model, and the manufacturer of each component were decided, the design was sent out for procurement and scheduled for delivery to the project site during the construction window. Advance procurement of all components proved to

be successful, with all equipment arriving before or within the first week of construction. A collection of photographs showing assembly of the system components is included in Appendix A.

3.1.2. Slab Preparation

3.1.2.1. The concrete containment slab for the treatment system required 28 days to cure. Advance procurement of a local subcontractor was necessary to ensure that the slab was ready for system installation as soon as the equipment and materials arrived on the site. A local engineering company was hired to inspect the steel reinforcing prior to pouring concrete, and to take quality control samples during placement of the concrete. The concrete slab was poured the week of 12 July 1996.

3.2 CONSTRUCTION CHRONOLOGY

3.2.1. Extraction Wells

3.2.1.1. Although the original project scope called for the use of three wells for the demonstration testing, the plan was changed to utilize only two wells just before start of construction. The wells had already been installed and developed during the previous studies, so they only required installation of pumps and piping. Both wells were expected to produce 25 gpm in order to operate the treatment system at the design flow rate. After the wells were connected and pumping started, it was discovered that Well 66 would not produce more than 13 to 15 gpm of flow. This reduced flowrate was factored into the demonstration testing. Piping to the wells was completed 19 July 1996, and the pumps were installed and connected 9 August 1996.

3.2.2. Ozone Generator

3.2.2.1. The ozone generator was delivered to the site and set up on 23 July 1996. The liquid oxygen tank for the generator was delivered and set up 25 July 1996 by Linweld, a local oxygen supplier.

3.2.3. Contactors

3.2.3.1. The first three contactors were delivered on 26 July 1996 and the first contactor was set the same day. Piping to the first three contactors was completed 29 July 1996.

The last three contactors were delivered 30 July 1996 and were set and plumbed by 2 August 1996.

3.2.4. Activated Carbon Vessels

3.2.4.1. Three carbon vessels were delivered to the project site on 1 August 1996. They were placed on the pad and connected 7 August 1996.

3.2.5. Reverse Osmosis Unit

3.2.5.1. A reverse osmosis (RO) treatment unit was leased to produce deionized make-up water for the hydrogen peroxide solution. The reverse osmosis unit was delivered and set up 8 August 1996. The unit was leased from Culligan's local distributor.

3.2.6. Sodium Thiosulfate Feed System

3.2.6.1. This system consisted of a chemical metering pump, a mixing tank and a mixer, level controls and the associated piping. The assembly was installed during 5 to 8 August 1996.

3.2.7. Chemical Feed Pumps

3.2.7.1. Six chemical feed pumps were set up and connected to supply and delivery tubing on 5 and 6 August 1996.

3.2.8. Piping and Fittings

3.2.8.1. All piping and fittings for the Peroxone system were connected by the end of the week of 8 August 1996. This included the stainless steel ozone delivery lines, the well water delivery lines, the effluent piping, and the RO effluent water piping.

3.2.9. Water System

3.2.9.1. The potable water system was connected on 7 August 1996. The piping was tested that day for leaks and repaired as required.

3.2.10. Power

3.2.10.1. The power company set the main power supply pole and the three-phase transformers during 29 July 1996 to 2 August 1996. Power conduit and wiring from the control panel was run to the equipment on the pad the week of 5 August 1996. Power connection to the system was completed 6 August 1996.

3.2.11. Instrumentation and Controls

3.2.11.1. Instrumentation wiring was pulled in with the power wiring the week of 5 August 1996. The instrumentation was connected and tested the week of 12 August through 17 August 1996.

3.3 STARTUP

3.3.0.1 Startup of the Peroxone system was divided into three efforts; clean water testing, debugging, and optimization. Clean water testing is discussed below. Debugging and optimization are discussed in Section 4.0.

3.3.1. Clean Water Testing

- **3.3.1.1.** As individual portions of the system were completed they were tested prior to startup. First the conveyance piping was connected to the contactors. The piping and the contactors were then filled with clean water and hydraulically tested for leaks. Then all tanks, piping and tubing on the pad were filled and checked prior to debugging. The carbon vessels were filled with water and the air pressure in each vessel was bled off until they flowed smoothly. The connections were then completed between each GAC vessel to check for leaks under operating pressure. This took place 12 through 14 August 1996.
- **3.3.1.2.** Once the tanks, piping, tubing, and pumps were checked and all repairs completed, the system was turned over to the operator for debugging and optimization testing.

4.0 PEROXONE SYSTEM TESTING PROGRAM

4.1 INTRODUCTION

4.1.0.1. This section describes the testing program implemented at the Peroxone groundwater treatment system. The testing approach and analytical methods used in the program are first described, followed by the experimental conditions and results obtained from each of the optimization and demonstration testing programs. Finally, a mathematical model describing the destruction of TNT, TNB, and RDX with ozone/hydrogen peroxide is proposed. The model is then calibrated using the optimization and the demonstration testing results.

4.1.0.2. The following legend has been used for figures in this section:

INF - Influent Concentration

C1 - Effluent Concentration from Contactor 1

C2 - Effluent Concentration from Contactor 2

C3 - Effluent Concentration from Contactor 3

C4 - Effluent Concentration from Contactor 4

C5 - Effluent Concentration from Contactor 5

C6 - Effluent Concentration from Contactor 6

4.2 TESTING APPROACH

4.2.0.1. The overall testing program extended over a period of 12 weeks. After the plant was constructed and all the equipment was installed, the demonstration plant operators conducted three primary tasks: (1) System Debugging, (2) System Optimization, and (3) System Demonstration. The following is a brief description of each task, and they are discussed in more detail later in this section.

4.2.1 Task 1 — System Debugging

4.2.1.1 During this task, which extended over two weeks, the plant pumps and chemical feed systems were started up at a low flow rate (approximately 10 gpm) using tap water, and checked for any water or chemical leaks. The system was also checked for malfunctions of chemical feed equipment and shut-down alarms. After the leaks and malfunctions were adjusted, the flowrate through the plant was continuously increased until the design flow of 25 gpm was reached. The plant was then operated at the design flowrate for a period of two days. During this period all water and chemical feed equipment were checked for operational stability. Tracer testing was also conducted during this phase to characterize the hydraulic residence time distribution of the system. A summary of significant problems and their resolution is discussed in Section 4.5.5.

4.2.2 Task 2 — System Optimization

4.2.2.1 During this task, which also lasted for two weeks, process optimization testing was conducted using water from each of the two test wells. Process optimization involved operating the system at various ozone doses and hydraulic retention times, collecting water samples from the effluent of each of the six contactors, as well as from the wall taps along the water depth of the first contactor, and analyzing them for ozone residual and explosives. The applied ozone dose tested ranged from 38 mg/L to 115 mg/L. The flowrate tested ranged from 13 gpm to 25 gpm. Steady-state conditions were reached (a minimum of 3 hydraulic retention times) before any operational parameter was changed. These optimization tests were used to determine the operating conditions that would result in the reduction of the target contaminants to the desired effluent quality.

4.2.3 Task 3 — System Demonstration

4.2.3.1 During this task, which was conducted over a period of eight weeks, the system was operated under two sets of conditions using water from New TRW Well only.

System demonstration involved operating the system at constant conditions over an extended period of time, collecting water samples from the effluent of each of the six contactors, and analyzing them for ozone residual, pH, oxidation-reduction potential (ORP), and explosives concentrations. These tests served to demonstrate that the system can achieve the anticipated performance on a long-term basis.

4.3 ANALYTICAL METHODS

4.3.0.1. Samples were taken from the Peroxone system influent, the effluent from each ozone contactor, and the effluent of the granular activated carbon (GAC) contactors on a daily basis. The following analyses were routinely conducted on these samples during the demonstration project:

- explosives
- nitrate
- · ozone residual
- hydrogen peroxide residual
- oxidation reduction potential (ORP)
- pH
- · temperature.

4.3.0.2. Samples were packaged in insulated containers, cooled with ice, and shipped to GP Environmental Labs in Gaithersburg, MD for analysis. A total of 15 explosives contaminants were reported including the three target compounds: TNT, TNB, and RDX. The mass sum of all the compounds analyzed for with EPA Method 8330 and reported by GP Environmental Labs was referred to as Total Nitrobodies. Nitrate samples were routinely taken from the influent and each contactor effluent. These samples were analyzed by GP Environmental Labs using EPA Method 9056. Ozone residual analyses were conducted on site using Standard Method 4500-O₃B Indigo colorimetric method. A known volume of Indigo Reagent II was drawn into a 10-mL gas-tight glass syringe. The remaining volume in the syringe was filled with the sample being analyzed.

absorbance of the mixture at 610 nm was then determined with a Hach DR-700 colorimeter. The hydrogen peroxide residual was measured using the method described by Masschelein et al. (1977). Oxidation-reduction potentials were performed using proposed Standard Method 2580 (ORP). An Orion Model 9678BN oxidation-reduction probe and Orion model 920 ion selective electrode meter were used. pH analyses were conducted on site using a Hach EC-10 portable pH meter and probe with automatic temperature compensation. The temperature of samples was measured using an alcohol thermometer graduated in 1 degree centigrade increments and was recorded during the determination of the oxidation-reduction potential. Tracer tests using Fluosilicic Acid were conducted at process flow rates of 13 and 25 gpm. Fluoride analyses were conducted during tracer testing using Standard Method 4500F. An Orion fluoride probe, Model 9609BN, and Orion model 920 ion selective electrode meter were used.

4.3.0.3. In addition, numerous analyses were conducted by GP Environmental Labs on a less frequent basis. The analyses conducted and methods used are listed below:

•	Volatile Organic Compounds	EPA Method 8260
•	Semi-Volatile Organic Compounds	EPA Method 8270
•	Iron, Calcium, Magnesium, Manganese	SW846, EPA Method 6010
•	Nitrate, Nitrite and Sulfate	EPA Method 9056
•	Carbonate, Bicarbonate, Ammonia,	
	& Phosphorous, Total Kjeldahl Nitrogen	Standard Method 4500
•	Total Suspended Solids, Total Dissolved Solids	Standard Method 2540
•	Alkalinity	Standard Method 2320
•	Total Organic Carbon	Standard Method 5310

4.3.0.4. It should be noted that all analytical results were proven to be reliable. The QA/QC data for the project are listed in the Independent Evaluator's report.

Masschelein, W.; M. Denis, and R. Ledent, "Spectrophotometric Determination of Residual Hydrogen Peroxide", Journal of Water & Sewage Works, pp. 69-72 (August, 1977).

4.4 GROUNDWATER QUALITY

4.4.0.1. During the beginning of the optimization testing, groundwater samples were collected and analyzed for various general physical/mineral water quality parameters, as well as an array of volatile organic chemicals (VOC). The results of the general/mineral analyses are listed in Table 4-1. Both waters can be characterized as relatively high alkalinity, high hardness waters. The results suggest that Well #66 water had a substantially lower organic content than New TRW Well water.

Table 4-1

General Physical/Mineral Groundwater Quality Characteristics

		Val	ue
Parameter	Unit	New TRW Well	Well #66
Alkalinity	mg/L as CaCO ₃	311	326
Nitrate	mg/L	1.41	9.51
Ammonia	mg/L	0.29	13.6
Calcium	mg/L	63.6	82.5
Iron	μg/L	< 52	< 52
Magnesium	mg/L	10.7	16.8
Manganese	mg/L	0.637	0.564
Total Phosphorous	mg/L as P	0.301	0.668
Total Dissolved Solids	mg/L	452	BKN
Total Organic Carbon	mg/L	5.32	1.92
pH	_	7.0	7.0

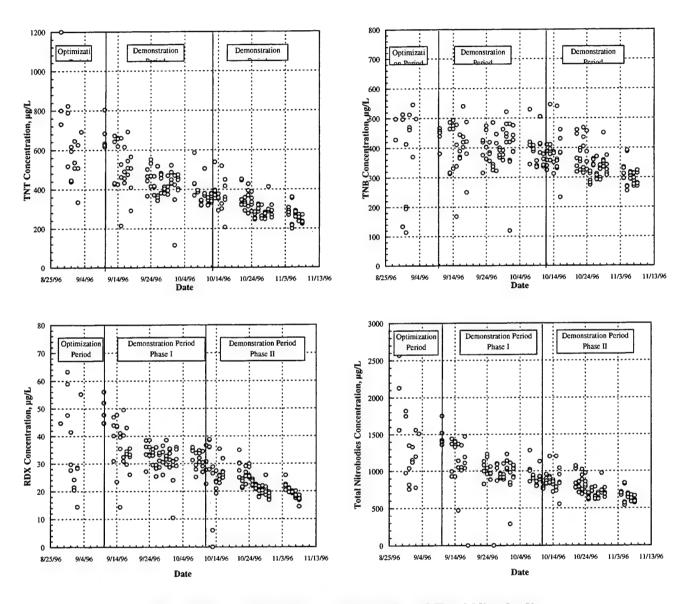
BKN: Broken Sample Vial

4.4.0.2. The types and concentrations of the synthetic organic chemicals (SOCs) analyzed for during the optimization testing period are listed in Table 4-2. The results show that waters from New TRW Well and Well #66 did not contain VOCs above the compound-specific detection limits.

4.4.0.3. In addition, during the optimization testing period and the demonstration testing period, the influent water to the Peroxone treatment system was analyzed daily for explosives, including the target contaminants of TNT, TNB, and RDX. The average concentration and range of each of these compounds, as well as the sum of Total Nitrobodies measured in New TRW Well and Well #66 waters are listed in Table 4-3 for each of the optimization and demonstration testing periods. The results are also plotted in

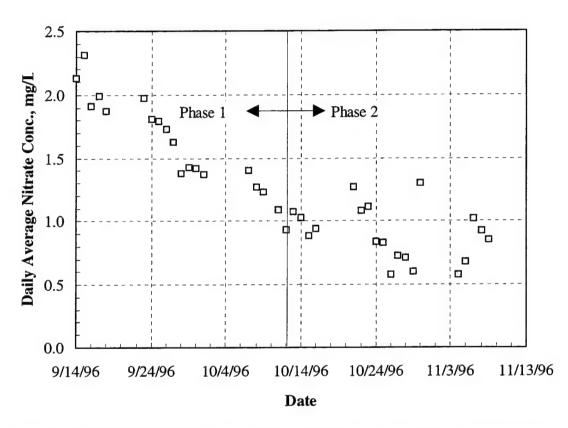
Figure 4-1 on the following page. The analytical results showed that the contaminant concentrations varied significantly throughout the testing period. This wide variation impacted the ability to interpret the Peroxone system performance data since reaching steady-state conditions was virtually impossible. In addition, the data presented in Table 4-3 and Figure 4-1 show that the influent concentrations of all contaminants decreased substantially from the beginning of the project (during the optimization period) to the end of the project (Phase II demonstration period). This should also be considered when comparing the performance of the Peroxone system under the different conditions of the optimization period and the demonstration period. For optimum data analysis, no average influent concentrations were used, but rather each effluent concentration was coupled with its corresponding influent value. In order to facilitate the comparison of system performance under different influent contaminants' concentrations, the performance was expressed in terms of percent removal and not as effluent concentration values.

4.4.0.4. In addition, the analytical results show that the concentration of nitrate in the groundwater decreased through the testing period. Figure 4-2 shows a plot of the daily average concentration of nitrate in the New TRW Well water during Phases 1 and 2 of the demonstration period. The results show that the daily average nitrate level continuously decreased from a high of 2.3 mg/L at the beginning of Phase 1 of the demonstration period to a low of 0.6 mg/L at the end of Phase 2 of the demonstration period. It should be noted that this decrease in influent nitrate level should not have had any impact on the process performance since no reaction between ozone and nitrate is expected because nitrate is the highest oxidation state for nitrogen.



Raw Water Levels of TNT, TNB, RDX, and Total Nitrobodies

Figure 4-1



Nitrate Concentrations in the New TRW Well Water During the Demonstration Period

Figure 4-2

Table 4-2

Types and Concentrations of SOCs in New TRW Well and Well #66 Waters
(All Concentrations Were Below the Indicated Detection Limits)

Chemical	Detection Limit (µg/L)	Chemical	Detection Limit (µg/L)	I Chemical	Detection Limit (μg/L)
1,2,4-Trichlorobenzene	10.5	Benzo[b]fluoranthene	10.5	1,1,2,2-Tetrachloroethane	e 5
1,2-Dichlorobenzene	10.5	Benzo[g,h,i]perylene	10.5	1,1,2-Trichloroethane	5
1,3-Dichlorobenzene	10.5	Benzo[k]fluoranthene	10.5	1,1-Dichloroethane	5
1,4-Dichlorobenzene	10.5	Benzyl alcohol	10.5	1,1-Dichloroethene	5
2,4,5-Trichlorophenol	52.5	bis(2-Chloroethoxy) methane	10.5	1,2-Dichloroethane	5
2,4,6-Trichlorophenol	10.5	bis(2-Chloroethyl) ether	10.5	1,2-Dichloropropane	5
2,4-Dichlorophenol	10.5	bis(2-Chloroisopropyl) ether	10.5	2-Butanone	10
2,4-Dimethylphenol	10.5	bis(2-Ethylhexyl)phthalate	10.5	2-Chloroethylvinyl ether	10
2,4-Dinitrophenol	52.5	Butyl benzyl phthalate	10.5	2-Hexanone	10
2,4-Dinitrotoluene	10.5	Chrysene	10.5	4-Methyl-2-pentanone	10
2,6-Dinitrotoluene	10.5	di-n-Butylphthalate	10.5	Acetone	10
2-Chloronaphthalene	10.5	di-n-Octylphthalate	10.5	Benzene	5
2-Chlorophenol	10.5	Dibenzofuran	10.5	Bromodichloromethane	5
2-Methylnaphthalene	10.5	Dibenz[a,h]anthracene	10.5	Bromoform	5
2-Methylphenol	10.5	Diethylphthalate	10.5	Bromomethane	10
2-Nitroaniline	52.5	Dimethyl phthalate	10.5	Carbon tetrachloride	5
2-Nitrophenol	10.5	Fluoranthene	10.5	Chlorobenzene	5
3,3'-Dichlorobenzidine	21	Fluorene	10.5	Chloroethane	10
3-Nitroaniline	52.5	Hexachlorobenzene	10.5	Chloroform	5
4,6-Dinitro-2-methylphenol	52.5	Hexachlorobutadiene	10.5	Chloromethane	10
4-Bromophenyl-phenylether		Hexachlorocyclopentadiene	10.5	cis-1,3-Dichloropropene	5
4-Chloro-3-methylphenol	10.5	Hexachloroethane	10.5	Dibromochloromethane	5
4-Chloroaniline	10.5	Indeno[1,2,3-cd]pyrene	10.5	Ethylbenzene	5
4-Chlorophenyl phenyl ethe	r 10.5	Isophorone	10.5	Methylene chloride	5
4-Methylphenol	10.5	N-Nitroso-di-n-propylamine	10.5	Styrene	5
4-Nitroaniline	52.5	N-nitrosodiphenylamine	10.5	Tetrachloroethene	5
4-Nitrophenol	52.5	Naphthalene	10.5	Toluene	5
Acenaphthene	10.5	Nitrobenzene	10.5	trans-1,2-Dichloroethene	5
Acenaphthylene	10.5	Pentachlorophenol	52.5	trans-1,3-Dichloropropen	e 5
Anthracene	10.5	Phenanthrene	10.5	Trichloroethene	5
Benzoic acid	52.5	Phenol	10.5	Vinyl Acetate	10
Benzo[a]anthracene	10.5	Pyrene	10.5	Vinyl chloride	10
Benzo[a]pyrene	10.5	1,1,1-Trichloroethane	5	Xylenes (total)	5

 $Table \ 4-3$ Concentrations of TNT, TNB, RDX, and Total Nitrobodies (in µg/L) in the Test Groundwater

	TNT	TNB	RDX	Total Nitrobodies
O		IND	KDA	Mittobodies
Optimization Task - New TRW W		10	10	10
No. of Samples	13	13	13	13
Average	733	453	54.1	1624
Std Deviation	168	124	10.7	398
Median	692	456	52.0	1520
Range	517 - 1200	134 - 711	41.5 - 74.4	978 - 2570
Optimization Task - Well #66			***************************************	
No. of Samples	9	9	9	9
Average	515	398	24.7	1083
Std Deviation	99	195	5.1	270
Median	508	461	27.6	1120
Range	335 - 645	114 - 711	14.4 - 29.7	755 - 1560
Demonstration Task Phase 1 - New	v TRW Well			
No. of Samples	93	93	93	93
Average	437	397	33.0	1005
Std Deviation	97	67	6.1	190
Median	423	395	32.9	972
Range	114 - 692	119 - 540	10.4 - 49.5	285 - 1470
Demonstration Task Phase 2 - Nev	v TRW Well			
No. of Samples	100	100	100	100
Average	312	346	22.5	758
Std Deviation	64	57	6.1	135
Median	297.5	336.5	21.8	729
Range	198 - 538	233 - 546	0.01 - 38.8	535 - 1200

4.5 SYSTEM DEBUGGING

4.5.1. Objectives

- **4.5.1.1.** The objectives of this two-week task were as follows:
 - Start up the demonstration plant
 - Ensure that all its components were fully operational
 - Calibrate all chemical feed systems
 - Test all alarms and emergency shut-down systems
 - · Check for leaks and malfunctions.
- 4.5.1.2. A description of the tests that were conducted in this task is described below.

4.5.2 System Startup

- **4.5.2.1.** Following the initial hydraulic testing done after construction was complete, tap water was pumped into the system at a flowrate of 25 gpm to fill up the six contactors with water.
- 4.5.2.2. The water was then adjusted to a flow rate of 10 gpm. The ozone system was turned on, and ozone was fed to the six contactors at 40 percent of capacity. Soap-Bubble tests were conducted on all gas-phase pipe connections outside the ozone generator, ozone monitor, and ozone destruction unit. While ozone was being fed to the system, the hydrogen peroxide feed system to the six contactors was turned on. The peroxide system was checked for any hydrogen peroxide leaks. Any leaks discovered in the ozone system or the hydrogen peroxide system resulted in shutdown and draining of the system, and the leaks were repaired. This test was repeated until both feed systems were void of detectable leaks.

4.5.2.3. After all system components were checked for leaks, tap water flow rate was increased gradually to 25 gpm, accompanied by a corresponding increase in the ozone generator setting and hydrogen peroxide feed rates to deliver the design doses of 330 mg/L ozone and 108 mg/L hydrogen peroxide. The system was operated under these conditions for a period of 30 minutes during which a final leak check was conducted on all system components. These procedures were repeated three times over a period of 2 days until all ozone and hydraulic leaks were corrected.

4.5.3 Equipment Calibration

- **4.5.3.1.** The following instruments and monitoring equipment were calibrated during this task:
 - Influent water flowmeter
 - Hydrogen peroxide metering pumps
 - Ozone monitor
- **4.5.3.2. Influent Water Flowmeter.** The influent flow meter was calibrated with tap water using a 55-gallon polyethylene drum. A total of three (3) indicated flow rates were evaluated: 10, 18, and 25 gpm. A constant flow rate was allowed through the meter. The water was diverted from the effluent of the first contactor through a flexible hose to the drain. After 10 minutes of steady flow, the water was diverted into the 55-gallon calibration drum. Time was kept using a stopwatch until the 50 gallon mark was reached. The ratio of 50 gallons divided by the fill time (in minutes) constituted the actual flowrate value in gpm. This test was repeated in triplicate for each of the three flow rates. Once the calibration curve was developed, the "actual" flow rate, instead of the "indicated" flow rate, was used in all subsequent testing.
- **4.5.3.3.** Hydrogen Peroxide Metering Pumps. Hydrogen peroxide metering pumps were calibrated according to the manufacturer's recommendation: the pump stroke was adjusted to a level that produced the desired output at an approximate speed setting of 60 percent. The output per stroke was then calculated by measuring the volume drawn

from a 1-liter graduated cylinder over a 100-stroke period. This procedure was repeated for each pump. The approximate required pump setting during testing was set by calculating the pump stroke rate required to produce the desired output. This output was then verified with a 1-Liter graduated cylinder on a daily basis.

4.5.3.4. Ozone Monitor. The ozone monitor was factory-calibrated by the manufacturer at the beginning of the study.

4.5.4 Alarm Checks

- **4.5.4.1** The Peroxone treatment system contained several operational safety alarms, including the following:
 - Low process flow alarm
 - · Overflow alarm on the first contactor
 - · Containment pad spill alarm
 - Three chemical feed tank low-level alarms
 - Ozone generator failure plant shutdown alarm
 - Ozone destruct failure alarm
 - Numerous ozone generator alarms.
- **4.5.4.2.** All of the above alarms were checked prior to system startup.

4.5.5. Summary of Problems

- 4.5.5.1. During debugging, several minor problems were identified and corrected:
 - The influent gas lines to the contactors kept filling with water that backed up through the rotameters into the tubing; water almost flowed into the header piping. To remedy this problem the gas tubing was lengthened and run up above the top of the contactor to prevent water from backing up the tubing higher than the water level in the contactor vessel.
 - Several CPVC fittings on the off-gas analyzer tubing connections began to dissolve from contact with the ozone gas mixture. These were replaced with teflon or stainless steel fittings.

- 3. The metering pumps were inconsistent in flowrate. The problem was diagnosed as a loss in prime to the pumps. This was remedied by lowering the feed piping from the hydrogen peroxide day tanks and installing back-pressure control valves on the pump intake lines. This correction prevented a break in suction to the pumps, and prevented them from losing their prime.
- 4. The ozone generator had two failure episodes where the control panel showed a high DC voltage alarm. This problem was traced to the main power source which was delivered at 500 volts rather than the 480 volt service requested. The over-voltage burned out several components in the control panel which resulted in the need to bring a representative from the manufacturer to the site for repairs and calibration. Repairs were made and the inlet voltage was adjusted to a proper operating level within the DC transformer that fed the ozone generator vessel. Repairs to the generator were made 27 and 28 August 1996.
- 5. The north well pump (Well No. 66) kept shutting down with an overload. The motor starter was adjusted to a higher amperage trip-out and restarted. The pump ran without interruption, but the influent groundwater stream was filled with air bubbles after a few minutes of operation. The water level was checked in the well and it revealed that the well was not producing enough water to maintain a flow of 25 gpm as expected. The well was pressurized to improve the yield and the pump was lowered slightly. The remedy for this well was to reduce the flowrate to about 15 gpm to maintain a sustained flow.
- 6. The rotameters controlling the gas flow into the contactors were beginning to cloud up and could not be read easily. The acrylic bodies were not holding up to the concentration of ozone in the feed gas. This was remedied by replacing the original rotameters with glass-bodied units.
- 7. The gas flow into the contactors was not producing the fine bubble mist in the water that was anticipated. Closer inspection revealed that the manufacturer had sent the wrong kind of gaskets to seal the diffuser stones to the gas header in the contactors, and that two of the twelve stones were cracked and defective. This resulted in ozone leaking around the connections and forming

large bubbles in the tanks. The problem was remedied by replacing the defective stones and gaskets.

4.6 SYSTEM OPTIMIZATION

4.6.0.1. The objective of the system optimization phase was to run the Peroxone system under varying conditions of water source, water flow rate (i.e., varying reaction times), and ozone doses to determine the impact of these variable conditions on the destruction of TNT, TNB, RDX, and total nitrobodies through the system. The following is a discussion of the experimental conditions used and results obtained.

4.6.1 Experimental Conditions

4.6.1.1. The experimental conditions evaluated during the two-week optimization program are outlined in Table 4-4. A total of 10 tests were conducted. Six of these tests (Tests #1 through #4 and Tests #9 and #10) were conducted on New TRW Well water, while the remaining four tests (Tests #5 through #8) were conducted on Well #66 water. The flow rates tested were 13 gpm, 18 gpm, and 25 gpm, which equate to average hydraulic retention times (HRT) of 46 minutes, 33 minutes, and 24 minutes, respectively, in each of the six contactors. The total applied ozone dose ranged from a low of 228 mg/L (38 mg/L per contactor) to a high of 690 mg/L (115 mg/L per contactor). To achieve these doses, the percent ozone in the oxygen feed gas ranged from a low of 3.3 percent to a high of 10.5 percent. The applied ozone dose was varied by changing the ozone concentration in the feed gas while maintaining a constant feed gas flow rate. Note that the ozone generator was designed to deliver 55 mg/L ozone per contactor at 25 gpm flow rate. Therefore, in order to increase the applied ozone dose to greater than 55 mg/L, the water flow rate had to be decreased below its design value of 25 gpm, which in turn increased the HRT value through the contactors. Therefore, for ozone doses greater than 55 mg/L, two variables - ozone dose and contact time - had to be varied simultaneously, which is not ideal for an optimization testing program. Therefore, when analyzing the results of the study, it is important that comparisons be made between tests that differed by only one variable at a time.

4.6.1.2. As shown later in this report, two of the independent variables, hydrogen peroxide dose and water source, had little to no effect on the performance of the Peroxone process for explosives treatment within the range of values tested. The only two remaining independent variables were 1) water flow rate, and 2) ozone dose (which is a direct result of changing the percent ozone in the feed gas stream). Therefore, it was realized that simultaneously changing these variables during the optimization testing was not an ideal experimental approach. However, as mentioned above, the limitation of the ozone generator capacity forced the project team to lower the flow rate in order to achieve a higher ozone dose. In order to overcome this shortcoming, the project team relied on mathematical modeling as opposed to direct analysis of the results. Thus, an empirical mathematical model was developed specifically for this project. The model, which is presented and discussed later in this report, focused on both the ozone dose and the hydraulic behavior (including the water flow rate) in interpreting and sorting through all the results of the optimization and demonstration tasks. Once the model was calibrated with the experimental results, it was then used to optimize the design of the 1000-gpm facility. The project team believes that this approach eliminated the concern over the impact of simultaneous variation of the ozone dose and water flow rate through the system on the ability to interpret the experimental results.

4.6.1.3. During Tests #1 through #8, the hydrogen peroxide feed rate was varied with the ozone dose in order to maintain a mass Peroxone ratio of approximately 0.3 (i.e., 0.3 mg hydrogen peroxide per mg of ozone transferred to the water). This ratio was based on the stoichiometry of reaction between ozone and hydrogen peroxide. However, the ozone residual concentration measured in the effluent of each contactor was substantially higher than expected (greater than 1 mg/L). Accordingly, the Peroxone ratio during Tests #9 and #10 had to be increased to approximately 0.54 and 0.47 respectively, in order to maintain the ozone residual in the effluent of each contactor at less than 1 mg/L. There is no explanation at this point as to why this ratio is significantly higher than the commonly used stoichiometric ratio of 0.3. However, it is important to note that the ozone doses

All H₂O₂/O₃ ratios or Peroxone ratio presented in this report are based on a mass ratio of hydrogen peroxide dose to transferred ozone dose. A mass ratio of 0.3 mg/mg is equivalent to a molar ratio of 0.42 mole/mole.

used in this treatment system are almost two orders of magnitude higher than those used in conventional ozone applications in drinking water treatment, where the bulk of the industry's understanding of ozone/hydrogen peroxide reaction chemistry was developed. It is likely that the reactions at such high ozone doses may vary from those experienced at the low ozone doses, resulting in an increase in the optimum Peroxone ratio.

4.6.2 Experimental Results

4.6.2.1. The results obtained during the optimization period are listed in Appendix B.² The data show that TNB was the critical compound in that it was the most difficult to oxidize compared to TNT or RDX. Table 4-5 shows a summary of the results for New TRW Well. The objective of this table is to show the impact of ozone dose and HRT on the average percent removal of TNB through each of the six contactors. For example, Tests #1 and #10 had similar average transferred ozone doses to each contactor (within 10% difference). However, the contact time through each contactor in Test #10 was 46 minutes compared to 24 minutes in Test #1. Although the contact time was doubled while maintaining the same transferred ozone dose, the average percent removal only increased from 39 percent to 49 percent. This is primarily due to the fact that each contactor is completely mixed as will be shown and discussed later in this section. It is also noted that the ozone transfer efficiency varied from a low of 62% to a high of 82%.

4.6.2.2. On the other hand, comparison of Tests #4, #9, and #10 shows the impact of increased transferred ozone dose on TNB removal at a constant average HRT of 46 minutes through each contactor. As the dose was increased from 31 mg/L, to 42 mg/L, to 80 mg/L, the percent TNB removal increased from 36%, to 49%, to 62%, respectively.

As noted earlier, the QA/QC results for explosives analysis are included in the Independent Evaluator's report for ESTCP titled: "Peroxone Demonstration: Performance and Cost Evaluation"

Table 4-4

Experimental Conditions Used During the System Optimization Program

Test #	Well	Flow gpm	HRT min	Total Applied O ₃ Dose mg/L	Percent Ozone	H ₂ O ₂ Dose mg/L	Peroxone Ratio* (mg/mg)
1	New TRW	25	24	360	10.1%	87	0.39
2	New TRW	18	33	390	7.9%	90	0.30
3	New TRW	18	33	510	10.2%	108	0.30
4	New TRW	13	46	690	10.5%	135	0.27
9	New TRW	13	46	228	3.3%	101	0.54
10	New TRW	13	46	336	4.9%	119	0.46
5	#66	18	33	258	5.5%	54	0.31
6	#66	18	33	390	7.9%	90	0.31
7	#66	18	33	510	10.2%	108	0.29
8	#66	13	46	690	10.5%	135	0.27

^{*} The Peroxone Ratio is calculated as the ratio of hydrogen peroxide dose (in mg/L) to the transferred ozone dose (in mg/L)

Table 4-5
Summary of the Optimization Results for TNB Removal From New TRW Well Water

Test #	Ave. Transferred O_3 Dose/Chamber mg/L	Ozone Transfer Efficiency	Ave. HRT min	Average C _{eff} /C _{inf}	Average % TNB Removal
1	37	62%	24	0.61	39%
2	48	74%	33	0.62	38%
3	57	67%	33	0.35	65%
4	80	70%	46	0.38	62%
9	31	82%	46	0.64	36%
10	42	75%	46	0.51	49%

4.6.2.3. The results from Tests #2 and #3 are unusual in that the changes in TNB removal do not reflect the changes in the transferred ozone dose and/or hydraulic retention time. For example, comparing the operating conditions of Test #2 to those of Test #1 show an

increase in the average transferred ozone dose from 37 mg/L to 48 mg/L per contactor with a parallel increase in the hydraulic contact time from 24 minutes to 33 minutes. However, there was no change in the average percent removal of TNB. Similarly, while the average transferred ozone dose and average contact time in Test #3 were significantly lower than those in Test #4, the average percent TNB removal was higher in Test #3. Examination of the raw data sheets from Tests #2 and #3 shows a substantial scatter in the TNB removal data, which may explain the observed anomalies in the summary results presented in Table 4-5.

- **4.6.2.4.** These observations are important because they have direct implications to the design of the full-scale Peroxone treatment system. For example, based on the percent removals listed in Table 4-5, Table 4-6 lists various configurations of Peroxone treatment system required to achieve 99% removal of TNB. It is important to note that the configurations listed in Table 4-6 are simulated based on the average percent TNB removal observed during the optimization task.
- **4.6.2.5.** Some interesting observations are made from the configurations listed in Table 4-6. The first and fourth configurations have virtually identical total hydraulic residence times, which means the same size contactor. However, increasing the number of chambers from five to nine, and reducing the contactor HRT from 46 minutes to 24 minutes, increased the contactor efficiency and reduced the required total transferred ozone dose from 400 mg/L to 333 mg/L, approximately 17% reduction in required ozone capacity.
- **4.6.2.6.** Four optimization tests were also conducted on Well #66 water. The summary of these results are listed in Table 4-7. The results from Test #6 were highly scattered, and are thus not listed in Table 4-7.

Table 4-6
Simulated Peroxone Treatment System Configurations
That Would be Required to Achieve 99% Removal of TNB

Number of Chambers	HRT/Chamber minutes	Total HRT hrs	Transf. Ozone Dose/Chamber mg/L	Total Transf. Ozone Dose, mg/L
9	24	3.6	37	333
10	46	7.7	31	310
7	46	5.4	42	294
5	46	3.8	80	400

Note: The values listed in this table were estimated based on linear extrapolation of the experimental results reported in Table 4-5.

Table 4-7
Summary of the Optimization Results for TNB Removal
From Well #66 Water

Test #	Ave. Transferred O ₃ Dose/Chamber mg/L	Ave. HRT min	$\begin{array}{c} \textbf{Average} \\ \textbf{C}_{\text{\tiny eff}} / \textbf{C}_{\text{\tiny inf}} \end{array}$	Average % TNB Removal
5	27	33	0.82	18%
7	60	33	0.74	26%
8	80	46	0.38	62%

4.6.2.7. Due to the anomalies observed in Tests #2 and #3 during New TRW Well testing (as discussed earlier in paragraph 4.6.2.3), the results of Tests #5 and #7 cannot be reliably used to compare the performance of the Peroxone treatment system on Well #66 water to that on New TRW Well water. However, the operational conditions of Test #8 (i.e., transferred ozone dose and average contact time) using Well #66 water were similar to those of Test #4 using New TRW Well water. The corresponding TNB removal in the two waters was identical at 62%, suggesting that the performance of the Peroxone treatment system was independent of the water quality differences between New TRW Well and Well #66. It is interesting to note that the results suggest that the HRT had a more significant impact on TNB removal than the ozone dose.

4.7 SYSTEM DEMONSTRATION

4.7.0.1. Based on the results of the optimization testing, two sets of operating conditions were selected for the demonstration testing. In addition, after discussions with the various project members, it was decided that the demonstration testing was only to be conducted on New TRW Well water (due to its low yield, and the similarity in the Peroxone performance for explosives' oxidation in both waters).

4.7.1 Experimental Conditions

4.7.1.1. For the first four weeks of the demonstration period (Phase I), the Peroxone system was operated under conditions predetermined to achieve the target water quality goals of 0.002 mg/L of each of TNT, TNB, and RDX, and 0.3 mg/L of Total Nitrobodies. The operating conditions for Phase I demonstration testing are listed in Table 4-8.

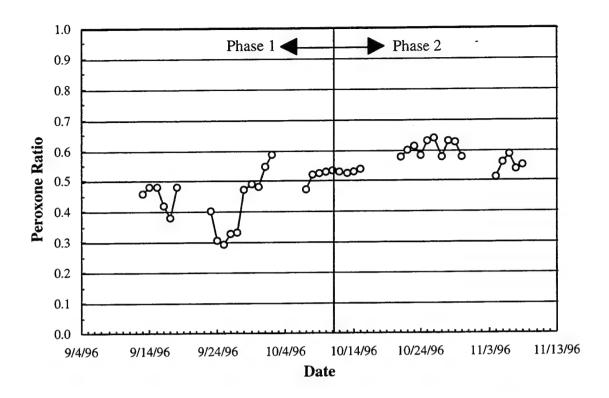
During the second four weeks of the demonstration period (Phase II), the Peroxone system was operated at higher flowrate (i.e., lower contact time), and lower ozone dose than those used in Phase I. The experimental conditions for Phase II testing are also listed in Table 4-8. Based on the system optimization results, it was clear that the conditions used in Phase II were not going to achieve the target effluent concentrations. However, the project team decided to evaluate these conditions with the idea that a hybrid system of a PEROXONE process operated under these conditions followed by a GAC adsorption process for complete contaminants removal may actually be more cost effective than a PEROXONE process alone. However, it is noted that this approach does not address the possible formation of oxidation by-products which may consume the GAC capacity more rapidly. During Phases I and II testing, the ozone transfer efficiency was approximately 78% and 76%, respectively.

Table 4-8
Operating Conditions During Phases I and II
of the Demonstration Testing Period

		Val	ue
Parameter	Unit	Phase I	Phase II
Well Number		New TRW	New TRW
Water Flowrate	gpm	13	25
Total Ave. Contact Time	hrs	4.6	2.4
Applied Ozone Dose	mg/L	100	58
	J	(95 to 115)	(55 to 60)
Transferred Ozone Dose	mg/L	78	44
	0	(72 to 92)	(42 to 47)
Transfer Efficiency	Percent	78%	76%
H,O, Dose	mg/L	35	25
11202 2000		(24 to 46)	(24 to 28)
Peroxone Ratio	mg/mg	0.45	0.57
1 Olohollo Radio	8'8	(0.29 to 0.59)	(0.51 to 0.64)

Note: numbers in parentheses represent minimum and maximum values.

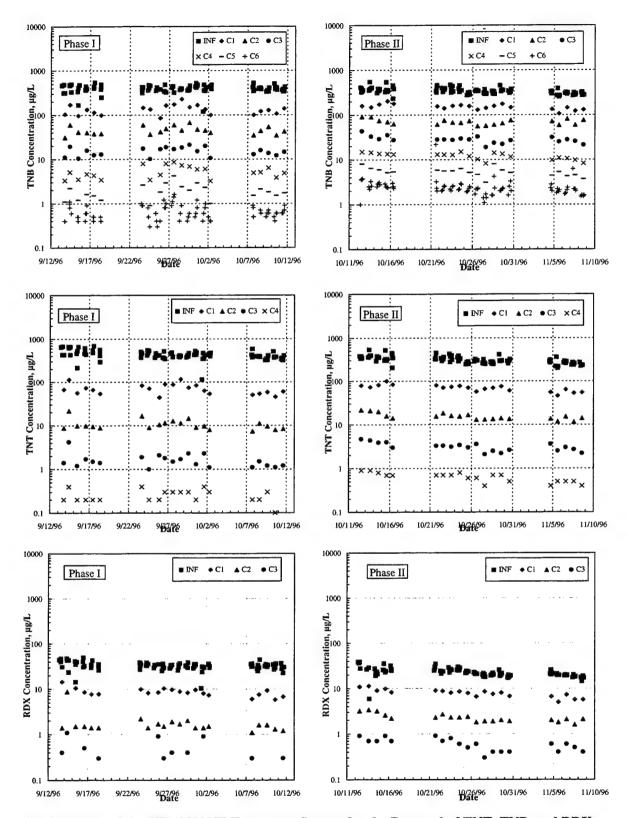
4.7.1.2. It should be noted that the Peroxone mass ratio during Phase I of the demonstration period was increased from a low of 0.29 to a high of 0.59 mg/mg. This is due to the fact that when the demonstration period started, the target Peroxone ratio was still at 0.3 (as stated in the RFP documents). However, it was then decided that the hydrogen peroxide dose (and thus the Peroxone ratio) should be increased until the ozone residual concentration in the effluent of the contactors was at levels less than 1 mg/L. Accordingly, the average Peroxone ratio was increased to approximately 0.55. Figure 4-3 shows the profile of the average Peroxone ratio (among the six contactors) throughout the demonstration period.



Peroxone Ratio During Phases 1 and 2 of the Demonstration Period
Figure 4-3

4.7.2 Experimental Results

4.7.2.1 Explosives Removal. The results of the demonstration period are summarized in Appendix C of this report. The data gathered during this period are shown graphically in Figure 4-4 for Phase I and Phase II of the demonstration period. The results show that with an average HRT of 4.6 hours and an average transferred ozone dose of 78 mg/L per contactor during Phase I of the demonstration period, all influent concentrations of TNT, TNB, and RDX were reduced to less than 2 μ g/L in the effluent of the Peroxone treatment system, with TNB being the most difficult compound to remove. When the contact time and transferred ozone dose were reduced to 2.4 hrs and 44 mg/L, respectively (in Phase II), the effluent concentrations of TNT and RDX were still lower than the target level of 2 μ g/L. The effluent concentration of TNB ranged from 2 to 4 μ g/L. While this was higher than its target level of less than 2 μ g/L, it represents significant removals considering the substantially lower ozone dose and contact time used. This result suggests that a hybrid system of a Peroxone process for partial contaminants removal followed by another



Performance of the PEROXONE Treatment System for the Removal of TNT, TNB, and RDX During Phase I and Phase II of the Demonstration Period

Figure 4-4

process (such as GAC adsorption) for final treatment may be more economically feasible than a stand-alone Peroxone process for complete treatment.

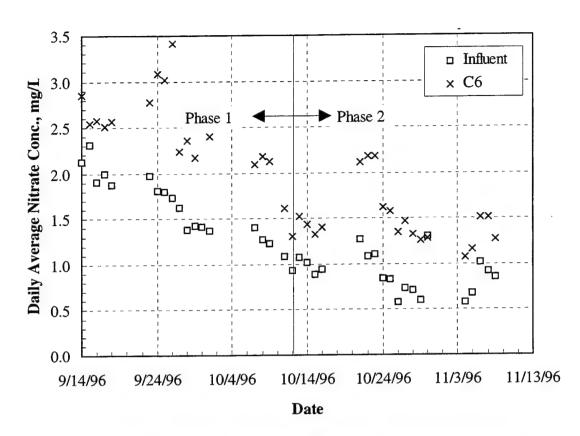
4.7.2.2. It is interesting to note that the removal of TNB (as well as the other compounds) did not substantially change throughout Phase I of the demonstration period, despite the fact that the Peroxone ratio increased from a low of 0.27 mg/mg to a high of 0.59 mg/mg. This suggests that the oxidation of the target contaminants was not limited by the concentration of hydroxyl radicals (or other highly reactive radicals) in the treatment process, but rather by the rate of reaction between these radicals and each of the target contaminants.

4.7.2.4. Nitrate Formation. The influent and effluent water to and from each contactor was also analyzed for nitrate concentration. The results of the nitrate analysis are shown in Figure 4-5 for the influent water and the last contactor (C6) effluent. The results clearly show an increase in the concentration of nitrate through the Peroxone treatment system. As a daily average value, the nitrate concentration increased by an average of 0.86 mg/L during Phase I testing period, and by 0.60 mg/L during Phase II testing period. It is clear that the higher ozone dose and contact time used during Phase I testing resulted in the higher formation of nitrate. There are two potential sources for the additional nitrate:

1. Oxidation of ammonia-nitrogen to nitrate-nitrogen via the following half-reaction:

$$NH_3 + 3H_2O \rightarrow NO_3 + 9H^+ + 8e^-$$

2. Oxidation of the nitrogen in the organic nitrobodies (i.e., TNT, TNB, RDX, etc.) to nitrate.



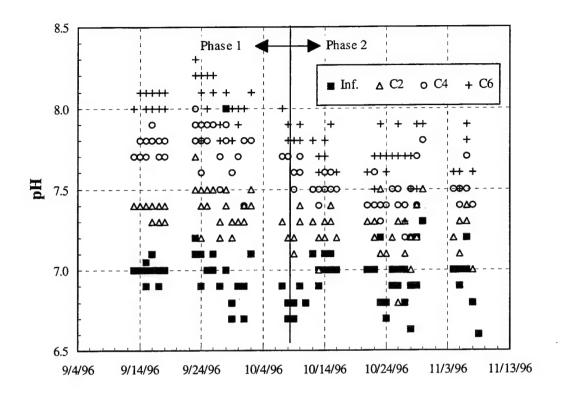
Formation of Nitrate Through the Peroxone Treatment System
Figure 4-5

4.7.2.5. A few measurements were made of the ammonia concentration in the influent and effluent waters to and from the Peroxone treatment system. The average concentration of ammonia in New TRW Well water was 0.29 mg/L, whereas the average concentration of ammonia in the effluent of the treatment system was approximately 0.17 mg/L. This translates into an equivalent increase in nitrate concentration of approximately 0.44 mg/L. However, it should be noted that the influent and effluent ammonia measurements were not made on the same day, and therefore, the calculated ammonia removal may not be accurate. If it is assumed that all the ammonia (0.29 mg/L) was converted to nitrate, the corresponding increase in nitrate concentration would be estimated at 1.06 mg/L (see chemical half-reaction in paragraph 4.7.2.4). Therefore, the oxidation of ammonia-nitrogen to nitrate-nitrogen may account for the majority, if not all of the increase in nitrate concentration in the water (due to the high solubility of ammonia in water, no significant volatilization of ammonia is expected).

4.7.2.6. The complete oxidation of organic nitrobodies can convert the organic nitrogen into inorganic nitrate-nitrogen. Assuming that complete oxidation did occur in the Peroxone treatment system, an influent TNT concentration of 0.5 mg/L would result in the formation of 0.41 mg/L nitrate.[‡] Therefore, considering all the other organic nitrobodies in the influent water, the oxidation of the organic nitrogen to inorganic nitrate-nitrogen can also account for all of the measured increase in nitrate concentration. Therefore, no conclusion can be made regarding the exact source of nitrogen that was converted to nitrate.

4.7.2.7. pH & ORP Measurements. Daily samples were collected from the effluent of each of the six contactors and analyzed for pH and oxidation-reduction potential (ORP). The profiles of the pH in the influent and effluent of the Peroxone treatment system are shown in Figure 4-6. The results show that the average pH of the influent groundwater was between 6.5 and 7.0. As the water went through each of the six contactors, the pH increased to 7.1, 7.3, 7.5, 7.6, 7.8, and 7.9, respectively. No specific testing was conducted to determine the cause of the pH drift. It may be due to CO₂ stripping from the groundwater during treatment, or a result of the reaction between ozone and hydrogen peroxide. The ORP results are listed in Appendix C. In general the ORP of the water increased from an average of 400 mV in the influent water to approximately 900 mV in the effluent of the sixth contactor. This increase in the ORP level is expected considering the high doses of oxidants (ozone and hydrogen peroxide) added to the water.

 $^{^{+}}$ $C_{7}H_{5}(NO_{2})_{3} + 17H_{2}O \rightarrow 3NO_{3}^{+} + 7CO_{2} + 39H^{+} + 36e^{-}$



Profile of pH in the Influent Water and Effluent Waters from Contactors #2, #4, and #6 During the Demonstration Period

Figure 4-6

4.8 FORMATION OF OZONATION BY-PRODUCTS

4.8.0.1. Ozonation of natural water is known to produce several inorganic and organic by-products. These include bromate (in bromide-containing waters), aldehydes, haloacetic acids, and other compounds. In order to determine the levels of ozonation by-products formed by the Peroxone process, two water samples were collected from the influent and effluent of the treatment system during Phase II of the demonstration period and analyzed for a wide range of organic compounds. The types and concentrations of the analyzed organic compounds in the two samples are listed in Table 4-9. The results show that, of the analyzed compounds, only one compound, formaldehyde at $11 \mu g/L$, was present in the effluent of the Peroxone system. Considering that formaldehyde is

highly biodegradable, however, it is anticipated that natural biodegradation of this compound will occur shortly after discharge of the treated water into the environment. Interestingly, trichlorotrifluoroethane (Freon) was measured at $66.3~\mu g/L$ in the New TRW Well water. However, this compound was removed by the treatment system to levels less than its detection limit of $0.5~\mu g/L$.

4.8.0.2. It should also be noted that the influent and effluent samples were analyzed for total organic carbon (TOC) concentration. The influent water sample had a TOC concentration of 2.2 mg/L, whereas the effluent sample had a TOC concentration of 0.8 mg/L. This represents approximately 64% removal of the organic carbon. The removal mechanism is believed to include the oxidation of the organic carbon to inorganic carbon (i.e., CO₂) as a result of the extremely high ozone doses added to the system, and the formation of elevated levels of the highly reactive free radicals.

Table 4-9

Types and Levels of Organic Chemicals in the Influent and Effluent of the Peroxone Treatment System

	Level, µg/L		Level,	μg/L
Chemical	Inf. Eff.	Chemical	Inf.	Eff.
Aldehydes:		Bromodichloromethane	<0.5	<0.5
Aetaldehyde	<1.0 <1.0	Benzene	< 0.5	< 0.5
Butanal	<1.0 <1.0	Bromobenzene	< 0.5	< 0.5
Formaldehyde	<5.0 11	Bromochloromethane	< 0.5	< 0.5
Glyoxal	<1.0 <1.0	Bromomethane	< 0.5	< 0.5
M-Glyoxal	<1.0 <1.0	cis-1,2-Dichloroethene	< 0.5	< 0.5
Pentanal	<1.0 <1.0	Chlorobenzene	< 0.5	< 0.5
Propanal	<1.0 <1.0	Carbon tetrachloride	< 0.5	< 0.5
Haloacetic Acids:		cis-1,2-Dichloropropene	< 0.5	< 0.5
Bromochloroacetic acid	<1.0 <1.0	Bromoform	< 0.5	< 0.5
Bromodichloroacetic acid	1 < 1.0 < 1.0	Chloroform	< 0.5	< 0.5
Chlorodibromoacetic acid		Chloroethane	< 0.5	< 0.5
Dibromoacetic acid	<1.0 <1.0	Chloromethane	< 0.5	< 0.5
Dichloroacetic acid	<1.0 <1.0	Dibromochloromethane	< 0.5	< 0.5
Monobromoacetic acid	<1.0 <1.0	1,2-Dibromo-3-Chloropropar	ne <1.0	<1.0

Table 4-9

Types and Levels of Organic Chemicals in the Influent and Effluent of the Peroxone Treatment System (Continued)

	Lev	el, μg/L		Level,	μg/L
Chemical	Inf.	Eff.	Chemical	Inf.	Eff.
Monochloroacetic acid	<2.0	<2.0	Dibromomethane	<0.5	<0.5
Tribromoacetic acid	<1.0	<1.0	Dichlorodifluoromethane	< 0.5	< 0.5
Trichloroacetic acid	<1.0	<1.0	1,2-Dibromoethane	< 0.5	< 0.5
Volatile Organic Compound	ts:	Ethylb	enzene	< 0.5	< 0.5
1,1,1,2-Tetrachloroethan	e<0.5	< 0.5	Hexachlorobutadiene	< 0.5	< 0.5
1,1,1-Trichloroethane	< 0.5	< 0.5	Isopropylbenzene	< 0.5	< 0.5
1,1,2,2-Tetrachloroethan	e<0.5	< 0.5	Methylene Chloride	< 0.5	< 0.5
1,1,2-Trichloroethane	< 0.5	< 0.5	m+p-Xylenes	< 0.5	< 0.5
1,1-Dichloroethane	< 0.5	< 0.5	Methyl tert-butyl ether	< 5.0	< 5.0
1,1-Dichloroethene	< 0.5	< 0.5	Naphthalene	< 0.5	< 0.5
1,1-Dichloropropene	< 0.5	< 0.5	n-Butylebenzene	< 0.5	< 0.5
1,2,3-Trichloropropane	< 0.5	< 0.5	n-Propylbenzene	< 0.5	< 0.5
1,2,4-Trichlorobenzene	< 0.5	< 0.5	Tetrachloroethene	< 0.5	< 0.5
1,2,4-Trimethylbenzene	< 0.5	< 0.5	p-Isopropyltoluene	< 0.5	< 0.5
1,3-Dichlorobenzene	< 0.5	< 0.5	sec-Butylbenzene	< 0.5	< 0.5
1,3-Dichloropropane	< 0.5	< 0.5	Styrene	< 0.5	< 0.5
1,4-Dichlorobenzene	< 0.5	< 0.5	trans-1,2-Dichloroethene	< 0.5	< 0.5
2,2-Dichloropropane	< 0.5	< 0.5	tert-Butylbenzene	< 0.5	< 0.5
2-Chlorotoluene	< 0.5	< 0.5	Trichloroethene	< 0.5	< 0.5
4-Chlorotoluene	< 0.5	< 0.5	Trichlorotrifluoroethane	66.3	< 0.5
trans-1,3-Dichloroproper	ne<0.5	< 0.5	Toluene	< 0.5	< 0.5
Trichlorofluoromethane			Vinyl Chloride	< 0.3	< 0.3

4.9 MATHEMATICAL MODELING

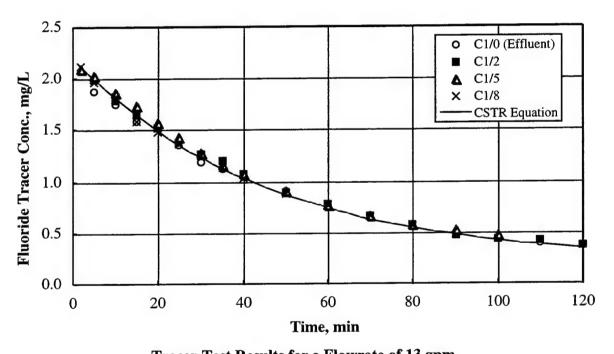
4.9.0.1. The scope of work for this project did not include the development of a mathematical model for the Peroxone treatment system. However, Montgomery Watson believes that such a model can be an effective tool for optimizing the design of any future large-scale Peroxone treatment system for the removal of TNT, TNB, and RDX from contaminated groundwaters. Such an optimized design results in a cost-effective treatment system.

4.9.1. Characterization of System Hydraulics (Tracer Testing)

- **4.9.1.1.** In order to develop a mathematical model for a continuous flow process, such as the Peroxone treatment system, it is imperative that the hydraulic residence time distribution of the system be fully characterized. This was accomplished by conducting two tracer tests on the first contactor at two water flowrates, 13 gpm and 25 gpm. The results of the tracer tests were then used to mathematically describe the hydraulic behavior of the Peroxone demonstration system.
- 4.9.1.2. Tracer Testing Methodology. Fluosilicic acid was used as the tracer chemical, with fluoride being the conservative tracer ion. A 25% Fluosilicic acid solution was purchased from VWR Scientific. A total of 22.3 grams of the tracer were diluted to 2 liters for both the 13 gpm test and the 25 gpm test. Based on a 79% fluoride content in Fluosilicic acid, the fluoride mass injected was 4.4 grams. The tracer solution was then injected through an injection port installed in the influent line to the first contactor, immediately before the water enters the top of the contactor. Tap water was pumped into the system during this test. The oxygen flowrate through the contactor was maintained at 1.5 scfm. However, no ozone was added to the influent stream in order to prevent possible interference with the fluoride analytical method.
- **4.9.1.3.** At time zero, the tracer solution was injected into the influent water stream. Water samples were then collected from three taps along the depth of the first contactor (at 2 ft, 5 ft, and 8 ft from the bottom of the contactor), as well as from the effluent of the contactor at various time intervals. The sampling was continued over a period of time equivalent to three HRTs of the contactor. In addition, samples of the influent were collected throughout the testing period to obtain a good estimate of the background fluoride concentration in the water. Using an ion-selective electrode, all samples were analyzed on site for fluoride concentration.

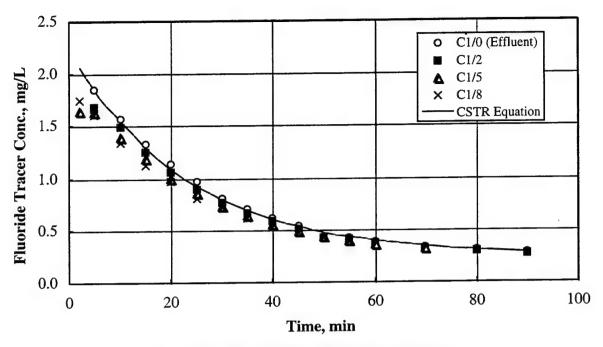
4.9.1.4. Tracer Testing Results

The results of the two tracer tests are shown in Figures 4-7 and 4-8. Overlayed on each graph is the theoretical tracer result that would be obtained if the contactor is simulated by a completely stirred tank reactor (CSTR) of equal hydraulic retention time. The CSTR model line is virtually on top of the experimental results obtained from all taps sampled. This shows that, for all practical purposes, each contactor in the Peroxone system behaved as a CSTR.



Tracer Test Results for a Flowrate of 13 gpm

Figure 4-7



Tracer Test Results for a Flowrate of 25 gpm

Figure 4-8

4.9.2. Model Development

4.9.2.1. With the hydraulic behavior of each contactor in the Peroxone system well characterized, a basic rate equation is required to complete the model development. Based on our experience with other oxidation reactions, a pseudo first-order reaction in explosive concentration as a function of time is a likely representation for the destruction of each of TNT, TNB, and RDX in the Peroxone system. In addition, the reaction rate constant is assumed to be proportional to the transferred ozone dose. Therefore, the resulting rate equation is expressed as follows:

$$rate = \frac{dC}{dt} = -k D^m C \tag{4-1}$$

where, k = basic reaction coefficient, (mg/L)^{-m} (min)⁻¹,

m = empirical constant,

D = transferred ozone dose, mg/L, and

C = concentration of target contaminant, μg/L (i.e., TNT, TNB, or RDX)

4.9.2.2. The mass balance equation on a CSTR operating under steady-state conditions is:

$$C_{inf} - C_{eff} + (rate)\tau = 0 ag{4-2}$$

where, C_{inf} = influent contaminant concentration, $\mu g/L$,

 C_{eff} = effluent contaminant concentration, $\mu g/L$,

 τ = average hydraulic retention time in the contactor, minutes.

4.9.2.3. Substituting Equation 4-1 into Equation 4-2, and deriving an expression for C_{eff} gives Equation 4-3 describing the performance of each of the six contactors in the Peroxone system:

$$C_{eff} = \frac{C_{inf}}{\left(1 + k D^m \tau\right)} \tag{4-3}$$

4.9.2.4. This model suggests that the effluent concentration of TNT, TNB, or RDX from any of the six contactors can be calculated if its influent concentration is known, along with the transferred ozone dose to the contactor, the contactor average hydraulic retention time, and the two model constants, k and m.

4.9.3 Calibration of Model Parameters

- **4.9.3.1.** In the design of a full-scale system, all model parameters are known except for the basic reaction coefficient 'k' and the empirical constant 'm'. Therefore, the experimental results obtained during the optimization and demonstration testing programs were used to estimate the values of 'k' and 'm' for each of TNT, TNB, and RDX in the Peroxone process.
- **4.9.3.2.** To achieve this, all the experimental results obtained in this project were tabulated. The data included the following parameters:

- Water flow rate
- Average hydraulic retention time through each contactor
- Transferred ozone dose to each contactor
- Measured influent concentration of TNT, TNB, and RDX to each contactor
- Measured effluent concentration of TNT, TNB, and RDX from each contactor.

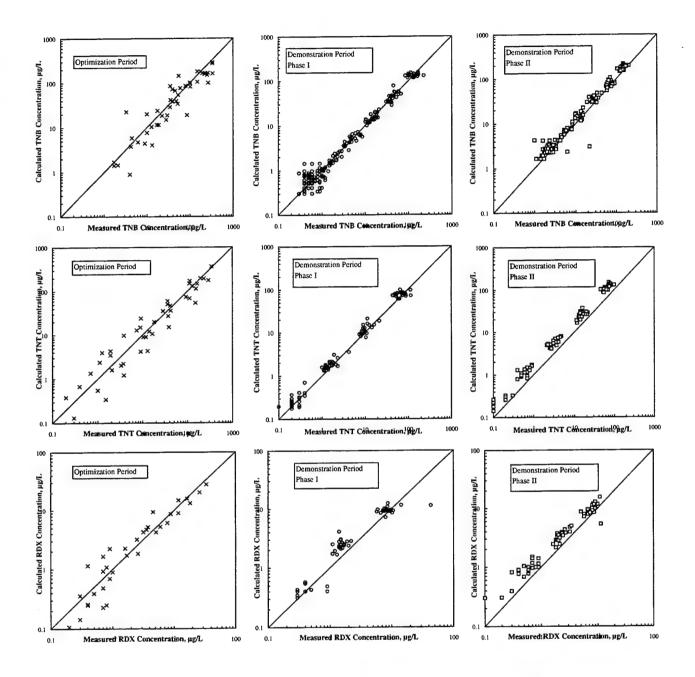
4.9.3.3. Using the values of the transferred ozone dose, hydraulic retention time, and influent contaminant concentration, Equation 4-3 was used to calculate the effluent concentration of TNT, TNB, and RDX for an assumed value of the basic reaction coefficient, k, and the empirical constant, m, for each compound. The calculated concentrations were then compared to the measured values. Using the SOLVER macro in Microsoft EXCEL, the optimum k and m values resulting in the minimum sum of the square of the error between the calculated and measured concentrations were determined for each contaminant. These values are listed in Table 4-10. It is important to note that these values are only applicable to the destruction of TNT, TNB, and RDX in Grand Island groundwater, and may vary significantly with changes in water quality and water source.

Table 4-10

Estimated Values of the Basic Reaction Rate Constants and Empirical Constants for the Oxidation of TNT, TNB, and RDX with Peroxone

Compound	k	m
TNB	0.0152	0.237
TNT	0.00569	0.662
RDX	0.0544	0.000

4.9.3.4. The quality of fit between the calculated concentrations (using the proposed model), and the measured concentrations from the effluent of each of the six contactors during the optimization and demonstration period are shown in Figure 4-9. The lines in Figure 4-9 are not linear regression lines through the data, but rather the "perfect fit"

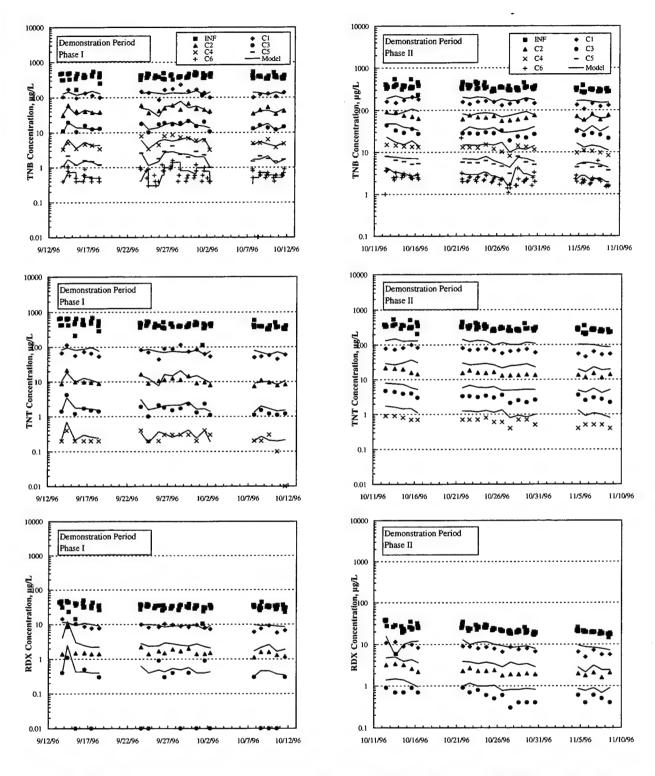


Comparison Between Measured and Model-Calculated Concentrations of TNB, TNT, and RDX During the Optimization Period and the Demonstration Period

Figure 4-9

lines. In other words, the lines represent the ideal situation where the model-calculated values are equal to the measured values. If all the data points fall on the "perfect fit" lines, then the model is considered to be the "perfect" model to represent these data.

- **4.9.3.5.** The model fit to the results of the demonstration testing is also presented in Figure 4-10 as plots of measured concentrations and model-calculated concentrations as a function of time. These graphs also show that the model was able to well represent the removal of TNB, TNT, and RDX during Phase I of the demonstration period, as well as that of TNB during Phase II of the demonstration period. However, the graphs show that the model underestimated the removal of TNT and RDX during Phase II of the demonstration period.
- **4.9.3.6.** No explanation for this underestimation can be given at this time. Nevertheless, the model is still a useful tool for the design of larger scale Peroxone treatment systems for the following reasons:
 - The model well predicted the removal of the critical design compound, TNB,
 in all the tests conducted at the various contact times and ozone doses
 - The use of the model for TNT and RDX removal would, in the worst case, result in a conservative design, thus maintaining the required removals of these compounds.
- 4.9.3.7. It should be noted that the value of the empirical constant 'm' for RDX was estimated by the model at zero suggesting that RDX removal through the Peroxone process is independent of ozone dose. This is clearly not realistic since the removal efficiency of all compounds increased with increasing ozone dose. This model behavior is primarily due to the lack of sufficient high-concentration RDX data for a wide range of ozone doses and contact times (all RDX data are below 40 μ g/L in the optimization testing, and below 16 μ g/L in the demonstration testing compared to greater than 250 μ g/L for TNT and TNB).



Model Fit to the Measured Concentrations of TNB, TNT, and RDX During Phase I and II of the Demonstration Period

Figure 4-10

4.9.3.8. The plots presented above in Figure 4-9 show some scatter around the perfect-fit lines. This scatter is due to several known and unknown factors involved in experimental work and mathematical modeling such as the following:

Experimental Errors. In any field (or laboratory) testing program, experimental errors are inevitable. These include errors in the measurement of operational parameters such as water flowrate, air flowrate, ozone dose, hydrogen peroxide dose, etc.

Analytical errors. All field and laboratory analyses conducted on the project include some level of analytical error that is attributed to instrument calibration, analytical technique, etc.

Variations in the Influent Concentrations of the Target Contaminants. Significant variations were measured in the groundwater concentrations of TNT, TNB, and RDX. Since the model is based on calculating the effluent concentrations as a function of the influent concentrations, these variations in the groundwater concentration levels have a direct impact on the model's ability to accurately predict the effluent concentrations. It is noted that, in the calibration of the model, the average groundwater concentrations measured during each test were used as the influent concentrations to the first contactor.

Empirical Nature of the Model. The model was developed empirically, and is not based on any known fundamental chemical reactions between the target contaminants, natural constituents of the groundwater, and the various oxidants produced as a result of the reaction of ozone with hydrogen peroxide. Therefore, it is understandable that the model will not predict the "exact" effluent concentration of each contaminant under all conditions.

Nonideal Hydraulic Flow Regimes in the Contactors. Based on the two tracer tests conducted, each contactor was modeled as a CSTR. However, ideal mixing conditions are only theoretical. Therefore, there is always some variation between the actual

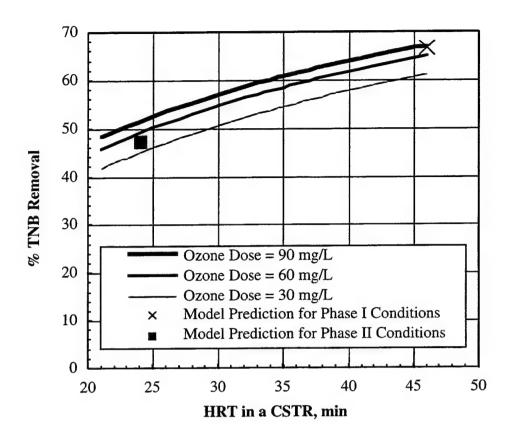
hydraulic conditions in the contactor and those in an "ideally" mixed contactor. These variations can result in slight discrepancies in the model predictions.

4.9.3.9 It is important to note that the proposed model is purely empricial and is not based on any fundamental analysis of the chemical reactions taking place in this process. Therefore, the model should be used with caution and should not be extrapolated to operating conditions (i.e., ozone dose and hydraulic retention times) outside the limits of the conditions used in this project. In addition, the performance of the Peroxone process is highly dependent on water quality. Therefore, the estimated model parameters can only be used to estimate the removal of TNT, TNB, and RDX from the Grand Island groundwater, and should not be extrapolated to other waters.

4.9.4 Sensitivity Analysis

4.9.4.1 Figure 4-11 shows a model sensitivity plot depicting the model-calculated percent TNB removal with the Peroxone process in one contactor as a function of the HRT and transferred ozone dose in that contactor. The plot clearly shows that the process performance, as interpreted by the proposed model, has low sensitivity to either dose or HRT. In other words, the plot suggests that substantial increases in either ozone dose or HRT would result in small increases in % TNB removal. This is more apparent with the impact of the ozone dose on TNB removal where an increase from 60 mg/L to 90 mg/L (a 50% increase in dose) resulted in a modest increase of only 2% removal of TNB through the contactor.

4.9.4.2 The above analysis explains the difference in performance between Phase I and Phase II testing conditions. To illustrate this difference, the predicted percent TNB removals under each set of conditions are superimposed on Figure 4-11. Due to the low sensitivity of the process to either ozone dose and HRT, the model shows that increasing the ozone dose from 44 mg/L (Phase II) to 78 mg/L (Phase I), and increasing the contact time from 24 minutes (Phase II) to 46 minutes (Phase I) would result in an increase in the percent TNB removal from 47% (Phase II) to 66% (Phase I).



Impact of HRT and Transferred Ozone Dose on the Percent Removal of TNB in Each Contactor as Predicted by the Proposed Empirical Model

Figure 4-11

4.10 IMPLICATIONS FOR DESIGN

4.10.0.1. The results of the demonstration testing program have shown that TNT, TNB, and RDX can be removed from contaminated waters with the Peroxone treatment system, with TNB being the critical compound for the determination of the design criteria (i.e., most difficult to oxidize). The availability of a mathematical model for the Peroxone system provides an additional tool for design of the planned 1,000 gpm treatment system. With this simple model, design engineers can quickly simulate various conditions of influent concentrations, hydraulic retention times (i.e., treatment system sizes), and ozone doses and determine initial configurations of the full-scale system for the treatment

of TNT, TNB, and RDX to desired effluent quality. Section 6.0 presents a proposed design configuration for the full-scale Peroxone treatment system.

5.0 PEROXONE SYSTEM DEMOBILIZATION

5.1 REVISION TO WORK PLANS

5.1.0.1. The Work Plans (Montgomery Watson, 1996) for this project called for shutdown and dismantling of the Peroxone treatment system after completion of the demonstration period. Each of the reactors was to be disconnected and shipped to a US Army storage facility to be named after shutdown. The pumps, mixers, electronic instruments, and control panels were to be warehoused at CAAP. The remaining equipment that was leased or rented would be returned and the piping, wiring, and power lines would be torn up and hauled away as scrap. The concrete pad would then be broken up and hauled to the landfill.

5.1.0.1. At the conclusion of the demonstration period, however, it was decided by USAEC that the system would be useful for further testing in the near future. Therefore, the contactors and the connecting piping were left in place, and only the equipment that may be weather sensitive was removed and warehoused.

5.2 SHUTDOWN

5.2.0.1. The demonstration period was completed 8 November 1996. After the last gallon of groundwater was treated through the system, the well pumps were turned off and clean water from the hydrant was diverted into the system to flush the contactors clean. The hydrogen peroxide feed pumps were shut off and the sodium thiosulfate feed system was also turned off. The ozone generator was shut down when flow from the wells was stopped, and the lines were purged with oxygen from the oxygen feed system. After several hours of flow with clean water, the water was shut off and the contactor tanks were shut down. Each of the contactor tanks was drained into the pad and pumped from the sump into the effluent tank. After all the contactors were emptied, the hydrogen peroxide day tanks were emptied and rinsed. The softened water tank was also emptied as was the sodium thiosulfate tank. All tanks were rinsed and all water was collected and pumped through the carbon vessels for final treatment. The piping on the pad was drained and anything that was subject to freeze damage was drained.

5.2.0.2. The system was turned off just before a cold front moved into the Grand Island Area on 10 November 1996. Though the treatment system had been shut down and drained, some water was still in the 2-inch conveyance lines from the wells and froze

before the lines could be drained. The well pumps were removed and all valves were opened to allow drainage back into the wells when the lines thawed.

5.3 DECOMMISSIONING

5.3.0.1. To decommission the facility, the water supply was shut down, the power was turned off and disconnected, the unused chemicals were returned to the respective suppliers, the telephone was disconnected, and the office trailer was hauled away. All local utilities were notified that the site was closed.

5.4 DISMANTLING AND STORAGE

5.4.0.1. As discussed above, the Army decided to leave the treatment system at the CAAP for possible future use. It was decided that the contactors, tanks, and the secondary containment pad would remain in place. The demobilization efforts changed from complete dismantling and storage of the system to removal and storage of pumps, motors, electronics and return of rented equipment. The two well pumps, seven chemical feed pumps, mixers, meters, sump pump, transfer pump, hoses, effluent pump, and calibration equipment were all stored at the Cornhusker Facility in Building S-6 under the direction of Tom Jamieson, the Facility Administrator. Table 5-1 shows a list of items stored.

5.4.0.2. The contactors, piping, tanks, tubing, valves, and equipment supports were left in place on the pad. All tanks were drained prior to leaving the site.

5.4.0.3. The ozone generator was dismantled, crated, and shipped back to the leasing company with the control panel, the ozone destruct unit and the small air compressor used as a nitrogen source. The reverse osmosis unit was likewise returned, and the carbon vessels were drained and sampled. The GAC was tested by the vendor (Calgon Carbon) and was determined to be nonhazardous. The GAC was transported to the Laidlaw landfill in Utah (RCRA subtitle C landfill). The remaining equipment was kept inside a fenced, locked area around the Line 2 assembly buildings.

Table 5-1
Stored Equipment Inventory

Item	Quantity
Rotameters and stainless steel connection piping	6
Box of wire	1
Little Giant sump pump with level switch	1
	7
LMI chemical metering pumps 2 inch in-line static mixer	1
	1
1/2 inch glass rotameter	_
Effluent pump	1
Paddlewheel flow meters (Signet)	2
Extraction well pumps (Grundfos)	2
Buckets (2.5 gallons)	7
10 ft. long by 2 in. dia. flexible hoses (camlock)	2
Fire hydrant backflow preventer attachment	1
Bubble wrap packing in boxes	12
Broom	1
Mop	1
Garden Hoses (50 ft.)	2
Trailer power cord - 4 conductor, 100 ft.	1
20 ft. braided stainless steel hose 1 in. dia.	1
3/8" copper tubing - 25 ft.	1
Buckets (5 gallons)	2
3/8" plastic tubing - 25 ft	1
Box of miscellaneous CPVC fittings	1
Hand operated drum pump	1
Electric Mixers	3
Lab Equipment, boxes/containers	5

6.0 DEVELOPMENT OF FULL-SCALE PEROXONE SYSTEM

6.1 INTRODUCTION

- 6.1.0.1. This section discusses the recommendations for a full-scale Peroxone system based on the results of the demonstration testing program. The full-scale system developed and presented in this section is based on the Peroxone technology tested at the CAAP. No effort was made to evaluate alternative designs such as baffled or packed-bed contactors, using ozone bubble recombination, or other diffuser types or to evaluate the Peroxone technology in combination with other technologies (such as UV/ozone and GAC technologies) for a more economical treatment of explosives-contaminated groundwater.
- **6.1.0.2.** A conceptual Process Flow Diagram (PFD) and preliminary capital and operations and maintenance (O&M) cost estimates for the recommended full-scale system are included.
- **6.1.0.3.** As part of this effort, computer simulations and input from equipment vendors were used to determine the optimum treatment system with respect to performance, capital cost, operational cost, flexibility, and ease of operation.

6.2 SYSTEM SCALE-UP AND DEVELOPMENT PROCESS

- **6.2.0.1.** The scale-up and development effort can be categorized as a five-step process outlined below. Each step is described in detail later in this section.
 - (1) Scale up of the Peroxone contactors to 1,000 gpm capacity.
 - (2) Use the Peroxone model developed in Section 4.0 to conduct computer simulations for a full-scale system (1,000 gpm flow rate per the Contract requirements).
 - (3) Evaluate the system configurations generated from the model simulations for technical feasibility, cost effectiveness, and ease of system operation and maintenance.
 - (4) Select a configuration for the full-scale design.
 - (5) Present preliminary capital and O&M cost estimates for the selected system.

6.2.1. Model Limitations

6.2.1.1. The Peroxone system model (Section 4.0) was used in the scale-up process as a tool which allowed for a quick relative comparison of numerous reactor configurations and oxidant doses. It is not intended to serve as the only tool for design of a full-scale system. The model is reliable within the boundary conditions which include the reaction kinetic parameters, contaminant type and concentration range, minimum and maximum applied ozone dosage, ozone transfer efficiency, and hydrogen peroxide dosage based on the Peroxone ratio. However, like other empirical models, there is little certainty in the accuracy and reliability of the model under conditions that are outside the boundary conditions of the data used to calibrate the model.

6.3 CONTACTOR SCALE UP

- **6.3.0.1.** The contactor vessel used for the demonstration testing was a cylindrical tank 3 feet in diameter with a 10-foot side wall depth. It was determined through tracer testing that each contactor was completely mixed and that it acted as a continuously stirred tank reactor (CSTR).
- **6.3.0.2.** Effluent sampling and analyses from individual contactors showed that all contactors provided approximately equal percent destruction of contaminants indicating that the contactor design was independent of the influent concentrations or oxidation chemical doses. Equal percent destruction in all contactors suggests that each contactor can be considered as a single CSTR for the purpose of the system design and scale up.
- **6.3.0.3.** Assuming that contactors for the full-scale system can be considered as CSTRs, and that the number of CSTRs per contactor equals one, the contactor scale-up process simply involves selecting cylindrical tanks with diameter to side wall depth ratios similar to those provided for the demonstration testing. It is noted that preliminary cost analysis showed that the cost of utilizing cylindrical tanks for the 1000 gpm plant was comparable to that of a concrete contactor with multiple chambers. Therefore, the use of cylindrical steel tanks is not necessarily a recommendation at this time, but only an option for cost estimation purposes.
- **6.3.0.4.** The initially selected tank diameter varied from 4 feet to 12 feet while the side wall depths ranged from 12 feet to 36 feet. Each combination was evaluated for holding

capacity, ozone transfer efficiency (assumed at 90%), number of tanks required for 1,000 gpm system, and area requirements to hold the required number of tanks for a full-scale system. The evaluation showed that selection of tanks with a diameter less than 6 feet would necessitate too many tanks for a 1,000 gpm system and cause excessive head loss in the system. Thus, tanks with a diameter less than 6 feet were deleted from evaluation.

6.3.0.5. Montgomery Watson's experience suggests that no appreciable mass transfer between ozone and the liquid phase is realized beyond 23- to 25-foot side wall depth. In order to maintain a minimum side wall depth to diameter ratio of 3, this meant eliminating all tanks with side wall depth greater than 25 or a diameter greater than 8 feet.

6.3.0.6. Tanks passing the initial selection criteria included those with diameter between 6 and 8 feet and side wall depths ranging between 18 and 25 feet, and were retained for the simulation process. Table 6-1 shows the tank combinations that were used for computer simulations.

Table 6-1
Selected Tank Configurations

Tank Diameter (feet)	Side Wall Depth (feet)
6	18
6	19
6	20
6	21
6	22
6	23
6	24
6	25
7	21
7	22
7	23
7	24
7	25
8	24
8	25

6.4 PEROXONE MODEL SIMULATIONS

6.4.1. Model Development

- **6.4.1.1.** For each simulation, the total flow to the system was fixed at 1,000 gpm and the influent concentrations were assumed to be 400 μ g/l TNB, 600 μ g/l TNT, and 200 μ g/l RDX. These compound-specific concentrations resulted from sampling and analyses of the groundwater during the demonstration testing. The system flow rate and the influent concentrations were assumed constant during each simulation.
- **6.4.1.2.** For each simulation, the target effluent concentration for each contaminant was set at $2.0 \,\mu\text{g/L}$ or less.
- 6.4.1.3. Certain parameters in the model were provided with preset values including: (1) ozone transfer efficiency = 90%, (2) peroxide-to-ozone ratio = 0.5, and (3) number of continuously stirred tank reactors (CSTRs) per contactor = 1. These values were assumed constant during each simulation. Although the transfer efficiency measured during project was consistently below 85%, it is believed that increasing the sidewater depth from 10 ft to greater than 18 ft would increase the ozone transfer efficiency to greater than 90%. This is based on the project team's experience with the design of ozone contactors for water treatment plants where greater than 95% ozone transfer efficiency is achieved with 20-ft side water depth.
- **6.4.1.4.** The minimum applied ozone dosage tested during the testing program was approximately 30 mg/L, and this was used as a boundary condition for the model. This means that even though the model suggests that it may be possible to achieve effluent goals with an ozone dosage less than 30 mg/L, this condition was not simulated. Similarly, the maximum applied dosage during the demonstration testing was 115 mg/L, and this was used as another boundary condition for the model.
- **6.4.1.5.** Simulations were conducted with 2, 3, and 4 parallel trains. In other words, the first simulation called for two trains with a flow of 500 gpm through each train. For the second simulation, the total flow was split evenly into three 333-gpm trains, and so on. A single treatment train would require too many contactors in series or such a large capacity ozonation system that it would be cost prohibitive. For this reason, no simulations were conducted for a single treatment train system. Similarly, a Peroxone

system with greater than 4 treatment trains would require too many contactors and may not be practical from an operational standpoint.

6.4.1.6. Head loss is a consideration in design of treatment systems with multiple reaction tanks and gravitational flow. The head loss increases with an increase in the number of tanks per treatment train. A higher head loss in the treatment train requires either a system design in which the first reaction tank is the tallest with consecutive tanks having a smaller side wall depths so that the liquid can flow by gravitational head or increasing the inter-connecting pipe size. Changing the side wall depth means that each reaction vessel would produce a different percent removal efficiency. An extra large pipe would increase the total height of the tank and impact the system cost. To minimize the impact of head loss on the system design, the total number of tanks per treatment train was limited to eight and this criterion is based on experience gained from the demonstration system design.

6.4.1.7. Table 6-2 shows the various combinations that were simulated during the full-scale Peroxone system development process. Note that the simulations were conducted for each tank configuration described in Table 6-1.

Table 6-2
Simulated Configurations

Number of Trains	Contactors per Train	Total Number of Contactors	Applied Ozone Dose per Contactor (mg/L)
2	6	12	30 -115
2	7	14	30 - 115
2	8	16	30 - 115
3	6	18	30 - 115
3	7	21	30 - 115
3	8	24	30 - 115
4	6	24	30 - 115
4	7	28	30 - 115
4	8	32	30 - 115

6.4.2 Peroxone Model Simulation Results

6.4.2.1. Simulation results indicate that for 6-foot or 7-foot diameter tanks, none of the combinations of number of trains, number of contactors per train, and applied ozone dosage within the model boundary conditions were capable of treating influent groundwater to the desired effluent quality. Results of these simulations are thus not included for discussion. Table 6-3 shows the simulation results using an 8-foot diameter and 24-foot side wall depth contactor. Note that only the combinations capable of meeting the desired effluent quality within the model boundary conditions are shown on the table.

Table 6-3
Simulation Results: 8-Foot Dia And 24-Foot Swd Contactor

Run No.	Number of Trains	Contactors per Train	Total Number of Contactors	O ₃ Dose per Contactor (mg/L)	Cumulative O ₃ Demand (lb./day)
1	3	7	21	85	7,140
2	3	8	24	38	3,648
3	4	7	28	30	2,520

6.4.2.2. A typical output from the model simulations is shown on the following page. The model predicts the number of contactors required in each train to meet the effluent quality, lists the pounds per day of ozone required for each train, and estimates the total daily ozone requirement of the system. The model also predicts the effluent concentration of each contaminant from individual contactors. As an example, for the model simulation with 4 trains, seven contactors are required in each train for a total of 28 contactors and the total cumulative ozone demand of the treatment system is 2,520 pounds per day to meet the effluent quality.

6.5 EVALUATION OF PEROXONE SYSTEM MODEL SIMULATIONS

6.5.0.1. This section evaluates the results obtained from the model simulations described above. The purpose of this evaluation is to weigh the technical effectiveness of each system configuration against capital and O&M costs and operational strategy.

FULL-SCALE PEROXONE DESIGN MODEL FOR EXPLOSIVES REMOVAL

Developed by Issam Najm, Ph.D. Applied Research Department, Montgomery Watson. 1997

٠		-
	(INPUT BOLD CELLS)	
	Influent Water Quality Conditions:	
	Total Water Flow Rate =	1000 gpm
	Number of Parallel Trains =	4
	Water Flow Rate/train =	250 gpm
	TNB Influent Concentration =	400 µg/L
	TNT Influent Concentration =	600 µg/L
	RDX Influent Concentration =	200 µg/L
	Contactor Configuration:	
	Side Water Depth, H =	24 ft
	Contactor Diameter, W =	8 ft
	ntactor Cross-Sectional Area, L =	50.3 sq. ft
	Contact Time/Chamber =	36 min
	No. of CSTRs/Chamber =	-
	Applied Ozone Dose/Chamber =	30 mg/L
	Ozone Trans. Efficiency =	% 06
	ansfered Ozone Dose/Chamber =	27 mg/L
	H,O, Dose/Chamber =	13.5 mg/L

Kinetic	tic Parameters.		
	TNB	TNT	RDX
~	0.015	0.015 0.006	0.054 (mg/L) ^{-m} (min) ⁻ⁿ
ш	0.237	0.662	0.000

Note: Model based on the following equation: $dC/dt = - k D^{m} C$

					CON	CONTACTOR		IN EACH	H TRAIN	NI			
Parameter	Unit	1	2	3	4	5	9	7	8	6	10	11	12
Cumulative Contact Time	mim	36	72	108	144	180	217	253	289	325	361	397	433
Cumulative Applied Ozone Dose mg/L	mg/L	30	09	90	120	150	180	210	240	270	300	330	360
Cumulative Ozone Consumption/ Ibs/day	lbs/day	90	180	270	360	450	540	630	720	810	006	066	1080
Total Cumulative Ozone Consum Ibs/day	lbs/day	360	720	1080	1440	1800	2160	2520	2880	3240	3600	3960	4320
Effluent TNB Concentration	µg/L	182	82.8	37.7	17.1	7.8	3.5	1.6	0.7	0.3	0.2	0.1	0.0
Effluent TNT Concentration	µg/L	213	75.4	26.7	9.5	3.4	1.2	0.4	0.1	0.1	0.0	0.0	0.0
Effluent RDX Concentration	µg/L	<i>L</i> 9	22.8	7.7	5.6	6.0	0.3	0.1	0.0	0.0	0.0	0.0	0.0

6.5.1. Technical Evaluation

6.5.1.1. For evaluation of technical feasibility, only the three combinations shown in Table 6-3 are considered. Each configuration is capable of meeting the effluent quality goals and is thus technically effective. However, the ozone demand for Run No. 1 (three trains and seven contactors per train) is significantly higher compared to Run No. 2 and No. 3. This higher ozone demand would necessitate a much larger and a very different type of ozonation which is generally not used for hazardous waste treatment facilities. For this reason, Run No. 1 is deleted from further discussions.

6.5.2. Cost Evaluation

6.5.2.1. Table 6-4 presents the capital cost, annual O&M costs, and the 20-year present worth cost for the two configurations under consideration.

Table 6-4
Preliminary Cost Estimates

	Number of Trains	Contactors per Train	System Capital Cost	O&M Cost	System 20-Year Present Worth
2	3	8	\$14,160,000	\$1,113,000	\$26,924,000
3	4	7	\$13,447,000	\$906,000	\$23,837,000

6.5.2.2. The cost estimates are for a complete groundwater extraction, conveyance, and treatment facility including extraction wells, groundwater conveyance piping network, influent storage, treatment facilities, effluent storage and discharge, ozonation systems, and a chemical feed system. Cost for the treatment system also include a structural pad, a metal building, electrical and instrumentation, and civil and mechanical work.

6.5.2.3. The costs are based on parametric cost estimates with a +50% to -30% accuracy. Cost data were obtained from equipment vendors, Means Building Construction Cost Data 1995, the Environmental Restoration Unit Cost Book, and other available sources.

Some of the constraints related to the full-scale system costs are included on Table 6-6 as footnotes.

- **6.5.2.4.** As shown on Table 6-4, both the capital and O&M costs are reduced with an increase in the number of trains. The primary reduction in the full-scale system capital cost results from reduced ozone demand and hence a smaller ozone generation system which is the primary cost for the Peroxone system.
- **6.5.2.5.** The full-scale system O&M costs are primarily dependent on the total ozone demand of the system with minor contributions from hydrogen peroxide demand, system labor requirements, and other operational activities. Since the total ozone demand of the system decreases with increasing number of trains, the annual O&M costs for the full-scale system is reduced as the number of trains is increased.
- **6.5.2.6.** Present worth costs were calculated over a 20-year project duration using an 6 percent annual compound interest rate.

6.6 FULL-SCALE PEROXONE SYSTEM

- 6.6.0.1. Selection of a full-scale system would require detailed evaluation of site conditions, influent concentrations, contactor design options, and other considerations described earlier in this section. Based solely on the model simulations and evaluation results presented in this section, a system with 4 parallel trains (i.e., flow evenly split into 4 trains) should provide the most technically feasible and cost-effective system for treatment of explosives-contaminated groundwater at the CAAP. The selected configuration also provides other benefits that are discussed below. However, it should be emphasized that the proposed design criteria is only one of many viable alternatives. A more detailed and comprehensive cost analysis is required in order to fully optimize the design of the treatment plant.
- **6.6.0.2.** Splitting the flow evenly into 4 parallel trains provides flexibility in the system operation. An individual treatment train can be removed from operation without adversely affecting the entire treatment process. With 4 trains, removal of an individual train reduces the total treatment capacity by only 250 gpm.
- **6.6.0.3.** A treatment process with multiple trains provides flexibility to construct the system in phases. A multi-phase construction program would eliminate the need for a

large capital investment up front and still satisfy regulatory requirements for a treatment system. Treatment trains can be added later as desired.

6.6.0.4. A multi-train treatment process also offers the benefit of reducing the treatment capacity in phases toward the end. Early removal of unneeded capacity will result in O&M cost savings without affecting the treatment process.

6.6.1. Full-Scale Peroxone System Design Criteria

- 6.6.1.1. Preliminary design criteria for a hypothetical extraction and treatment system are included on Table 6-5. The design criteria for the Peroxone treatment process are based on the flow requirements and treatment goals stated in the SOW. The design criteria for the groundwater extraction and conveyance system is hypothetical and will depend on site conditions, site geology and hydrogeology, as well as state and local codes and guidelines. It is noted that a LOX system was assumed as an oxygen source for ozone generator. Preliminary cost estiamtes have shown that the total annual cost of a PSA or VSA system is comparable to, if not slightly higher than, a LOX feed system.
- **6.6.1.2.** Table 6-5 is not intended as an exhaustive inventory of materials, but summarizes major components of the full-scale treatment system.
- **6.6.1.3.** Figure 6-1 presents a conceptual Process Flow Diagram (PFD) for the recommended full-scale Peroxone system. Note that several other supporting equipment and components would be required for a full-scale Peroxone treatment system which are not shown on the PFD.

6.6.2. Full-Scale Peroxone System Cost Estimates

6.6.2.1. A detailed preliminary capital and O&M cost estimate for the recommended full-scale Peroxone system is presented in Table 6-6. These costs estimates are not intended for use as a construction estimate.

Full-Scale Peroxone System Conceptual PFD

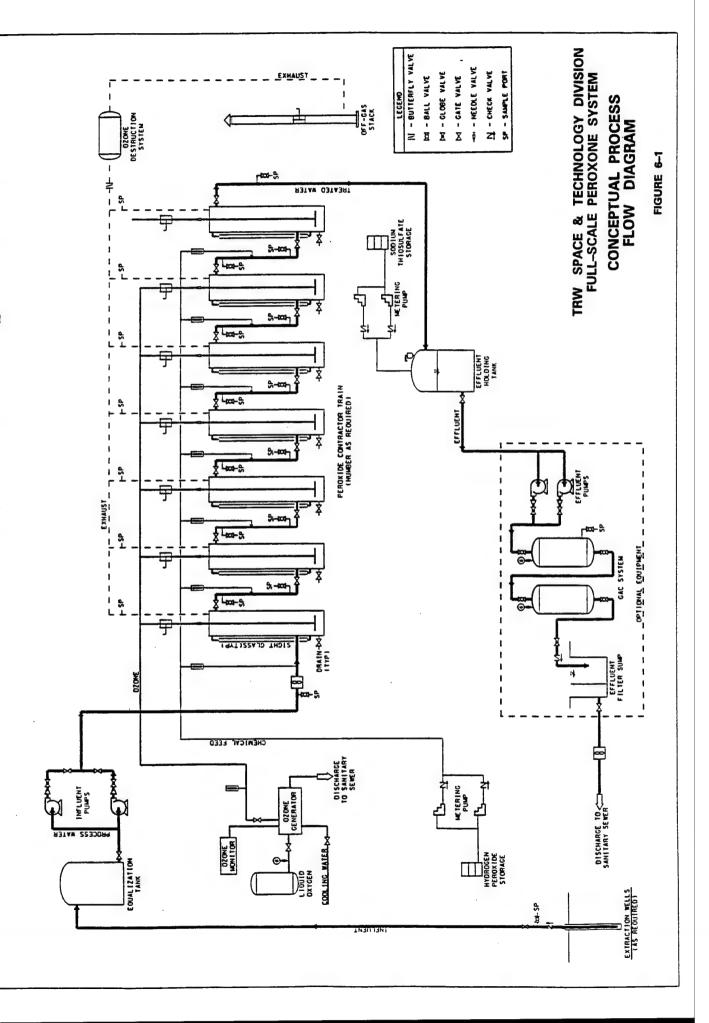


Table 6-5

Full-Scale Peroxone System Conceptual Design Criteria

Equipment	Description	Criteria	Comments
Extraction Well	Number of wells Well casing	40 6-inch (min)	Number of wells varies based on the site hydrogeology and groundwater yield rates.
Well Head	Number Pipe Type Vault Type	40 (one for each well) Heat traced carbon steel pipe Concrete, flush with surface	Number of wellhead equals number of wells installed. Each well provided with magnetic meter to control flow rate.
Extraction Pump	Number Type Capacity	40 (one for each well) Submersible, electrical 25 gpm each	Pump number and capacity depends on the site conditions. Total head and horsepower will vary depending on well locations.
Conveyance Line	Type Location	HDPE, double contained Buried 3 feet (min) below ground surface	Provide freeze protection and leak detection system. Pipe size will depend on flow from individual wells and piping layout.
Influent Flow Meter	Range Number Type	100-1000 gpm 1 Magnetic	Indication of total flow to the treatment system. 'Assumed that 10% of the wells will be in the discharge mode at a given time.

Table 6-5

Full-Scale Peroxone System Conceptual Design Criteria (Continued)

Equipment	Description	Criteria	Comments
Equalization Tank	Number Capacity Tank Material	2 10,000 gallon each High density polyethylene	Two 10,000 gal tanks provide flexibility in system operation while minimizing down time. HDPE suitable material for long-term protection.
Influent Pump	Number Type Capacity	5 (four plus one standby) Centrifugal, end suction 250 gpm each	One pump for each train allows flexibility in system operation with one standby to minimize system operation down time.
Ozone Contactor	Number of contactors Type Capacity Size Material	28 (7 per train) Unpacked column 10,000 gal (each appr.) 8 feet diameter, 24 feet high 316 SS	36 minute retention time per contactor. 316 SS for long-term protection against ozone corrosion. Provide manway at the top and bottom side.
Effluent Storage Tank	Capacity Number Type	10,000 gal 1 High density polyethylene	10 minute retention time at 1,000 gpm flow for thiosulfate mixing.
Effluent Pump	Number of pumps Type Capacity	3 (two plus one standby) Centrifugal, end-suction 500 gpm (each)	Two pumps allow flexibility in system operation while minimizing down time.

Table 6-5

Full-Scale Peroxone System Conceptual Design Criteria (Continued)

Equipment	Description	Criteria	Comments
Effluent Flow Meter	Range Type Number	10-1000 gpm Magnetic 1	Indication of total discharge from the treatment system. Other NDEQ limitations may apply for discharge monitoring.
Ozone Generator	Capacity Ozone Dosage (each vessel)	2,520 lb/day 30 mg/l at 10% ozone by weight	See "Full-Scale Peroxone Model" for calculations details.
Hydrogen Peroxide System	Daily Capacity Applied dosage	1,135 lb/day of 35% solution 15.0 mg/l (each contactor)	
Sodium Thiosulfate System	Daily Capacity Applied dosage	84 lb/day of pure solution 7 mg/L per mg/L of residual ozone	

Table 6-6
Preliminary Cost Estimate for 1,000 gpm Peroxone System⁽¹⁾

Item/Description	Quantity	Unit	Unit Cost	Total Cost
DIRECT CAPITAL COSTS				
General				
Contractor Mobilization ⁽²⁾		lump sum	\$100,000	\$100,000
Contractor Demobilization (2)		lump sum	\$50,000	\$50,000
Treatment System Pad ⁽³⁾	556	cubic yard	\$230	\$127,778
Excavated Soil Disposal ⁽⁴⁾	9,228	ton	\$50	\$461,421
2.10u.u.u.u 00.1.2.1speess.	2,===		Subtotal	\$739,199
Groundwater Extraction System				
Vertical Extraction Wells ⁽⁵⁾	40	each	\$15,000	\$600,000
Conveyance Pipe (6)	50,000	linear foot	\$35	\$1,750,000
Conveyance Tipe	20,000		Subtotal	\$2,350,000
Groundwater Treatment System				
Equalization Tanks	2	each	\$12,000	\$24,000
Influent Transfer Pumps	5	each	\$12,500	\$62,500
Automatic Pressure Filters	4	each	\$12,500	\$50,000
SS 316 Contactors ⁽⁷⁾	28	each	\$35,000	\$980,000
Ozonation System ⁽⁸⁾		lump sum	\$2,000,000	\$2,000,000
Chemical Feed System ⁽⁹⁾		lump sum	\$50,000	\$50,000
Effluent Holding Tank	1	each	\$12,000	\$12,000
Effluent Transfer Pumps	3	each	\$15,000	\$45,000
Polishing GAC Vessels (optional equipment	- see footnote 10)			
			Subtotal	\$3,223,500
		Total Direct C	Capital Costs (DCC)	\$6,313,000
INDIRECT CAPITAL COSTS ⁽¹¹⁾				
Equipment Installation (10% of DCC)		lump sum		\$631,300
Mechanical Piping/Accessories (10% of DCC)		lump sum		\$631,300
Electrical and Instrumentation (18% of DCC)		lump sum		\$1,136,340
Civil/Site Improvements (10% of DCC)		lump sum		\$631,300
Building/Facilities (6% of DCC)		lump sum		\$378,780
Design/Engineering (15% of DCC)		lump sum		\$946,950
Permitting and Approvals (2% of DCC)		lump sum		\$126,260
Construction Management (8% of DCC)		lump sum		\$505,040
Contractor's Fee (4% of DCC)		lump sum		\$252,520
Contingency (30% of DCC)		lump sum		\$1,893,900
		Total In	direct Capital Costs	\$7,133,690
		TOTAL CA	PITAL COSTS ⁽¹⁶⁾	\$13,447,000

Table 6-6
Preliminary Cost Estimate for 1,000 gpm Peroxone System
(Continued)

Item/Description	Quantity	Unit	Unit Cost	Annual Cos
OPERATION AND MAINTENAN	CE COSTS			
Ozonation System ⁽¹²⁾	12	per month	\$38,430	\$461,160
Chemical Additives	12	per month	\$15,000	\$180,000
Electrical Power (13)	150	kw-hr	\$0.08	\$105,120
Labor ⁽¹⁴⁾	2,920	per hour	\$30	\$87,600
Analytical Cost (15)	12	per month	\$5,000	\$60,000
General Maintenance	12	per month	\$1,000	\$12,000
		ANNUAL (O&M COSTS ⁽¹⁶⁾	\$906,000
PRESENT WORTH				
Interest Rate = 6%	Project Life = 20 Years	20-YEAR PRES	ENT WORTH ⁽¹⁶⁾	\$23,837,000

- 1 Parametric cost estimate based on standard engineering practice and costing methods. Refer to Section 6.0 of the Report for items not included in the cost estimate. Accuracy of cost estimate is within the +50% to -30% range.
- 2 Single mobilization and demobilization assumed for the treatment system construction.
- 3 A 200-foot x 100-foot x 1-foot thick concrete slab on footings with a 1-foot containment berm.
- 4 50 percent of excavated soil disposed at a Subtitle D (non-hazardous) landfill.
- 5 A 6-inch diameter well, average vertical depth 30 feet bgs, 15-foot SS screen. Cost includes drilling, installation, well-head completion, development, pump electrical, and controls. Number of wells will vary depending on site hydrogeology and groundwater yield rates.
- 6 Assumes 2-inch double-contained HDPE pipe. Total pipe length will vary depending on well locations, pipe routing and layout, and the treatment system siting.
- 7 Each contactor 8 feet in diameter and 24 feet tall; SS 316 shell material with a top and bottom manway; no packing included.
- 8 Vendor quote (Ozonia, Lodi, NJ) for a complete system including liquid oxygen storage and feed, ozone generators, nitrogen generator and feed system, demisters, preheaters, residual ozone destruct units, vent gas blowers, and power supply and control systems.
- 9 Complete chemical feed system including chemical storage, day tanks, chemical feed pumps and piping, and control systems.
- # GAC vessels may be required by NDEQ. Cost not included in the estimate.
- # Parametric cost estimate based on standard engineering practice and costing methods.
- # Vendor quote (Ozonia, Lodi, NJ) based on per pound of ozone generated, excluding labor and chemical additives.
- # Excluding cost for the ozonation system which is included in item 12 above.
- # One operator, 8 hours per day, 7 days a week at \$30 per hour.
- # Analysis for pH, oil & grease, explosives, general minerals, and other parameters.
- # Cost values rounded to the nearest \$1,000.

7.0 SUMMARY, CONCLUSIONS, & RECOMMENDATIONS

7.1 INTRODUCTION & SYSTEM DESIGN

- 7.1.1. This project was aimed at demonstrating the applicability of the Peroxone process (i.e., Ozone with Hydrogen Peroxide) for the remediation of explosives-contaminated groundwater at the Cornhusker Army Ammunition Plant (CAAP) in Grand Island, Nebraska. The primary contaminants were TNT, TNB, and RDX. The measured concentration of each of these contaminants in the groundwater varied from 114 μ g/L to 1200 μ g/L for TNT, 114 μ g/L to 711 μ g/L for TNB, and 0.01 μ g/L to 74 μ g/L for RDX. The treated water concentration goal for each contaminant was set at 2 μ g/L.
- 7.1.2. The Peroxone demonstration plant design criteria, developed by the project Technical Advisory Board, was based on bench-scale and pilot-scale testing conducted by the US Army Corps of Engineers (USACE) at the Waterways Environmental Station (WES). The plant consisted of three main parts. The first part was a groundwater extraction system drawing water from two wells at CAAP. The second part was the main Peroxone treatment process which consisted of six (6) 12-foot high stainless-steel contactors operated in series (the sidewater depth in each contactor was approximately 10 ft). Hydrogen peroxide was added to the influent stream to each contactor while an ozone-rich gas stream was bubbled through each contactor via two stone diffusers installed at the bottom of each contactor. The third part was a Granular Activated Carbon (GAC) treatment process intended to capture any contaminants present in the effluent of the Peroxone treatment process prior to discharging the water into a nearby ditch. The GAC treatment process consisted of three GAC vessels operated in series.
- 7.1.3. The Peroxone treatment process was designed to treat a maximum groundwater flow rate of 25 gpm at a maximum applied ozone dose of 55 mg/L in each of the six contactors. This results in a maximum total applied ozone dose of 330 mg/L. At the design flow rate of 25 gpm, the average hydraulic residence time (HRT) in each contactor was 24 minutes for a total HRT of 144 minutes. The hydrogen peroxide feed system was designed to provide sufficient hydrogen peroxide to result in a Peroxone weight ratio of 0.3 mg/mg. The Peroxone weight ratio is the ratio of applied hydrogen peroxide dose (expressed in mg/L) to the transferred ozone dose (expressed in mg/L).

7.2. SYSTEM TESTING PLAN

- **7.2.1.** The treatment train was operated for a total of 14 weeks. During the first two weeks, debugging of the treatment processes and equipment was conducted. During the next four weeks, an optimization task was conducted during which the Peroxone process performance for contaminants destruction was evaluated under varying conditions of ozone dose, contact time, and water source. During the final eight (8) weeks of the testing schedule, a demonstration task was conducted during which the system was operated under two sets of conditions for a period of 4 weeks each.
- **7.2.2.** During the first phase of the demonstration task, the system was operated at an average flow rate of 13 gpm (which corresponded to an average HRT of 46 minutes in each contactor), an average transferred ozone dose of 78 mg/L, and an average Peroxone ratio of 0.45 mg/mg. During the second phase of the demonstration task, the system was operated at an average flow rate of 25 gpm (which corresponded to an average HRT of 24 minutes in each contactor), an average transferred ozone dose of 44 mg/L, and an average Peroxone ratio of 0.57 mg/mg.
- **7.2.3.** The performance of the treatment process was monitored on a daily basis. Water samples were collected from the effluent of each of the six contactors, as well as from the effluent of the GAC process, and transported to GP Laboratories in Gaithersburg, MD.

7.3. SYSTEM PERFORMANCE

7.3.1. The experimental results obtained showed that TNB was the most difficult compound to remove with the Peroxone process, followed by RDX, and finally by TNT which was the most readily removed compound. However, the results of the project showed that the Peroxone system was not capable of achieving the target explosives' removals at the design dose of 330 mg/L and a total contact time of 144 minutes (2.4 hours). In order to achieve the target water quality goals, the contact time was increased to 276 minutes (4.6 hours) by reducing the groundwater flow rate into the system from 25 gpm to 13 gpm, and the applied ozone dose was increased to 600 mg/L. With an ozone transfer efficiency of approximately 78 percent, the transferred ozone dose was approximately 470 mg/L.

- **7.3.2.** This project also demonstrated that the Peroxone ratio had to be increased from the design value of 0.3 mg/mg to approximately 0.5 mg/mg in order to maintain a low ozone residual in the effluent water and a high ozone transfer efficiency. Therefore, at the transferred ozone dose of 470 mg/L and a Peroxone ratio of 0.5 mg/mg, the required hydrogen peroxide dose was 235 mg/L divided equally among the six contactors.
- 7.3.3. While a high transferred ozone dose of 470 mg/L and a long contact time of 4.6 hours were required to meet the effluent water quality goal of 2 µg/L for each individual contaminant, a lower transferred ozone dose of 265 mg/L and a shorter contact time of 2.4 hours removed TNB to an effluent concentration of 2 to 4 µg/L, while achieving complete removals of TNT and RDX. Since the cost of the Peroxone process is highly impacted by the required ozone dose, this finding suggests that a hybrid treatment system of a Peroxone process for partial explosives removal, followed by a polishing treatment process (such as GAC adsorption) for removing the remaining explosives, may be far more cost effective than a stand-alone Peroxone process designed for complete explosives removal. However, it is noted that this approach does not address the possible formation of oxidation by-products which may consume the GAC capacity more rapidly.

7.4. MODEL DEVELOPMENT

- **7.4.1.** In order to develop the design criteria for a 1000-gpm Peroxone treatment process, Montgomery Watson developed an empirical model to simulate the removal of TNT, TNB, and RDX by the Peroxone process. The model contained two empirical coefficients which were estimated by fitting the model calculated removals to those measured through each contactor. The model was successful in simulating the optimization task results and those of the first phase of the demonstration task. However, it somewhat underestimated the removals of TNT, TNB, and RDX measured during the second phase of the demonstration task.
- **7.4.2.** It is emphasized that the mathematical model developed in this project is purely empirical, and is limited to the ranges of concentrations, doses, and contact times evaluated in this project. In addition, its accuracy is highly dependent on the hydraulics of the contactor, as well as the quality of the source water. Therefore, the model should be used with caution when estimating the removals of TNT, TNB, and RDX with the Peroxone process.

7.5. DESIGN & COST OF FULL-SCALE SYSTEM

7.5.1. Based on the results of the testing program, the empirical model was used to develop a preliminary design criteria and cost of a 1000-gpm Peroxone treatment system for removing TNT, TNB, and RDX from CAAP groundwater. There are several design configurations that are applicable for this plant. One of the configurations was selected for this plant, and is listed in Table 7-1. It should be noted that the design criteria is highly dependent on the influent concentrations of TNT, TNB, and RDX. For the purposes of this design, the groundwater concentrations of TNT, TNB, and RDX were assumed at 600 μ g/L, 400 μ g/L, and 200 μ g/L, respectively. The target effluent concentration of each contaminant was set at 2 μ g/L.

Table 7-1

Conceptual Design Criteria of 1000-gpm Peroxone Treatment Plant

Parameter	Unit	Value
Total Water Flow Rate	gpm	1,000
Number of Parallel Trains		4
Flow Rate per Train	gpm	250
Number of Contactors per Train		7
Total Number of Contactors	_	28
Contactor Type	stainless-steel cylindrical columns	
Contactor Diameter	ft	8
Side-water Depth	ft	24
Contact Time per Contactor	min	36
Total Contact Time	min	252
Ozone Dose per Contactor	mg/L	30
Total Ozone Dose per train	mg/L	210
Ozone Capacity	lbs/day	2,520
Ozone Transfer Efficiency*	%	90
Peroxone Ratio	mg/mg	0.5
Total Hydrogen Peroxide Dose	mg/L	95
Hydrogen Peroxide Capacity	lbs/day	1,135

^{*} The ozone transfer efficiency of 90% was assumed based on the project team's experience with the design of ozone contactors with such side water depth.

7.5.2. The plant consists of four (4) parallel trains, each with a capacity of 250 gpm. Each train included seven (7) stainless-steel cylindrical contactors in series. Each contactor had a diameter of 8 ft and a side-water depth of 24 ft. At a flow rate of 250 gpm per train, the estimated contact time through each contactor is estimated at 36 minutes, for a total contact time of 252 minutes. The applied ozone dose to each contactor was estimated at 30 mg/L for a total of 210 mg/L of water treated, which translates into a required ozone generation capacity of 2,520 lbs/day. An ozone transfer efficiency of 90% was assumed. With a Peroxone ratio of 0.5 mg/mg, the required hydrogen peroxide dose was thus estimated at 95 mg/L, which translates into a total required hydrogen peroxide consumption of 1,135 lbs/day.

7.5.3. Based on the above design criteria, a preliminary capital and O&M cost estimates were developed for the 1000-gpm Peroxone treatment system. The cost breakdown is summarized in Table 7-2. The total system capital cost is estimated at \$13,447,000 and the annual Operations & Maintenance costs are estimated at \$906,000/yr. Assuming an amortization period of 20 years and a 6% cost of money, the total annual cost is estimated at \$2,079,000/yr. The 20-yr present worth of the system is estimated at \$23,837,000.

Table 7-2
Preliminary Cost Breakdown
for the 1,000-gpm Peroxone System

ITEM/DESCRIPTION	COST
Direct Capital Cost	
General	\$739,200
Groundwater Extraction System	\$2,350,000
Treatment System	\$3,223,500
Total Direct Capital Costs (DCC)	\$6,313,000
Indirect Capital Costs	\$7,133,690
Total Capital Cost	s \$13,447,000
Amortized Capital Costs (8%; 30 yrs)	\$1,195,000
Annual O&M Costs	\$906,000
Total Annual Cost	\$2,079,000
Total Cost of water	\$3.95/1000 gal
20-year Present Worth	\$23,837,000

7.5.4. It should be noted that the capital cost includes \$600,000 for the construction of a total of 40 wells, and \$1,750,000 for conveyance piping. These wells are required because the maximum individual well capacity was estimated at 25 gpm. If hydrogeological studies at CAAP determine that wells can deliver significantly higher flow rates, significant savings can be realized by reducing the number of wells and length of piping required. In addition, due to the preliminary nature of the cost estimate, the indirect capital cost estimate includes approximately \$1,900,000 in capital cost contingency.

7.6. CONCLUSIONS

- 7.6.1. While this project demonstrated that TNT, TNB, and RDX can be reliably removed from groundwater using the Peroxone process, the amount of ozone and hydrogen peroxide needed, as well as the required contact time, are higher than initially anticipated. At CAAP, the required transferred ozone dose was estimated at 470 mg/L, with a required hydrogen peroxide dose of 235 mg/L and a contact time of 4.6 hours. These are high values when compared to ozone doses and contact times required for conventional groundwater remediation of typical organic contaminants. Due to these high chemical doses and high contact time, the total annual cost of a Peroxone treatment system designed to treat 1000 gpm of CAAP groundwater was estimated at \$2,079,000/year.
- **7.6.2.** However, this project also demonstrated that substantially lower chemical doses and lower contact time can achieve near complete removals of TNB, which was the most difficult contaminant to remove. This suggests that a hybrid treatment system of a Peroxone process for partial explosives removal, followed by a polishing treatment process (such as GAC adsorption) for removing the remaining explosives, may be far more cost effective than a stand-alone Peroxone process designed for complete explosives removal. However, it is noted that this approach does not address the possible formation of oxidation by-products which may consume the GAC capacity more rapidly than anticipated. Therefore, the concept of the hybrid system should first be tested before a conclusion can be made about the cost effectiveness of such a system.

7.7. RECOMMENDATIONS

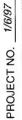
- **7.7.1.** The following is a list of recommendations developed as a result of the outcome of this project. The objective of these recommendations is to possibly further minimize the overall system cost.
- 7.7.2. Applicability of a Hybrid System Design. It is recommended that a desktop study be conducted to evaluate the hybrid process design alternative discussed above, develop the optimum design criteria for each of the two processes (i.e., Peroxone and GAC), and verify whether this hybrid system will result in a minimum total system cost. Using the empirical Peroxone process model developed in this project and various GAC adsorption models presented in the literature, system design optimization should be feasible. If such models are not available, simple laboratory studies can be conducted using CAAP groundwater samples to evaluate the adsorption of TNT, TNB, and RDX onto various types of GAC. It was noted that an identical recommendation was included in the WES report in order to help meet the desired system performance criteria while minimizing the overall system cost.
- 7.7.3. Evaluation of Alternative Peroxone Design Criteria. As indicated earlier, the design parameters for the Peroxone system used in this project were set by USAEC. There are various other modes of ozone application during water treatment. Considering that the performance of an ozonation process is highly dependent on the mode of ozone application and contactor hydraulics, it is recommended that a study be conducted to evaluate various Peroxone system design criteria and come up with the most cost effective design.
- 7.7.4 Confirmation of CAAP Site Results. The performance of the Peroxone process is dependent on the water quality of the groundwater being treated. This project was conducted at a single site, and evaluated the remediation of explosives from one groundwater source. The chemical dose requirements and reaction kinetics are known to be function of the background organic matrix of the water being treated. Therefore, before the results of this study are extrapolated to other sites, it is recommended that the performance of the Peroxone process be tested at other sites using other groundwater sources to confirm whether or not such high chemical doses and contact times are also required for the treatment of other waters.

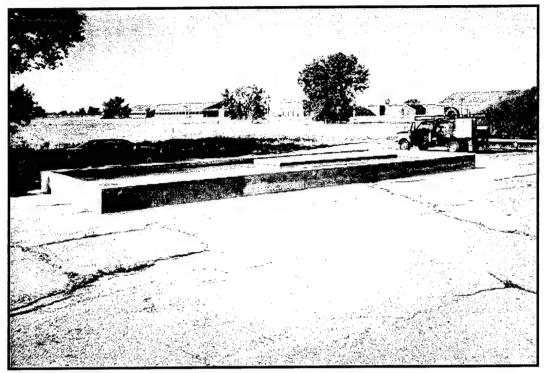
7.7.5 Challenging the 2 μ g/L Discharge Limit. The basis for the minimum concentration requirement of 2 μ g/L for each of TNT, TNB, and RDX set in the RFP is not known, and may not be based on scientific information regarding the health effects of these contaminants. It is noted that the chemical doses and contact time (i.e., system size) required to achieve an effluent TNB concentration of 4 μ g/L were virtually half those required to meet the 2 μ g/L limit. Therefore, it is our recommendation that this limit be challenged by conducting a wide review of all available information on the health effects of TNB in water. If the TNB discharge limit can be raised to 5 to 10 μ g/L, the cost of the treatment process may be substantially reduced by as much as 50 percent.

7.7.6 Conducting a more detailed cost estimate. The cost estimate developed in this report is a budgetary estimate. A more comprehensive engineering estimate should be developed in order to get a more accurate estimate of the treatment plant cost.

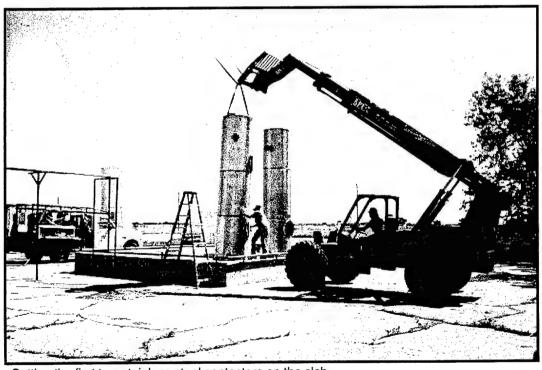
Appendix A

Peroxone System Construction Photographs





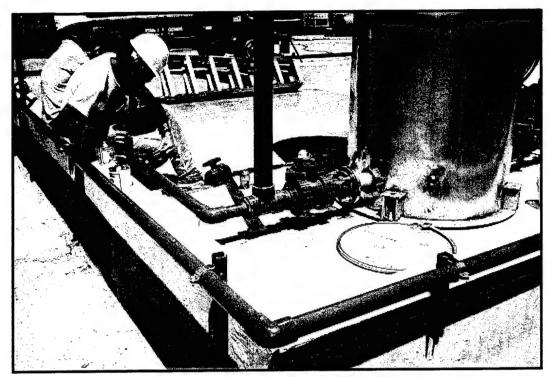
18 Foot x 48 foot secondary containment slab ready for Peroxone system.



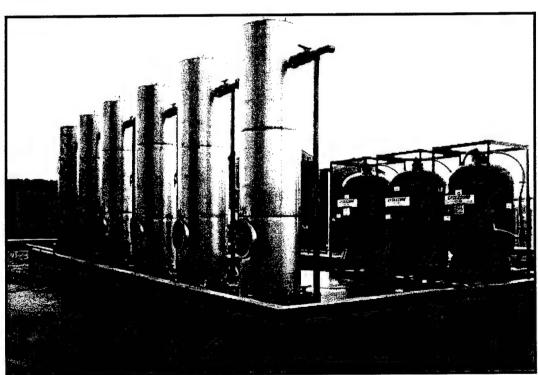
Setting the first two stainless steel contactors on the slab.



CORNHUSKER ARMY AMMUNITION PLANT GRAND ISLAND, NEBRASKA PHOTOS 1 & 2

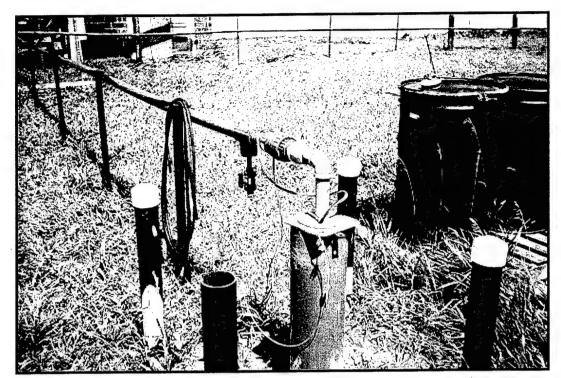


Workers connecting distribution piping from extraction wells into contactor number 1.

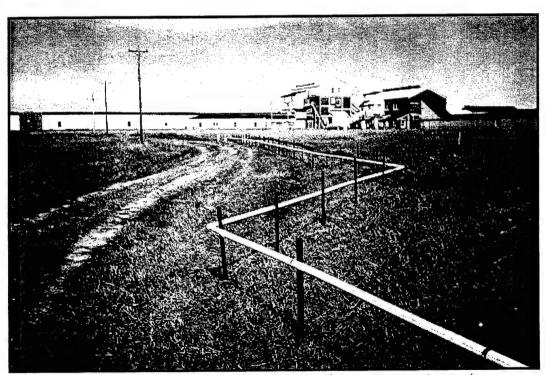


All contactors set and 3" connection piping completed. 3 Carbon vessels delivered and set on pad (right).



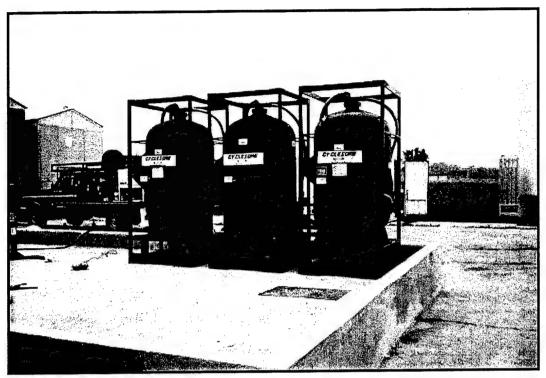


Completed wellhead for new TRW well showing 2" conveyance piping and sample port.

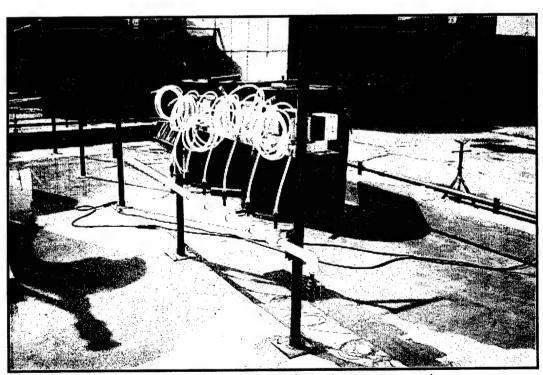


Water distribution system from fire hydrant (yellow object) to treatment system pad.



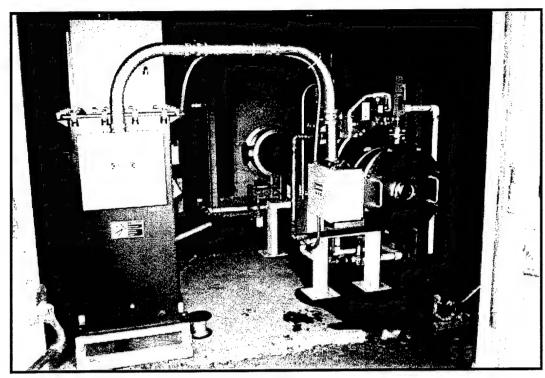


Three 1,000 pound Carbon vessels rented from Calgon Corporation.

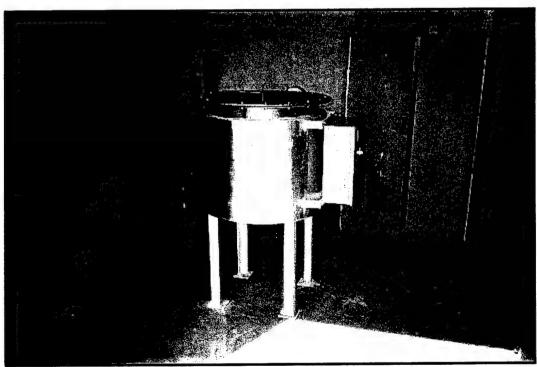


Six hydrogen peroxide chemical feed pumps, one for each contactor vessel.



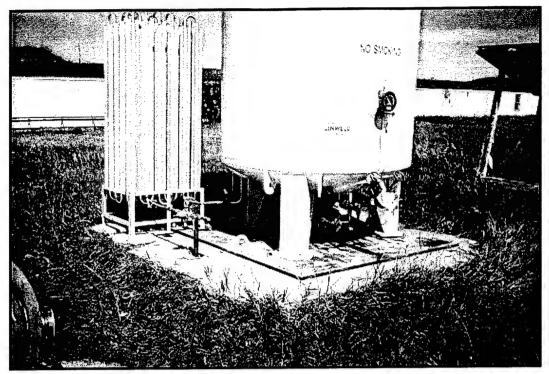


10 lb/day ozone generator (right) with power supply and control panel (left).

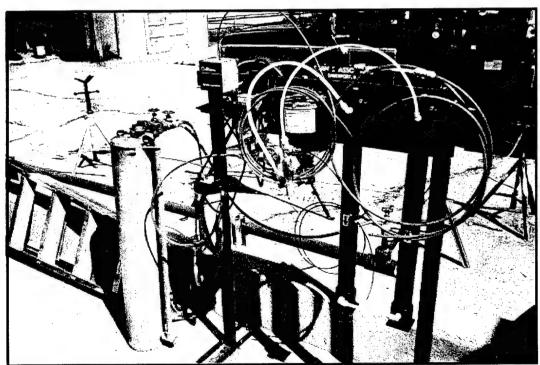


Ozone destruct unit, not yet connected to off-gas piping.



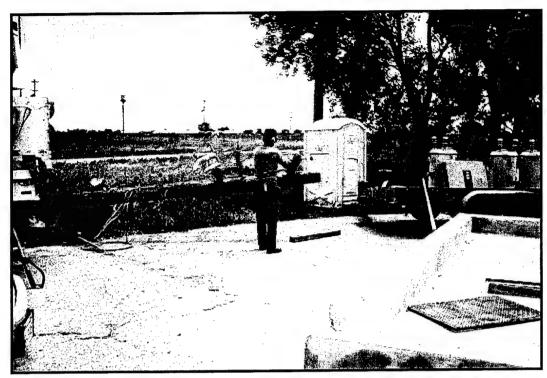


Leased liquid oxygen storage tank (right), oxygen evaporator (left), supplied by Linweld Oxygen.

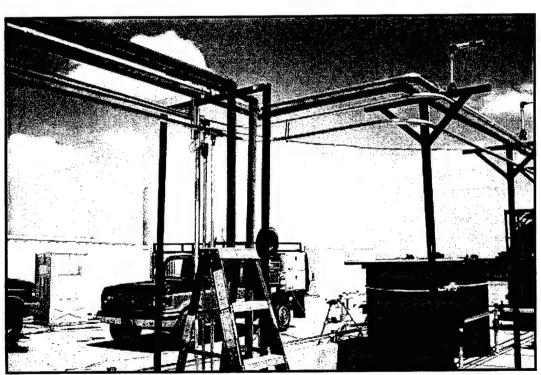


Reverse osmosis water purification unit. Carbon filter canister (left), R.O. units 2" gray vertical piping (right center).





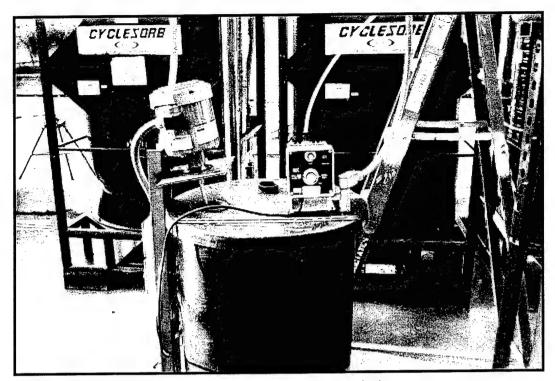
Power company installing 480 volt, three phase transformers with pole.



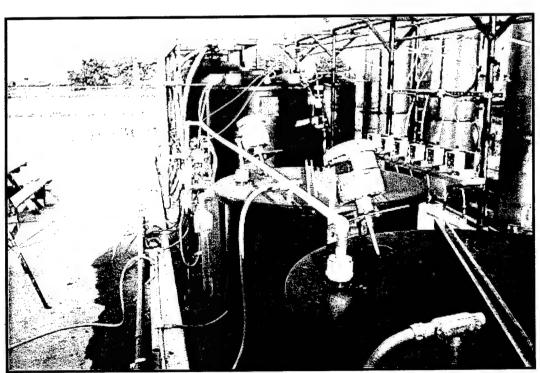
Conveyance piping, power conduits, influent water line coming from building housing the ozone generator to the treatment pad.

CORNHUSKER ARMY AMMUNITION PLANT GRAND ISLAND, NEBRASKA PHOTOS 13 & 14

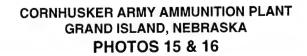


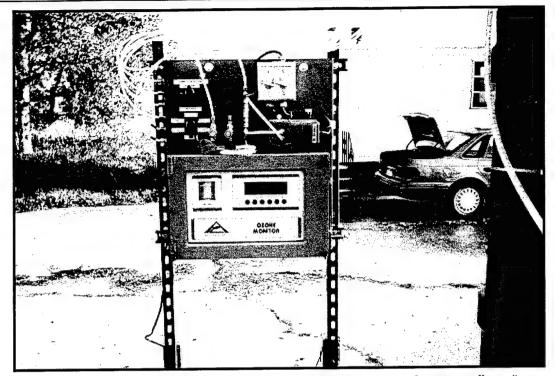


Sodium thiosulfate day tank with chemical metering pump and mixer.

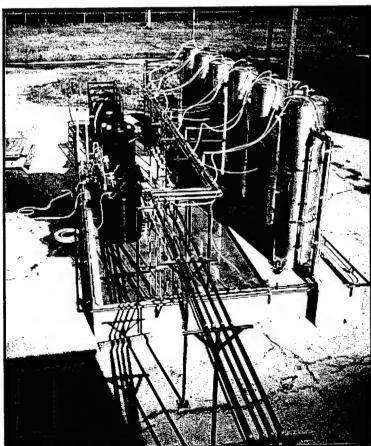


Hydrogen peroxide day tanks (right-center foreground) with mixers and R.O. water supply connections.





Ozone analyzer, showing connection tubing from ozone delivery lines. Contactor off-gas lines and oxygen supply line.



Completed peroxone demonstration system.

CORNHUSKER ARMY AMMUNITION PLANT GRAND ISLAND, NEBRASKA PHOTOS 17 & 18



Appendix B

Peroxone System OptimizationTesting Data

				Contactor	Contactor Contactor Contactor	Conforter							- 2		, ,,	A! 4 ¢		,	Al 7.6				
		Water		Applied 7	Contactor Applied Transferred Peroxide	Peroxide		Ozone			Total		dinitro-	dinitro-	dinitro-	dinitro-	2-Nitro-	3-Nitro-	donitro-	4-Nitro-		Nic	
Date Test Los	Test Location Well			Ozone	Ozone	Dose P	Dose PEROXONE	Residual	TNB TNT		RDX Nitrobodies	Nitrate	benzene	toluene	toluene	tofuene	toluene		tolucae	toluene	НМХ	benzene	Tetry
		(mdg)	(min)	(mg/L)	(mg/L)	(mg/L)	Ratio	(mg/L) ((µg/L) (µg/L) (µg/L)	L) (µg/L)	(µg/L)	(mg/LN)	(µg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)
8/28/96 0201 II	INFI	25.0							428 731		1560	5.71	2.2	BOL	BQL	BQL	BOL	BQL	331	BQL	10.8	BOL	14.3
8/28/96 0201 II	INF2 I	25.0							711 1200	00 73	2570	3.9	3.3	BQL	BQL	BQL	BQL	BQL	538	BQL	25.4	BQL	8
8/28/96 0201 []	INF3 I	25.0							498 800		2130	2.88	BQL	BQL	BQL	317	BQL	BQL	416	BQL	14.5	BQL	13.9
8/28/96 0201 C	C1/2 1	25.0	23.9	0.09	37.0	14.5	0.39	2.7	564 789		1410	0.207	9.0	9.4	8.0	BQL	BQL	BQL	BQL	BQL	18.3	BQL	2.1
8/28/96 0201 C	C1/4 1	25.0	23.9	0.09	37.0	14.5	0.39	2.2	429 548	8 57	1120	0.187	0.5	0.4	0.7	BQL	BQL	BQL	BQL	BQL	78.9	BQL	2.3
8/28/96 0201 C	C1/6 1	25.0		0.09	37.0	14.5	0.39	2.6			1800	0.374	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	147	BQL	2.8
8/28/96 0201 C	C1/8 1	25.0		0.09	37.0	14.5	0.39	2.1			1530	0.382	0.4	0.2	0.4	BQL	BQL	BQL	BQL	BQL	13	BQL	8.1
8/28/96 0201 C	C1/0 1	25.0	23.9	0.09	37.0	14.5	0.39	1.8	330 342		719	3.22	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	11.9	BQL	Ξ
8/28/96 0201 C	C2/0 1	25.0		0.09	35.0	14.5	0.41	1.2		1.12	292	1.98	BQL	BOL	BQL	BQL	BQL	BQL	BQL	BQL	0.7	BQL	0.1
8/28/96 0201 C	C3/0 1	25.0		0.09	38.0	14.5	0.38	2.8	_		146	1.53	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	4.7	BQL	0.2
8/28/96 0201 C	C4/0 1	25.0		0.09	37.0	14.5	0.39	2.0		2 1.6	6.99	1.08	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	3.1	BQL	BQL
8/28/96 0201 C	C5/0 1	25.0	23.9	0.09	35.0	14.5	0.41	1.2	34.3 3.8		41.7	8.	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	2.7	BQL	BQL
8/28/96 0201 C	C6/01 1	25.0	23.9	0.09	37.0	14.5	0.39	2.7			22.3	0.302	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	2	BQL	BQL
8/28/96 0201 C	C6/02 1	25.0	23.9	0.09	37.0	14.5	0.39		26.1 1.3	3 0.6	29.9	0.384	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	6.1	BQL	BQL
8/28/96 0201 G	GAC1 1										BQL	0.675	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL
8/31/96 0202 1	INFI	18.0										1.9	1.5	BQL	14.4	BQL	BQL	BQL	246	BQL	10.4	BQL	∞
8/31/96 0202 C	C1/2 1	18.0	33.2	65.0	53.0	15.0	0.28	1.5	154 122		299	2.22	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	10.2	BQL	0.3
8/31/96 0202 C	C1/4 1	18.0	33.2	65.0	53.0	15.0	0.28	4.1				2.21	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	7.6	BQL	0.4
8/31/96 0202 (C1/6 1	18.0	33.2	65.0	53.0	15.0	0.28	Ξ	206 156		384	2.19	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	6.5	BQL	0.3
8/31/96 0202 (C1/8 1	18.0		65.0	53.0	15.0	0.28	1.0				2.15	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	6.7	BQL	0.4
8/31/96 0202 (C1/0 1	18.0		65.0	53.0	15.0	0.28	8.0				2.13	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	5.3	BQL	BQL
8/31/96 0202 (C2/0 1	18.0		65.0	46.0	15.0	0.33	2.9				2.32	0.1	BQL	1.8	BQL	BQL						
8/31/96 0202 (C3/0 1	18.0		65.0	48.0	15.0	0.31	4.9	10 8.9			2.35	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	0.5	BQL	BQL
8/31/96 0202 (C4/0 1	18.0		65.0	48.0	15.0	0.31	3.0		9 0.3	11.5	3.12	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	0.4	BQL	BQL
8/31/96 0202 (C5/0 1	18.0		65.0	41.0	15.0	0.37	2.0				2.37	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	-	BQL	BQL
0202	C6/01 1	18.0		65.0	51.0	15.0	0.29	8.0			, 7.3	2.53	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	0.7	BQL	BQL
	C6/02 1	18.0	33.2	65.0	51.0	15.0	0.29					2.64	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL
0202	GAC3 1								BQL BQL		BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BOL	BQL
0203	INFI	18.0								823 59		1.98	2.3	BQL	22.6	BQL	BQL	BQL	369	BQL	14.7	BQL	11.5
0203	INF2	18.0										1.92	<u>∞</u> .	BQL	23	BQL	BQL	BQL	230	BQL	15.9	BQL	9.5
	INF3 1	18.0									1750	19.4	BQL	BQL	BQL	BQL	BQL	BQL	374	BQL	14.1	BQL	15.2
8/30/96 0203 (C1/2 1	18.0		85.0	0.99	18.0	0.27	2.7			200	2.26	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	∞	BQL	9.0
8/30/96 0203 (CI/4 1	18.0		85.0	0.99	18.0	0.27	2.2			362	2.24	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	6.9	BQL	6.0
8/30/96 0203 (C1/6 1	18.0		85.0	0.99	18.0	0.27	5.6				2.32	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	3.8	BQL	0.3
8/30/96 0203 (C1/8 1	18.0		85.0	0.99	18.0	0.27	2.1	_	Ξ.	_	2.17	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn
8/30/96 0203 (C1/0 1	18.0		85.0	0.99	18.0	0.27	8.1			•	2.36	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	10.7	BQL	6.0
8/30/96 0203 (C2/0 1	18.0		85.0	58.0	18.0	0.31	1.2		_		2.4	BQL	BQL	BQL	BQL	BQL	BQL	4.6	BQL	3.1	0.3	BQL
8/30/96 0203 (C3/0 1	18.0		85.0	26.0	18.0	0.32	2.8		7.4 0.7	•	2.46	BQL	BQL	BQL	BQL	BQL	BQL	2.5	BQL	1.2	0.2	BQL
8/30/96 0203 (C4/0 1	18.0		85.0	56.0	18.0	0.32	2.0				2.5	BQL	BQL	BQL	BQL	BQL	BQL	3.7	BQL	8.0	0.2	BQL
8/30/96 0203 (C5/0 1	18.0	33.2	85.0	52.0	18.0	0.35	1.2	Brkn Bı	Brkn Brkn	n Brkn	2.65	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn	Brkn

				Conf	Contactor Contactor Contactor	tartor Cor	nfactor							1.3	3.4.	26. 1	Amino 46			A Amino 26				
		Š	Water Cont	actor App	Contactor Applied Transferred Peroxide	sferred Per	roxide		Ozone			Total		dinitro-	dinitro-	dinitro-	dinitro-	2-Nitro-	3-Nitro-	4-Autuno-2,0 donitro-	4-Nitro-		Nitro-	
Date Test I	Test Location Well						Jose PE	Ä		TNB T	TNT RD		es Nitrate		toluene	toluene	toluene	toluene	toluene	tofuene	toluene	HMX	henzene	Tetryl
		3)	m) (mdg)	(min) (mg	(mg/L) (m)	(mg/L) (n	(mg/L)	Ratio	(mg/L)	(µg/L) (µg/L) (µg/L)	(L) (µg/l	(hg/L)	(mg/LN)	(µg/L)	(µg/L)	(µg/L)	(hg/L)	(µg/L)	(µg/L)	(hg/L)	(µg/L)	(µg/L)	(µg/L)	(µg/L)
8/30/96 0203	C6/01	31 1	18.0 33	33.2 85	85.0 52	52.0	18.0	0.35	2.7	4.7	1 BOL		2.71	BOL	BOL	BOL	BOL	BOL	BOL	8.7	BOL	BOL	BOL	BOL
0203	C6/02	31					18.0	0.35		4.3 B(BOL BOL	L 4.8	2.74	BOL	BOL	BOL	BOL	BOL	BOL	0.5	BOL	BOL	BOL	BOL
0203	GAC3	_								BQL B(_	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BOL	BQL
9/3/96 0204	INFI		13.0									1510	7.92	1.9	BQL	BQL	BQL	BQL	BQL	242	BQL	7.5	BQL	8.01
9/3/96 0204	C1/2		13.0 46	46.0 11:	115.0 87	87.0 2	22.5	0.26	2.0	180	126 16	330	2.82	BQL	BOL	7	BQL	8.0						
9/3/96 0204	C1/4		13.0 46	46.0 11:	115.0 87	87.0 2	22.5	0.26	2.7		114 15		5.11	BQL	BQL	9.9	BQL	9.0						
9/3/96 0204	C1/6		13.0 46	46.0 11	115.0 87	87.0 2	22.5	0.26	2.2		131 16	349	5.99	BQL	BQL	7.3	BQL	9.0						
9/3/96 0204	C1/8	-	13.0 46	46.0	115.0 87	87.0 2	22.5	0.26	2.0	169	131 17	325	6.36	BQL	BQL	7.6	BQL	9.0						
9/3/96 0204	C1/0		13.0 46	46.0 11	115.0 87		22.5	0.26	1.7				10.9	BQL	BQL	7.7	BQL	0.7						
9/3/96 0204	C2/0		13.0 46	46.0 11	115.0 79	79.0	22.5	0.28	3.5	59.6	18.2 3.6	5 85.2	9.71	BQL	BQL	3.6	BQL	0.2						
9/3/96 0204	C3/0		13.0 46	46.0 11	115.0 87		22.5	0.26	2.5	_	_	_	8.8	Brkn	Brkn	Brkn	Brkn	Brkn						
9/3/96 0204	C4/0		13.0 46	46.0 11	15.0 74	74.0	22.5	0.30	4.2	11.1			9.94	BQL	BQL	-	BQL	BQL						
9/3/96 0204	C5/0		13.0 46	46.0 11	115.0 7		22.5	0.30	2.8			L 4.8	7.94	BQL	BQL	9.0	BQL	BQL						
9/3/96 0204	C6/01		13.0 46	46.0 11			22.5	0.28	3.0	_			9.05	BQL	BQL	BQL	BQL	BQL						
9/3/96 0204	C6/02		13.0 46	46.0 11	115.0 79	79.0	22.5	0.28		1.6 B(BQL BQL		8.59	BQL	BQL	BQL	BQL	BQL						
9/3/96 0204	GAC3	_										_	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL
9/2/96 0205	INFI	2 I	18.0								536 21		20	BQL	BQL	∞	BQL	BQL	BQL	87.3	BQL	5.6	BQL	20.8
9/2/96 0205	INF2	2 1	18.0								626 29		8.52	BQL	BQL	21.2	BQL	BQL	BQL	109	BQL	6.9	24.9	31.4
9/2/96 0205	C1/2	2	18.0 33	33.2 43	43.0 3	34.0	0.6	0.26	6.0				8.94	BQL	BQL	8.9	BQL	BQL						
9/2/96 0205	C1/4	2	18.0 33				0.6	0.26	6.0				7.36	BQL	BQL	1.2	BQL	BQL	BQL	BQL	BQL	6	BQL	BQL
9/2/96 0205	C1/6	2					0.6	0.26	6.0				8.81	BQL	BQL	1.5	BQL	BQL	BQL	BQL	BQL	11.3	BQL	BQL
9/2/96 0205	C1/8	2					0.6	0.26	6.0	411 2	243 15		8.79	BQL	BQL	1.2	BQL	BQL	BQL	BQL	BQL	8.5	BQL	BQL
9/2/96 0205	C1/0	2					0.6	0.26	0.7				9.81	BQL	BQL	7	BQL	BQL						
9/2/96 0205	C2/0	2					0.6	0.33	2.9				7.41	BQL	BQL	3.9	BQL	BQL						
9/2/96 0205	C3/0	2				29.0	0.6	0.31	2.2				68.9	BQL	BQL	2.9	BQL	BQL						
9/2/96 0205	C4/0	2				24.0	0.6	0.38	3.8		16.6 0.7	7 57.5	8.42	0.1	BQL	BQL	BQL	BQL	BQL	BQL	BQL	Ι.Ι	BQL	BQL
9/2/96 0205	C5/0	2					0.6	0.38	2.5				10.2	BQL	BQL	2.5	BQL	BQL						
9/2/96 0205	C6/01	2					0.6	0.35	2.7	46.9 2	.5 BQL	J. 51.1	8.09	BQL	BQL	1.7	BQL	BQL						
9/2/96 0205	C6/02	2	18.0 33	33.2 4.	43.0 2	26.0	0.6	0.35				_	8.89	BQL	BQL	BQL	BQL	BQL.	BQL	BQL	BQL	0.7	BQL '	BQL
		2									_	_	10.1	BQL	BQL	BQL	BQL	BQL						
8/31/96 0206	INFI	2	0.81								439 28			-	BQL	21.5	BQL	BQL	BQL	120	BQL	5.5	BQL	26.1
8/31/96 0206	INF2	2	18.0											1.4	BQL	25.9	BQL	BQL	BQL	143	BQL	7	BQL	33.7
8/31/96 0206	C1/2	2				52.0	15.0	0.29	6.0				37.2	BQL	BQL	1.7	BQL	BQL						
8/31/96 0206	C1/4	2	18.0 33			52.0	15.0	0.29	=		21.3 1.6	•	11.8	BQL	BQL	1.7	BQL	BQL						
8/31/96 0206	C1/6	2	18.0 33	33.2 6	65.0 5	52.0	15.0	0.29	1.2		21.2 1.9	•	11.8	BQL	BQL	1.7	BQL	BQL						
8/31/96 0206	C1/8	2				52.0	15.0	0.29	6.0	_	_	_	12.1	Brkn	Brkn	Brkn	Brkn	Brkn						
8/31/96 0206	CI/0	2				52.0	15.0	0.29	0.7	3			12.8	BQL	BQL	BQL	BQL	BQL	BOL	BQL	BQL	0.3	BQL	BQL
8/31/96 0206	C2/0	2				45.0	15.0	0.33	2.4			3 34.2	12.7	BQL	BQL	1.7	BQL	BQL						
8/31/96 0206	C3/0	2				49.0	15.0	0.31	1.5				_	BQL	BQL	0.5	BQL	BQL						
8/31/96 0206	C4/0	2	18.0 3:	33.2 6	65.0 4	44.0	15.0	0.34	2.8	Brkn B	Brkn Brkn	kn Brkn	13.3	Brkn	Brkn	. Brkn	Brkn	Brkn						

					ontootor C	Contrader Contactor Contrador	Soutoclear							-	7.0	76								
		ż	Water Co	ontactor #	Applied Tr	Contactor Applied Transferred Peroxide	Peroxide		Ozone			Total		1,3- dinitro-	2,4- dinitro-	7	2-Anuno-4,6 dinitro-	t,o- - 2-Nitro-	. 3-Nitro-	4-Amuno-2,6 donitro-	2,6- 4-Nitro-		Nitn-	
Date	Test Location Well							岁		TNB TN	TNT RDX	z	es Nitrate		toluene							HMX	benzene	Tetry
		3	(mdg)	(min)	(mg/L)	(mg/L)	(mg/L)	Ratio	(mg/L)	(µg/L) (µg/L) (µg/L)	/L) (µg/L	-) (µg/L)	(mg/LN)	(ug/L)	(µg/L)	(µg/L)	(hg/L)	(Hg/L)	(µg/L)	(µg/L)	(μg/L)	(hg/L)	(µg/L)	(µg/L)
8/31/96	0206 C5/0	2	0.81	33.2	65.0	37.0	15.0	0.41	3.8	5.4 2.			12.9	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	0.7	BOL	BQL
8/31/96	0206 C6/01	2	18.0	33.2	65.0	52.0	15.0	0.29	1.2	I BQL	JL BQL	L 1.3	12.7	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	0.3	BQL	BOL
8/31/96	0206 GAC31	2								BQL BQL	JL BQL	L BQL	BQL	BQL	BQL	BQL	BQL	BQL		BQL	BQL	BQL	BQL	BQL
96/18/8	0206 GAC32	2								BQL BQL	_	L BQL	0.236	BQL	BQL	BQL	_			BQL	BQL	BQL	BQL	BQL
96/1/6	0207 INF1	2	0.81							-			5.42	BQL	BQL	13.9				103	BQL	5.2	BQL	27.5
96/1/6	0207 INF2	2	0.81							-		_	4.3	BQL	BQL	11.8	_	_		113	BQL	5	BQL	22.6
9/1/6	0207 INF3	2	18.0										6.37	BQL	BQL	6.5	_	_	_	89.7	BQL	5.3	BQL	24.8
9/1/6	0207 C1/2	2	0.81	33.2	85.0	0.79	18.0	0.27	1.5	200 13	134 15		5.95	0.1	BQL	9.0	BQL	_		BQL	BQL	8.8	BQL	8.0
9/1/6	0207 C1/4	2	0.81	33.2	85.0	0.79	18.0	0.27	2.0	187 11			4.48	BQL	BQL	BQL	_			BQL	BQL	4.9	BQL	0.5
9/1/6	0207 C1/6	2	0.81	33.2	85.0	0.79	0.81	0.27	2.0	203 12	123 9.5	342	5.06	BQL	BQL	BQL		BQL		BQL	BQL	6.5	BQL	0.5
96/1/6	0207 C1/8	2	0.81	33.2	85.0	0.79	18.0	0.27	8.	Brkn Br	Brkn Brkn	n Brkn	5.46	Brkn	Brkn	Brkn	Brkn	_			Brkn	Brkn	Brkn	Brkn
9/1/6	0207 C1/0	2	0.81	33.2	85.0	0.79	18.0	0.27	1.0				5.46	BQL	BQL							6.3	BQL	6.0
96/1/6	0207 C2/0	2	0.81	33.2	85.0	57.0	0.81	0.32	3.6				5.75	BQL	BQL			_				2.8	BQL	BQL
96/1/6	0207 C3/0	2	0.81	33.2	85.0	64.0	18.0	0.28	2.2				9.9	BQL	BQL			Ξ.				2.2	BQL	BQL
96/1/6	0207 C4/0	2	0.81	33.2	85.0	58.0	18.0	0.31	4.6				7.57	BQL	BQL							0.2	BQL	BQL
96/1/6	0207 C5/0	2	18.0	33.2	85.0	50.0	18.0	0.36	4.5				5.73	BQL	BQL							BQL	BQL	BQL
96/1/6	0207 C6/01	2	0.81	33.2	85.0	62.0	18.0	0.29	2.2	4.1 BC	BQL BQI	L 4.1	6.07	BQL	BQL			BOL				BQL	BQL	BQL
9/1/6	0207 C6/02	7	18.0	33.2	85.0	62.0	18.0	0.29			_		4.31	BQL	BQL			_				BQL	BQL	BQL
9/1/6	0207 GAC3	2											4.48	BQL	BQL							BQL	BQL	BQL
9/2/96	0208 INFI	2	13.0										7.16	BQL	BQL							3.8	BQL	BQL
9/7/96	0208 INF2	2	13.0											0.9	BQL							7.6	BQL	23
96/2/6	0208 C1/2	2	13.0	46.0	115.0	0.06	22.5	0.25	8.					BQL	BQL			_				4.4	BQL	BQL
9/2/96	0208 C1/4	7		46.0	115.0	0.06	22.5	0.25	2.5					0.2	BQL							3.3	BQL	BQL
9/2/96	0208 C1/6	2		46.0	115.0	0.06	22.5	0.25	2.0		97.3 5	260	7.41	0.2	BQL	BQL		BOL	, BQL	BQL	BQL	3.6	BQL	BQL
96/2/6	0208 C1/8	7		46.0	115.0	0.06	22.5	0.25	N. 1.8	168	102 6.1			BQL	BQL							3.8	BQL	BQL
9/2/96	0208 C1/0	2	13.0	46.0	115.0	0.06	22.5	0.25	1.4					BQL	BQL		, BQL					5.3	BQL	0.4
9/2/96	0208 C2/0	7		46.0	115.0	78.0	22.5	0.29	3.5		14.1 0.9			BQL	BOL						BQL	1.2	BQL	BQL
9/2/6		7		46.0	115.0	85.0	22.5	0.26	3.0					BQL	BQL			_				6.0	BQL	BQL
9/2/6		7	13.0	46.0	115.0	71.0	22.5	0.32	4.3	Brkn Br	_			Brkn	Brkn							Brkn	Brkn	Brkn
9/2/6	0208 C5/0	7	13.0	46.0	115.0	0.97	22.5	0.30	2.2		0.2 BQL			BQL	BQL			BOL			BQL	BQL	BOL	BQL
9/2/96	0208 C6/01	7	13.0	46.0	115.0	79.0	22.5	0.28	3.3	_				BQL	BQL						BQL	BQL	BQL	BQL
9/10/6	0209 INFI	_	13.0								803 56			BQL	BQL				BQL			13.9	BQL	10.3
9/10/6	0209 INF2	_	13.0											BQL	BQL							8.2	BQL	9.2
9/10/96	0209 INF3	_	13.0									_		BQL	BQL						_	10	BQL	6
9/10/96	0209 C1/2	_	13.0	46.0	38.0	32.0	6.91	0.53	0.0	280 2/		_	1.94	BQL	BQL							01	BQL	0.7
9/10/96	0209 C1/4	_	13.0	46.0	38.0	32.0	6.91	0.53	0.0	• •		Ī	2.12	0.2	BQL							11.4	BQL	8.0
96/10/6	0209 C1/6	_	13.0	46.0	38.0	32.0	16.9	0.53	0.0	269 2			2.19	BQL	BQL			_			_	9.3	BQL	0.7
96/01/6	0209 C1/8	_	13.0	46.0	38.0	32.0	6.91	0.53	0.0			_	2.15	0.1	BQL			_	_			10.9	BQL	8.0
96/01/6	020	_	13.0	46.0	38.0	32.0	16.9	0.53	0.0	260 2:		3 583	2.21	0.1	BQL	8.0	_	_	BOL	, 2.8		8.8	BQL	9.0
9/10/96	0209 C2/0	_	13.0	46.0	38.0	32.0	16.9	0.53	0.2		152 75		2.29	BQL	BQL	_	BQL	_			_	9.2	BQL	0.5

Appendix C

Peroxone System Demonstration Testing Data

		Fetryl	(P8/L)	를 ;	10.2	6.9	0.4	BQL	BOL	,	BQL	BOL		BQL	BQL	BQ.	15 E	x0 0	5,0	6.0	0.5			BQL			вог			BOL		BOL	,		BOL	BQL	BQL	BQL	BOL	BQL	BQL	5	8.2	9.8	BQL
	Nira	enzene	_	BOL	BOL	BQL	BQL	BQL	BOL	,	BQL	BOL	,	BQL	BQL	일	80E	7 2	2 2	2 2	BQL			BQL			BQL			30 10		Bol	,		BQL	BQL	BQL	BQL	BOL	BQL	g :	<u> </u>	를 달	BQL	BQL
		нмх ь	3	2 5	3 2 2	4.5	6.2	8:	-		BQL	BOL		BQL	BQL	BQE	BQL:		1.0	4.5	6.4			3.2			4.			=		8.0			0.5	0.4	BOL	BQL	BQL	BQL	gor.	1.7	6.5	7.1	9.6
	.Nim.	toluene		<u> </u>	30F	BQL	BQL	BQL	BQL		BQL	BOL		BQL	BQL	2 2	30 E	2 2	2 2	2 2	BQL			BQL			BQL			BQE		BOL			BOL	BQL	BQL	BQL	BQL	BQL	BQ.	F 5	g F	BQL	BQL
	Amino.2 6. 4	nitrotoluene		80F	BOL	BQL	BQL	BQL	BQL	,	BQL	BOL		BQL	BQL	BQL	BQL	10 P	שלב הלב	BOI.	BQL			BQL			BQL		;	BOL		BOL	,		BQL	BQL	BQL	BQL	BOL	BQL	9 G	2 2	BQL BQL	BOL	BQL
	-Nitro- 4-	stuene di	(µg/L)	ਰੂ ਤੋਂ ਕੋ	30F	BQL	BQL	BQL	BQL		BQL	BOL		BQL	BQL	BOL	BQL	7 E	70	BOL.	BQL			BQL			BQL			BQL		BOL			BQL	BŲL	BQL	BQL	BOL	BOL	BQL BQL	2 3	10 10 10 10 10 10 10 10 10 10 10 10 10 1	BQL	BQL
	-Nitro- 3	ioluene t	_	9 2 3	BOE 25	BQL	BQL	BQL	BOL		BQL	BOL		BQL	BQL	BOL	80r	2 3	2 2	305	BOL			BQL			BQL		;	BQL		BOL	,		BOL	BQL	BQL	BQL	BOL	BOL	BQL	2 2	BQL	BQL	BQL
	1.3-Dinitar-24-Dinitar-26-Dinitar-2-Amina-46-2-Nitar-3-Nitar-4-Amina-26-4-Nitar-	dinitratoluene toluene taluene dinitratoluene toluene HMX benzene Tetryl	(µg/L)	\$6.5	198	129	ВОГ	BQL	BQL		BQL	BOL		вог	BQL	BOL	BQL 103	761	6/1	125	0.7			BQL			0.3			BQL		BOL	,		BQL	BOL	BQL	BQL	BQL	BOL	BQL	174	145	165	5.2
	Dinitro- 2	whene di	(µg/L)	BQL BQL	BOL	BQL	BŲL	BQL	BQL		BQL	BOL		BQL	BQL	BQL	BQL 201	מלך	2 2	BOL	BQL			BQL			BQL			BQL		BOL			BQL	BQL	BQL	BQL	BŲL	BQL	BQL	70 E	BOL BOL	BQL	ВОГ
	nitro-2.6-	ine to																	14.5	10.8	0.2			BQL			BQL		;	BQL		BOL	,		BQL	BOL	BQL	BQL	BQL	BQL	30F	15.6	8.11	12.9	5
	ro-2.4-Di	e toluene	٦	13.2			BQL	BQL	BOL		BQL	BOL																																	
	1.3-Dinit	benzene	٦ .	2 2 2 3 3	90	BQL	BQL	BQL	BQL		BQL	BOL		BOL	BQL	BOL	10 E	3 2	2 2	BOL	BQL			BOL			BQL		1	S S		BOL	,		BQL	BQL	BQL	BQL	BQL	BOL	10g	4. 2	2 =	1.5	BQL
		Nitrate	(mg/L N)	2.14	2.61	2.69	2.92	2.86	2.86		2.99	2.82		3.06	3.16	3.12	5.37	5.75	77.7	2.38	2.16			2.32			2.54			2.72		2.8			2.89	3.09	2.85	2.91	3.84	2.84	2.83	2.18	2.56	2.18	2.56
	Total	TNT TNB RDX Nitrohydies Nitrate	(µg/L)	1370	1440	929	192	43.1	13.9		3.5	=		0.4	0.4	0.4	BG:	1410	1400	926	335			93.5			26.5			9.0		1.5			5	1.5	1.2	9.0	BOL	BOL	BOL	000	1140	1370	179
		RDX N	ug/L)	43.9	46.9	30.9	14.3	4.	0.4		BQL	BOL		BQL	BQL	BQL			43.4	23.4	43.4			8.7			Ξ			BQE		BOL	,		BQL	BOL	BQL	BQL	BOL	BQL	당:	1.1	35.4	39.6	9.01
		TAB.	(µg/L) (µg/L) (µg/L)	\$ 5	486	316	<u> </u>	31.1	Ξ		3.3	Ξ		0.4	4.0	0.4	BOL S	67.5	404	330	170			8			19.5		;	2.5		2.2			8.0	Ξ	-	9.0	BOL	BOL	를 등	338	410	478	9.96
		FE :	(µg/L)	433	672	432	67.2	90	*		0.2	BOL		BQL	BOL	BQ.	BQL	8 3	610	426	11			21.6			4.2		;	4.0		0.1			BOL	BOL	0.2	BQL	BQL	BOL	BQL	462	527	999	56.2
	Contactor	Peroxide	(mg/L)																																										
	Contactor Co		(mg/L) ((86 X	96	95 96	93	89	55	06	96	93						68	8	È	68	98	2	68	\$8	i	98 80	83	78		=	36 30		84								78
		Ο.	틧																																										
			(%)				2 7	1.7	2.0 1.7	6.1	2.0	1.8	2.2	1.7	6.						0.1	-	<u>:</u>	6.0		1	6.0	13	:	7.7	1.5	67		9.	0.1		4.1								1.5
	Contactor	Ozone	(%)	324			251 1.5	514 1.7	313 1.7		532 2.0	521 1.8		427 1.7	67		***	330	376	2	226 1.0	133		216 0.9	316 13		212 0.9	205 1.3		516 1.2	207 1.5	944 1.9		97 1.6	219 1.0		184 1.4		301			211	430	425	
	Oxidation Contactor Reduction Off-gas	Potential Ozone	(m) (%)	17 324											6.1			13 330	370				-																14 301			113 211		14 425	255
	Contactor	Sample Potential Ozone	(°C) (inV) (%)	=			18 251	17 514	313		532	521		427	6.1		:			:	226	133		216	316			205		316	207	44		195	219		184				:			4	255
	Temperature Oxidation Contactor of ORP Reduction Off-gas	Sample Potential Ozone	("C) (mV) (%)	7.0 17 324			251	514							61		:	2	77	:	7.2 13 226	1.4	riging and	7.4 13 216	74 14 216		212	7.7 14 205		13 516	7.8 14 207	14 944		14 195	8.1 14 219		8.1 14 184		7		:	23	7	7.1 14	13 255
	Ozone of ORP Reduction Offeas	Residual pH Sample Potential Ozone	(mg/L) ("C) (mV) (%)	=			7.2 18 251	0.2 7.4 17 514	7.5 313		7.7 532	0.3 7.9 521		0.1 8.0 427	18:00		:	2	77	2	0.0 7.2 13 226	17 14 933	0.2	0.0 7.4 13 216	74 14 216	0.1	7.6 212	0.0 7.7 14 205	0.3	7.7 13 516	0.0 7.8 14 207	7.9 14 944	0.0	8.0 14 195	0.0 8.1 14 219	0.0	0.0 8.1 14 184	0.4	8.1 14		:	23	7	7.1 14	7.2 13 255
	Operations Temperature Oxidation Contactor Sample Ozone of ORP Reduction Off-gas	Time Residual pH Sample Potential Ozone	(mg/L) ("C) (mV) (%)	71 0.7 00.00			15:52 0.0 7.2 18 251	15:48 0.2 7.4 17 514	15,44 0.2 7.5 313	18:00	15:38 0.6 7.7 532	15:32 0.3 7.9 521	18:00	15:27 0.1 8.0 427	18:00	CRAD	01.00	06:18	14:16		08:20 0.0 7.2 13 226	10:48 0.0	15:31 0.2	08:28 0.0 7.4 13 216	0.0	15:33 0.1	0.0 7.6 212	13:45 0.0 7.7 14 205	15:36 0.3	0.3 7.7 1.3 516	13:52 0.0 7.8 14 207	0.6 2.7 7.9 14 944	11:00 0.0	0.0 8.0 14 195	09:05 0.0 8.1 14 219	11:05 0.0	14:10 0.0 8.1 14 184	15:45 0.4	09:15 0.0 8.1 14	GACI		8.1 0.7	10:59 6.9 14	15:43 7.1 14	0.0 7.2 13 255
	Operations Temperature Oxidation Contactor Sample Ozone of ORP Reduction Off-gas	Time Residual pH Sample Potential Ozone	(#g/L) ("C) (mV) (%)	7.0	Ē	INFI	0.0 7.2 18 251	C2/0 15:48 0.2 7.4 17 514	0.2 7.5 313	C3/0 18:00	0.6 7.7 532	C50 15:32 0.3 7.9 521	C5/0 18:00	C640 15:27 0.1 8.0 427	C6/0 18:00		GAC3	51 07	INF 14:16 70 14	INF	CIA 08:20 0.0 7.2 13 226	CIA 10:48 0.0	CI/O 15:31 0.2	C2/0 08:28 0.0 7.4 13 216	10:54 0.0	C2/0 15:33 0.1	C3/0 08:40 0.0 7.6 212	13:45 0.0 7.7 14 205	C3/0 15:36 0.3	08:45 0.3 7.7 13 516 10:59 0.1	C4/0 13:52 0.0 7.8 14 207	15:39 0.6 08:56 2.7 7.9 14 944	C5/0 11:00 0.0	14:00 0.0 8.0 14 195	C640 09:05 0.0 8.1 14 219	C6/0 11:05 0.0	C6/0 14:10 0.0 8.1 14 184	C6/0 15:45 0.4	GAC3 09:15 0.0 8.1 14		GAC2	08:45	INF 10:59 6.9 14	INFI 15:43 7.1 14	09:00 0.0 7.2 13 255
	Average Operations Temperature Oxidation Confactor PEROXONE Sample Sample Oxone of ORP Reduction Officeas	Time Residual pH Sample Potential Ozone	(mg/L) ("C) (my/L) (%)	INFI 16:00 7:0 17	0.46 INFI	0.46 INFI	C1/0 15:52 0.0 7.2 18 251	0.46 C2/0 15:48 0.2 7.4 17 514	C2/0 18:00 C3/0 15:44 0.2 7.5 313	0.46 C3/0 18:00	C4/0 15:38 0.6 7.7 532	0.46 C50 15:32 0.3 7.9 521	0.46 C5/0 18:00	0.46 C6/0 15:27 0.1 8.0 427	0.46 CG/0 18:00	0.46	0.46 GAC3	INT 08:18 7.0 13	0.48 INFI 14-16 70 14	048 INFI	0.48 C1A 08:20 0.0 7.2 13 226	CIA 10:48 0.0	0.48 CI/O 15:31 0.2	0.48 C2/0 08:28 0.0 7.4 13 216	C2/0 10:54 0.0	0.48 C2/0 15:33 0.1	0.48 C3/0 08:40 0.0 7.6 212	C3/0 1037 0.0 C3/0 13:45 0.0 7.7 14 205	0.48 C3/0 15:36 0.3	C4/0 08:45 0.3 7.7 13 516 C4/0 10:59 0.1	0.48 C4/0 13:52 0.0 7.8 14 207	C4/0 15:39 0.6 C5/0 08:56 2.7 7.9 14 944	0.48 C5/0 11:00 0.0	C5/0 14:00 0.0 8:0 14 195 C5/0 15:41 0.2	0.48 C6/0 09:05 0.0 8.1 14 219	0.48 C6/0 11:05 0.0	0.48 C6/0 14:10 0.0 8.1 14 184	0.48 C6/0 15:45 0.4	0.48 GAC3 09:15 0.0 8.1 14	9.48	0.48 GAC2	11 0.7 CF:80 1-1N1	0.48 INFI 10:59 6.9 14	0.48 INFI 15:43 7.1 14	CIA 09:00 0.0 7.2 13 255
	Average Operations Temperature Oxidation Confactor PEROXONE Sample Sample Oxone of ORP Reduction Officeas	Dose Ratio Location Time Residual pH Sample Potential Ozone	(mg/L) ("C) (mV) (%)	42.5 0.46 INFT 16:00 7.0 17	42.5 0.46 INF1	42.5 0.46 iNFi	42.5 0.46 C1/0 15:52 0.0 7.2 18 251	0.46 C2/0 15:48 0.2 7.4 17 514	42.5 0.46 C2/0 18;00 42.5 0.46 C3/0 15;44 0.2 7.5 313	42.5 0.46 C3/0 18:00	42.5 0.46 C4/0 15:38 0.6 7.7 532	42.5 0.46 C5/0 15:32 0.3 7.9 521	42.5 0.46 C5/0 18:00	42.5 0.46 C6/0 15:27 0.1 8.0 427	42.5 0.46 CG/0 18:00	42.5 0.46	42.5 0.46 GAC3	41.1 U.46 INF1 U8:18 /.U 13	41.1 0.40 INT. 14.16 70 14	41.1 048 INFI	41.1 0.48 C1/0 08:20 0.0 7.2 13 226	0.48 CI/O 10:48 0.0	41.1 0.48 C1/0 15:31 0.2	41.1 0.48 C2/0 08:28 0.0 7.4 13 216	0.48 C2A 10.54 0.0 0.48 C2A 13.15 0.0 74 14 215	41.1 0.48 C20 15:33 0.1	0.48 C3/0 08:40 0.0 7.6 212	41.1 0.48 C3/0 10.37 0.0 41.1 0.48 C3/0 13:45 0.0 7.7 14 205	41.1 0.48 C3/0 15:36 0.3	0.48 C4/0 08:45 0.3 7.7 13 516 0.48 C4/0 10:59 0.1	4i,1 0.48 C4/0 13:52 0.0 7.8 14 207	0.48 C4/0 15:39 0.6 0.48 C5/0 08:56 2.7 7.9 14 944	41.1 0.48 C5/0 11:00 0.0	0.48 C5/0 14:00 0.0 8:0 14 195	41.1 0.48 CKN0 09:05 0.0 8.1 14 219	41.1 0.48 C6/0 11:05 0.0	41.1 0.48 C6/0 14:10 0.0 8.1 14 184	41.1 0.48 CG/0 15:45 0.4	41.1 0.48 GAC3 09:15 0.0 8.1 14	41.1 0.48	41.1 0.48 GAC2	37.1 0.48 INFI 08:45 7.0 13	37.1 0.48 INF1 10:59 6.9 14	0.48 INFI 15:43 7.1 14	37.1 0.48 C1/0 09:00 0.0 7.2 13 255
Average Average	Transferred Hydrogen Average Operations Temperature Oxidation Confactor Ozone Peroxide PEROXONE Sample Sample Oxone of ORP Reduction Off-gas	Dose Ratio Location Time Residual pH Sumple Potential Ozone	(mg/L) (mg/L) (mg/L) (%)	92 42.5 0.46 INFT 10.00 7.0 17	92 42.5 0.46 INF1	92 42.5 0.46 INFI	92 42.5 0.46 CHØ 15:52 0.0 7.2 18 251 0.0 42.5 0.46 CHØ 18:00	92 42.5 0.46 C.20 15.48 0.2 7.4 17 514	92 42.5 0.46 C2/0 18:00 92 42.5 0.46 C3/0 15:44 0.2 7.5 313	92 42.5 0.46 C340 18:00	92 42.5 0.46 C4/0 15:38 0.6 7.7 532 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5	92 42.5 0.46 C5/0 15:32 0.3 7.9 521	92 42.5 0,46 C5/0 18:00	92 42.5 0.46 C640 15.27 0.1 8.0 427	92 42.5 0.46 CG/0 18:00	92 42.5 0.46	92 42.5 0.46 GAC3	85 41.1 U.48 INF1 U8.18 7.0 13	95 41:1 0.46 INFT 14:16 70 14	85 41.1 048 INFI	85 41.1 0.48 C1/O 08:20 0.0 7.2 13 226	85 41.1 0.48 CI/O 10.48 0.0	85 41.1 0.48 CI/O 15:31 0.2	85 41.1 0.48 C2/0 08:28 0.0 7.4 13 216	85 41.1 0.48 C2A 10.54 0.0 85 41.1 0.48 C2A 13.35 0.0 74 14 216	85 41.1 0.48 C2/0 15:33 0.1	85 41.1 0.48 C3/0 08:40 0.0 7.6 212	85 41.1 0.48 C3/0 10.5/ 0.0 85 41.1 0.48 C3/0 13:45 0.0 7.7 14 205	85 41.1 0.48 C3/0 15:36 0.3	85 41.1 0.48 C4/0 08:45 0.3 7.7 13 516 85 41.1 0.48 C4/0 10:59 0.1	85 41.1 0.48 C4/0 13:52 0.0 7.8 14 207	85 41.1 0.48 C4/0 15:39 0.6 85 41.1 0.48 C5/0 08:56 2.7 7.9 14 944	85 41.1 0.48 C5/0 11:00 0.0	41.1 0.48 C5/0 14:00 0.0 8:0 14 195	85 41.1 0.48 C6/0 09:05 0.0 8.1 14 219	85 41.1 0.48 C6/0 11:05 0.0	85 41.1 0.48 C6A0 14:10 0.0 8.1 14 184	85 41.1 0.48 CG/0 15:45 0.4	85 41.1 0.48 GAC3 09:15 0.0 8.1 14	85 41.1 0.48	85 41.1 0.48 GAC2	77 37.1 0.48 INFI 08:45 7.0 1.3	77 37.1 0.48 INF1 10.59 6.9 14	37.1 0.48 INFI 15:43 7.1 14	77 37.1 0.48 C1/0 09:00 0.0 7.2 13 255
Average Average	Applied Transferred Hydrogen Average Operations Temperature Oxidation Contactor Ozone Ozone Peroxide PEROXONE Stannle Sannle Ozone of ORP Reduction Officers	Dose Dose Dose Ratio Location Time Residual pH Sumple Potential Ozone	(mg/L) (mg/L) (mg/L) ("G) ("A)	42.5 0.46 INFT 16:00 7.0 17	92 42.5 0.46 INF1	92 42.5 0.46 INFI	92 42.5 0.46 CHØ 15:52 0.0 7.2 18 251 0.0 42.5 0.46 CHØ 18:00	115 92 42.5 0.46 C2/0 15/48 0.2 7.4 17 514	42.5 0.46 C2/0 18;00 42.5 0.46 C3/0 15;44 0.2 7.5 313	115 92 42.5 0.46 C3/0 18:00	115 92 42.5 0.46 C4/0 15.38 0.6 7.7 532	115 92 42.5 0.46 C5/0 15:32 0.3 7.9 521	115 92 42.5 0.46 C5/0 18:00	115 92 42.5 0.46 C640 15.27 0.1 8.0 427	115 92 42.5 0.46 CG/0 18:00	115 92 42.5 0.46	115 92 42.5 0.46 GAC3	100 83 41.1 0.48 INF1 08.18 7.0 13	100 03 41.1 0.46 INF1 14.16 70 14	100 85 41.1 048 INFI	100 85 41.1 0.48 CI/O 08:20 0.0 7.2 13 226	100 85 41.1 0.48 CI/0 10.48 0.0	100 85 41.1 0.48 CI/O 15:31 0.2	100 85 41.1 0.48 C2.0 08:28 0.0 7.4 13 216	100 85 41.1 0.48 C2.0 10.54 0.0 10.0 10.0 10.0 10.0 10.0 10.0 10.0	100 85 41.1 0.48 C20 15.33 0.1	100 85 41.1 0.48 C3/0 08:40 0.0 7.6 212	100 85 41.1 0.48 C3/0 13.45 0.0 7.7 14 205	100 85 41.1 0.48 C3/0 15:36 0.3	100 85 41.1 0.48 C4/0 08:43 0.3 7.7 13 516 100 85 41.1 0.48 C4/0 10.59 0.1	100 85 41.1 0.48 C4/0 13.52 0.0 7.8 14 207	100 85 41.1 0.48 C4/0 15:39 0.6 100 85 41.1 0.48 C5/0 08:56 2.7 7.9 14 944	100 85 41.1 0.48 C5.0 11.50 0.0	100 85 41.1 0.48 C5/0 14:00 0.0 8.0 14 195	100 85 41.1 0.48 C6/0 09:05 0.0 8.1 14 219	100 85 41.1 0.48 C6/0 11:05 0.0	100 85 41.1 0.48 C640 14:10 0.0 8.1 14 184	100 85 41.1 0.48 CG/0 15:45 0.4	100 85 41.1 0.48 GAC3 09:15 0.0 8.1 14	100 85 41.1 0.48	100 85 41.1 0.48 GAC2	95 77 37.1 0.48 INF1 08:45 7.0 13	95 77 37.1 0.48 INF1 10.59 6.9 14	95 77 37.1 0.48 INFI 15:43 7.1 14	95 77 37.1 0.48 C1/0 09:00 0.0 7.2 13 255
Average Average	Transferred Hydrogen Average Operations Temperature Oxidation Confactor Ozone Peroxide PEROXONE Sample Sample Oxone of ORP Reduction Off-gas	Dose Dose Dose Ratio Location Time Residual pH Sumple Potential Ozone	(myl.) (myl.) (myl.) ((myl.) (%)	92 42.5 0.46 INFI 10:00 7.0 17	115 92 42.5 0.46 INFI	115 92 42.5 0.46 INFI	92 42.5 0.46 CHØ 15:52 0.0 7.2 18 251 0.0 42.5 0.46 CHØ 18:00	115 92 42.5 0.46 C200 15.48 0.2 7.4 17 514	92 42.5 0.46 C2/0 18:00 92 42.5 0.46 C3/0 15:44 0.2 7.5 313	115 92 42.5 0.46 C3/0 18:00	92 42.5 0.46 C4/0 15:38 0.6 7.7 532 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5	115 92 42.5 0.46 C5/0 15:32 0.3 7.9 521	115 92 42.5 0.46 C5/0 18:00	115 92 42.5 0.46 C640 15.27 0.1 8.0 427	115 92 42.5 0.46 CG/0 18:00	115 92 42.5 0.46	115 92 42.5 0.46 GAC3	100 83 41.1 0.48 INF1 08.18 7.0 13	95 41:1 0.46 INFT 14:16 70 14	100 85 41.1 048 INFI	100 85 41.1 0.48 CI/O 08:20 0.0 7.2 13 226	85 41.1 0.48 CI/O 10.48 0.0	100 85 41.1 0.48 CI/O 15:31 0.2	100 85 41.1 0.48 C2.0 08:28 0.0 7.4 13 216	85 41.1 0.48 C2A 10.54 0.0 85 41.1 0.48 C2A 13.35 0.0 74 14 216	100 85 41.1 0.48 C20 15.33 0.1	100 85 41.1 0.48 C3/0 08:40 0.0 7.6 212	85 41.1 0.48 C3/0 10.5/ 0.0 85 41.1 0.48 C3/0 13:45 0.0 7.7 14 205	100 85 41.1 0.48 C3/0 15:36 0.3	85 41.1 0.48 C4/0 08:45 0.3 7.7 13 516 85 41.1 0.48 C4/0 10:59 0.1	100 85 41.1 0.48 C4/0 13.52 0.0 7.8 14 207	85 41.1 0.48 C4/0 15:39 0.6 85 41.1 0.48 C5/0 08:56 2.7 7.9 14 944	100 85 41.1 0.48 C5.0 11.50 0.0	85 41.1 0.48 C5/0 14:00 0.0 8:0 14 195 85 41.1 0.48 C5/0 15:41 0.2	100 85 41.1 0.48 CAN 09.05 0.0 8.1 14 219	100 85 41.1 0.48 C6/0 11:05 0.0	100 85 41.1 0.48 C640 14:10 0.0 8.1 14 184	100 85 41.1 0.48 CG/0 15:45 0.4	85 41.1 0.48 GAC3 09:15 0.0 8.1 14	100 85 41.1 0.48	100 85 41.1 0.48 GAC2	95 77 37.1 0.48 INF1 08:45 7.0 13	95 77 37.1 0.48 INF1 10.59 6.9 14	77 37.1 0.48 INFI 15:43 7.1 14	95 77 37.1 0.48 C1/0 09:00 0.0 7.2 13 255
Average Average	Applied Transferred Hydrogen Average Operations Temperature Oxidation Contactor Ozone Ozone Peroxide PEROXONE Stannle Sannle Ozone of ORP Reduction Officers	e Dose Dose Ratio Location Time Residuat pH Sumple Potential Ozone	(gpm) (mg/L) (mg/L) (mg/L) (%)	115 92 42.5 0.46 INFT 16:00 7.0 17	1 13 115 92 42.5 0.46 INFI	1 13 115 92 42.5 0.46 INF1	115 92 42.5 0.46 C1/0 15:52 0.0 7.2 18 251	1 13 115 92 42.5 0.46 C2/0 15,48 0.2 7,4 17 514	115 92 42.5 0.46 C240 18:00 115 92 42.5 0.46 C3/0 15:44 0.2 7.5 313	1 13 115 92 42.5 0.46 C340 18:00	115 92 42.5 0.46 C4/0 15.38 0.6 7.7 532	1 13 115 92 42.5 0.40 C470 16500 1 13 115 92 42.5 0.46 C5/0 15:32 0.3 7.9 521	1 13 115 92 42.5 0.46 C5/0 18:00	1 13 115 92 42.5 0.46 CK/0 15:27 0.1 8.0 427	1 13 115 92 42.5 0.46 CG/0 18:00	1 13 115 92 42.5 0.46	1 13 115 92 42.5 0.46 GAC3	1 13 100 63 41.1 0.48 INF1 08:18 7.0 13	100 03 41.1 0.46 INFT 14.16 70 14	1 13 100 85 41.1 048 INFI	1 13 100 85 41.1 0.48 C1/0 08:20 0.0 7.2 13 226	100 85 41.1 0.48 CI/0 10.48 0.0	13 100 85 41.1 0.48 C1/0 15.31 0.2	i 13 100 85 41.1 0.48 C2/0 08:28 0.0 7.4 13 216	100 85 41.1 0.48 C2.0 10.54 0.0 10.0 10.0 10.0 10.0 10.0 10.0 10.0	1 13 100 85 41.1 0.48 C2.0 15.33 0.1	1 13 100 85 41.1 0.48 C3/0 08:40 0.0 7.6 212	100 85 41.1 0.48 C3/0 13.45 0.0 7.7 14 205	1 13 100 85 41.1 0.48 C3/0 15:36 0.3	100 85 41.1 0.48 C4/0 08:43 0.3 7.7 13 516 100 85 41.1 0.48 C4/0 10.59 0.1	1 13 100 85 41.1 0.48 C4/0 13:52 0.0 7.8 14 207	100 85 41.1 0.48 C4/0 15:39 0.6 100 85 41.1 0.48 C5/0 08:56 2.7 7.9 14 944	1 13 100 85 41.1 0.48 C5/0 11:00 0.0	100 85 41.1 0.48 C5/0 14:00 0.0 8.0 14 195	1 13 100 85 41.1 0.48 CK40 09.05 0.0 8.1 14 219	1 13 100 85 41.1 0.48 C600 11:05 0.0	1 13 100 85 41.1 0.48 C6/0 14:10 0.0 8.1 14 184	1 13 100 85 41.1 0.48 CG/0 15:45 0.4	1 13 100 85 41.1 0.48 GAC3 09:15 0.0 8.1 14	1 13 100 85 41.1 0.48	1 13 100 85 41.1 0.48 GAC2	13 95 77 37.1 0.48 INFI 08:45 7.0 1.3	95 77 37.1 0.48 INF1 10.59 6.9 14	1 13 95 77 37.1 0.48 INFI 15:43 7.1 14	95 77 37.1 0.48 C1/0 09:00 0.0 7.2 13 255

(3-Dinitro-2,4-Dinitro-2,6-Dinitro-2,Amino-4,6-2-Nitro-3-Nitro-4-Amino-2,6-4-Nitro-Nitro-Perroren taluene tinterne dinitrotoluene toluene dinitrotoluene toluene HMX benzene Teteyl (1987) (1987) (1987) (1987) (1987) (1987) (1987) (1987) (1987)		BOL	,		,																8.3																				
					BOL			BOL	,		PO.	,		BOL	BOL	BQL	BOL	BOL	BQL	BOL	를 다 다	į		BOL			BQL			BQL			BOL			BQL	B 2	10 E	BQL	BQL	BQL
		6.			1			6.0			0.5	,		BOL	BOL	BQL	BOL	10.9 10.9	7.2	7.3	9.4	!		30			4.			6.0			BQL			BQL	10 E	1 1 2 1 3 1 1 1 1	BQL	BOL	BQL
ino-2,6- 4- atoluene te		BOL			BOL	,		BOL	ì		BOL	: '		BOL	BOL	BQL	BOL	E CL	BOL	BQL	BQL	ř		BOL			BQL			BQL			BQL			BQL	BQE	10 10 10 10 10 10 10 10 10 10 10 10 10 1	BOL	BOL	BQL
		BOL			BQL			BOL	1		BOI.	ļ		d	BQL	당	당 :	글글	5 75	70	BQL BOIL	1		BOL			BQL			BQL			BQL			BOL	<u> </u>	BOL	301	301	3QL
c dinitr																																									
e toluene (µg/L)		BQL			BOL			BOL			BOL			BOL				1 G			BOL S			BOL			, BQL			BOL			BOL			L BQL				L BQL	
e toluene (ug/L)		BQL			BOL	•		BOL			BOL	í		BOI	BOL	BQL	BQL	d d	BQL	BOL	BQL	'n		BOL			BOL			BQL			BQL			BQL	HOL BOL	ğ	80	BQL	В
2-Amino-4,6 dinitrotoluen (µg/L)		BQL			BOL	,		BOL	}		BOL	,		BOL	BQL	BQL	BQL BQ:	12 5	202	133	127 ROI	,		BOL	,		BQL			BQL			BQL			BQL	BQE	10g	BQL	BQL	BQL
2,6-Dinitro- toluene (µg/L)		BQL			BQL			BOL	ř		BOL	ì		BOL	BQL	BQL	BQL	gor Bor	BQL	BQL	BQL	,		BOL	,		BQL			BQL			BQL			BOL	10g	80 E	BQL	BQL	BQL
4-Dinitro-		BQL			BQL	,		BOL	ļ		BOI.	i i		BOL	BQL	BQL	BQL	10.6	16.9	13.1	II 0	,		BOL	,		BQL			BQL			BOL			BQL	BOL	BOL	BQL	BQL	BQL
3-Dinitro-2 benzene (µg/L)		BQL			BQL	,		BOL	i Y		BOL	l Y		BOL	BQL	BQL	BOL	10 G	2.4	BQL	BQL	ľ		BOL			BQL			BQL			BQL			BOL	10g	BG F	BOL	BQL	1.2
		2.56			2.7			2.5	ŀ		2.7	i		2.75	2.56	2.42	2.46	1.73	2.02	2.01	1.89			2.41			2.64			2.63			2.77			2.96	2.47	2.4	2.2	2.17	2.28
Total TNT TNB RDX Nitrobodics Nitrate (1921.) (1921.) (1921.) (1921.)		54.4			13			4.6			1.7			0.5	0.4	0.4	0.5	1020	1350	1050	1010	i		53.1			19.4			5.7			9:1			9.4	9.0	13	BQL	BQL	1.2
RDX (ug/L)		1.5			BQL	,		BOL	ř		BOL			BOL	BQL	BQL	10 E	40.3 10.3	49.5	30.9	32.3			5.			0.5			BOL			BQL			BOL			BQL		BQL
TNT TNB RDX (9g/L.) (9g/L.) (9g/L.)		1.1			10.5			3.5			1.2				0.4	9.4		370			389			40			15.8			4.6			9.1				9.0			BOL.	
		6.6			1.2			0.2			BOL			BOL	BQL	BQL	BQL	456	614	490	433			8.0			1.7			0.2			BOL			BQL	108	4.0	BQL	BQL	BQL
or Contactor ed Measured Peroxide (mg/L)																																									
Transferred Ozone (mg/L)	78	77		78	76		11	77		74	28		78	79	:	77					92	:	79	76		79	76	92	2	7.3	79		75	79		75	F				
Contactor Off-gas Ozone (%)	1.5	1.6		5	1.7		971	9.1		1.8	51		5.	4.		1.6					1.1		1.7	61		1.7	6.1	1.3	-	2.2	1.7		2.0	1.7		2.0	-	2			
Oxidation Reduction Potential (mV)	311	545	995	831	646	871	850	926 879	934	936	534	842	895	889	920	910	606	307	403		949		837	828	1	942	936	640	7+6	943	626		870	834		939	760	•	311		
Temperature of ORP Sample (°C)	4 2	4 E	4	2 2	7	4	52	<u> </u>	7	z :	<u> </u>	7	= :	<u> </u>	4	14	4 :	13	15		2	:	11	13		9	13	ž	2	13	20		23	20		13	2	=	16		
	7.1	7.1	1.4	7.4	97	1.6	7.6	7.7	7.8	7.7	7.9	7.9	8.0	0.8	0.8	 %	0.8	7.1	7.0		7.3	:	7.1	7.4	1	7.3	7.7	7.6	9.	7.9	7.8		œ.	7.9		œ	9	0.0	8.2		
Ozone Residual pH (mg/L)	0.0	0.3	0.2	0.4	0.4	0.5	9.0	0.7	0.9	0.7	0.1	0.3	0.4	0.0	0.5	0.5	0.8	0.0			0	0.3	0.4	0.0	0.7	9.6	0.8	9 7	2.1	1.2	₹ =	1.9	0. 0	3 =	2.1	0.	7	1 1	0.0		
Operations Sample Time	11:04	16:00	11:12	14:26	09:22	11:32	14:35	06:50	11:42	15:15	09:50	11:50	15:26	0000	11:55	15:35	16:45	08:15	13:43		08:17	10:55	13:52	08:30	10:58	14:03	08:40	11:03	16:26	08:47	14:21	16:36	08:57	14:42	16:42	09:25	11:15	16:49	09:33		
	CIA	C1/0	C2/0	C250	C3/0	C3/0	0.3	C 28	C4/0	C470	S 5	C5/0	CSAO	2 2	C6/0	C6/0	C640	NFI.	INF	E E	INF.	CIVO	CIO	220	C20	23.00	C3/0	Q (2)	C30	C4/0	2 S	C4/0	C58	CSAO	C5/0	Cevo	CRAD	Cen Cen	GAC3	GACI	GAC2
Hydrogen Average Peroxide PEROXONE Sample Dusc Ratto Location (mg/L)	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42	0.42
tydrogen Peruxide PE Dusc (mg/L)	37.1	37.1 37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	37.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1	32.1
Transferred Hydrogen Ozone Peroxide Dose Dose (mg/L) (mg/L)	rr rr	r r	11	רר רר	: 1	11	t :	: 1	7.1	77	t t	11	77	: 1:	11	11	77	: %	92	92	92 Y	92	92	ę <u>9</u>	92	9 2	92	2 2	2 22	92	92	92	92	76	92	26	٤ ٤	2 %	76	92	76
Applied Tra Ozone Dose (mg/L) (95 95	95 95	9.5	25 X	; %	95	56 5	95 95	95	56	26 S	95	95	S S	95	95	95	r 36	86	86	86 86	86	86	8 %	86	% %	86	86 8	s 26	86	s s	86	% 8	86	86	86	86 8	8 %	86	86	86
		e -					m -	m	<u></u>	<u> </u>	2 2	<u> </u>	13	<u> </u>	==	13	e :	2 2	2	3	5 5	==	2 5	2 5	2	5 5	: 2	= :	2 2	83	==	2	<u> </u>	2 = 2	5	5	<u>∽</u> :	2 52	=	<u>e</u>	13
Process Well Flow Rate (gpm)		E E	13	2 2	2 2	13	Ξ:	2 2		<u></u> :	_ :	. ==	1	_ =	. =	<u>-</u>	_ :	_	-				_ :		- -		. -			_		_			_	_			_	_	_
	96	2 × 3c	1 96	9 %	36	1 96	96	 9 9	96	1 96	9 9	96	96	9 9	96	96	1 96	- 12	1 96	1 96	1 96	96	96	96	96		96	96/	1 96	96	96	96	1 96/	196	96	96/	96/	96,	96/	96/	9/16/96
Date	96/51/6	9/15/96	9/15/96	9/15/96	9/15/96	9/15/96	9/15/96	9/15/96	9/15/96	9/12/96	9/15/96	9/15/96	9/15/96	9/15/96	9/15/96	9/15/96	9/15/96	96/91/6	9/16/96	9/16/96	9/16/96	96/91/6	96/91/6	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	96/91/6	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	9/16/96	9/16

1		Tetryl (µg/L)		8.1 V 0	. ~!	4			EQ.	3			BQL			BQL			BOL			5	200	BQL		BQL	٠ :	: =	9	0.3			BQL		;	BQL			BQL			BQL			BQL.	30F	BQL BQL
	ė	rene Tel		BOL 8		BQL			BOT. B				BQL B			BQL B			BOL					BQL B		BQL B		g g					BQL B			B TOB			BQL			BQL B					202
	Ž	dX ben: (L) (ng		9.9 B(0				BQL			BQL B			BQL B				BOL B			BQL B		4.6 B					9. 9.			S:0			BQL			BŲL					80 E
	itro-	tolucne HMX benzene Tetryl (ug/L) (ug/L) (ug/L)		BOL 9					BOH.				80F			BQL B			BQL B			100				BQL B							BQL			BQL BQL			BQL			BQL E					80°E
	2,6- 4·N	iene tok		in a	i mi	æ																																									
	-Amino-	initrotofue (µg/L)		B 5	BOL	BQL			ROL	2			BOL			BQL			BQL			Č	E C	BQL		BQL	2 2	3 G	BCL	BOL			BQL			BQE CE			BOL			BQL			BQL	BQE	고 고
	Nitro- 4	toluene di (µg/L.)		2 2	BOL	BQL			ROIL	2			BQL			BQL			BQL			EO.	BOL S	BQL		BQL	2 2	B E	BQL	BQL			BQL		;	BOL			BQL			BQL			BQL	BQL	15 E
	Nitro- 3	(µg/L) (10 E					BOI.				BOL			BQL			BQL			IO8	BOL	BQL		BQL	2 2	10 10 10 10 10 10 10 10 10 10 10 10 10 1	BQL	BQL			BQE BQE			BQL			BQL			BQL			BQL	BQL	BQL BQL
	1,3-Dinitro-2,4-Dinitro-2,6-Dinitro- 2-Amino-4,6- 2-Nitro- 3-Nitro- 4-Amino-2,6- 4-Nitro-	and a								2.			7			٦.			BŲL			=	10E	, ,		BQL	2 2	. 2	73	7.			BQL		;	BOL			BQL			BQL			귥	궁:	80°L
	2-Amin	dinitrotolus (µg/L)	1	143	9	BQ			ROL	ž			BOL			BQL			Ä			2	3 2	B		æ:	2	. 12	7	æ			ĕ		ì	ñ			ĕ			ă			ě	ě.	ā ā
	-Dinitro	(pg/L)		5 5	10E	BOL			BOI	7		:	BQL			BOL			BQL			O	30F	BQL		BOL	2 2	10g	BQL	BQL			BOL			BOL			BQL			BQL			BQL	BQL	BQL BQL
	initro- 2,0	toluene (µg/L)		<u> </u>	. 2	BQL			ROI.	1			BOL			BOL			BQL			108	충청	BQL		JÒE :	12.5	3 =	9.8	QF.			BOL		;	RQL BQL			BQL			BQL			BQL	JQL.	BQL BQL
	ro-2,4-D			- =	- 4	9			æ	i			Ä																																		
	1,3-Dinit	benzene (µg/L)	1	2 2	BOL	BOL			BOL	2			BQL			BQL			BQL			BOI	BOL 5	BQL		BQL	2 2	걸	-	BQL			BQL		i	BQL			BQL			BQL			BQL	108	BQL BQL
		Nitrate (mg/L N)		10.2	2.03	2.26			2.46	i			2.45			2.43			2.22			356	2.72	2.17		3.07	8 6	8.	2.02	2.41			2.11		;	2.35			2.44			2.48			2.7	2.61	2.47
	Total	itrohodies (μg/L) (R 9	470	192			20.6			:	<u>4</u>			4.5			1.7			Š	0.4	0.5		BQL	3 2	, i	BQL	BQL			BOL			PQF			BQL			BQL			BQL	BQL	BQL BQL
	-	Z		34.4		1.7			4				BOL			BQL			BQL			5	BOL	궁		BOL							4.			6.3			BQL			BQL			BQL	당:	BQL BQL
		INB RI		4.10					37.8				12.6 B			43 B			1.5 B			0.5 B				BQL B			250				38.2			2			3.4			1.2					0.4
		TNT TNB RDX (µg/L) (µg/L)		¥ %					9.5			:	1.5			0.2			0.2			BOI	BOL	BQL		BQL		265	292	53.8			5 .			4:			0.2			BQL			BQL	BQL	BQL BQL
Contactor	casured	Peroxide (mg/L)	ł			24.7		23.6	25.1		25.1		20.0	25.5		22.6	30.4	t'na	21.1		28.0	24.4	Š	24.4						22.5	22.9		23.8	23.8	į	27.3	23.8		25.5	15.		28.1	10.5	7.7	23.8	į	22.9
	_					_		•	_		_		•	_		•	22		_		_	20		_						18	6/		6/	78					79	07		==	2		62	:	æ
r Contactor		Ozone (mg/L)				77	i	2	ž		=		76	2 0		79	ř	-	77		≅	-	•	200						1	7		7	7	•	30	30		_	-	•	3 C	,	•	-	•	
Contactor	Off-gas	Ozone (%)				-		9.1	57		5.	;	9.	1.5		9.1	4	2	90			9	2	1.5						1.7	1.6		9.	1.7	;	1.5	1.5		1.6	1		1.5	4	4.	1.6	:	1.7
Oxidation	Reduction	Potential (mV)		318		267	;	243	254		252		266	528		481	103	100	877		211	700	2	214		369	613	420		547	587		247	784		226	395		119	K74		533	366	000	848	į	818
																_			_		_						_	_		_	10		_	٠,		_	1			_		-			**		_
Temperature	of ORP			52		4		2 2			91		2	17		14	-		14		17	7		17		2 2		4			2 15		5	4 16		<u> </u>	5 17		20	-		0			_		8.0
	эис	Residual pH (mg/L)		7.0		0.2 7.1		0.0 7.2			0.0		0.1	0.1 7.6	0.1	7.7 1.0	0.8		0.5 7.8		0.0 8.0	0.1		0.0 8.1	0.0	0.0 7.9	2	7.0		0.3 7.2	0.3 7.2		0.2 7.3	0.4 7.4		97 03	0.1 7.6	0,1	0.3 7.8	0.2 7.7		0.1 8.0	0.5		0.3 8.1		
suo				ve																								6																			2 2
Operations		Time		13.08		08:26	10:40	13:11	08:43	10:52	13:19	15:35	10.55	13:26	15:40	80:60	10:58	15:44	81:60	10:11	14:03	1000	11:05	14:10	15:57	09:45	3	61:60		06:59	06:30	11:15	07:20	09:50	11:17	08:53	10:11	11:19	07:43	08:36	11:23	07:59	10:01	11:33	08:14	09:02	11:24
	Peroxide PEROXONE Sample	Location	į	Z Z	INF.	CIV	Q :	2 5	200	C2/0	C2/0	C20	5 5	30	C3/0	C4/0	3 5	CAS C	C5/0	C5/0	CSV	25	C6/0	C6/0	CKO	GAC3	E N	Ž	NF	CIO	2 2	C1/0	3 2	C20	C2/0	2 8	C3/0	C3/0	C450	3 8	C4V	CSAO	C.550	25.5	CGO	C6/0	080
Average	OXONE	Ratio		0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	28	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.38	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48 0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48
co Av	le PER(a. a.
Average Average Transferred Hydrogen	Peroxi	Dose (mg/L)	9.00	30.4	30.4	30.4	30.4	30.4	30.4	30.4	30.4	30.4	30.4	30.4	30.4	30.4	50.4 4.05	30.4	30.4	30.4	30.4	30.4	30,4	30.4	30.4	30.4	38.7	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2	38.2
Average	Ozone	Dose (mg/L)	į	2 2	2	62	5	£ 2	5 2	2	52	2	5 5	. 2	70	2	2 2	20,00	62	79	76	2 2	: 2	52	62	£ 5	2 2	2 2	6/	62	2 2	16	67 P	3 2	79	2 2	79	79	2 1	2 2	2	79	\$ \$	2 2	. 62	97	2 2
Average A Applied Tra		Dose (mg/L) (ć æ	20	86	86	æ 3	8 26	86	86	86	e e	86	86	86	20 00	2 86	86	86	86	× %	2 25	86	86	86 8	90	. %	86	86	× ×	86	98 8 95 8	8 86	26	e e	86	86	86	× ×	. 86	86	86	e e	. %	86	36 36 36 36
Ave				. 5																												-		-						===				2 2			2 2
	Process	Well Flow Rate (gpm)	3		===	13	≘ :	= :	2 2	2	23	Ω:	2 2	2 2	13	Ξ:	2 2	2 2	=	13	13	2 2	2 2	13	13	≘ :	2 2	2 2	Ξ.	≏ :	2 2	2	= =		E :		- 13	13	s :	_ =	. =	-	_ = =		. =	-	
				96	1,06	1 96/	96/	1 96/	964	96/	96/	96/	1,96	1 96/	96/	96/	96/	706	1 96/	1 96/	1 96/	96/	96/	1 96/	1 96/1	96/1	70%	9/18/96	1/96	3/96	1 96/81/6	3/96	1 96/8	1 96/8	1 96/81/6	1 96/81/6	9/18/96	9/18/96	96/81/6	9/18/96	9/18/96	9/18/96 1	1 96/8//6	1 96/81/6	96/81/6	9/18/96	9/18/96
		Date		96/11/6	9/11/96	9/11/6	96/11/6	9/11/6	96/11/6	96/11/6	9/11/96	9/12/96	9/11/96	9/17/96	96/11/6	96/11/6	96/1/6	9/11/6	96/11/6	9/11/6	96/11/6	96/11/6	9/11/6	96/11/6	9/11/6	96/11/6	0/18/06	81/6	9/18/46	9/18/6	9/18/96	96/81/6	9/18/96	9/18//6	31/6	31/6	81/6	31/6	31/6	3/6	81/6	81/6	1/6	11/6	: 1/6	81/6	1/6

l		1. (J.		ş -	.2	7.1	۲. ۱	œ.			BQL			POI.	ì			BQL			BQL			Ö	, ₁	공	5	글 :	3 5	, 5	6.7	e 2	. 5			BQL			BQL			BOIL	ڍ			BQL		
	ė	ene Tetryl L) (µg/L)																			BQL B				BOLB						BQL 1					BOL B			BQL B			BOI B				BQL		
	Nitro-	K benzene J (µg/L)				BQL					BOL			HOL				BOL																												BQL B		
		ic HMX		12.9			. 73				. 2.5			1.7				. 0.7			0.4				108								. 			- E			1.0.8			• 0						
	2,6- 4-Nim	(µg/L) (µg/L) (µg/L)	N/A	B G	BOL	BOL	10 E	B O			BOL			BOIL				BOL			BQL				BOL				BOL				BQL			BQL			BQL			IOR				BQL.		
	4-Amino	dinitrotofu (µg/L)	08	10g	BQL	BOL	BQL	BQL BQL			BQL			BOL				BQL			BQL			BOL	BOL	BQL	BQL	108	BO!	BQL	BQL	10g	BQL			BQL			BQL			BOI	ż			BOL		·
	- 3-Nitro-		BOIL					EÇ.			BOL			BOL				BQL			BQL				BOL		BQL.				BOL					BOL			, BQL			POI				n BQL		
	2-Nitro	tolucne (ug/L)	ION I	E E	BQL	BQL	3 2 3 3	Ę,			BOL			BOL	,			BOL			BQL			BOL	BOL	BOL	BG	<u> </u>	B C	BCI	BOL	B 5	, E			BQL			BQL			BOIL	2			B OL		
	1,3-Dinitro-2,4-Dinitro-2,6-Dinitra-2-Amino-4,6-2-Nitro-3-Nitro-4-Amino-2,6-4-Nitro	dinitrotoluene toluene (µg/L)	Ca	₹ ₹	95.7	123	129	BQL			0.4			BOL	ļ			BOL			BQL			BOL	BOL	BQL	BQL	E E	BOL	125	140	37	BQL			BQL			BQL			ROIL	370			BQL		
	.6-Dinitro-	toluene e (µg/L)	2	BQL	BOL	BQL	BQE	BUL			BQL			BOL				BOL			BQL			BOL	BOL	BOL	BQL	BQL BQL	BOL BOL	BOL	BOL	BQF	BQL			BQL			BQL			BOIL	i d			BQL		
	4-Dinitro-2	toluene (µg/L)	5	2 2	8.7	11.4	6.H.	10°E			BOL			BOL				BQL			BŲL			BOL	BOL	BQL	BQL	2 G	30F	12	12.5	9 =	BQL			BQL			BQL			BOI	2			BQL		
	3-Dinitro-2,	henzene (µg/L)	108	BQL	4.	BQL	BQL BQL	EQ.			BQL			BOL				BOL			BQL			BOL	BOL	BOL	BQL	10 E	BOL BOL	BOL	BOL	BOL BOL	BOL			BŲL			BQL			108	2			BQL		
		Nitrate mg/L N)	264	1.93	1.92	2	2.04	2.29			2.54			2.71				1.86			2.84			2.89	2.62	2.72	2.86	3.42	3.16	1.79	1.99	1.77	2.32			2.31			2.6			9.6	;			2.88		
	Total	TNB RDX Nitrohodies (µg/L) (µg/L) (Q.	1130	823	666	0801	246			81.2			20.7				6			e			-	1.7	6.0	6.0	10 E	BOL B	960	1230	1040	220			48.8			13			41	;			6.0		
		RDX P	E L	38.4	27.1	33.3	36.1	6.9			2.2			BOL	,			BOL			BQL			BOL	0.7	BQL	BQL	2 G	BOL P	33.3	38.5	32.9	8.2			4.			BQL			ROL	i L			BQL		
		TNB (µg/L) (OB	₹ 1	315	375	425	4			9.65			17.6				7.9			5.6			_	_	6.0	6.0	10 E	B 2	362	474	460	135			36.6			10.2			7.7	1.0			6.0		
		TNT (ug/L)	0	20.5	366	443	465				16.5			6				6.4			BQL			BOL	BOL	BQL	BOL	80 E	g 5	415	155	531	71.1			6			-			0.0	4			BOL		
Contactor	Measured	Peroxide (mg/L)						27.3	28.3		28.3		29.3	28.8		28.3		29.3	20.3		29.3		27.8	27.3		29.3							23.6	;	24.6	24.4	2	0.42	24.4		24.0	24.4	5.4.7	22.4		24.0	20.3	
Confactor Cor	2	Ozone Per (mg/L) (n						2	75		19		73	72		71		70	72		22		73	72		73							08	į	2	62	ē	ž.	80		80	11	:	11		80	9	}
Contactor Co		Ozone (%)					,	5.5	2.0		2.7		2.2	2.3		2.4		2.5	23	1	2.3		2.2	2.3		2.2							9.1	:	-	1.7	-	2	9.1		9.1	=	e.	1.8		9.1	9	:
		Potential (mV)	476	286		415		287	668		931		917	936		904		943	417	;	006		283	068		806	į	273		273	į	308	262	9	097	254	970	843	250		570	000	Š	106		225	300	
- 43		Sample Pr ("C)	<u>~</u>	13		91		2 9	17	17	15	٠ ي	<u> </u>	12	91	81	22	9 :	9 2	. 22	11	9	<u>.</u>	s 5	2	22	<u></u>	11		13	!	2	13	:	<u>c</u>	13	3	<u>e</u>	13		20	2	4	17		12	ž	ì
Tem			9	7.2		7.1	:	7 2	7.4	2.5	7.5	7.5	27	: 2	1.7	7.8	7.9	2.8	. O 8	8.0	8.1		8.2	28	8.2	8.2	£	7.4		6.9		7.1	7.0	6.9	2 2	7.2	7.2	0.4	7.4	1.4	7.7	7.7	9.7	7.8	6.7	7.8	7.7	8.1
	Ozone	Residual pH (mg/L)				-							_ S					9 8						90			9.						0.0	0.0	0.0	0.0	10 6	3 5	0.0	0.0	0.2	9.0	63	0.4	0.5	0.0	0.0	0.0
Operations		Time Re	08:36			14:57	00.00	12:11	14:46	16:07	10:10	11:53	14:35	09:57	11:24	14:21	15:41	09:45	11:13	15:32	01:60	11:05	13:52	08:57	10:54	13:46	15:12	08:41		08:28		13:09	09:44	11:30	14:47 16:44	09:36	11511	14:30	09:25	10:57	13:55	16:15	10:42	13:44	16:01	10:60	10:30	15:52
Мo		Location	CACS		INFI			210					8 8					C4/0						080				GAC3	GAC2			NF I			S 5		C2/0			C3/0	C3/0	980		CAVO			0,50	CSAO
Average	型	Ratio	048		0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40	0.40			0.40		0.30	0.30	0.30	0.30	0.30	0.30	0.30	0.70	0.30	0.30	0.30	0.30	0.30	0,30	0.30	0.30	0.30	0.30
	roxide PEF	Dose (mg/L)		28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	28.8	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0
Average Average Transferred Hydrogen		Dose E (mg/L) (m						72 72									72							7 7																						6¢ 6	£ 5	2 2
Average Ave Applied Trans		Dose D (mg/L) (m	***	88	86			8 88					oo o						8 8 8					86					8 8			86 8 66 8			86.86			× ×				e e				86 B		
Av				3.																																_					_	_				_	_	
	Process	Well Flow Rate (gpm)	-	2	2	≘ :	= :	2 2	2	13	13	2 :	2 2	2 2	13	13	23	≘ :	2 2	2	13	13	2 :	2 2	2	13	<u>:</u>	= :	2 5	2 2	13	2 2	2 2	Ξ:	2 2	13	= :	2 2	2 2	13	Ξ:	2 2	2 2	2 2	2	2 3	2 5	2 2
			-	-		- .	 e ·		-	- 5	- 4					1 9		<u> </u>			- 4	- 9	. 1			9	9	٠ -	 e •		- 9			1 9		- 2	<u> </u>	9 4		1 9	<u>.</u>	9 1	9 9		9	92	 9 y	. 1
		Date	9/18/06	9/23/96	9723796	9/23/96	9/23/96	9723/96	9723/96	9/23/96	9/23/96	9723/96	9/23/46	9/23/96	9723/96	9/23/96	9723796	9/23/96	9723790	9/23/96	9723796	9723796	9723/96	9723796	9/23/96	9723/96	9723796	9/23/96	967576	9/24/96	9724/96	9/24/96	9/24/96	9/24/96	9/24/96	9724/96	9724796	9724/96	9/24/96	9/24/46	9/24/96	9/24/96	9/24/96	9/24/96	9/24/96	9/24/96	9724/96	9/24/96

		Tetryl (µg/L.)	E	5	5	BOL	귏.	_ a			BQL			3	2			BQL			BQL			3	2			5	BQL	G.	<u> </u>	4 5 5		6.3	5.1	4.0			BQL			BQL			BOL	,		5	2
	Nim-	henzene Tet (µg/L) (µg	BOL BC		BQL BC			BOT S						70				BQL B(BQL B(1			BQL B			BOLB						-		BQL B			BQL B			BOLB				9 70g
	ž	1X benz (1) (1)						6.1 BG			4.2 BC			9				<u>-</u>			BOL B			200							BQL B				3.5 B				œ. œ.						0.6 B				7
	ģ	cne HMX (L) (µg/L)						BQL BQL A						5				BQL			BQL B			70				BQL B				3 10 10 10 10 10 10 10 10 10 10 10 10 10			10g				BQL			BQL			BOL				9 10 10 10
	6- 4-Ni	ne toluene (µg/L)	98	BQL	æ	8	¥ ;	ž =	í ĕ	¥ ¥	Ä			ž	á			ĕ			ĕ			ì	á			ğ	ă	ă	ă ă	ăă	ă	ă	æ i	ā			Ď			m							
	Amino-2	itrotoluc (µg/L)	BOL	BQL	BQL	BQL	F 1	2 2	2 2	BCL	BQL			200	2			S S			BQL			5	2			BQL	BOL	중	80 60 60 60 60 60 60 60 60 60 60 60 60 60	300	BQ.	BOL	BQL	2			BQL			BQL			BOL	,		Š	3
	-Nitro- 4-	toluene di (µg/L)	BOL	BQL	BQL	BQL.	702	2 2	10 E	BOL	BOL			20	1			BQL			BQL			3	2			BQL	BQL	EG.	BQL BQL	BOL	BQL	BQL	BQL	2			BQL			BQL			BOL	,		3	2
	2-Nitro- 3	(µg/L.)	BOL	BQL	BQL	BQL	2 2	2 2	BOL 2	BOL	BOL			100	7			BQL			BQL			3	2			BOL	BQL	BQL	BOL	10 G	BQL	BQL	BOL	2			BQL			BQL			BOL	,		Į Č	a la
	L3. Dinitro-2,4. Dinitro-2,6. Dinitro-2. Amino-4,6-2. Nitro-3. Nitro-4-Amino-2,6-4-Nitro-	dinitrofoluene fatuene datuene dinitrofoluene toluene HMX kenzene Tetryl (192L) (192L) (192L) (192L) (192L) (192L)	BOL	BOL	BQL	BOL	10°E	801	= =	91	BQL			100	1			BQL			BQL			Den	2			BQL	BQL	BOL	80F	92.7	9.7.6	122	27	a C		į	BQL			BQL			BOL	,		104	J.
	5-Dinitro- 2-	(ug/L)	BOL	BQL	BQL	BQL	10 F	2 2	BOL	BOL	BOL			100	2			BQL			BQL			170	בלב			BŲL	BQL	BQL	<u>5</u>	BOL	BQL	BOL	BQL BGI	J G		į	BQL			BQL			BOL	,		100	J.
	I-Dinitro- 2,0	(µg/L)	BOL	BQL	BQL	BQL	PQE	8.8	6.7	10.9	BOL			to d	2			BQL BQL			BQL			100	2			BQL	BQL	B QL	BQL BOIL	8. 5.	8.8	11.2	6.6	a de		;	BOL			BQL			BOL	,		100	a VE
	3-Dinitro-2,	henzene (µg/L)	BOL	BQL	BOL	BQL BQ:	7 2	3 5	BOL	BOL	BQL			104	2			BQL			BQL			100	T)			BQL	BQL	BQL	BOL	BOL	BŲL	BQL	_ G	2			BOL			BQL			BOL	,		100	ב
		~	2.87	3.02	3.21	3.25	90 1	6 8	161	<u>7</u>	2.31			,	6			2.53			2.65			5	19.7			2.84	3.07	3.13	3.02	1.75	1.68	1.71	<u>~</u> 3	DC:		į	2.23			2.88			2.59			111	3.23
	Total	TNT TNB RDX Nitrobradies Nitrate (12g/L) (12g/L) (12g/L) (12g/L) (12g/L)	8'0	0.5	0.3	0.3	BQL 864	904	196	896	142			7 7 7	O'CC		,	21			4.6			9	0.1			9.0	0.4	0.3	9.7	10E	BQL	BQL	JOE 10 I	1			BOL			BQL			BOL	,		io	ב ב
		KDX (BOL	BQL	BOL	BQL	2 2	24.1	32.9	35.2	8.4				3			6.0			BQL			100	2			BQL	BOL	BOL	<u> </u>	25.3	28.2	35.9	29.5	5			2			0.3			BOL	,		29	1
	1	TNT TNB RDX (pg/L) (pg/L) (pg/L)	9.0	0.5	0.3	0.3	, S	55. X14	384	379	85.4			¥ 17	?		!	11			4.4			-	9			9.0	0.4	0.3	0.4	4 4	336	485	325	Ē			49.2			8.8			7.9			c	
	1	(Pg/L)	BOL	BQL	BOL	10 E	, E	46.	=	416	44.2			3 01	C)		;	2.1			0.7			DO	3			BQL	BQL	BOL	BQE	368	358	517	343	0.00		:	<u>-</u>			<u>~</u>			0.3			104	a The
Contactor	Measured	Peruxide (mg/L)	24.4		23.6						26.4		24.4	34.6	9.	24.0	;	24.4	24.4		24.0		19.5	24.0	0.43	18.3		24.4		24.0					2	0.02	26.4	;	24.0	24.8		22.4	2	8.02	24.4		25.6	910	3
Confactor	7	Ozone (mg/L)	9		08						08		79	70	2	08	i	2	=		11		2	S	8	98		6/		ş					9	9	77	i	92	74	:	78	ř	ę	11		27	95	٤
		(%)	97		9.1						1.6		1.7	1.3	3	1.6		1.7	1.5		œ. —		9.	31	9	1.6		1.7		9.1					ž	0.7	2.7	1	2.8	3.0		2.6		7.8	2.7		2.9	3.6	17
	Reduction	Potential (mV)	212		220		377	*00	286		283		273	723		265	;	616	247		920		235	737	£77	220		316		215	340	429		396	900	430	686	į	156	886	2	882	9	Ē	925		972	990	Mus
9	OCORP	Sumple ("C)	=		15	:	= =	2	14		9		14	2	:	15	;	4	15		4		=	3	Ē	15		13		9	2	2 2		11	2	2	15	:	5	2	:	13	2	<u>e</u>	12		15	2	1
Ę		- 1	7.8	7.8		2.5	9.7	2	7.0		7.2		7.2	7.4	ţ	7.5	1	7.6	7.7		7.8		7.9	0	0.0	3.1		8.0		8.2	0	7.0		7.1	,	3	7.3	i	7.5	75	2	9.7	ř	97	7.9		7.9	9	2.0
	Ozone	Kesidual pH (mg/L.)	0.0	0.1	0.0	9	6.0				0.0	0.0	0:0	9 6	03	0.0	0.0	0.7	00	0.0	8.0	0.7	0.0	0.0	3 6	0.0	0.0	0.0	0.5	0.0	0.0	0.0			:	: :	1.5	2.1	2 5	0.6	=	1.0	2 :	2.6	2.9	1.7	67	5.4	3 =
		Time	08:55	10:18	13:12	15:41	10.06	0.00	16:59		09:55	11:59	16:55	00:30	11:52	16:46	17:29	09:20	16:42	17:26	90:60	11:47	16:42	17:26	11.35	16:24	17:17	08:37	11:22	16:22	17:07	10:22		14:32	91.01	11:36	14:23	60:91	95:50	14:14	16:03	09:35	11:20	15:58	81:60	11:15	13:55	90.00	11:08
	Sample	Location	CKO	C6AD	C6/0	Cero	LAC.	Z Z	INF	INF	CIVO	CIVO	010	3 5	C2/0	C200	C20	0.30	C3/0	C3/0	C4/0	C4/0	C470	040	250	CSVO	C5/0	Cevo	Cevo	Cevo	282	INF	INFI	INFI	INF:	CIS	C1/0	C1/0	22.0	82	C2/0	C3/0	C3/0	3 5	C40	C4/0	C470	3 3	CSO
Avcrage	뿌	Katio	0.30	0.30	0.30	0.30	0.30	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.29	0.00	0.29	0.29	0.29	0.29	0.29	0.29	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33	0.33
Average Hydrogen	Penxide P	(mg/L)	24.0	24.0	24.0	24.0	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	316	23.6	23.6	23.6	23.6	23.6	23.6	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2	25.2
Average Average Average Applied Transferred Hydrogen		Dosc (mg/L)	52	79	20	2 8	2 8	2 2	2 98	08	98	08	08	08	2 08	80	£ :	08 9	2 2	80	08	80	2	e s	2 5	8 8	08	80	80	2	2 5	2 12	11	11	F :		11	77	F F	: :	77	11	<i>t</i> :	: :		11	11	2 5	: 1:
Average / Applied Tr		Dose (mg/L)	86	86	86	86	8 60	0 20	86	86	86	86	86	86 80	86	86	86	80 8	86	86	86	86	86	e e	0 0	2 86	86	86	86	86	8 3	801	801	801	801	802	108	108	801	108	108	801	801	8 8	108	108	801	80 9	108
				5	2	<u> </u>	2 2	2 =	: =	: ::	13	~	<u> </u>	2 5	2 2	13	<u> </u>	<u> </u>	2 2	2	2	13	<u></u>	<u> </u>	2 2	2 22	13	13	13	<u> </u>	<u> </u>	2 22	13	13	<u>s</u> :	2 2	2	<u> </u>	<u> </u>	2 2	· =	13	€ :	2 2	3 5	= =	2 :	<u> </u>	2 5
	Ē ;	Well Flow Kate (gpm)	_	_	_			-		_	_	_				_	_				_	_	_				_	_	_	_			_	_			_	_				_					_		
		- 1	96/	96/	96/1	96/	130 140 140 140 140 140 140 140 140 140 14	96/	96/	96/	96/	96/9	96/9	964	96/	96/9	96/3	9/25/96	9725/96	9/25/96	9025/96	9/25/96	9725/96	9/25/96	06/67/6	9/25/96	9/25/96	9025796	9/25/96	90/52/6	9/25/96	9/26/96	9126/96	9/26/96	9726/96	9/26/96	9/26/96	9/26/96	9726/96	9726/96	9/26/96	9726/96	9726/96	9726/96	9726/96	9/26/96	96/92/6	9/26/96	9/26/96 9/26/96
	-	Date	9/24/96	9/24/96	9/24/96	9/24/96	06/67/6	96/57/6	9/25/96	9/25/96	9725/96	4725746	9725796	9672576	9725/96	9/25/96	9/25/96	972	9725	9/25	9725	9725	972	276	7/16	9725	9725	973	972	206	206	9726	9726	9726	9726	926	9724	9726	926	700	4724	9724	972	716	900	972	2/6	26	216

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The contine	The continue The		Ann	ied Transfer	ed Hydrogen			Oncration		Ten				ctor														
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No. 10. No.	No.	2	2		25.2	0.33	GAC3	10:44	0.0	8.2		38		BQ	BOL				BOL	BQL	BQL	BOL	BOL	BQL				5
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6 53 33 33 33 34 </td <td> No. 1</td> <td>~</td> <td>2</td> <td></td> <td>25.2</td> <td>0.33</td> <td>C3/0</td> <td>99:18</td> <td>2.5</td> <td>7.4</td> <td></td> <td></td> <td></td> <td></td> <td>11</td> <td></td> <td></td> <td></td> <td>BQL</td> <td>BQL</td> <td>BQL</td> <td>BQL</td> <td>BQL</td> <td>BQL</td> <td></td> <td></td> <td></td> <td>Š.</td>	No. 1	~	2		25.2	0.33	C3/0	99:18	2.5	7.4					11				BQL	BQL	BQL	BQL	BQL	BQL				Š.
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		Tetryl (µg/L)	BQL			BOL	BQL	BQL	BQL	BQL	BOL	BQL	1.0	S. 5	6.5	, ,	6.0			BOL				BQL			BQL			POI	2			10 E	BQL			5.0	6.7						BOL			108		
	Nitro-	benzene (µg/L)	BOL			BOL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	200	2 5	BOL	BQL			BOL				BQL			BQL			Od	2			ğ 2	B I	•	BQL	BQ	BO F	BQL	BQL	BOL			BOL))		70	2	
		(µg/L)	BOL			BUL	BQL	BQL	BQL	BOL	BQL	BQL	30 v	0.0	5.7	6.4	4.7			2.2				-			9.0			E/O	1			2 G	BQ 1		BQ	5.5	63	5.4	6.7	5.2			20	1		-	•	
	4-Nitro-	(µg/L)	BQL			BOL	BOL	BQL	BQL	BOL	BQL	BOL	80F	2 2	2 2	10g	BQL			BOL				BQL			BQL			S	2		į	ğ 2	걸		BQL	306	BOL	BQL	BQL	BQL			BOL			2	7	
	-Amino-2,6-	(µg/L)	BQL			BQL	BQL	BQL	BQL	BQL	BOL	BQL	BQL	1) A	200	BOL	BOL			BOL			į	BOL			BQL			108	200		į	2 2	BQL BQL		BOL	2 2	gor Bor	BQL	BQL	BQL			BOL	ļ V		Ö	-	
	-Nitro- 4	(ug/L)	BQL			BOL	BQL	BQL	BQL	BQL	BQL	BOL	BQL	2 2	2 2	BOL	BQL			BOL				BÖ			BQL			20	2			2 2	10g		BQL	10 E	10g	BQL	BQL	BQL			BOL	i Y		2	<u>}</u>	
	2-Nitro-	(ug/L)	BQL			BQL	BQL	BQL	BQL	BUL	BOL	BQL	BOL	200	200	BOL	BQL			BOL			į	BQL			BQL			IO8	1			<u> </u>	BOL		BOL	10 B	BOL	BOL	BQL	BOL			BOL	í		IOB))	
	1,3-Dinitro-2,4-Dinitro-2,6-Dinitro-2.Amino-4,6-2-Nitro-3-Nitro-4-Amino-2,6-4-Nitro-	dinitronatuene toluene taluene dinitrotoluene toluene HMX benzene Tetry (pg/L.) (pg/L.) (pg/L.) (pg/L.) (pg/L.) (pg/L.) (pg/L.)	BQL			BOL	BOL	BQL	BQL	BQL	BQL	BOL	94.1	5 5	926	100	BQL			BQL			į	BQL			BQL			IOR	1		į	10 E	BQL		BQL	2 2	107	106	108	BQL			BOL	i,		NO.	1 7 1	
	-Dinitro- 2	(J/B/L.)	BQL			BQL	BQL	BQL	BQL	BQL	BQL	BOL	BQL	P.C.	10 E	BOL	BQL			BOL				BQL			BQL			POL	1			BOL BOL	BOL		BOL	971	BOL	BQL	BQL	BQL			BOL	ļ		BOIL	2	
	Dinitro-2,6	(hg/L)	BQL			BQL	BQL	BQL	BQL	BQL	BQL	BOL		0.01	9.5	1.4	BQL			BOL				BQL			BQL			BOIL	2			20.00	BQL		BQL	30F	811	11.3	12.3	BQL			BOL	,		BOIL	1	
	Dinitro-2,4-	henzene (c (µg/L) (BQL			BQL				BQL			BQL BQL	2 2	BOL	BQL	BQL			BOL				BQL			BQL			ROL	2			J 0	BOL		BOL	BOL	BOL	BQL	BQL	BQL			BOL	ļ		BCI	2	
			2.46			2.22	2.16	2.58	2.49	2.75	2.64	2.64	<u> </u>	5.7	95	4.	1.86			2.02			;	2.26			2.4			3.6	1			207	2.47		3.06	13	1.45	1.36	1.49	1.74			184			3 18	2	
	Total	INI INB KDX NIROMONICS NIRALE (Hg/L) (Hg/L) (Hg/L) (Hg/L)	2.4			0.4	9.0	6.0	6.0	BQL	BQL	BQL	5885	100	916	001	238			86.3			;	25			7.8			,				4.0	9.0		BQL	1330	1030	1040	1140	267			8.65			17.5	:	
		N XOX	BQL			BQL	BQL	BQL	BQL	BQL	BQL	BQL S	32.5	30.5	29.7	38.5	8.3			7				0.4			BQL			ROL	2		3	2 2	10 m		80L	28.6	31.4	33.6	35.4	9.6			4			EO.	1	
		ug/L) (2.4					6.0	6.0	BQL	BQL	BOL	359	358	376	462	150			67.5				21.3			6.9			,	4		;	4.0	9.0		BQL:	5 5	418	84	480	891			47.5			15.7	:	
		(Hg/L) (Hg/L) (Hg/L)	BQL			BQL	BQL	BQL	BQL	BQL	BQL	BQ	374	30.4	394	469	74.2			14.6			;	2.3			0.3			ROL	ž		3	2 2	BQL		BQL	429	44	432	486	84.2			9.1			-	?	
Confactor	Measured	(mg/L)	37.7		33.0	34.8		35.4									36.0	,	0'00	40.2		48.8	,	38.4	38.4		39.0		38.4	36.0	900	33.5	9	40.8	36.6							38.4	;	38.4	36.6		42.7	10,0	:	
	-	(mg/L)	75	i	4	75		75									11	P	ę	92		78	1	11	11		76		76	7.	2	11	ì	9	27							70	ı	-	79		9/	32	?	
		(%)	2.2	,	2.3	2.2		2.2									2.0		<u>.</u>	2.1		6:1		2.0	2.0		2.1		2.1	22	1	2.5		1.7	2.2							- 00		2.0	<u>~</u>		2.1	0		
	Reduction	(mV)	952	9	926	988		913		320		į	427	415	ì		868	570	Š	952		952	1	947	876		914		933	927	ŧ	948	ces	5.00	306		238	075	421			289	***	282	305		888	873	;	
Temperature	of ORP	(°C)	13	5	2	=		9		61		:	4	1	:		13	2	2	53		11	:	13	8		14		11	1	3	11	:	5	9		2 :	<u>e</u>	81			15	;	17	16		20	77	:	
12		ı	8.0	i	6.7	<u>~</u>		8.0		7.8		,	8.9	63	3		7.0	7	e e	7.3		7.2	į	4.4	7.5		9.7		7.6	7.8	2	1.7	6	8.0	7.8		pc :	6.0	6.9			7.1	;	7.	7.3		7.3	7.4	:	
	Ozone	(mg/L)	1.5	0.1	2 9	0.7	6.0	0.1	0.7	0.3							0.4	9.0	0.0	9.0	0.7	9.0	8 5	6.0	5	0.3	0.5	9.0	20 2	5 0	13	13	4.6	C'0	0.6	0.7	0.0					0.0	9.0	0.3	00	0.9	0.2	0.0	0.7	
Operations	Sample	all l	95:60	11:50	15:52	09:41	11:38	13:52	15:57	00:60			09:48	91.36			09:42	10:43	16:54	06:30	10:38	14:16	16:43	97:60	14:02	16:40	09:14	10:28	13:54	09:06	10:20	13:38	16:32	10:14	13:26	16:29	08:20	10:37	15:00			91:01	4 5	14:36	09:50	11:42	14:24	06:00	11:39	
	Sample	COCAHOD	CS/O	C5/0	S S	C640	C6/0	C6/0	CeAD	GAC3	GACI	GAC2	Z Z	Z Z	N N	INF	C1/0	212	010	C2/0	C2/0	020	973		Cao	C3/0	C4/0	C4V0	8 8	200	CSA	C5/0	CSAO	80	C6/0	C6/0	GAC3	Z Z	INFI	INF	INF	CIV	0 G	2 5	C280	C2/0	C20	92.5	C3V0	
Average	Peroxide PEROXONE Sample	Katio	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.49	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0.48	0,48	
Average Hydrogen	Peroxide 1	(mg/L)	35.4	35.4	35.4	35.4	35.4	35.4	35.4	35.4	35.4	35.4	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.0	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.0	36.6	
Average Average Applied Transferred Hydrogen	Охоне	(mg/L)	75	75	5 X	75	7.5	27	7.5	75	75	25	27 25	5 %	5 22	75	75	27 25	2 2	75	75	75	22	۲ ۲	22	75	75	75	52 X	2 %	22	75	25 5	c	7.5	75	52 5	6 %	92	9/	92	9/	92	۶ ۶	2 92	9/	92	6 %	36	
Average Applied T	Ozome	(mg/L)	001	00 5	8 8	001	001	00	100	100	8	001	00 5	3 2	8 8	00	100	9 9	8 9	100	100	00 5	00 5	8 8	8 9	100	100	001	8 8	3 2	8 8	901	901	8 5	8 2	8	9 5	8 5	001	100	001	100	<u>6</u>	8 8	8 8	100	001	8 9	100	
,			2	<u>e</u> :	<u> </u>	2	13	3	13	2	<u>~</u> :	≘ :	<u> </u>	2 2	2 2	2	13	<u> </u>	2 ==	=	=	<u>e</u> :	<u> </u>	<u> </u>	3 5	: 2	13	=	<u> </u>	2 =	2 2	13	2 2	2 2	2	13	<u> </u>	2 2	==	13	13	<u>e</u>	<u>n</u>	5 5	2 5	=	<u>e</u> :	<u> </u>	2 2	
	Æ .	well riow kate (gpm)	_			_	_	_	_	_	_					_	_			_	_					_	_	_				_				_		. .		_	_	_				_	_		. _	
		nanc w	9/28/96	9/28/96	9/28/96	9/28/96	9/28/96	9/28/96	9/28/96	9/28/96	9728/96	9/28/96	9/29/96	90/00/0	96/62/6	9129796	9/29/96	9/29/96	90/6/16	9729/46	9729746	9179716	9729796	9/29/96	96/67/6	96/62/6	9/29/96	9//62/6	9/29/96	96/67/6	9729/96	96/62/6	9/29/96	96/67/6	90/67/6	96/67/6	9/29/96	96/08/6	9/30/96	9/30//96	9/30/96	9/30/96	9/30/96	9/30/46	96/06/6	9/30/96	9/30/96	9/30/96	9/30/96	
	2	2	972	472	972	972	2/6	972	9/2	472	472	26	276	7 60	972	2/6	972	76	26	2/6	26	7/6	76	76	26	472	7/6	36	36 6	2 6	276	472	3/6	÷ 9	3/6	3/6	76	6 6	6	16	16	76	76	74.	. 6	16	.76	5 6	. 76	

ì		Je G	ij		7,			BQL			=	BQL	7.	ನ :	۔ چ	ج ,	, _	4.9	5 20	3			BQL BQL			BŲL			ROIL	3		;	RQL		,	BOL B	, 2	35	,	<u>ئ</u> ے ج	ļ, r-	~	_	.7	10 E	3		BŲL
	÷	ene Tetryl			T BOL																																											
	Nitro	X benzene			BQL			BOL						3 G				BOL					BQL			BOL			IOR				305			1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1									글 등			BOL
	ė	toluene HMX			. 0.5			, BQL							2 2				9.9				. 1.7			. 1.2			90				PACE.												5.7			. 1.7
	2,6- 4-Nin	sene toluer	ĺ		BQL			BQL				BOL		8QL									BQL			BQL			BOIL				PQF												10g			BQL
	. 4-Amino	dinitrotoluc			BQL			BQL			BOL	10g	BOL	10g	200	BOL	BQL	BOL	2 2	2			BQL			BQL			BOL))		Š	ğ		Š	108	BQL	BQL	10 E	BOL BOL	BOL	BQL	BOL	BQL	ig ig	7		BOL
	3-Nitro	(oluene			BQL			BQL			BOL	BOL	BOL	30 G	202	BOL 2	BQL	BOL	2 2	2			BOL			BQL			BOL	,		3	P.C.		2	2 G	BQL	BOL	<u> </u>	5 5	E S	BQL	BQL	BOL	<u> </u>	3		BQL
	- 2-Nitro	toluene (uv.)			BQL			BOL			BOL	BQL	BOL	BQ1	3 5	BOL	BOL	BOL	2 2	2			BQL			BQL			BOL	7		29	T) I		2	10g	BQL	BQL	2 2		10°	BOL	BQL	BOL	<u> </u>	3		BOL
	1.3-Dinitro-2,4-Dinitro-2,6-Dinitro- 2-Amino-4,6- 2-Nitro- 3-Nitro- 4-Amino-2,6- 4-Nitro-	dintratoluene toluene toluene dinitrotoluene toluene HMX benzene (1994) (1994) (1994) (1994) (1994)			BQL			BQL			BOL	BOL	BQL	30E	33.1	70.6	108	91.9	88.2	ì			BQL			BQL			BOL	ì		0	JOS		704	BOL BOL	BQL	BQL	2 E	03	8.66	100	80.7	79.3	80.7 BOIL	2		BQL
	2,6-Dinitro-	(ne/L)			BQL			BQL			BOL	BQL	BQL	10 E	BOL 2	BOL	BQL	BQL	2 2	1			BOL			BQL			BOL	i i		770	פלר		200	BOL BOL	BOL	BQL	2 5	BOL	BOL	BQL	BQL	BQL	10 E	7		BQL
	4-Dinitro-	toluene (ug/L)			BQL			BQL			BOL	BQL	BQL	10 EQ	4	30	11.5	30 30	, [2	i Y			BQL			BQL			BOL	ļ		100	300		100	BQL BQL	BQL	BQL	BO L	10g	11.3	10.5	9.5	6.3	9.9 BOI.))		BQL
	3-Dinitro-2	benzene (ug/L)			BQL			BQL			BOL	BQL	BOL	30 E	10g	BOL	BQL	80F	200	ì			BOL			BQL			BOL	ļ		Ca	100		100	10g	BQL	BQL	2 2	BQE	BQL	BQL	BQL	BQL	BQL BQL	7		BQL
	_	-			2.45			2.6			2.4	2.46	2.92	143	1.42	1.32	1.32	1.38	88			:	2.22			2.03			2.3			3.45	C+:7		100	2.29	2.62	2.67	2.53	2.52	2.36	2.14	18.1	2.33	3.36			3.75
	Total	TNT TNB RDX Nitrohwlies Nitrate (11g/L) (14g/L) (14g/L) (14g/L) (11g/L)			9.9			2.9			0.5	9.0	1.2	8.0	285	814	1110	845	211				57.9			24.5			7			11	7		Š	90	8.0	- 3	<u> </u>	3.1	1060	1030	923	116	1080)		51.3
		RDX (ug/L)			BQL			BQL			BOL	BQL	BOL	2 2	10.4	23.9	30.9	30.7	*. ×			:	4.			6.0			BOL	,		2	2		7		BQL	를 당	2 2	P of	35.9	35.1	31.6	29.1	31.6	ŧ		1.5
		TNT TNB RDX (µg/L) (µg/L)			5.9			2.9						8.0 10.0		358	480	353	135				45.			20.1			9			11			2				\$ 5		445	426	388	386	90.8	1		40.1
İ.	_				0.2			BOL			BOL	BQL	BQE	2 5	=	347	470	348	63.4				7.6			2.3			0.4			POI	2		IOB	g F	BQL	BOL	4	4.	456	448	403	396	53.5	l		90
Superposit		Peroxide (mg/L)	19.6		37.2	30.5		33.5	35.6	0.05	36.0		36.6						42.1		42.7		47.7	38.4		54.9		36.6	39.6		42.7	17.8	9.75	39.0	41.6	?	38.4								57.9		0.0	43.9
Confector		Ozone (mg/L)	11		78	72		75	12	=	75		75						8		92	i	-	75	!	80		0	11		78	76	2	74	22	:	74								80)		22
Continutor	Off-gas	Ozone (%)	2.0		<u>5:</u>	2.4		2.2	3 6	C.2	2.2		2.2						1.7		2.1	•	6.2	2.2	1	1.7		1.7	2.0		1.9	,	·	2.3	01	ì	2.3								1.7	i		6.1
Oxidation	Reduction	Potential (mV)	817		178	916		630	909	Š	887		842	233	420		400		403		932		/99	616		943	;	242	199		263	886	à	913	203	}	897	310	67		311		295		282	i	280	290
Temperature	of ORP	Sample ("C)	21		2	21		4	5	1	7		21	7	91		20		20		<u>~</u>	:	2	20		15	;	50	11		20	. 5	2	70	2	1	21	91	2		2		13		2	!	4	4
۰			7.5		1.7	7.7	i	7.9	7 6		8.0		6.7	7.9	6.7		6.9		7.1		7.1		ŧ.	7.3		7.6		97	7.8		7.4	7.9	<u>:</u>	7.9	0		8.0	9			7.1		1.1		7.3		7.2	7.5
	Ozone	Residual pH (mg/L)	0.3	9.0	0.3	0.5	0.5	6.0	- 2	4.	9.0	0.5	6 6	0.0					0.1	0.3	9.0	0.0	0.3	0.4	0.1	0.7	= :	0.0	0.1	0.3	0.0	0.0	0.8	0.7	6.8	0.2	0.5	6.0	2						0.0	0.0	0.0	0.0
Operations	Sample	Time	14:07	16:46	08:56 11:36	13:58	16:43	08:48	13-50	16:37	08:40	11:29	15:47	08:24	10:20		15:30		09:51	11:16	15:25	16:37	11:12	15:17	16:34	66:36	80:11	15:00	09:11	11:04	14:58	08:05	10:59	03:07	77:01	10:52	14:41	16:16	09:50		08:03		11:10		07:57	08:54	10:46	91:10
		Lixation	C3VD	C3VD	C42	C4/0	C4/0	95	3 5	CSVO	CKM	C6/0	9 5	GAC3	INF	INFI	INF	Z 2	CIV	C1/0	CIO	2 5	C200	C2/0	C2/0	0,62	0,50	2 5	C40	C4/0	C4/0	5 5	CSVO	C5/0	200	C6/0	CKO	CEAD	CACI	GAC2	INF	INF	N.	E N	CIA	CIA	C1/0	C2/0
Average	Peroxide PEROXONE Sample	Ratio	0.48	0.48	0.48	0.48	0.48	0.48	6 48	0.48	0.48	0.48	0.48	0.48	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.55	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59
Average Hydrogen	Peroxide F	Dosc (mg/L)	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.6	36.0	36.6	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	42.1	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7
Average Average Average Applied Transferred Hydrogen	Ozone	Dosc (mg/L)	92	92 2	e %	9/	92	9 %	92	92	92	92	0 %	2 92	7.7	11	11	: F	11	11	F F	= 1	: 1:	11	$\iota\iota$	11	F :	: =	: L	11	11		1	11	11	11	11	F F	: 1	11	78	78	82 8	8/ P	ó 56	78	78	25
Average Applied T	Ozone	Dose (mg/L)	001	901	3 5	001	901	3 5	8 8	901	100	90 5	8 9	8 8	001	100	00 5	8 8	901	001	9 2	8 8	8 8	100	100	00 :	8 9	3 2	9	001	90 5	8 8	901	<u>8</u>	8 9	001	001	9 5	00	001	001	001	8 8	8 8	3 8	00	001	00
<			5	13	2 2	13	£ :	2 2	2 2	2	13	£ :	2 5	: ==	13	2	<u> </u>	2 5	13	13	5 5	2 2	2 2	=	13	£ :	<u> </u>	2 2	2 :	13	<u> </u>	2 22	: 2	13	3 2	13	2	2 2	: =	13	13	<u>s</u> :	<u> </u>	2 2	2 22	13	2	2
	Pro	Well Flow Rate (gpm)	_			_			_	_					_	_			_	_			- 	_	_					_			_		_	_	_		_	_	_	_			-	_	_	_
		Dake W	9/30/6	96/06/6	96/08/6	9/30/96	9/30/96	071070	9/30/96	96/06/6	9/30/96	9/30/96	96/00/6	96/00/6	96/1/01	96/1/01	10/1/96	96/1/01	10/1/96	96/1/01	90/1/00	06/1/06	96/1/01	96/1/01	96/1/01	96/1/01	96/1/01	0/1/00	10/1/96	96/1/01	96/1/01	96/1/01	96/1/01	96/1/01	96/1/01	96/1/01	10/1/96	96/1/01	96/1/01	96/1/01	10/2/96	96/2/01	10/2/96	10/2/96	10/2/96	10/2/96	10/2/96	10/2/96
		ا ۵	9/30)E/6	9/30	9/30	8/6	0770	96/6	9/3	9/3	16/6	0/16	9/30	10/	10/	/01	2	10/1	10/	0		10/1	10/	10/	0	9 5	0	10/	10	/01	10	10/	/01	10	10/	10	701	101	10/	10%	701	70.	100	10/2	10%	107	10/.

ı		j	를 걸			BQL		BQL		BOL	Ļ		라 ?	3 5	BQL	ಸ		3	5.9	7	S	ಕ			BQL			3	DQL DQL		;	BQL BQL			PQL PQL		3	ಕ್ರಕ	Ę,	BOL	Ę,		5.3	5.3	5.7	272	7.5	
		ė	(Hg/L) (Hg/L) (Hg/L) (Hg/L)																						BQL B				1			8 7 8				-				BQL B						BOL BOL		
	i	Nitro-) (µg/			BQL		BQL		L BOL			PQL							BQL																												
		4 5	(ug/			X		, 0.4		BOL			BQL					5.7		, 6.3					1.5				9.0			. 0.5			a gor					BOL								
		2,6- 4-Nitr	rene norden (μg/L			BQ.		BQL		BOL					BOL				BQL						BQL				2			PQE			17					BQL						10 10 H		
		4-Amino-	(µg/L)			BQL		BQL		BOL	,		BQL	BOL B	BQL	BOL		BOL	BQL	BQL	30 E	BQL			BQL			Š	פער			BQE.		2	D C		n d	E CL	BQL	BQL	ВОГ		BQL	BQL	DG BOT	<u>ફ</u> ફૂ	BQL	-
	T. C.	- 3-Nilro-	(µg/L)			80F		BQL		BOL			BQL		BQL					BOL					BQL				2			η R								BQL				BOL		E E		
İ	o Miles	Z-Nille	(µg/L)			BOL		BQL		BOL			를 급	90	BQL	BOL		BOL	BOL	BOL	B 5	B D			BQL			S	2		2	PA PA		Š	3		ICH	중	BQL	BQL	BQL		BQL	BOL	BC	B CL	BQL	
	Ampar 46	L.3-Unittio-2,4-Diffito-2,0-Diffito-2-Afritio-4,0-2-Nitro-3-Nitro-4-Amino-2,6-4-Nitro-Nitro-Nitro-Paramente de la la la la la la la la la la la la la	(µg/L)			BQL		BQL		BQL			3 2 3	80F	BQL	BQL		112	106	110	16	BOL			BQL			100	374		776	709		Ç	200		ROI	BQL BQL	BQL	BQL	BQL		8.06	94.3	76.7	85.5	BQL	
	Pinite.	tohon.				H)		BQL		BQL			BQL	BOL	BQL	BQL		BOL	BQL	BQL	BQL	RQL			BQL			100	1		100	T) N		200	200		EC!	BQL	BQL	BQL	BQL		BQL	BQL	BQL	1 1 1 1 1 1 1	BQL	
	Diales	toluene	(µg/L)		į	BQL		BQL		BQL			BQL	BOL	BQL	BQL		11.2	11.4	89	10.1	T)			BQL			JOB	1		2	200		100	2		BOI	15 15 15	BQL	BQL	BQL		Ξ	Ξ	9.2	14	BQL	
	1. Dinitar.	S-Dinitin-2 Jenzene	- 1			BQL		BQL		BOL			BQ1	BOL	BQL	BOL		BOL	BQL	BQL	10 E	E D			BQL			2	ž		3	200		00	2		BOIL	10 T	BQL	BQL	BOL		BQL	BQL	BQL	2 2 2 3	BQL	
	-		(mg/LN)			3.98		4.38		4.39		į	5.56	5.56	5.49			1.34	1.27	1.49	9.5	78.1			1.85			200	0.1		č	10.7		;	7.7		206	2.26	2.09	1.97	2.35		1.24	1.08	1.33	138	1.72	
	Total	TNT TNR RDX Nitrobylics Nitrate	(µg/L)		;	12.5		3.9		-		;	0.0	0.4	0.4	0.4		1020	1000	1280	828	2			44.7			2			3	0.0		-	2		0	0.8	0.5	0.5	BQL		926	940	900	940	186	
		RDX	(ng/L)		i	ĘČ.		BQL		BQL		1	10 E	B 5	BOL	BQL		35.2	34.6	35.9	34.2	c			Ξ			03	3		2	2		100	3		ROL	B C		BQL			27.8	33.8	25.1	43.9	7.5	
		Ĭ,	(HEA.) (HEA.) (HEA.)		,	9.01		3.2		-			0.5					419	408	529	ž 5	3			34.7			13.0	1		9.7	r.		-				0.8		0.5			388			379		
					:	2		0.3		BQL		-	20 20	BOL L	BQL	0.5		427	429	585	369	20.7			7.4			=	:		ć	7.0		104	2		ROL	BQL	BQL	BQL	BOL		394	393	385	380	53.3	
	Contactor	Peroxide	(mg/L)		0.0	43.9	0.0	8.8	0.0	35.4		00	42.7	0.0							3.5	63.9	0.0		35.3		0.0	110		0.0	33.0	6,55	0.0	14.0	200	0.0	346		0.0								40.5	
	Contactor	Ozone	(mg/L)		í	Ē.		62		76		i	8/								ř	ę			75			77	2		ž	2		74	t		75	2									92	
	Confactor Off-use		(%)		:	<u>«</u>		8 0.		2.1			5:								9	ć.			2.0			2.0			ç	0.7		11	i		61	:									6.1	
	Oxidation	Potential	(ImV)		281	767	354	916	539	524		925	811	856		235		432			649	749			106			610	•		900	2		00%			842	!					427		459		730	
	Temperature of ORP	Sample	Đ.		9 9	2	11	7	15	4	:	12	=	4	:	13		14			7	ŧ			13			-	:		12	2		7	:		4						7	:	9		7	
		Ħ	- 1		7.4	7	9.7	7.8	7.8	8.0	:	œ :	×i	- -	i	œ.		6'9			7	3			7.3			7.5	1		11	3		2 2	2		0.8						8.9		6.7		7.0	
	Ozone	Residual pH	(mg/L)	0.0	0.0	3 3	0.0	ē ē		0.1	0.4	0.5	7.0	0.2		0.0					90	80			0.8	13		-	1.2		-	2 !		=	4		80	1.2		ć	0.0						0.4	0.4
	Operations			08:41	10:28	08:35	10:00	07:02	09:35	96:38	08:28	09:28	08:22	09:15				14:58			14.46	16:28			14:37	16:25		14:12	16:21		14.36	91:91		14.14	16:10		14:02	16:04		13.43	13:47		10:02	:	15:42		09:34	11:30
		Location		C2/0	072	C3/0	C3/0	C470	C470	CSAO	CS/O	CSO CSO	8 8	C6/0	CAN	GACS	GAC2	INFI	INF	ENE.	INT.	CIO	CIA	CIVD	C2/0	C2/0	072	300	C3VD	C3/0	C3/0	C40	C4/0	Ş 5	CSAD	C5/0	Q Q Q	C6/0	C6/0	0,90	GACI	GAC2	INFI	INF.	INFI INFI	INF.	CIA	Q C
	Average FROXONE	Ratio		0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.59	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.47	0.52	0.52	0.52	0.52	0.52	0.52
	۵.	Dose	(mg/L)	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7	45.7	34.9	34.9	34.9	24.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34.9	34,9	39.8	39.8	39.8	39.8	39.8	36°8
Average	Applied Transferred Hydrogen Ozone Ozone Peroxide		(mg/L)	28	78	2 20	28	8 %	78	38	8.	× 9	8 %	3/2	78	8 2	2 82	74	74	4.	ŧ 7	. 4	74	74	74	74	7 7	7 7	74	74	4 7	7.	74	74	7.	74	4 7	74	74	7 :	4 4	. 4	11	11	F F	: ::	77	11
Average	Applied 1	Dose		901	2 2	8 8	00	8 8	8 8	001	00 5	8 8	8 8	100	00 5	8 9	8 8	86	86	80 3	0 00	86	86	86	86	86	86 9	86	86	86	26 S	2 25	86	8° 8	86	86	86 66	86	86	86 8 85	× ×	2 28	86	86	86 3 6	8 %	86	86
	Process	w Rale	(md8)	2	2 2	2 2	13	<u> </u>	2 2	13	<u>e</u> :	2 2	2 2	=	<u>s</u> :	2 2	2 2	13	13	<u> </u>	2 2	2 22	13	13	Ξ	<u> </u>	2 2	2 2	2	2	<u> </u>	2 5	2	2 2	2 2	5 5	2 2	2	£ :	<u> </u>	<u> </u>	2 22	2	£ :	2 2	2 2	2	2
	څه	Well Flow Rate	3	_			_			_				_				_	_				_	_	_				_	_			_					_	_				_	_			_	_
		Dake	- 1	10/2/96	10/2/96	96/2/01	96/2/01	10/2/96	10/2/96	96/2/01	10/2/96	96/2/01	10/2/96	10/2/96	10/2/96	10/2/96	10/2/96	96/1/01	96/1/01	96/1/01	107786	96/1/01	96/1/01	96/1/01	96/1/01	96/1/01	10/7/96	96/2/01	96/1/01	96/1/01	90/1/01	96/1/01	96/1/01	96/1/01	96/1/01	96/1/01	10/7/96	10/7/96	96/1/01	96/1/01	96/2/01	96/1/01	10/8/96	96/8/01	96/8/01	10/8/96	10/8/96	10/8/96

l		Tetryl (HE/L)		108	2		BOI.	3			BQL			BQL			j)	JO.	Š Š	<u>5</u>	를 걸	, 5 5	6.8	8.4	5.6	7.5	. E	1		Š	Z)		-	J)		1	B QE			BOL			BOL	BQL	BQL BQL	BQL	ב ב
	Nitro-	henzene Te (µg/L) (µg		BOIL			BOI. B				BŲLB			BQL B													2 2							- - - - - - - - - - - - - - - - - - -	-		BQL B			BQL					BQL		
	Ž	HMX hen (µg/L) (µg					00				0.6 B			BQL B								BOLB		7.4 B			0.3				×.			8.0			BQL -			BOL					BQL		
	-o.ni	(ug/L) (ug		Brit 3			BOI. 0				BOL			BQL B									13 13				2 2				2			i So			BQL z			BQL 1					BOL		
	2,6- 4-N	iciie toh																																													
	-Amino-	initrotoluc (µg/L)		108	3		ROR	2			S S			BQL			BQL	BQL	108	BOL	2 2	200	BOL	BQI	BQL	<u> </u>	2 2	ž		Š	a S			PQE PGE			BQL			BQL			BO	BQ	PQ.	90	2
	Nitro- 4	(µg/L)		S	2		EOI	2			BQL			BQL			BQL	BQL	BQL	BQL	200	BOL	BOL	BQL	BQL	<u>2</u>	2 2	2		2	J N			RQE FQ			BQL			BQL			BOL	BQL	80L	BQL	
	2-Nitro	(ug/L)		i d	2		BOL	2			BQL			BQL			BQL	BQL	BOL	BCL	2 2	BOL	BOL	BQL	BQL	BQL	2 5	Ž		3	בל בל			BQL BQL		i	BOL			BQL			BOL	BQL	BOL	BQL	
	1,3-Dinitro-2,4-Dinitro-2,6-Dinitro-2-Amino-4,6-2-Nitro-3-Nitro-4-Amino-2,6-4-Nitro-	dinitrototuene toluene (toluene dinitrotoluene toluene HMX benzene Teiryl (1981.) (1981.) (1981.) (1981.) (1981.) (1981.)		108	2		BOIL	2			BQL			BQL			BQL	BQL	BQL	BOL	20 20	10E	93.7	7.07	93.1	82.8	PO.	מאַ		3	g g		į	BQL B		į	BOL			BQL			108	BQL	BQL	3 gG	n n
	2,6-Dinitro- 2	(µg/L)		IO B	2		BOIL	ý			BQL			BQL			BQL	BQL	BOL	10g	30 E	BOL	BOL	BQL	BQL	g 2	2 2	2		2	Š		į	7Ď9			BŲL			BQL			BOL	BQL	BQL	BQL	מלר
	4-Dinitro-	(µg/L)		EC I	2		SOF	ž			BQL			BQL			BQL	BQL	BQL	BOL	BQL BQL	BOL	11.6	12.2	13.1	= 3	5 2	2		ŝ	P C		į	BQL			BOL			BQL			BOL	BOL	BQL	BQ 5	EQL.
	Dinitro-2,	henzene (µg/L)		IO B	2		ROI.	2			BOL			BQL			BQL	BQL	BQL	BQL	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	BOL B	2.3	2.3	61	× :	0.1	2			50E		į	EQ.			BQL			BQL			BOL	BOL	BQL	BQL	<u> </u>
		_		78			*				2.04			2.2			2.18	2.13	2.18	2.23	3.27	2.95	1.2	1.2	1.18	1.25	F. 1	75.1			97		:	1.92		;	2.05			2.11			2.13	2.11	2.15	2.12	71.7
	Total	TNT TNB RDX Nivehodies Nivate (ug/L) (ug/L) (ug/L) (ug/L)		3 83	200		18.3				8.8			2.1			BQL	BQL	0.7	9.0	10 E	BOL BOL	808	811	857	262	716	6		,	7.08			15.7			9.9			1.8			0.5	9.0	0.7	9.0	E E
		(DX Nii		4	2		ROF	ž			BQL			BQL			3QL	BQL	BQL	BOL	BQE BQE	BOL P	32	34.7	33.7	30.5	35.8	7.			<u>e</u>			BQL			BQL			BQL			BOL	BQL	BQL	BOL	3
		TNT TNB RDX (ug/L) (ug/L)		43.8			150				2			2.1									334	340	356	335	127	(7)			23.2		;	13.7			6.3			<u>8</u> .			0.5	9.0	0.7	9.0	3
		TNT (Hg/L) (=			3	1			0.2			BQL			BQL	BQL	BQL	BQ	<u> </u>	BO 1	318	339	345	320	014	1.00		,	9.6		:	1.2			0.3			BQL			BOL	BOL	BQL	BQL	בלה מלה
Confactor	Measured	Peroxide (mg/L)	39.2	30 %	5.2.0	39.2	30.8		39.8		39.8	39.2		38.5		3/.1	39.8		38.5								7 05	33.7	39.0		39.7	39.0	:	40.4	39.0		40.4	30.0		40.4		39.0	39.7		39.7		
Confactor Co	-	Ozone Pe (mg/L) (76	ž	2	75	75	2	7.5		75	26		74	ř	c	76		77								¥	9	08	i	6	67	i	70	79		Ş.	S	ì	61	i	£	11		79		
Contactor		(%)	6.1	0,0	2	2.0	2.0	2	2.0		2.0	6.1		2.1		0.2	1.9		8.1								9	<u>.</u>	9.1	!		1.7			1.7		1.7	9	1	1.7	:	1.7	90		1.7		
		Potential (mV)	593	ÇUÐ	70	867	80	700	268		903	825		884	908	680	843		608		279		434		426		200	930	893		816	908		922	855		913	808		106	1	863	898		865	÷	1
			21	2	2	17	2	:	20		=	2	:	=		<u></u>	=		81		10		4		91		2	2	91	:	13	17		13	20		12	12	:	13	!	11	12	!	11	:	=
Tem		- 1	7.0	,,	1	7.2	7.5	2	7.4		7.7	1.7		7.8		7	6.7		7.8		7.8		8.9		2.9		0.0	2.	6.9	;	7.2	7.1		7.4	7.4		7.5	3,6	2	1.7		7.6	7.8		7.8	;	2.8
	Охопе	Residual pH (mg/L)	7.0	0.5			9.0					0.7				0.4	0.5	9.0			0.0						90	2 6	0.4	0.5	9.0	0.5	0.7	0.8	0.7	0.7	8.0	8.0	8.0	9.0	0.7	0.0	0.7	0.5	0.4	9.0	0.0
Operations		Time R	14:58	16:44	11:28	14:51	16:39	11:24	14:40	16:32	85:80	11:20	16:27	08:50	91:19	61.91	08:36	11:11	14:03	16:14	08:22		09:32		14:22		200.33	11-37	14:15	16:30	09:16	14:07	16:26	06:10	13:45	16:23	08:59	11:25	16:20	08:53	11:21	13:28	08:40	11:17	13:19	16:14	08:25
Ĉ		Location	CIA	CIA		C2/0	23.0		C3A0			S S S	C4/0	C5/0	CSAB	200	C6/0	C6/0	C6/0	C6/0	GAC3	GAC2	INF	INF	INF	E E	12.	2 2	CIV	CIAO	07.0	3	C2/0	9 S	C3/0	C3/0	C4/0	242	£ 5€	CSAO	C5/0	0,50	Con Con	C6/0	CKO	CGN	GACS
Average	핒	Ratio Lo	0.52	0.52			0.52					0.52				0.52					0.52			0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52
	roxide PE.	Dose (mg/L)	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	39.8	30.8	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1	40.1
Average Average Average Applied Transferred Hydrogen	Ozone Per	Dose D		11									77						11			: =										: 1:	11	11	77	77	11	7.		11	11	<i>t</i>	: =	11	11	77	77 77
age Ave	ne OA																					200						86 86			 26. 33			 86 90				86 9	6 26	86	86	86 9	6 3	20	38	86	86 86 86 86
Average Applied		ate Dose	8	86	8 %	86	86 8	6 26	86	86	86	86 86	86	86	85	× ×	86	6	6	6	5 6	, 5	, 5 ,	5	2	5	J 6	, 0								-											
	Process	Well Flow Rate (gpm)	=	<u>e</u> :	2 2	£	2 2	2 2	2 2	13	2	2 2	2 2	13	<u> </u>	2 2	2	2	2	2	<u> </u>	2 2	2 2	2	13	€:	Ξ:	2 2	2 22	13	= =	2 2	13	2 2	2 2	13	13	≏ :	2 5	2	=	= =	2 2	2	13	13	5 5
		- 1	_			1 9				- 5	- 2			1 9			. ~	- 4	- 9	<u>-</u>	. ـ			- 9	- 9	- 9	٠ -		 2 ys	9,	9 9		92	96		1 94	1 96			1 92	1 96				96	96 1	 96 97
		Date	10/8/96	10/8/96	10/8/96	96/8/01	10/8/96	10/8/96	10/8/96	96/8/01	96/8/01	10/8/96	96/8/01	96/8/01	96/8/01	96/8/01	10/8/96	10/8/96	10/8/96	10/8/96	96/8/01	10,8796	10/9/96	96/6/01	10/9/96	10/4/96	96/6/01	96/6/01	96/6/01	10/9/96	10/9/96	96/6/01	10/9/96	96/6/01	10/9/96	10/9/96	96/6/01	96/6/01	96/6/01	96/6/01	96/6/01	96/6/01	96/6/01	96/6/01	10/9/96	96/6/01	96/6/01 10/6/40

	logo.	(ug/L)		5.7	4.9	7.4	4.5	6.9	BQL				BQL			3	3			BQL				RQL P			BOL	BQL	BQL	BOL	BOL	27	6.0	BOL	BQL	BQL			BQL			ICA	2			BQL			BOL	,		
1	Nitro-	(ug/L) (s			BQL				BQL 1				BQL				2			BQL				- - - - - -				BQL			PQ.			BOL I					BQĹ			EQ.	3			BQL			BQL	,		
1	Z XW	(Hg/L) (s		7.2				6					1.7			=				BOL				PQL BQL			BOL	BQL	BQL	BQL	BOL	6.01	2.7	3.8	10.1	1 .			7			80	0.0			0.5			9.4			
	Nitro-	(µg/L) (µ		BŲL		BQL		BOL	BOL				BQL			Š	2			BQL				2									2 2	BG L	BQL	BQL			BQL			ROI	ž Ž			BQL			BQL	,		
,	2-Amino-4,6- Z-Nitro- 3-Nitro- 4-Amino-2,6- 4-Nitro- dinitroloment teluene teluene dinitroloment teluene BMX Sentano	(pg/L) (1				BQL			BQL				BQL				מלה			BQL.				IZ CE			Š.	BQL	lQL	BQL	JOE :	10 TO 10 TO	<u> </u>	5 5	BQL	3QL			BQL			IOI	ş			ВОГ			BQL	,		
4 4 4 4 4	tro- 4-Ami	L) (F)																													BOLB								BQL B			EO.				вот в			BOL			
Ž,	iro. 3-Ni	(µg/L) (µg/L)		BQL BQL		BOL BOL			BQL BQL				BOL BOL				מלה מלה			BQL BQL				BUL BUL						BQL BC				BOL BC					BQL BG			BOI BC				BQL B(BOL BO			
, v	14,6 2-N	, (µg																																																		
, Amin	dinitrolol	(µg/L)		62.1	80.3	76.2	84.5	99.2	BQ				BQL			104	1			BQL			č	EQ.			BQ	BQL	BOL	BOL	BOL	68.4	727	46	78.4	90			BQL			IOR	3			BQL			BOL			
, Diplica	2,6-Dinitro	(pg/L)		BQL	BQL	BQL	BOL	BQL	BQL				BQL			100	2			BQL				מלר			BQL	BOL	BQL	BOL	BOL	BQL BQL	200	BQL	BOL	BQL			BQL			ROI	ž			BOL			BQL	,		
Diedir	4-Dimitro-	(µg/L)		11.3	9.3	9.4	6.9	11.2	BQL				BQL			100	2			BQL			2	J.			BQL	BQL	BOL	BOL	BOL.	12.6	10.7	8.4	10.7	BQL			BQL			ROL	2			BQL			BQL	,		
Piediro, 3	1,3-Diatro-2,4-Diatro-2,6-Diatro-2-Amino-4,6- 2-Nitro-3-Nitro-4-Amino-2,6- 4-Nitro- heazene inhene tahene diatronolamen tahene ahenen diatronolamen tahene			2.4	6:1	9.	BQL	1.9	BQL				BQL			100	2			BQL			7	T T T			BQL	BQL	BQL	BQL	BOL	52	1.7	BQL	BQL	BQL			BQL			ICH ICH ICH ICH ICH ICH ICH ICH ICH ICH	ב ב			BQL			BQL	,		
-	Nitrate by	_		0.944	0.872				<u>4</u>				1.54			773				1.36			:	151			1.82	1.84	1.93	1.95	2.45	1.05	0.802	0.804	60.1	1.31			1,46			33	}			97			1.73			
-	rotal	(µg/L) (m		1130				606					45			2				3.9			:	2			0.5	0.4	9.0	9.0	BQL		09/		996	208			53.7			8 91	90			5.3			2.1			
-	TNT TNR RDX Nitrohydies	1g/L) (p		34.4		_		36					2			100	2			BQL			,	מלב								36.6							1.2			0,1				BQL.			BQL	,		
	EN E	18/L) ()		202			414	365	2				34.1			2				3.8				2				0,4				361	378	341	426	138			42.1			5 77	}			8.			1.7			
	TAT	(HgA.) (HgA.) (HgA.)		505	368	380	379	380	45.1				7.9			=	3			0.1			3	2			BQL	BQL	BQL	BOL	BQL	<u>4</u> 5	328	318	1	29.7			8.4			12	<u> </u>			BQL			BQL	,		
Contactor	Peroxide	(mg/L)							40.4		38.3		40,4		40.4	41.6	ç	39.0		40.4		39.0	3	40.4	40.4		40.4		39.7							34.3	43.3		38,5		39.2	38.5	200	37.8		39.8	20.3	77.25	39.8		38.5	
Contactor		(mg/L)							-		75		7.1	i	74	97	6	74		79		78	į	Ĉ.	73	<u>:</u>	79		74							75	80		74	ì	22	7	2	7.5		9/	ž	ę	75		75	:
Contactor Off-eas 7		(%)							1.5		2.0		<u>e</u> :	;	2.1		3	2.1		1.7		2.0	:	2	2.2		1.7		2.1							2.0	9.		2.1		2.0	7.0	9	2.0		1.9	9	1.7	2.0		2.0	
Oxidation ((mV)		433		430			922		918		914		<u>8</u>	910		920		927		912	9	Š	894	;	864		810		270	433	424			931	906		924		915	803	2	851		930	8	F.	200		880	
2	Sample			4		15			=		2		12	:	٤	2	:	91		=		9	:	=	91	!	=		91		6 :	4	11	:		13	11		13	;	<u></u>	=	2	2		13	2	<u>•</u>	=		2	ļ
Tem		- 1		6'9		6.9			7.1		1.1		7.3		4.4	16	3	7.5		9.7		7.7		9.	2.8	!	7.9		7.8		0.8	œ.	80	1		6'9	7.0		7.2	;	7.2	7.3	3	7.4		1.7	7 1	ę.	7.8		7.8	
Ozone	Residual nH	(mg/L)							9.0	9.8	9.0	9.0	0.7	8.0	6.5		80	0.7	0.7	6.0	0.7	0.7	9.0	9 6	90	0.5	0.5	9.0	0.3	0.4	0.0					0.7	0.7	0.5	9.0	0.7	6.4	Ç 0	8.0	9.0	0.7	9.0	8.0	0.0	0.5	9.0	0.4	
Operations				09:53		15:07			09:47	11:38	14:58	16:16	09:34	134	14:45	71:01	11:30	14:36	16:05	91:60	11:27	14:25	00:91	11-22	14:17	15:56	08:42	11:14	14:04	15:51	08:24	91:01	14:53			09:47	14:45	16:15	09:37	11:45	14:35	01:01	11:40	14:20	90:91	90:60	11:35	16:04	08:53	11:30	13:53	
		- 1	GAC2				- N			CIV	CIVO	CIV	C2/0	520	2 2		C3/0	C3/0	C3/0	C4/0	C4/0	C470	CAO CAO	3.5	CSAO	CSAO	C6/0	C6/0	0/92	CKN	GAC3	INC	INFI	INFI	INF	C1/0	200	CIV	C270	020	9 S	38	C3/0	C3/0	C3/0	C4V0	C40	3 5	CS/O	CSVO	C5/0	
Average Hydrogen Average Peruxide PEROXONE Sample	Ratio L		0.52	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53			0.55	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	
Average Hydrugen A: Peroxide PER	Dose					40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	40.4	
ge Av Ted Hyu			4	4	4	4	4	4	4	4	4	4	4 ,	4 .	4 4	. 4	. 4	4																																		
Average Average Average Applied Transferred Hydrugen Ozone Ozone Peroxide		-	11	11	11	77	11	11	77	77	77	11	77	11	: F	: F		11	77	11	11	11	E :	: 1		77	77	77	11	11	11	2 %	9, 9	76	76	92	92	76	92	92	9,7	6 2	76	26	16	76	25 %	76	76	76	76	. ;
			86	86	86	86	86	85	86	86	86	86	86	e :	86 8	6 8	2 28	86	86	86	86	86	3° 3	8 3	2 25	86	86	86	86	86	9 0	8 8	\$ 3	86	86	86	8 8	86	86	86	5 6 9	8 3	8 %	86	86	85	86	2 20	35	86	86	1
Process	Well Flow Rate	(gpin)	2	2	2	2	2	2	<u>e</u> :	2	<u>=</u>	2	<u> </u>	<u> </u>	= =	2 2	2 2	13	5	13	13	2	<u> </u>	2 =	2 2	: 2	13	13	13	13	<u>=</u> :	= =	2 5	==	2	<u>e</u> :	2 5	2	2	<u>=</u> :	<u> </u>	2 2	2 2	13	13	13	2 2	2 2	=	=	13	:
_	Well E		-	-	-	_	_	_		_	_	_						-	_	-	-	-			-	-	-	-	-	_					-			-	-	-				-	-	-				_	-	,
	Date	- 1	10/9/96	96/01/01	10/10/96	96/01/01	10/10/96	96/01/01	96/01/01	96/01/01	10/10/96	10/10/96	96/01/01	96/01/01	96/01/01	06/01/01	96/01/01	10/10/96	96/01/01	10/10/96	10/10/96	96/01/01	96/01/01	10/10/96	96/01/01	96/01/0	10/10/96	96/01/01	96/01/01	10/10/96	96/01/01	10/11/96	10/11/06	10/11/96	96/11/0	96/11/01	10/11/96	96/11/01	96/11/01	96/11/01	96/11/01	10/11/06	10/11/96	10/11/96	10/11/96	10/11/96	96/11/01	06/11/01	10/11/96	10/11/96	96/11/08	

				Tetry	(ne/L)	BOL	BOL	800	BOL	BOL	,	
			Nitro-	benzene	(III)	BOL	BOL	BOL	BOL	BOL		
				НМХ	(Tyan)	0.2	BOL	BOL	BQL	BOL		
			· 4-Nitro-	e toluene	(ne/L)	BOL	BOL	BOL	BOL	BOL		
			Amino-2,6	nitrotolucn	(ne/L)	BOL	BOL	BOL	BOL	BOL		
			3-Nitro- 4-	toluene di	(ug/L)	BQL	BŲL	BOL	BQL	BOL		
			-Nitro-	oluene	(uv/L)	BQL	BQL	BOL	BOL	BOL		
			1,3-Dinitro-2,4-Dinitro-2,6-Dinitro- 2-Amino-4,6- 2-Nitro- 3-Nitro- 4-Amino-2,6- 4-Nitro-	TNT TNB RDX Nitrobodies Nitrate henzene toluene doluene dinitrotoluene toluene dinitrotoluene toluene HMX henzene Tetryl	(mg/L) (mg/L) (ug/L) (ug/L) (ug/L) (mg/L) (ug/L) (ug/L) (ug/L) (ug/L) (ug/L) (ug/L) (ug/L) (ug/L) (ug/L)	BOL	BQL	BQL	BQL	BOL		
			5-Dinitro- 2-	tolucne dir	(ns/L)	BQL	BQL	BQL	BUL	BQL		
			4-Dinitro- 2,0	tofuene	(ug/L)	BQL	BQL	BQL	BQL	BQL		
			-Dinitro-2,	cnzcne	(ug/L)	BQL	BQL	BQL	BQL	BQL		
			1,3	Nitrate h	mg/L N)	971	1.56	1.66	1.64	1.79		
			Total	itrohodies	(hk/L) (i	8.0	9.0	0.7	6.0	BQL		
				RDX N	ug/L)	30L	BQL	BQL	BQL	BUL		
				EN.	ux/L) (9.0		0.7		BQL		
				TNT	(µg/L) (BQL	BQL	BQL	BŲL	BQL		
		Confactor	Measured	Peruxide	(mg/L)	39.2		39.2				
		Oxidation Contactor Contactor Contactor	Transferred Measured	Ozone	(mg/L)	75		92				
		Contactor	Off-gas	Ozone	(%)	2.0		1.9				
		Oxidation	Reduction	Potential	(mV)	875		820		283		
		emperature	of ORP Re	Sample	()()	Ξ		<u>*</u>		=		
		Γ-		Residual pH Sample		7.9		7.8		7.9		
-			Ozone	Residua		0.5	0.5	0.3	0.3	0.0		
		Operations	Sample	Time		08:40	11:25	13:40	15:53	08:25		
			Peroxide PEROXONE Sample Sample	Lxcation		C6/0	CWO	C6/0	C6/0	GAC3	GACI	GAC2
		Average	PEROXONI	Ratio		0.53	0.53	0.53	0.53	0.53	0.53	0.53
	Average	Hydrogen	Peroxide	Dose	(mg/L)	40.4	40.4	40.4	40.4	40.4	40.4	40.4
	Average	Transferred	Ozone	Dose	(mg/L)	92	92	92	92	92	9/	36
	verage	phlicd	Ozone	Dose	mg/L)	86	86	86	86	86	86	×
	<	*	Process Ozone Ozone Peroxide Pl	Flow Rate	(gpm) (mg/L) (mg/L)	13	13	5	5	2	2	13
				Well		- 4	1 9	- 9	1 9	- 9	1 9	-
				Date		96/11/01	10/11/96	10/11/9	10/11/9	10/11/6	10/11/9	10/11/96

			Tetryl	(µg/L)	BQL	BQL	BQL	BQL	5.4	BQL		BQL	,	BQL		BQL		BQL		BQL	9	6.4	6.9	7	6.3	-			BOL	,			BQL				BQL									
		Nitro:	benzene	(µg/L)	BQL	BQL	BQL	BQL	BOL	BQL		BQL		BQL		BQL		BQL		BQL		-	BOL				BQL				BQL															
			HMX	(hg/L)	8.5	7.3	æ	8.9	9.7	5.7		2.8		9.1		-		9.0		BQL	BQL	0.3	BQL	BQL	BQL	BQL	9.1	6.5	10.8	9.4	7.3	6.1			3.3				œ:				1.2			
		4-Nitro-	toluene	(hg/L)	BQL	BQL	BQL	BQL	BOL	BQL		BQL		BQL		BQL		BQL		BQL			BOL				BQL				BQL															
		3-Nitro- 4-Amino-2,6- 4-Nitro-	dinitrotoluene	(µg/L)	BQL	BQL	BQL	BQL	BOL	BQL		BQL		BQL		BQL		BQL		BQL			BOL				BQL				BQL															
		3-Nitro- 4-	totuene di	(µg/L)	BQL	BQL	BQL	BQL	BOL	BOL		BQL		BQL		BQL		BQL		BQL			BOL	,			BQL				BQL															
				(µg/L)	BQL	BOL	BQL	BQL	BOL	BQL		BQL		BQL		BQL		BQL		BQL			BOL	ı			BQL				BQL															
		1,3-Dinitro-2,4-Dinitro-2,6-Dinitro-2-Amino-4,6- 2-Nitro-	toluene dinitrotoluene toluene	(µg/L)	103	92	87.4	89.8	80.3	BQL		BQL		BQL		BQL		BQL		BQL	72.4	69	86.5	78.8	73.8	BQL			BOL				BQL		•		BQL									
		-Dinitro 2-	ofuene din	(µg/L)	BQL	22.5	BQL	BQL	BOL	BQL		BQL		BQL		BQL		BQL		BQL			BOL				BQL				BQL															
		-Dinitro-2,6	toluene	(µg/L)	14.2	11.4	12.1	12.3	11.2	BQL		BQL		BQL		BQL		BQL		BQL	10.3	7.6	11.7	11.3	10.7	BQL			BOL				BQL				BQL									
		-Dinitro-2,4	benzene	(µg/L)	BQL	BQL	BQL	BQL	œ: 	BQL		BQL		BQL		BQL		BQL		BQL			BOL	,			BQL				BQL															
		<u></u>			0.953	0.998	0.873	16.0	0.934	Ξ		1.25		1.25		1.28		1.33		1.22	1.36	1.29	1.37	1.96	1.56	1.77	1.2	1.06	1.04	1.08	-	1.29			1.37				1.4				1.42			
		Total	RDX Nitrobodic: Nitrate	(µg/L) (mg/L N	954	825	896	864	832	256		119		50.7		17.3		ó		-	3.6	4	3,8	BQL	BQL	BQL	832	840	1200	927	406	241			119				40				16.7			
			RDX N		38.5	35.9	36.1	38.8	27.8	10.9		3.2		6.0		0.3		BQL		BQL	26.4	BQL	BQL	28.8	5.9	= .4			3.4				0.7				BQL									
			TNB	ug/L)	410	324	388	339	358	091		91.5		43.5		15.1		2 .		-	3.6	3.7	3.8	BQL	BQL	BQL	357	378	546	396	410	150			91.3				33.1				14.6			
			TNT	(µg/L) (µg/L) (µg/L)	380	332	366	377	340	78.9		21.6		4.7		6.0		0.3		BQL	354	370	538	396	393	72.7			20.8				4.4				6.0									
	Contactor	Measured	Peruxide	(mg/L)						25	25	24	25	24	24	24	24	23	23	24		24										24	;	2	24		24		25		24		24		25	
	Oxidation Contactor	Reduction Measured	Potential	(mV)	424					575		889		924		932		929		888				255			427		422			870	į	6/5	941		913		930		931		943		903	
	Temp.		Sample	(၁၀)	<u>«</u>					<u>ec</u>		11		11		17		11		81				91			15		<u>«</u>			4	;	61	4		<u>∞</u>		15		<u>«</u>		4		61	
	_	\$	Έ		7.1					7.1		7.3		7.4		7.5		9.7		7.8				<u>«</u>			6.9		7.0			7.0		7.	7.0		7.2		7.3		7.3		7.4		7.5	
		Охипс	Residual	(mg/L)						0.3	0.4	0.4	9.0	0.4	9.0	9.0	0.7	9.0	0.7	0.4		9.0		0.0								0.3	9.0	0 0	0.7	0.7	0.5	0.5	8.0	8.0	0.5	0.7	8.0	0.4	0.4	9.0
	Operations		Time		13:41					13:26	15:02	13:14	14:59	13:07	14:56	12:58	14:52	12:50	14:48	12:40		14:45		12:25			20:22		15:13			10:00	11:56	50:51	08:60	11:53	14:57	16:28	09:35	11:49	14:51	16:25	09:28	11:45	14:43	16:21
		Sample	Location		INF	INFI	INFI	INFI	INFI	C1/0	C1/0	C2/0	C2/0	C3/0	C3/0	C4/0	C4/0	C5/0	C\$/0	C6/0	C6/0	0/9O	C6/0	GAC3	GACI	GAC2	INFI	INFI	INFI	INFI	INFI	C1/0	C1/0	2 2	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0
	Average	Peroxide PEROXONE Sample Sample	Ratio		0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52
Average	Hydroger	Peroxide	Dose	(mg/L)	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.6	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7
Average Average Average	Applied TransferredHydroger Average	Ozone	Dose	(mg/L)	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	46,4	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2
Average	Applied	Ozone	Dose	(ng/L)	09	09	09	9	09	9	9	09	9	09	9	09	9	09	9	9	9	60	60	9	09	09	99	09	09	99	9	09	90	2 8	. 8	8	9	99	60	09	9	60	99	99	9	9
		Process	Well Flow Rate	(gpm)	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0
		-	Vell Fi	-	_	_	_	_	_	_	_	_	_	_	_	_	_	_	_			_	_	_	_	_	_	_	_	_	_	_	_		. –	_	_	_	_	_	_	_	_	_	_	-
			Date		10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/12/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96

			Tetryl	(hg/L)	BOL	,			BQL	BQL	BQL	BQL	BQL	5.2	5.2	5.7	5.9	9.6	0.5				BQL				BQL				BQL			i	BUL			Da	3 5	2 2	BOL BOL	BOL	4.4	7	3.6	4.6
		Nitro-	cuzene	(µg/L)	BOL				BQL	BQL	BQL	BQL	BQL	3.9	BQL	BQL	BQL	BQL	BQL				BQL				BQL				BQL			- ;	n N			Ö	2 6	2 2	B 2	BOL	BQL	BQL	BQL	BQL
			НМХ	(нg/L) (нg/L)	BOL				BQL	BQL	BQL	BQL	BQL	4	5.1	5.5	5.7	8.5	~				3				9.1				8.0				0.0			Od	2 5	3 5	BOL	BOL	4.5	5.2	2.4	4.5
		-Nitro-	toluene	(µg/L)	BOL				BQL				BQL				BQL				BQL			9	BQL			108	30	BO I	BOL	BOL	BQL	BQL	BQL	BQL										
		1,3-Dinitro 2,4-Dinitro 2,6-Dinitro 2-Amino 4,6. 2-Nitro 3-Nitro 4-Amino 2,6. 4-Nitro	toluene dinitrotoluene toluene HMX benzene Tetryl	(µg/L)	BOL				BQL				BQL				BQL				BQL			3	ng Cr			Od	BO 10	2 2	BOL	BOL	BQL	BQL	BQL	BOL										
		-Nitro- 4	oluene d	(hg/L)	BOL				BQL				BQL				BQL				BQL			č	Ž,			IOB	3 G	2 2	BOL	BOL	BQL	BQL	BQL	BQL										
		Nitro- 3		(hg/L) (BOL				BQL				BQL				BQL				BQL				a Cr			10	3 G	2 2	BOL 25	BOL	BQL BQL	BQL	BQL	BQL										
		ino-4,6-2	toluene dinitratoluene toluene	(μg/L) (BOL				BQL	BQL	BQL	BQL	BQL	19	62.4	70	6.07	1.69	BQL				BQL				BQL				BQL			Š	17 CF			ĮO.	10 E	2 2	BOL	30F	63.1	78.7	23	24
		tro-2-An	e dinitr	1																																			a _:		נ נ		د ،	_	ب	ے
		-2,6-Dini	toluen	(µg/L)	BOL				BQL	BQL	BQL	BQL	BQL	BOL	BQL	BQL	BQL	BOL	BQL				BQL				BQL				BQL			Č	a C			IOR IOR	BOL B	2 2	log Bor	BOL	10g	BQL	BQL	BQL
		2,4-Dinitro	toluene	(µg/L)	BOL				BQL	BQL	BQL	BQL	BQL	10.7	9.6	10.7	10.9	8'01	BQL				BQL				BQL				BQL			č	D/L			O	200	2 0	BOL 2	BOL	8.6	12.1	œ: œ:	6
		3-Dinitro	benzene	(hg/L)	BOL				BQL	BQL	BQL	BQL	BQL	4.	BQL	BQL	BQL	BQL	BQL				BQL				BQL				BQL			Š	D C			IOA	10 E	2 2	30F	BOL	- e-	9.1	0.7	BQL
				- 1	7.6				1.43	1.58	1.55	1.53	1.74	1.08	10.1	0.999	1.03	-	1.27				1.29				1.42				1.45			:	34.			1 33	51	27	5	164	0.869	0.894	0.852	0.896
		Total	TNT TNB RDX Nitrohodie: Nitrate	(mg/L) (µg/L) (µg/L) (µg/L) (mg/L)	6.7				2.5	2.2	3.3	5.6	BQL	718	842	878	820	853	270				101				35.2				15.3			;	0.7			3 6	2.9) P	2 2	BOL	74.5	1200	785	731
			RDX Ni	ug/L)	BOL				BQL	BQL	BQL	BQL	BQL	27.3	19.2	23.6	23.2	20.8	6				3.2				0.7				BQL			Č	1			0	2 5	2 2	BOL	BOL	25.8	35.3	25.5	22.9
			TNB	(µg/L) (6.5				2.5	2.2	3.3	5.6	BQL	312	386	386	351	381	173				75.1				53				13.7			,	c			č	2.9	} =	2.8	BOL	331	540	361	335
			TNT	(µg/L)	0.7				BQL	BQL	BQL	BQL	BQL	292	355	376	352	357	82.2				6.61				3,9				0.8				7:0			IOE	BO E	2 2	BOL	BOL	301	518	326	301
	Oxidation Contactor	Reduction Measured	Peroxide	(mg/L)	24		24		24		24								25		25		24		24		25		25		24		24	;	67	7	\$ 7	35	3	24	1					
	Oxidation	eduction	Potential	(IIIV)	925		903		988		870		258	425		405			895		959		926		106		940		904		934		922		416	8	306	878	2	850	2	279	463		422	
	Temp. C	of ORP R	0	(ၞ	7		20		4		70		258	15		91			15		11		15		<u>~</u>		15		81		15		<u>sc</u>	:	2	ę	07	×	2	20	1	15	2 2		91	
			핕		7.5		7.6		7.6		1.7		7.8	7.0		7.1			7.1		7.2		7.2		7.3		7.3		7.4		7.5		7.6	;	0	r	9.7	7.7	:	7.8	?	7.8	7.0		7.1	
	s	Ozone	Residual	(mg/L)	0.5	0.7	0.5	9.0	9.0	9.0	0.4	0.4	0.0						0.5	0.7	0.4	0.5	0.5	0.7	0.5	0.7	0.8	6.0	0.5	0.7	9.0	0.7	0.5	9.0	0.0	6.0	5 6		90	6 6	0.5					
	Operations	Sample	Time		81:60	11:42	14:35	16:14	90:60	11:32	14:24	16:18	08:45	10:01		15:30			09:53	11:51	15:24	16:25	09:46	11:48	15:20	16:23	09:34	11:42	14:50	16:21	09:11	11:40	14:40	16:19	60:90	2 3	16.16	08.48	11:33	4.02	16:12	08:38	10:08		15:10	
		Sample	Location Time		C5/0	C5/0	CS/0	C5/0	C6/0	C6/0	C6/0	C6/0	GAC3	IN E	E E	Ξ	E N	IN I	C1/0	C1/0	0/10	C1/0	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C49	C3/0	250		CKA	990	990	0,90	GAC3	INF	INF	EN	I'A
	Average	ROXONE	Ratio		0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.52	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	0.53	6.0	0.55	600	230	0.53	0.53	0.53	0.53	0.54	0.54	0.54	0.54
verage	Applied TransferredHydroger Average	Peroxide PEROXONE Sample Sample	Dose	(mg/L)	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.3	24.5	C.#2	24.5	24.5	24.5	24.5	24.5	24.7	24.7	24.7	24.7
Average Average	ısferredHy			(mg/L) (47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	47.2	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	46.5	5.04	40.5	40.3 46.6	344	46.5	46.5	46.5	46.5	46.1	46.1	46.1	46.1
Average Av	plied Trat			(mg/L) (t	09		09	9	09	09	09		09		09	09	09	09	09	09	09				60	60	09	09	09	09	09	09	99	E :	£ (S (2 9	3 9	2 9	9	2 9	9	3 3	9	99	09
Av	Αĥ	Process O	Rafe D	m) (mdg)	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	0.62	0.62	0.62	0.04	0.62	0.62	25.0	25.0	25.0	25.0	25.0	25.0
		Pro	Well Flow Rate	3)	2	2	2.	2	2	2	2	2	2	2	2	2	2	2	2	2	2	7		. 7	2	7		7	2	2	2	- 5	(4	_ '	- (_ (,,	-	
					96	1 96	1 96	1 96	1 96	1 96	96	96	1 96	1 96	1 96	1 96	1 96	1 96	y6 I	1 96	1 96	96	96	1 96	1 96	96	1 96	1 96	1 96	1 96	96,	1 96	96	96/	£ }	9	96	2 7	706	. y.	190	φ.		96/	196	96/
			Date		10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/13/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	10/14/96	* 100	10/14/96	10/14/06	10/14/96	10/14/96	96/51/01	10/15/96	10/15/96	96/51/01

		Terry	(µg/L)		0.7				0.5				0.3				BQL				BQL				BQL	BQL	BQL	BQL	BQL	7	5.9	5.2	5.2	60		6.4		0.2		BQL		BQL	e e	80E	,
		Nitro-	(ug/L)		10 E	i i			BQL				BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	9-10g	BOL	,	BQL		BQL		BQL		BQL		80F	,												
		HMX		,,	2.5				4.1				_				0.5				9.0				BQL	BQL	BQL	BQL	BQL	4.7	5.5	3.8	3,38	4.4		2.1		2		BQL		BQL		BQL BQL	
			(µg/L)		BOL	i i			BQL				BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BOL	,	BQL		BQL		BQL		BQL		BQL													
	,	1,3-Dinitro 2,4-Dinitro 2,6-Dinitro 2-Amino 4,6-2-Nitro 3-Nitro 4-Amino 2,6-4-Nitro benzene toltene toltene toltene toltene toltene toltene toltene	(µg/L)	ā	BOL	i i			BQL				BQL	BQL	BQL	BQL	BQL	BOL	BQL	BQL	BQL BOI	BOL	,	BQL		BQL		BQL		BQL	i	BQL BQL													
		Nitro- 4-A	(µg/L)		BOL				BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BOL	BOL	,	BQL		BQL		BQL		BQL	i	BQL BQL																	
	•	-Nitro- 3-		ı					BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BOIL BOIL	BOL	,	BQL		BQL		BQL		BQL		BQL BQL																	
		6-Dinitro-2-Amino-4,6- 2-Nitro- tolnene dinitratolsene tolnene	(µg/L)		BOL	ļ			BQL				BQL	BQL	BQL	BQL	BQL	73	63.2	1.09	59.4	BOL	,	BQL		BQL		BQL		BQL	ě	BQL BQL													
		itro-2-Ar re dinit	~		, ,				,				. 1				_1				د				_1	۔	_	۔	ر	_	٦			د. د		٦		۳.						ی ی	
		tro-2,6-Din c toluer							BQL				BQL				BQL .				BQL.				BQL.	, BQL	BQL	BOL					BOL BOL			BOL		BQL		L BQL		L BQL		708 1	
		n-2,4-Dinitre toluene			BOL	,			BQL				BQL	BQL	BQL	BQL	BQL	11.5	10.5	9.4	4.01 4.02	BOL	,	BQL		BQL		BQL		BQL		8 10 10 10 10													
		1,3-Dinitra benzene	(µg/L)	2	BOL	,			BQL				BOL				BQL				BQL				BQL	BQL	BQL	BQL	BQL	2.1	Ξ	1.7	<u>5</u> -	BOL	,	BQL		BQL		BQL		BQL		BQL BQL	
		Total RDX Nitrobodie: Nitrate	(µg/L) (mg/L N	0.00	1.12				1.17				1.21				1.28				1.35				1.3	1.3	1.36	1.37	1.48	0.956	0.977	0.921	0.902	1.09		1.18		1.21		1.25		1.37		8. 1 .	
		Total firobodic	(µg/L)	280	318				16				45				14.7				5.9				2.9	2.5	2.4	2.4	0.3	1040	552	952	848	278		82.4		32.8		14.2		5.5	•	3.2.5	
				24.7	6.6				5.6				6.0				BQL				BQL				BQL	BQL	BQL	BQL	BQL	31.8	27	25.7	26.8	83		2.2		0.7		0.2		BQL	9	80 L	
		TNB	(µg/L)	356	205				70.7				35.8				13.5				5.2				2.9	2.5	2.4	2.4	0.3	461	233	430	382	<u>=</u>		63.5		27.6		13.3		5.4	•	3.2.5	
		TNT		308	8				15.8				4				0.7				0.1				BQL	BQL	BQL	BQL	BOL	449	506	416	329	83.9		14.2		3		0.7		0.1	3	80 PG	
	Contactor	Measured	(mg/L)		25		24		24		24		25		24		24		24		24		24		24		24							24	24	24	24	24	74	24	34	24	24	24	
	Oxidation Contactor	Reduction Measured Potential Peroxide	(mV)		880		889		935		931		936	,	616		945		930		928		925		848		887		269	432				878		923		948		949		932		903	
	Temp.	Sample			13		91		13		91		13		11		13		11		12		11		13		81		<u>«</u>	13				13		13		13		13		13	:	13	
	_	o Ha	_		7.2		7.2		. 7.3		7.4		7.3		7.5		7.5		9.7		1.7		7.7		7.9		7.9		8.0	7.0				7.0		7.2		7.4		7.5		9.2	ì	9.7	
		Ozone Residual DH	(mg/L)		0.4	0.7	4.0	0.5	9.0	6.0	0.5	0.7	0.7	6.0	9.0	8.0	8.0	6.0	0.5	9.0	6.0	6.0	0.4	0.7	0.7	0.7	0.5	9.0	0.0					0.4	0.5	0.5	0.7	0.7	6.0	8.0	Ξ	0.7	6.0	0.7	
	Operations	Sample			10:00	12:40	15:07	16:35	09:53	12:37	14:59	16:32	09:42	12:35	14:49	16:29	08:55	12:33	14:42	16:26	08:50	12:30	14:34	16:23	08:42	12:25	14:26	16:20	08:28	07:56				07:48	81:60	07:37	09:15	02:20	09:12	07:24	60:60	07:08	50:60	06:58	
	5	Peroxide PEROXONE Sample Sample Ozone Dose Ratio Location Time Residua		1321	C1/0	0/10	C1/0	C1/0	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	C5/0	C5/0	CS/0	CS/0	C6/0	C6/0	C6/0	C6/0	GAC3	INFI	INFI	IN E	<u> </u>	0170	C1/0	C2/0	C2/0	C3/0	C3/0	C4/0	C4/0	C5/0	C5/0	C6/0	
	Average	Ratio		0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	Ϋ́Z	Y V	٧ Z	Y Z	ž	N A	٧	¥ Z	N A	¥	۲ ۲	Ϋ́Z	Ϋ́	¥ :	₹ ₹	
crage	droger A	eroxide PE) Dose	_	747	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	24.7	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0 25.0	
Average Average Average	$\overline{}$	Ozone Per Dose L		246.1	_			46.1	46.1	46.1	46.1	46.1	46.1	46.1	46.1	46.1		46.1	46.1	46.1	46.1	46.1	46.1	46.1	46.1	46.1	1.94	46.1	46.1	fonitor Dow			fonitor Dow			fonitor Dow	funitor Dow	funitor Dow	fonitor Dow	fonitor Dow	fonitor Dow			fonitor Dow fonitor Dow	
rage Aw	lied Tran.		_	09				60 4	60 4	60 4	60 4	60 4	60 4	60 4	60 4						60 4	60 4	4 09	60 4	60 4	60 4	60 4	4 09		60 fonite			60 t onie			60 fonit	60 fonit	60 funit		60 1onit	-	-	-	60 tonit 60 tonit	
Ave																													_																
	4	Process Well Flow Rate	(gpm)	. 050	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	
		Well		-		_	- 5	-	- 5	-	-	- 5	-	-	- 9	-	- 4	- 4	1	9	- 9	1 9	1 9	- 9	1 9	- 4	1 9	1 9	1 9	1 9	1 9	1 9	 _ v		1 9	1 9	9	- 9	- 4	1 9	- 9	1 9	- 9	 2 4	
		Date		10/15/96	96/51/01	96/51/01	96/51/01	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	96/51/01	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	96/51/01	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	10/15/96	10/16/96	96/91/01	96/91/01	10/16/96	96/91/01	10/16/96	10/16/96	10/16/96	10/16/96	10/16/96	96/91/01	10/16/96	96/91/01	10/16/96	96/91/01	

		benzene Tetryl	(µg/L)		BOL	y i			BOL	BOL	BOL	BOL	BOL	BQL	BOL	4.5	3.6	4.7	4.3	4.4	BOL	,			BQL				BQL				BQL			BOL				BOL					
	Nitro				BOL	,			BOL	BOL	BOL	BOL	BOL	BQL	BOL	BQL	BOL	BOL	BOL	BOL	BOL	,			BQL				BOL				핞			BOL	í			BOL	BOL	BQL	BQL	BQL	BQL
		HMX	(µg/L)		BOL	,			BOL	BOL	BOL	BOL	BOL	BQL	BOL	9.4	7.4	4.4	6.1	6.7	٠,				2.8				1.7				Ξ			0.7				BOL	BOL	BQL	BQL	BQL	8.9
	4.Nirm		(µg/L)		BOL	y y			BOL	BOL	BOL	BOL	BOL	BQL	BOL	BQL	BQL	BOL	BOL	BOL	BOL	,			BQL				BQL				BQL			BOL) '			BOL	BOL	BQL	BQL	BQL	BQL
	1.3.Dinito.2.4.Dinito.2.6.Dinito.2.Amino.46. 2.Nito. 1.Nito. 4.Amino.26. 4.Nito.	toluene dinitrotoluene toluene	(µg/L)		BOL				BQL	BOL	BQL	BOL	BQL	BQL	BOL	BQL	BQL	BQL	BOL	BQL	BQL	,			BQL				BQL				BQL			BOL				BOL	BOL	BQL	BQL	BQL	BQL
	3-Nirm. 4.	toluene di	(µg/L)		BOL	ļ			BQL	BOL	BQL	BOL	BQL	BQL	BOL	BQL	BQL	BQL	BOL	BOL	BQL	,			BQL				BQL				BQL			BOL	ļ			BOL	BOL	BQL	BQL	BQL	BQL
	2.Nitro-	toluene	(µg/L)		BOL	ļ			BQL	BOL	BQL	BOL	BOL	BOL	BOL	BQL	BQL	BQL	BOL	BQL	BQL	,			BQL				BQL				BQL			BOL	ļ			BQL	BOL	BQL	BQL	BQL	BOL
	-Amino-46-	toluene dinitrotoluene toluene	(µg/L)		BOL				BOL	BOL	BOL	BOL	BQL	BQL	BOL	74.3	62.4	53.3	67.1	58.4	BQL	•			BQL				BQL				BQL			BOL				BQL	BOL	BQL	BQL	BQL	64.7
	6-Diniern-2	toluene d	(hg/L)		BOL	,			BQL	BOL	BOL	BOL	BQL	BOL	BQL				BQL				BQL				BQL			BOL	,			BOL	BOL	BQL	BQL	BQL	BQL						
	4-Dinitro-2	toluene	(hg/L)		BQL				BQL	BOL	BQL	BQL	BQL	BQL	BQL	Ξ	7.6	6.8	10.4	9.5	BQL				BQL				BQL				BQL			BOL	,			BQL	BQL	BQL	BQL	BQL	10.4
	3-Dinitro-2	benzene	(µg/L)		BQL				BQL	1.7	1.3	_	1.7	BQL	BQL	,			BQL				BQL				BQL			BOL	,			BQL	BQL	BQL	BQL	BQL	1.2						
		RDX Nitrobodie: Nitrate	(µg/L) (mg/L N		1.95				2.14	2.14	2.14	2.32	1.92	1.73	1.97	1.14	1.14	Ξ	Ξ	90:1	1.43				1.62				1.76				1.94			2.08				2.15	2.14	2.25	2.18	2.25	0.853
	Total	irobodi	(µg/L)		5.7				2	2.1	2.5	5.6	0.3	0.3	0.4	1020	883	685	944	748	244				88.5				33				14.5			6.5				2.2	2	2.5	2.6	0.3	853
		RDX N	(µg/L)		BQL				BQL	29	25.5	22.4	26.8	25.4	oc				2.3				8.0				BQL			BOL	,			BQL	BOL	BQL	BQL	BQL	25.6						
		TNB	(µg/L) (µg/L)		9.6				2	2.1	2.5	5.6	0.3	0.3	6.4	467	404	314	435	332	158				67.5				27.4				12.7			5.6				2.2	2	2.5	5.6	0.3	386
		TNT	(µg/L)		0.1				BQL	427	369	276	393	312	73.2				.5.9				3.1				0.7			0.2				BQL	BOL	BQL	BQL	BQL	354						
	Reduction Measured	Peroxide	(mg/L)		25		36		56		25										23		24		25		25		24		24		25	36		26		26		26		26			
	Reduction	Potential	(mV)		930		914		905		006		329			380		423			363		805		920		941		940		894		935	033	3	116		016		886		897		260	325
1	of ORP	Sample	(၃)		10		13		9		13		6			2		15			5		16		12		91		13		16		=	~		=		91		=		91		10	13
	•	Ħ			7.5		7.6		7.5		7.7		7.6			8.9		7.2			7.1		7.2		7.1		7.5		7.3		7.4		7.4	7.3		7.2		7.7		7.7		7.9		7.8	6.7
,	Ozone		(mg/L)	Ξ	0.7	0.8	9.0	0.8	0.9	0.	0.7	0.8	0.0								0.1	0.5	0.4	0.4	9.0	0.8	9.0	0.7	0.7	Ξ	0.7	Ξ	0.7	2 0	0.0	0.4	0.7	0.4	9.0	0.5	9.0	9.0	0.7	0.0	
on citation of	Sample	Time		16:15	08:58	10:44	13:55	16:07	08:41	10:40	13:39	16:03	08:15			10:58		14:40			10:21	11:34	14:26	15:58	10:13	11:30	14:18	15:56	10:02	11:27	14:04	15:53	09:55	13.55	15:50	09:48	11:21	13:48	15:47	09:12	11:18	13:29	15:45	09:23	10:10
	Peroxide PEROXONE Sample Sample	Location		C4/0	CS/0	C5/0	C5/0	C2/0	C6/0	C6/0	C6/0	C6/0	GAC3	GACI	GAC2	INFI	INFI	INFI	INF	INF	C1/0	C1/0	C1/0	C1/0	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C5/0	C\$/0	CS/0	C5/0	C6/0	C6/0	C6/0	C6/0	GAC3	INFI
Avorage	EROXON	Ratio		09.0	09.0	09.0	09.0	09'0	09.0	09'0	09.0	09'0	09.0	09'0	09.0	19.0	19.0	19:0	0.61	0.61	19:0	19.0	19'0	0.61	0.61	0.61	19.0	19:0	0.61	19:0	19.0	0.61	19:0	190	19'0	0.61	19:0	0.61	19'0	19'0	0.61	19:0	19.0	19.0	0.58
Average	PeroxideP	Dose	(mg/L)	25.7	25.7	25.7	25.7	25.7	25.7	25.7	25.7	25.7	25.7	25.7	25.7	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.0	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	25.9	24.8
Average Average Average	Ozone		(mg/L)	42.9	42.9	42.9	42.9	42.9	42.9	42.9	42.9	42.9	42.9	42.9	42.9	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	45.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.4	42.7
Average	Ozone	Dose	(mg/L)	96	26	98	99	96	98	96	98	99	92	96	26	26	98	56	96	56	56	98	98	99	96	99	98	98	98	98	26	98	% %	2 %	. %	26	98	26	98	98	99	99	98	99	57
	Process	٥	(gpm)	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5
	-	Well F		_	_	-	_	_	_	_	_	_	_	-	_	-	_	_	-	_	_	_	_	_	_	-	_	_	-	-	-	_				_	_	_	_	-	-	-	_	_	-
		Date		10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/22/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	90/20/01	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/23/96	10/24/96

		Tetrvi	(µg/L)	4	5.5	4.7	5.1	0.4				BQL	BQL	BQL	BQL	BQL	4	3.8	4.3	5.7	5.2	0.3				BQL				BQL																
		Nitro-	Hg/L)	BOL	BOL	BOL	BQL	BQL				BQL	BÓL	BQL				BQL				BQL																								
		HMX	(µg/L) (µg/L) (µg/L)	4.2	7.9	6.7	7.5	4.9				2.7				9.1				Ξ				0.7				BQL	BQL	BQL	BQL	BQL	4.1	4.8	4.9	5.2	4.9	4.7				5.6				5.
		3-Nitro- 4-Amino-2,6- 4-Nitro- tolucne dinitrotolucne tolucne HMX benzene Tetryl	(hg/L)	BOL	BOL	BQL	BQL	BQL				BQL				BQL				BQL																										
		3-Nitro- 4-Amino-2,6- 4-Nitro- lotuene dinitrotoluene toluene	(µg/L)	BOL	BQL	BQL	BOL	BQL				BQL				BQL				BQL																										
	;	Nitro- 4-A luene dini	(µg/L)	BOL.	BOL	BOL	BQL	BQL				BQL				BQL				BQL																										
				BOL				BQL				BQL				BOL				BQL				BQL				BQL				BQL				BQ.										
		4,6- 2-N uenc tolu	ŝii)																																											
		1,3-Dinitro-2,4-Dinitro-2,6-Dinitro-2-Amino-4,6- 2-Nitro- benzene toluene toluene dinitrotoluene toluene	(µg/L)	55.2	74	63.4	69.7	BQL				BQL	BQL	BOL	BQL	BQL	51.3	49	57.7	53.4	55.1	BQL				BQL				BQL																
	i	2,6-Dinitre	(µg/L)	BOL	BOL	BQL	BQL	BQL				BQL	BQL	BOL	BQL	BOL	BOL	BQL	BQL	BQL	BQL	BQL				BQL				BQL																
	i	4-Dinitro	(µg/L)	5.6	11.8	10.1	9.11	BQL				BQL	BQL	BQL	BQL	BQL	8.7	9.6	9.6	9.3	9.8	BQL				BQL				BQL																
	i	,3-Dinitro-2 benzene	(Hg/L)	-	9.1	1.2	4.	BQL				BQL	BQL	BQL	BQL	BQL	1.7	1.3	4.	5:	9.1	BQL				BQL				BQL																
	•	Nitrate		0.866	0.833	0.787	0.866	Ξ				1.3				1.33				1.43				1.74				1.5	1.57	1.64	1.77	99'1	0.807	0.851	0.84	0.842	0.833	1.14				1.26				1.34
		Total RDX Nitrohodie: Nitrate	(µg/L)	712	974	784	810	256				86.9				35.2				11				7.2				2.1	2.2	2.9	2.8	0.3	612	879	645	703	631	240				94.5				33.1
		RDX		24.9	29.8	25.3	28.7	8.6				2.3				9.0				BQL				BQL				BQL	BQL	BQL	BQL	BQL	21.9	22.2	24.3	23.7	23.6	7.9				2.4				0.5
				326	456	352	352	165				66.5				29.6				15.1				6.3				2.1	2.2	5.9	2.8	0.3	275	313	289	320	282	157				72.9				28.1
		TNT	(µg/L) (µg/L) (µg/L)	287	387	321	334	77.4				15.4				3.4				8.0				0.2				BQL	BQL	BQL	BQL	BQL	245	274	254	284	250	70.3				9.91				6
	Temp. Oxidation Contactor	of ORP Reduction Measured Sannle Potential Peroxide TNT TNB	(mg/L)					27		24		25		25		24		24		24		25		56		26		56		56								26		56		26		26		25
	Oxidation	Reduction	(m)		382			300		620		897		925		920		945		940		931		016		885		743		116		335	366		398			376		381		968		884		921
	Temp. (Samule			œ			4		11		13		9		13		91		=		91		=		11		0		11		2	16		2			61		œ		4		12		15
		Ŧ			8.9			7.0		7.1		7.1		7.1		7.		7.3		7.4		7.4		7.5		7.5		1.6		7.7		7.7	6.9		7.0			7.1		7.1		7.2		7.2		7.4
		Ozone Residual	(mg/L)					0.0	0,4	0.3	0.4	0.3	9.0	9.0	9.0	0.8	1.2	0.7	0.8	0.8	1.0	0.7	0.8	0.5	0.7	0.5	9.0	0.3	0.7	0.5	9.0	0.0						0.2	0.3	0.2	0.2	0.4	0.7	0.4	0.5	0.7
	Operations	Sample			15:24			09:40	11:36	15:11	16:29	06:30	11:33	14:58	16:26	09:20	11:31	14:38	16:22	01:60	11:28	14:31	16;20	00:60	11:25	14:09	16:18	08:42	11:21	14:20	16:14	08:25	11:09		14:44			10:25	11:31	14:07	15:11	10:22	11:28	14:02	15:08	10:07
		Sample Sample Location Time		NE	INF	INF	ENE	C1/0	C1/0	C1/0	0/10	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	C5/0	C5/0	C5/0	C5/0	C6/0	C6/0	C6/0	C6/0	GAC3	INF	INF	INE	INE	IN.	CI/0	C1/0	C1/0	C1/0	C2/0	C2/0	C2/0	C2/0	C3/0
	Average	EROXONI Ratio		0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.63	0.63	69'0	0.63	0.63	0.63	0.63	0.63	0.63	69.0	0.63	0.63	0.63	0.63
Average	Applied TransferredHydroger Average	Peroxide PEROXONE Sample Sample Ozone Dose Ratio Location Time Residua	(mg/L)	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	24.8	26.9	26.9	26.9	26.9	26.9	26.9	26.9	26.9	26.9	26.9	56.9	26.9	26.9	26.9
Average Average Average	ransferredl	Ozone	(mg/L)	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.7	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8
Vverage	Applied T	Ozone	(mg/L)	2	57	23	22	57	57	23	27	22	23	27	23	22	23	23	57	57	27	57	57	57	2.2	2.2	2.1	57	57	2.1	2.1	23	27	22	22	57	57	57	57	57	57	57	57	22	23	57
		Process Ozone Well Flow Rate Dose	(mdg)		24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5
		Vell Fig.		_		_	_	_	_	_	_	_	_	_	_	_	_	_	_	_	_	_	_	-	_	-	_	_	_	_	_	_	-	_	_	-	_	_	_	-	_	-	-	-	-	_
		Date		1004/46	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/24/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96

			Tetryl	(µg/L)				BQL				BQL				BQL	4.6	4.6	4.6	3.6	4.5	9.0				BQL			O	2			BQL				BQL									
		Nitro-	benzene	(µg/L) (µg/L)				BQL				BQL				BQL	HQL	BQL	BQL	BQL	BQL				BQL		-	EO.	ý			BQL				BQL										
			HMX	(µg/L)				-				0.7				BQL	oc oc	4.7	5.3	ų	8.8	4.5				2.3			-	1			8.0				9.0									
		4-Nitro-	toluene	(hg/L)				BQL				BQL				BQL				BQL			BOI	ì			BQL				BQL															
		3-Nitro- 4-Amino-2,6- 4-Nitro-	toluene dinitrotoluene	(hg/L)				BQL				BQL				BQL				BQL			BOI	ì			BQL				BQL															
		3-Nitro- 4	toluene d	(µg/L)				BQL				BQL				BQL	BQL	BQL	BQL	BQL	BQL	HQL	BQL	BQL	BQL	BQL	BQL	BQL				BQL			ROI	ł			BQL				BQL			
		2-Nitro	i toluene	(µg/L)				BQL				BQL				BQL	BOL	BQL				BQL			BOI	ì			BQL				BQL													
		1,3-Dinitro 2,4-Dinitro 2,6-Dinitro 2-Amino 4,6 2-Nitro	toluene dinitratoluene toluene	(µg/L)				BQL				BQL				BQL	52.9	48.7	49.3	48.7	46.4	BQL				BQL			BOI	í			BQL				BQL									
		,6-Dinitro	toluene	(µg/L)				BQL				BQL				BQL	BOL	BQL	BOL	BQL	BQL				BQL			BOI	ļ			BQL				BQL										
		2,4-Dinitro-2	toluene	(µg/L)				BQL				BQL				BQL	8,8	8.4	9.8	80.86	8.2	BQL				BQL			BOI	í			BQL				BQL									
		.3-Dinitro	benzene	(µg/L)				BQL				BQL				BQL	BQL	BQL	BQL	BOL	BQL				BQL			HOL	ì			BQL				BQL										
			RDX Nitrobodie: Nitrate	(µg/L) (mg/LN				1.46				1.7				19.1	1.67	BQL	1.46	1.86	1.37	1.34	0.658	0.678	0.524	0.47	0.584	0.945				1.06			13	!			1.21				1.28			
		Total	Vitrobod	(hg/L)				13.5				9				6:1	2	2.3	2.5	0.3	0.3	BQL	722	773	768	730	728	133				73.1			38				9.11				5.2			
			RDX	(Hg/L)				BQL				BQL				BQL	22.5	20.9	22	20.7	20.9	9.9				8.1			9.0				BQL				BQL									
			TNT TNB	(µg/L) (µg/L)				6.11				5.2				1.9	2	2.3	2.5	0.3	0.3	BQL	339	362	359	344	341	130				95			3.0				10.2				4.6			
			TX	(hg/L)				9.0				0.1				BQL	285	324	319	298	301	57.3				13			3.6				9.0				BQL									
	Oxidation Contactor	_		(mg/L)		25		36		26		27		27		27		27										27		27		26	;	97	27	i	36		56		26		27		27	
		Reduction	-	(mV)		935		933		931		893		899		860		881		283			331		308			312		284		609		Suga	088		845		902		812		664		545	
	Temp.	of ORP	Sample	္		15		15		15		53		15		14		91		13			14		15			9		9		15		2	4		91		7		11		91		<u>×</u>	
			la H	<u>_</u>		7.4		7.5		7.5		7.6		9.7		7.7		1.1		7.8			6.9		7.0			7.1		7.2	_	8.9			73		7.4	_	7.4	_	7.5		9.7		7.7	_
		Ozone	_	(mg/L)	0.8	9.0	0.7	9.0	8.0	9.0	9.0	0.4	0.5			0.5		0.5	9.0	0.0														7.0			0.2	0.3	0.4	0.4	0.2	0.3				0
	Operations	Sample	Time		11:26	13:53	15:06	09:58	11:23	13:43	15:03	00:60	11:21	13:38	15:01	08:48	11:18	13:27	14:58	08:34			10:30		14:33			10:02	11:40	14:17	15:20	09:44	11:37	14:07	92-00	1:34	13:58	15:11	09:28	11:31	13:49	15:05	09:07	11:28	13:28	15:02
		Peroxide PEROXONE Sample Sample	Location Time		C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	C5/0	C5/0	C5/0	C5/0	Ce/0	C6/0	C6/0	C6/0	GAC3	GACI	GAC2	INFI	INFI	E	INFI	INFI	C1/0	C1/0	C1/0	C1/0	C2/0	22.0	C 200	030	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	C5/0	C5/0	C\$/0	C\$/0
	Average	EROXO	Ratio		0.63	0.63	0.63	0.63	0.63	0.63	0.63	0.63	0,63	0.63	0.63	0.63	0.63	6,63	0.63	0.63	0.63	0.63	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	95.0	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64	0.64
Average	Applied TransferredHydroger Average	Peroxide	Dose	(mg/L)	26.9	56.9	26.9	26.9	26.9	26.9	56.9	26.9	26.9	56.9	56.9	56.9	26.9	26.9	26.9	26.9	26.9	56.9	27.7	27.7	27.7	27.7	27.7	27.7	27.7	27.7	27.7	27.7	1.12	21.1	27.7	27.7	27.7	27.7	27.7	27.7	27.7	27.7	27.7	27.7	27.7	7.72
Average Average Average	Fransferre	Ozone	Dose	(mg/L)	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	42.8	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	45.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5	43.5
verage	\pplicd \	Ozone	Dose	(mg/L)	57	57	57	27	57	57	22	23	23	23	23	23	23	23	57	23	57	23	23	23	23	22	23	57	23	23	22	57	57	3 2	5 5	57	57	57	23	23	27	23	23	23	23	27
*			9	(gpm) (24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5
		_	Well Fi		_	_	_	_	-	_	_	_	-	-	-	-	_	_	_	-	-	_	_	_	-	-	-	-	-	_	-	-	_				-	_	-	_	-	-	-	-	-	-
			Date		10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/25/96	10/26/96	10/26/96	10/26/96	10/26/96	10/26/96	10/26/96	10/26/96	10/26/96	10/26/96	96/92/01	10/26/96	10/26/96	1005/06	10/26/96	10/26/96	10/26/96	10/26/96	10/26/96	96/97/01	10/26/96	10/26/96	10/26/96	10/26/96	10/26/96

			Tetry	(µg/L)	BQL	BQL	BQL	BQL	BQL	3.7	3.2	4.2	3.3	3.9	BQL				BQL			BQL	BQL	BQL	BQL	BQL	3.4	3,4	3	3	3.8	0.3														
		Nitro-	enzene	(hg/L)	BQL				BQL			BQL	BQL	BQL	BQL																															
			XWI :	(µg/L)	0.3	0.4	0.5	6.4	BQL	8.	3,4	5.4	5.2	4.5	4.6				2.3				Ξ				0.7				0.4			BQL	BQL	BQL	BQL	BQL	9.6	5.3	3.4	4.3	4.7	4.6		
		4-Nitro	toluene	(µg/L) (BQL				BQL			BQL	BQL	BQL	BQL																															
		3-Nitro- 4-Amino-2,6- 4-Nitro-	toluene dinificatoluene toluene HMX benzene Tetryl	(hg/L)	BQL				BQL			BQL	BQL	BQL	BQL																															
		-Nitro- 4-7	oluene din	(µg/L)	BQL				BQL				BQL				BQL.				BQL			BQL	BQL	BQL	BQL																			
				(hg/L)	BQL	вог	BQL				BQL			BQL	BQL	BQL	BQL																													
		1,3-Dinitro 2,4-Dinitro 2,6-Dinitro 2-Amino 4,6 2-Nitro	ĭ	(hg/L)	BQL	BQL	BQL	BQL	BQL	49.9	40	56.3	43.8	48.5	BQL				BQL			BQL	BQL	BQL	BQL	BQL	44.6	48.3	45.7	45.8	44	BQL														
		Dinitro 2-A	tucne dini	(µg/L)	BQL				BQL			BQL	BQL	BQL	BQL																															
		Dinitro-2,6-		(μg/L) (i	BQL	BQL	BQL	BQL	BQL	8.3	7.4	8.4	7.9	7	BQL				BQL			BQL	BQL	BQL	BQL	BQL	8.1	8.2	8.8 8.8	8.7	8.3	BQL														
		Jinitro-2,4-		(hg/L) (t		BQL F	BQL I	BQL		BQL	1.5	BQL	BQL	BQL	BQL				BQL			BQL	BQL	BQL	BQL	BQL	=	=	Ξ	Ξ	Ξ	BQL														
		<u></u>				1.27 B	1.36 B	1.44 E	1.33 E	0.649 E	0.712	0.732 E	0.766 E	0.804 E	1.07				1.17				2.57				1.32				1.51			1.46		1.53			0.645	0.723	0.719	0.755	0.734	1.01		
		Total	KDX Nitrobodie: Nitrate	(µg/L) (µg/L) (µg/L) (mg/L N	2	5.6	2.7	2.7	BQL	199	(11)	785 (059		221				74.5				22.4				9.3				3.5			1.4	Ξ	6.1	1.4	BQL	169	402	634	289	620	238		
		2	E Z	IB/L)	BQL	BQL	BQL	BOL	BQL	20.2	18.	21.7	20.1	17.8	9.8				6.1				0.3				BQL				BQL			BQL	BQL	BQL	BQL	BQL	19.2	19.3	20.4	21.9	20.2	7.5		
			9 3	Hg/L)	1.7	2.2	2.2	2.3	BQL	312	292	370	307	293	142				57.2				6.81				8.2				3.1			4.	Ξ	1.7	4.	BQL	336	341	298	327	162	157		
			INI INE	HB/L)	BQL	BQL	BQL	BQL	BQL	262	251	319	263	249	65.6				13.1				2.1				9,4				BQL			BQL	BQL	0.2	BQL	BQL	273	282	254	275	247	1.69		
	Contactor		Peroxide	(mg/L)	27		27								27		27		24		25		56		25		24		56		56	è	97	25		56								29	;	56
	Oxidation Contactor	Reduction Measured	=	(MV)	773		790		260	296		309			291		286		297		343		535		539		642		544		516	;	÷16	505		460		265	415		415			414		9
	Temp. (of ORP		5	11		11		15	12		13			12		4		13		15		12		15		=		9		=	2	2	01		15		01	13		7			2		17
	-	7	Ē		7.7		7.9		7.7	8.9		7.0			6.7		7.1		7.1		7.3		7.2		7.3		7.2		7.4		7.4	ì	9.	7.3		7.7		7.7	9.9		6.9			6.9		7.0
		Охопе	Kesidual	(mg/L)	0.3	0.3	0.2	0.3	0.0						0.0	0.1	0.0	0.1	0.0	0.1	0.0	0.5	0.1	0.3	0.2	9.0	0.2	0.2	0.2	0.4	0.1	- -	0.7	0.1	0.1	0.1	0.1	0.0						0.2	0.3	0.2
	Operations	Sample			08:31	11:24	13:16	14:57	08:15	10:32		16:33			10:24	14:54	16:23	17:00	09:45	14:52	16:04	16:58	09:32	14:49	15:56	16:56	09:23	14:46	15:35	16:53	91:60	14:43	16:50	00:60	14:34	15:08	16:48	08:50	11:04		15:46			10:33	12:09	15:16
	0	Sample	Location		0/9O	C4/0	C4/0	C6/0	GAC3	INFI	INFI	INFI	INF	INFI	C1/0	C1/0	C1/0	CI/0	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	C2/0	C5/0	3 3	C6/0	C6/0	C6/0	C6/0	GAC3	INF	INF	INFI	INF	F	0/10	C1/0	CIV
	Average	ROXONE	Kallo		0.64	0.64	9.64	0.64	0.64	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.63	0.63	0.63	69.0	6,63	0.63	0.63	69.0
Average	Applied TransferredHydroger Average	Peroxide PEROXONE Sample Sample	Dose	(mg/L)	7.72	7.72	1.72	27.7	27.7	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	24.3	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2
Average Average Average	ransferred		Dose	(mg/L)	43.5	43.5	43.5	43.5	43.5	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	42.2	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3
Verage	Applied 1	Ozone	Dose	(mg/L)	57	57	23	57	57	55	55	55	55	55	55	25	22	55	22	55	22	55	55	55	25	55	55	55	55	55	22	\$;	8 %	\$\$	\$\$	55	55	55	99	26	99	99	36	99	26	26
		Process	**	(mdg)	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5
			Well FI		-	_	_	_	-	_	-	_	_	_	_	_	-	_	_	_	-	-	_	-	_	_	-	-	-	-	-			-	-	_	-		-		-	-	-	_	_	-
			Date		96/92/01	10/26/96	10/26/96	10/26/96	10/26/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	96/12/01	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/27/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96

			Tetryl	(µg/L)		BOL	i,			BQL				BQL				BQL				BQL	BQL	BQL	BQL	BQL	3.9	3.1	3.4	3.3	3.7	0.3			03	1			BQL				BQL			
		Nitro-	benzene	(µg/L)		BOL	ļ			BOL				BQL				BQL				BQL	BQL	BQL	BOL	BQL	BOL	BQL	BQL	BQL	BQL	BQL		-	BOI))			BQL				BQL			
			HMX	(µg/L)		2.3				1.3				-				0.4				BQL	BQL	BQL	BQL	BQL	6.7	4	4.9	3.6	3.4	4.7			2.5	ì			-				8.0			
		4-Nitro-	toluene	(µg/L)	ı	BOL	ļ			BQL				BQL				BQL				BQL			BOIL	y Y			BQL				BQL													
		3-Nitro- 4-Amino-2,6- 4-Nitro-	dinitrotolucne	(µg/L)		BOL	ļ			BQL				BQL				BQL				BQL			BO	r y			BQL				BQL													
		3-Nitro- 4	toluene	(µg/L)		BOL	,			BQL				BQL				BQL				BQL			BOL	i			BQL				BQL													
			tofuene	(µg/L)		BOL	,			BQL				BQL				BQL				BQL	BOL	BQL	BQL			ROIL	1			BQL				BQL										
		1,3-Dinitro-2,4-Dinitro-2,6-Dinitro-2-Amino-4,6-2-Nitro-	toluene dinitratoluene toluene	(µg/L)		BOL	,			BQL				BQL				BQL				BQL	BQL	BQL	BQL	BQL	60.5	49.2	43	51.1	46.5	BQL			BOL	ļ			BQL				BQL			
		6-Dinitro-2	toluene d	(µg/L)		BOL	,			BQL				BQL				BQL				BQL			BOL	ì			BQL				BQL													
		,4-Dinitro.2,	tolucne	(µg/L)	-	BOL	,			BQL				BQL				BQL				BQL	BQL	BQL	BQL	BQL	10	8.8	8.3	8.5	8.4	BQL			BOL	1			BQL				BQL			
		3-Dinitro-2	benzene	(µg/L)		BOL	,			BQL				BQL				BQL				BQL	BQL	BQL	BQL	BQL	1.4	1.2	Ξ	Ξ	-	BQL			BOI.	1			BQL				BQL			
		_		mg/L N		1.22				1.34				1.35				1.31				1.06	1.26	1.28	1.7	1.54	0.37	269.0	0.649	0.658	99'0	0.911			1 02				1.09				1.25			
		Total	RDX Nitrohodie: Nitrate	(µg/L) (mg/L N		77.9				27.8				15.9				9.8				9.1	9.	4.2	2.3	BQL	970	728	702	739	736	264			83.4				25.7				14.9			
			RDX Ni			6.1				6.4				BQL				BQL				BQL	BQL	BQL	BQL	BQL	25.7	18.6	20.6	19.7	21.5	эc			2				0.4				BQL			
			TNB	ug/L)		60.3				23.6				14.2				oc				9.	9.1	4.2	2.3	BQL	450	348	335	358	354	175			64.7				22.1				13.4			
			TNT	(µg/L) (µg/L) (µg/L)		3,4				2.5				0.7				0.2				BQL	BQL	BQL	BQL	BQL	412	295	286	294	297	75.5			14	:			2.2				0.7			
	Contactor	Acasured	Peruxide	(mg/L) (26		26		56		26		32		28		27		27		27		27								25	i	67	25	ì	25		25		56		25		56	
	Oxidation Contactor	Reduction Measured	Potential	(mV)		920		930		943		939		935		922		888		106		890		922		410	370		406			385		966	606		874		933		920		942		878	
	ııb.	ORP	Sample	(°C)		13		15		13		15		23		91		13		9		0		9		13	13		4			91	:	5	14		4		3		15		13		2	
		jo	된			7.0		7.2		7.1		7.4		7.2		7.5		7.2		9.7		7.5		1.7		1.7	6.9		7.2			7.0		57	7.2	!	7.4		7.3		9.7		7.4		7.7	
		Ozone	Residual	(mg/L)	,	0.5	9.0	0.3	0.7	0.7	8.0	9.0	9.0	6.0	1.7	0.3	9.5	0.5	9.0	0.4	0.4	0.4	9.0	9.0	9.0	0.0						0.2	0.7	7.0	0.4	0.5	0.3	0.4	9.0	0.7	0.4	0.7	0.5	0.5	0.4	i
	Operations	ample			16.13	10:25	12:06	15:08	16:11	10:20	12:02	15:02	60:91	10:11	11:54	14:56	16:07	10:05	11:51	14:49	16:05	08:36	11:46	14:27	16:03	08:20	10:00		14:55			09:34	01:54	17:03	09:22	10:51	13:57	16:59	09:12	10:46	13:43	16:54	\$0:60	10:44	13:35	
	Ö	Peroxide PEROXONE Sample Sample	Location Time		9	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	C5/0	C5/0	C5/0	C5/0	C6/0	C6/0	C6/0	C6/0	GAC3	INFI	INF	INF	INF.	INFI	C1/0	213	2 2	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	;
	Average	ROXONE	Ratio		670	0.63	0.63	6.63	6,63	69.0	6.63	69'0	69.0	69.0	0.63	0.63	0.63	0.63	0.63	0.63	0.63	0.63	69.0	0.63	69.0	69.0	0.62	0.62	0.62	0.62	0.62	0.62	0.62	79'O	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	1
verage	Applied TransferredHydroger Average	eroxidePE	Dose	(mg/L)	17.7	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	27.2	26.9	26.9	26.9	26.9	56.9	26.9	56.9	56.9 26.9	26.9	26.9	26.9	26.9	26.9	26.9	26.9	56.9	26.9	56.9	26.9	
rerage A	nsferredH	Ozone P	Dosc	(mg/L) (13.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.3	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	43.1	
Average Average Average	plied Tran	Ozone O	Dose	(mg/L) (r				26	, 95	, 98	. 95	99		. 98	26		26	. 99	98	99	95	98	99	99		98	57	57	57	57	57	57	27	<i>i</i>		57	57	57	57	57	57	57	23	57	57	Š
Avi	Αb																	24.5	24.5	24.5			24.5	24.5		24.5	24.5	24.5	24.5	24.5	24.5			24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	3
		Process	Well Flow Rate	(gpin)	346	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24.5	24	24	24	24.5	24.5	24	24	24	24	24	24	24	24	24	7,	2,	ž ž	, 7	. 8	24	24	2,	24	24	2,	2,	2	2 2	
					7,1	1 96	1 96	1 96/	1 96/	1 96/	1 96/	1 96/	1 96/	1 96/	1 96/	96/	196/	1 96/	1,96	1 96/	1 96/	1 96/	96/	1,96/	1 96/	1/96	96/	96/	1 96/	1 96/	1 96/4	96/0	96/6	96/	1 90%	1 96/	96/4	1/96/1	1,96	1,96	1 96/6	96/6	96/0	96/6	96/6	11.71
			Date		21100001	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/28/96	10/29/96	10/29/96	10/29/96	10/29/96	10/29/96	10/29/96	10/29/96	10/29/96	96/6/01	10/29/96	96/62/01	10/29/96	10/29/96	96/62/01	10/29/96	10/29/96	10/29/96	10/29/96	10/29/96	1

Participal Participa		Nira-	HMX benzene Tetryl	(µg/L) (µg/L)	BQL BQL				BQL BQL		BOL BOL					BQL 3		BQL 2.9	BQL 3.3	BQL 3.9		BQL BQL	BQL BQL	BQL BQL	BQL BQL	BQL BQL	BQL BQL	BQL BQL	BQL BQL	_			BQL 4.5		_		BQL BQL		BQL BQL		BQL BQL		BQL BQL		
Participal Participa		Ž	MX ber	g/L) (µ																																									
Partial Part		itro																																											
Partial Part		no-2.6- 4-N	toluene tol																																										
Partial Part		ro- 4-Ami	ne dinitro																																										
Motion M																																													
Moring M		6. 2-Nitr	nc toluc	Λgμ)	BQI				BQ	BQI	BOI	. <u>B</u>	. <u>0</u>	80	BQ	BQ	BQ	BÖ	BÖ	BQ	BQ	BQ	BQ	BQ	BQ	BQ	BQ	BQ	80	BQ	B	OB 1	2 2	2 02	BQ		βŐ		80		BQ		BQ		
Moring M		2-Amino-4	dinitrotoluc	(µg/L)	BQL				BQL	BQL	BOL	BOL	BQL	BQL	BQL	41.7	30	43.8	36.9	42.4	BQL	BQL	BQL	BQL	BQL	BOL	BQL	BQL	BQL	BQL	45.6	53	7.90.7	49.7	BQL		BQL		BQL		BQL		BQL		
Marine M		2.6-Dinitro	toluene	(µg/L)	BQL				BQL	BQL	BOL	BOL	BQL	BQL	BOL	BQL	BOL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BOL BOL	BOL	BQL		BQL		BQL		BQL		BQL		
Marinary Marinary		.4-Dinitro	toluene	(µg/L)	BQL				BQL	BQL	BOL	BOL	BQL	BQL	BQL	7.6	1.7	∞.	7.8	8.9	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	7.5	9.5	10.1	8.7	BQL		BQL		BQL		BQL		BQL		
Marinary Marinary		3-Dinitro-7	benzene	(µg/L)	BQL				BQL	BQL	BOL	BOL	BQL	BQL	BQL	-	Ξ	_	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	-	Ξ:	4. 0	<u>:</u>	BQL		BQL		BQL		BQL		BQL		
Average Av				(mg/L N	1.28				1.3	1.24	<u>~</u>	1.34	1.44	1.24	1.46	0.658	3.85	0.672	0.646	0.689	0.704	0.73	1.12	1.02	1.28	1.37	1.25	1.27	1.23	1.22	89.0	0.545	65.0	0.531	9.676		0.822		1.05		1.05		0.992		
Process		Total	trobodie	(hg/L)	5.9				2	2	2.2	2.3	BQL	BQL	BQL	706	645	644	689	773	217	93.8	31.2	12.4	5.1	9.1	2.5	3.3	2	BQL	669	714	2/4	702	661		06		37.2		10.1		4.5		
Process			RDX Ni	(µg/L)	BQL				BQL	BQL	BQL	BOL	BQL	BQL	BQL	17.5	8.91	19.7	18.2	19.2	6.7	6.1	9'0	BQL 21.2	22.4	70.7	22	9'9		2		9.0		BQL		BQL									
Process				(hg/L)	8.8				2	2	2.2	2.3	BQL	BQL	BQL	342	315	307	326	374	146	76.2	26.9	11.5	5.1	1.6	2.5	3.3	2	BQL	330	312	230	33	133		71.9		31.4		7.6		4.4		
Average A			TNT	(hg/L) (0.1				BQL	BQL	BOL	BQL	BQL	BQL	BQL	289	268	255	292	319	99	13.4	2.6	0.5	BQL	BQL	BQL	BQL	BQL	BQL	284	306	200	281	54.9		13.5		3.6		0.4		0.1		
Average A		feasured	croxide	(mg/L)	26		56		36		56										27	24	25	30	25	23									24	25	24	24	24	24	24	23	24	24	
Average Average Average Average (Applied TransferredHydloga: Average Applied TransferredHydloga: Average Average Applied TransferredHydloga: Average Average Average Applied TransferredHydloga: Average Averag	-	eduction N	otential		884		825		865		826		299			339					316	156	930	904	915	973				300	435				6\$\$		930		949		937		922		
Average Aver		ORP R	mple P	(C)	13		15		4		91		2			=					=	10	01	6	6	6				1	4				15		4		4		4		7		
Average Average Average	ŧ	- 5			7.5		7.8		7.5		6.7		1.7			7.3					7.5	7.5	7.6	7.8	7.9	4.9					7.0				7.1		7.2		7.3		7.5		9.7		
Average Average <t< td=""><td></td><td>Ozone</td><td>csidual</td><td>(mg/L)</td><td>0.2</td><td>0.4</td><td>0.2</td><td>0.4</td><td>0.3</td><td>4.0</td><td>0.2</td><td>0.5</td><td>0.0</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>0.2</td><td>8.0</td><td>9.0</td><td>0.4</td><td>0.5</td><td>3.8</td><td></td><td>6.0</td><td></td><td>0.0</td><td></td><td></td><td></td><td></td><td>0.1</td><td>0.3</td><td>9.0</td><td>8.0</td><td>8.0</td><td>Ξ</td><td>0.7</td><td>1.2</td><td>9.0</td><td>Ξ</td><td></td></t<>		Ozone	csidual	(mg/L)	0.2	0.4	0.2	0.4	0.3	4.0	0.2	0.5	0.0								0.2	8.0	9.0	0.4	0.5	3.8		6.0		0.0					0.1	0.3	9.0	8.0	8.0	Ξ	0.7	1.2	9.0	Ξ	
Average Average Average Average Average Average Average Average Average Average Applied TransferredHydroger Average Applied TransferredHydroger Average Average <th< td=""><td></td><td></td><td>Time R</td><td></td><td>38:55</td><td>10:41</td><td>13:24</td><td>16:46</td><td>98:36</td><td>10:39</td><td>13:15</td><td>16:41</td><td>98:25</td><td></td><td></td><td>99:05</td><td></td><td></td><td></td><td></td><td>08:30</td><td>08:21</td><td>08:14</td><td>80:80</td><td>07:58</td><td>07:47</td><td></td><td>91:60</td><td></td><td>07:40</td><td>15:24</td><td></td><td></td><td></td><td>15:12</td><td>16:17</td><td>14:56</td><td>16:08</td><td>14:51</td><td>80:91</td><td>14:51</td><td>16:33</td><td>14:45</td><td>16:28</td><td></td></th<>			Time R		38:55	10:41	13:24	16:46	98:36	10:39	13:15	16:41	98:25			99:05					08:30	08:21	08:14	80:80	07:58	07:47		91:60		07:40	15:24				15:12	16:17	14:56	16:08	14:51	80:91	14:51	16:33	14:45	16:28	
Well Flow Raine Average Average Well Flow Raine Ozome Ozome (ghm) (mg/L) (mg/L) 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 41.6 1 24.5 56 41.6 1 24.5 56 41.6 1 24.5 56 41.6 1 24.5 56 41.6 <th< td=""><td></td><td>Op Sanvole S</td><td>ocation</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>GACI</td><td>GAC2</td><td></td><td>INFI</td><td>INFI</td><td>INFI</td><td>INF</td><td></td><td></td><td></td><td></td><td></td><td></td><td>C6/0</td><td></td><td></td><td></td><td></td><td>N.</td><td>Z Z</td><td>INF</td><td>C1/0</td><td>C1/0</td><td>C2/0</td><td>C2/0</td><td>C3/0</td><td>C3/0</td><td>C4/0</td><td>C4/0</td><td>C5/0</td><td>C5/0</td><td></td></th<>		Op Sanvole S	ocation											GACI	GAC2		INFI	INFI	INFI	INF							C6/0					N.	Z Z	INF	C1/0	C1/0	C2/0	C2/0	C3/0	C3/0	C4/0	C4/0	C5/0	C5/0	
Montage Average Applied Transferrer Applied Transferrer (glmn) (mgL) (mgL) 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 57 43.1 1 24.5 56 41.6 1 24.5 56 41.6 1 24.5 56 41.6 1 24.5 <		OXONE	Ratio L		0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.62	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.57	0.51	0.51	15.0	0.51	0.51	0.51	0.51	0.51	0.51	0.51	0.51	0.51	0.51	0.51	
Average Average Applied Transferred (gpm) (mgL) (m	rerage	oroger A coxide PER	Juse	ng/L)														23.9		23.9	23.9	23.9	23.9	23.9	23.9	23.9	23.9	23.9	23.9	23.9	23.6	23.6	23.0	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	23.6	
Process Well Flow Rate (gpm) 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.0 1 24.0	rerage Av	zone Per		- 1																																					46.0		46.0	46.0	
Process Well Flow Rate (gpm) 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.5 1 24.0 1 24.0	rage A	price trai		- 1																																							09	09	
	Avc			- 1																																									
		Proce	I Flow	(gp)	24.	24.	24.	24	24	24.	24.	24.	24.	24	24.	24	24	24.	24.	24.	24.	24.	24.	24.	24.	24.	24.	24.	24.	24.	24	24	3 2	24.	24.	24.	24.	24.	24	24	24	24	24	24	
					- 2	1 90	1 90	1 9	1 94	1 9	1 9	1 9	1 90	1 9	16 1	1 90	1 90	96	1 90	1 96	1 96	96	1 96	1 96	1 96	1 96	1 96	1 96	1 96	1 96	1 9			ءِ :	1 9	1 90	1 90	- 9	1 9	1 9	1 90	1 9	94	16 1	

		Tetryl	(µg/L)	BOL	BQL	BQL	3.7	٣	3.3	3.8	3.2	BQL				BQL			ROI	BQL	BQL	BQL	BQL	BQL	BQL	3.9	4.3	3.6	3.5	4.2	BQL														
	Nifro-	anzene	(µg/L) (µg/L) (µg/L)	BOL	BQL				BQL			ROI	BOL	BQL																															
	-	4MX b	ug/L) (BQL		BQL	5.4	4.7	~	œ. 7	S	3.1				6:				Ξ				0.7				BQL			S	BQL	BQL	BQL	BQL	BQL	BQL	9.6	6.5	7.9	9	5.1	4.2		
	4-Nitro-	toluene	(µg/L) (BOL			BQL	BQL	BQL	BQL	BQL	BQL				BQL			BOI	BQL																									
	3-Nitro- 4-Amino-2,6- 4-Nitro-	toluene dinitrotoluene toluene HMX benzene Tetryl	(µg/L)	BQL				BQL			ROI	BOL	BQL																																
	Nitro- 4-7	stuene din	(µg/L)	BQL				BQL			POI	BQL																																	
			(µg/L) (BQL				BQL			EQ.	BOL	BQL																																
	mino-4.6- 2	rotoluenc	(µg/L)	BQL	BQL	BQL	48.2	46.3	47.1	46.5	48.8	BQL				BQL			BOL	BOL	BQL	BQL	BQL	BQL	BQL	43	44.7	48.3	43.9	46	BQL														
	1.3-Dinitro 2,4-Dinitro 2,6-Dinitro 2-Amino -4,6- 2-Nitro-	toluene dinitrotoluene toluene	(µg/L)	BQL				BQL			ROL	BOL	BQL																																
	itro-2,6-	nc tu	- 1																	교.				7.				BQL			POI	BQL	BQL	BQL	BQL	BQL	BQL	Ξ.	11.2	8.3	10.7	7.7	7		
	0-2,4-Dir	toluene	(µg/L)	BOL	BQL	BQL	9.4	8.2	8.6	7.5	7.3	BQL				æ			R	Œ	B	B	æ	B	ĕ	=	=	œ	2	7	ĕ														
	3-Dinite	benzene	(hg/L)	BOL	BQL	BQL	6.0	Ξ	Ξ	8.0	<u></u>	BQL				BQL				BQL				BQL				BQL			ROI	BOL	BQL	BQL	BOL	BOL	BQL	1.3	1.3	Ξ	1.2	0.0	BOL		
	_	RDX Nitrobodie: Nitrate benzene	(µg/L) (mg/LN	1.05	1.02	1.3	69.0	0.578	0.687	0.759	0.71	1.02				1.08				1.07				1.07				Ξ			9000	1.16	1.34	1.23	1.78	1.16	1.37	1.05	1.05	-	-	0.987	1.22		
	Total	trobodic	(µg/L) (2	2	BQL	577	824	836	535	898	205				74				29				12.2				5.4			~	2.2	6:1	2.1	BQL	BQL	BQL	655	089	969	637	682	220		
		RDX Ni		BOL	BQL	BQL	21.1	5'61	20.9	19.7	9.61	20				œ:				0.4				BQL				BQL			0	BOL	BQL	BQL	BQL	BQL	BQL	19.2	8.61	20	19,3	19.6	7.3		
) (J/8t		2	BQL	270	387	391	254	268	901				98.8				25				=				5.3			×	2.2	1.9	2.1	BQL	BQL	BQL	305	314	317	294	314	145		
		TNT TNB) (7/8rl	BOL	BQL	BQL	218	354	329	861	215	45.6				11.7				2.5				0.5				0.1			2	BOL	BQL	BQL	BQL	BQL	BQL	266	278	290	258	284	63.7		
	Measured		(mg/L) (µg/L) (µg/L)									23		23		24		24		54		24		25		25		24	;	54	35	ì	24										27	;	23
	temp. Oxidation Comactor of ORP Reduction Measured	otential	(mV)			275	436		422			813		968		950		957		957		952		952		936		949		930	100		893		243			429		431			416	į	211
	of ORP R	Sample	(၁၀)			4	=		7			=		Z		=		4		=		15		=		2		9	;	2	0		15		oc			13		14			15	:	9
1	- 3	PH S				7.4	6.9		7.0			7.0		7.1		7.1		7.3		7.3		7.4		7.4		7.5		7.5		7.6	7.5	2	9.7		7.8			7.2		7.0			7.3	ì	7.1
	Охипс	Residual	(mg/L)			0.0						0.4	0.7	9.4	0.4	0.7	6.0	0.7	6.0	0.1	1.2	9.0	8.0	0.1	6.0	0.7	8.0	0.	= :	0.7	0.7	0.8	9.0	0.5	0.0								0.2	0.3	0.3
		Fine -				14:06	09:32		15:26			61:60	12:11	15:19	17:12	09:02	12:07	15:10	17:08	93:05	12:07	15:05	10:21	09:11	11:59	14:53	16:49	08:43	11:52	14:42	16:40	11:46	14:34	16:30	08:16			09:23		14:31			08:59	11.37	4:13
d	IHYGENGEL AVERAGE OPERATIONS PEROXIGE PEROXONE Sample Sample	Location Time		C6/0	C6/0	GAC3	INFI	INFI	INF	INFI	INF	C1/0	C1/0	C1/0			C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	C2/0	C5/0	C2/0	0.570	C6/0	C6/0	C6/0	GAC3	GACI	GAC2	INFI	INFI	INF	E	INFI	0/10	010	0170
	Nortage ROXONE	Ratio		0.51	0.51	0.51	0.56	0.56	95.0	95.0	95'0	95.0	0.56	95.0	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	0.56	95.0	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58
verage	Applied Transferred Hydroger Average Ozone Ozone Peroxide PEROXON	Dosc	(mg/L)	23.6	23.6	23.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	25.6	26.8	8.92	26.8	26.8	26.8	26.8	26.8	26.8
Average Average Average	Ozone P		(mg/L)	46.0	46.0	46.0	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9
rerage A	Applied 112 Ozune ((mg/L) (09	9	09	09	09	09	09	09	09	09	09	9	60	09	09	09	09	09	9	99	09	9	09	09	99	99	90	E 8	8 8	09	9	09	90	09	09	09	9	09	09	99	99	9
ě.	Princess O		(mdg)		24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0
	Pro	ell Flow	E)	24	2	2,	1 2	1 2	1 2	1 2	2.	1 2.	2	1 2	1 2	1 2	1 2	1 2	2	1 2	1 2	- 2	1 2	1 2	- 2	- 2	1	- 2	1 2	- 2		, ,	1 2	- 7	1 2	1 3	- 2	-	_	-	_	-	_	_	_
		Date We		96/	11/4/96	11/4/96	11/2/96	11/5/96	11/5/96	96/5/11	96/5/11	96/5/11	96/5/11	96/5/11	96/5/11	11/5/96	96/5/11	11/5/96	96/5/11	96/5/11	11/5/96	96/5/11	11/5/96	96/5/11	96/5/11	11/5/96	11/5/96	11/5/96	96/5/11	11/5/96	11/5/96	96/5/11	11/5/96	11/5/96	96/5/11	96/5/11	11/5/96	96/9/11	96/9/11	96/9/11	11/6/96	96/9/11	96/9/11	96/9/11	96/9/11
		۵		11/4/96	=	*	11/2	11/2	11/2	11/2	17.	11/	117	117	=	È	=	=	Ξ	11	11/	Ξ	Ξ	=	ì	È	Ê	Ξ	Ê	Ē	₫ 3	= =	=	Ė	Ξ	Ė	=	Ξ	Ξ	=	=	Ξ	=	=	=

PEROXONE Plant Demonstration Task Test Conditions and Results Demonstration Phase 2

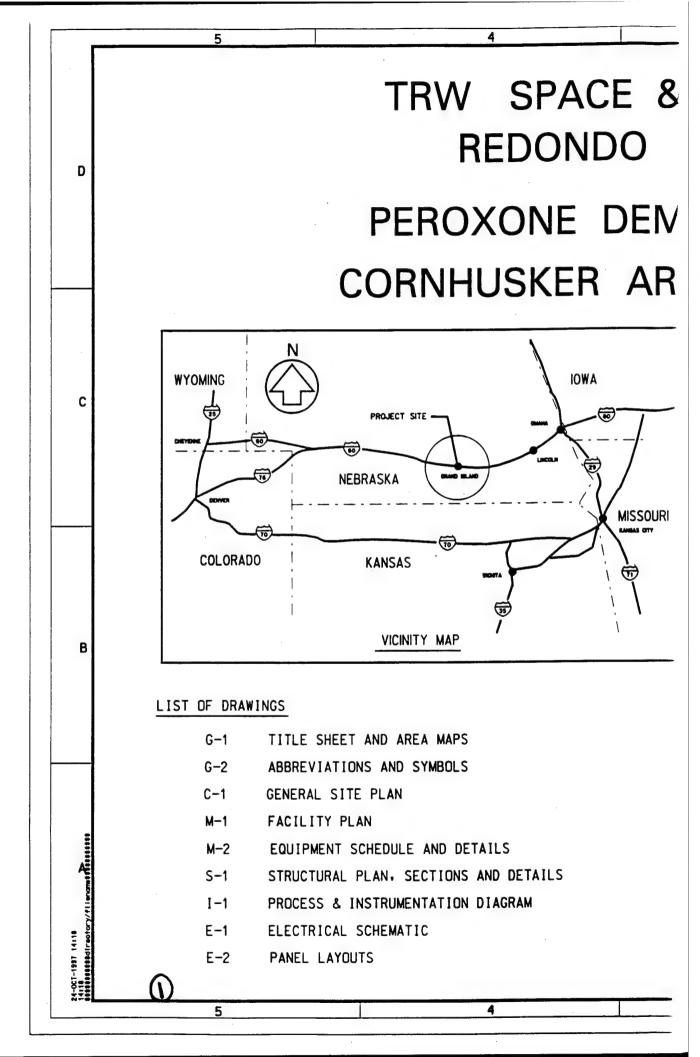
		Tetryl	(hg/L)		BQL				BQL	BQL	BQL	BQL	BQL	3.7	8.4	3.7	2.6	3.5	BQL	Š	D C	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	2.8												
	Nitro-		(hg/L)		BQL				BQL	BQL	BOL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	BQL	č	<u> </u>	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	BQL												
		HWX	(µg/L)		2.1				1.3				BQL				BQL				BQL	BQL	BQL	BQL	BQL	6.3	6.4	2.8	3,3	4	3.6	•	7	Ξ		BQL		BQL		BQL	BOL	BQL	BQL	BQL	5.3
			(HB/L)		BQL				BQL BQL	BQL	BQL	BQL	č	n P	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	BQL																			
	3-Nitro- 4-Amino-2,6- 4-Nitro-	tolucne dinitrotoluene toluene	(hg/L)		BQL				BQL BQL	BQL	BQL	BQL	Č	2	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	BQL																			
	Nitro- 4-A	duene dini	(µg/L)		BQL				BQL BQL	BQL	BQL	BQL	3	a P	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	BQL																			
			(µg/L) (BQL				BQL BQL	BQL	BQL	BQL	3	a CE	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	BQL																			
	0-4,6- 2	lucac to							رت.				⊒.				=				7.	7	J.	7.	BQL	37.6	45.7	40.5	-	37.2	.	,	BQL	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	42.2
	ro-2-Amin	din	(µg/L)		BOL				BQL	BQL	BQL	BQL					36.1		BQL																										
	o-2,6-Dinit		(hg/L)		BOL				BQL BQL	BQL	BQL	BQL		a de la composição de l	BQL		BQL		BQL		BQL	BOL				BQL																			
	1,3-Dinitro 2,4-Dinitro 2,6-Dinitro 2-Amino-4,6- 2-Nitro-		(Hg/L)		BQL				BQL	BQL	BQL	BQL	BQL	10.8	11.5	10.3	10.5	9.11	BQL	Č	BQE.	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	11.3												
	3-Dinitro	henzene	(ng/L)		BQL				BQL	BQL	BQL	BOL	BQL	6.0	1.3	Ξ	1.1	1.2	BQL	Š	PQF	BQL		BQL		BQL		BQL	BQL	BQL	BQL	BQL	6.0												
	-	Nitrate	(mg/L N		1.3				1.37				4.				1.45				1.53	1.56	1.52	1.46	2.09	0.92	0.918	0.942	0.935	0.911	1.14		57.1	1.27		<u>=</u>		1.31		1.5	1.51	1.37	1.67	1.98	0.873
	Total	RDX Nitrobadie Nitrate	(µg/L) (101				32.8				Ξ				4.3				1.7	8.1	2.1	2.4	BQL	582	673	613	583	630	184		74.9	30		10.2		4.5		6.2	2.2	2.2	2	BQL	663
		RDX N	(mg/L)		2.1				9.0				BQL				BQL				BQL	BQL	BQL	BQL	BQL	17.4	20	18.5	<u>«</u>	18.5	5.7	:	ē:	0.5		BQL		BQL		BQL	BQL	BQL	BQL	BQL	17.6
					81.5				27.9				9.01				4.3				1.7	œ.	2.1	2.4	BQL	270	313	288	276	297	122		97.60	25.7		6.7		4.5		6.2	2.2	2.2	2	BQL	314
		TNT	(µg/L) (µg/L)		15.2				3				0.5				BQL				BQL	BQL	BQL	BQL	BQL	236	270	248	235	257	52.4	:	4 .	2.7		0.5		BQL		BQL	BQL	BQL	BQL	BQL	269
ontactor	casured		(mg/L)		27		24		56		56		25		25		25		25		25		26								25	;	Q	25		22		25		25					
Oxidation Contactor		=	(mV)		912		926		925		938		942		930		935		925		886		988		242	427					573		919	945		915		616		735				250	429
Temp. Ox	of ORP Re	63	် ပို့		13		4		13		4		=		15		=		91		=		15		=	4					14	:	<u>+</u>	14		91		4		4				=	13
7	Jo	pH Sa			7.4		7.3		7.5		7.4		7.7		7.5		7.8		9.2		6.7		7.8		8.0	8.9					6.9		9:	7.2		7.4		7.5		9.7				9.7	9.9
	Охопе	_	(mg/L)	0.3	0.5	0.7	0.5	8.0	9.0	0.7	0.5	0.7	0.7	8.0	0.5	8.0	0.7	8.0	0.5	0.7	0.5	0.5	0.4	0.4	0.0						6.5	0.3	200	0.7	6.0	0.4	8.0	9.0	0.7	0.3	0.2			0.0	
Operations	ample (16:26	08:53	11:33	14:06	16:23	08:46	11:30	13:59	16:22	08:37	11:27	13:50	16:20	08:30	11:23	13:42	81:91	08:22	11:20	13:24	16:16	80:80	15:38					15:28	16:28	90:01	15:00	16:22	14:43	16:19	14:37	16:16	14:22	16:13			14:05	08:30
õ	Peroxide PEROXONE Sample Sample	Location Time		C1/0	C2/0	C2/0	C2/0	C2/0	C3/0	C3/0	C3/0	C3/0	C4/0	C4/0	C4/0	C4/0	CS/0	C5/0	C5/0	C5/0	0/9D	C6/0	C6/0	C6/0	GAC3	INFI	INF	INF	INFI	INE	C1/0	CIA	3 8	C3/0	C3/0	C4/0	C4/0	C5/0	C5/0	C6/0	C6/0	C6/0	C6/0	GAC3	IN I
Average	OXONE S	Ratio L		0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.58	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.54	0.55
	kide PER		5		26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	26.8	8.92	8.92	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0
Average Average Average Applied TransferredHydroger			(mg/L)	26.8																																									
e Avera	Ozone		(mg/L)	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	45.9	46.4	46.4	46.4	46.4	46.4	46.4	46.4	40.4	46.4	46.4	46.4	46,4	46.4	46.4	46.4	46.4	46.4	46.4	46.4	45.6
Average Applied	Ozone		(mg/L)	9	9	9	9	9	9	9	99	9	9	99	9	09	9	9	8	3	9	9	9	9	9	09	09	8	9	9	99	9	3 8	. 3	99	9	9	9	9	8	8	9	9	99	99
	Process	Well Flow Rate	(gpm)	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0
		Well F		-	-	_	_	_	_	-	_	-	-	-	-	-	-	_	-	_	_	-	_	-	-	-	-	_	_	-	-	_		. –	-	-	-	-	-	_	-	-	-	-	
		Date		11/6/96	11/6/96	11/6/96	96/9/11	96/9/11	96/9/11	96/9/11	11/6/96	96/9/11	96/9/11	11/6/96	11/6/96	11/6/96	11/6/96	96/9/11	11/6/96	11/6/96	11/6/96	96/9/11	11/6/96	96/9/11	96/9/11	11/7/96	117796	117796	117/96	96/1/11	11/7/96	11/7/96	96/1/11	11/7/96	11/7/96	11/7/96	11/7/96	1177/96	11/7/96	117/96	11/7/96	1177/96	1177/96	1177/96	11/8/96

PEROXONE Plant Demonstration Task Test Conditions and Results Demonstration Phase 2

			Tetryl	(µg/L)	2.8	2.4	2.9	3.6	BQL		BQL														
		Nitro-	benzene Tetryl	(hg/L)	BQL	BQL	BQL	BQL	BQL		BQL	BOL	BQL	BQL	BQL	BOL	BQL								
			HMX	(hg/L)	4.7	4	5.3	5.7	3.6		2.1		1.3		8.0		BQL		BQL	BQL	BQL	BQL	BOL	BQL	BQL
		4-Nitro-	toluene	(hß/L)	BQL	BQL	BQL	BQL	BQL		BQL														
		Amino-2,6-	toluene dinitrotoluene toluene	(µg/L.)	BQL	BQL	BQL	BQL	BQL		BQL														
		-Nitro- 4	oluene di	(µg/L)	BQL	BQL	BQL	BQL	BQL		BQL														
		-Nitro- 3		(µg/L)	BQL	BQL	BQL	BQL	BQL		BQL	BQL	BQL	BQL	BOL	BQL	BQL								
		1,3-Dinitro-2,4-Dinitro-2,6-Dinitro-2-Amino-4,6- 2-Nitro- 3-Nitro- 4-Amino-2,6- 4-Nitro-	toluene dinitrotoluene toluene	(µg/L)	31.3	37.5	39.8	38	BQL		BQL		BOL		BQL		BQL		BQL						
		-Dinitro-2	oluene d	(µg/L)	BQL	BQL	BQL	BQL	BQL		BQL														
		1-Dinitro-2,0	toluene	(µg/L)	9.5	10.5	11.2	9.01	BQL		BQL														
		3-Dinitro-2,	henzene	(µg/L)	9.0	-	Ξ	-	BQL		BÓL	BQL	BQL	BQL	BQL	BQL	BQL								
		_	Nitrate	- 1	0.858	0.84	0.875	0.827	1.05		1.17		1.2		1.19		1.27		1.28	1.31	1.25	1.25	1.47	1.36	1.25
		Total	RDX Nitrobodie: Nitrate	(µg/L) (mg/LN	809	570	594	586	194		94.6		25.2		9.5		3.7		1.5	1.6	5.	9.1	BQL	BQL	BQL
			RDX N	(µg/L)	4.4	16.7	18.5	17.2	5.7		2.1		0.4		BQL		BQL		BQL						
			TNB	ng/L)	323	270	280	278	130		76.4		21.3		8.3		3.7		2	9.	1.5	9.1	BQL	BQL	BQL
				(µg/L) (µg/L)	222	228	235	232	54.4		4		2.2		9.4		BQL		BOL	BQL	BQL	BQL	BQL	BQL	BQL
	onfactor	feasured	croxide	(mg/L)					23		25		25		25		25		56						
	Oxidation Contactor	of ORP Reduction Measured	Sample Potential Peroxide TNT	(mV)					610		945		956		955		945		006				253		
	Temp. O	ORP R	uple P	(°C)					12		=		=		9		10		2				oc		
	ī	Ē	pH Su						6,9		7.0		7.2		7.3		7.4		7.5				9.7		
		Ozone	Residual	(mg/L)					0.3	0.4	9.0	0.7	8.0	8.0	0.1	6.0	8.0	Ξ	0.5	0.4			0.0		
	Operations			٦					60:80	09:50	08:02	91:60	07:53	09:13	07:48	80:60	07:43	09:05	07:29	08:58			07:18		
	Ope	mple Sa	Location Time		NF.	INFI	INFI	INFI	C1/0 0	C1/0 0	C2/0 0	C2/0 0	C3/0 0	C3/0 0	C4/0 0	C4/0 0	C5/0 0	C5/0 0	C6/0 0	C6/0 0	C6/0	C6/0	GAC3 0	GACI	GAC2
	rage	XONE Sa	Ratio Low		0.55	0.55	0.55	0.55	0,55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 (0.55 G	0.55 G	0.55 G
ıge	Applied TransferredHydroger Average	Peroxide PEROXONE Sample Sample		2																					
Average Average Average	тефНуфго		Dose	(mg/L)	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0	25.0
Average	Transfer	Ozone	Dose	(mg/L)	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6	45.6
Average	Applied	Охопс	Dose	(mg/L)	09	9	9	9	09	99	9	9	9	9	9	09	9	9	9	09	99	9	9	9	9
		Process	Well Flow Rate	(mdg)	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0	24.0
			Well F		-	_	_	-	-	_	_	_	-	_	-	-	_	_	_	_	-	-	-	-	-
			Date		11/8/96	11/8/96	96/8/11	96/8/11	11/8/96	96/8/11	96/8/11	96/8/11	11/8/96	11/8/96	11/8/96	11/8/96	11/8/96	11/8/96	11/8/96	96/8/11	11/8/96	11/8/96	96/8/11	11/8/96	11/8/96

Appendix D

Peroxone System As-Built Drawings



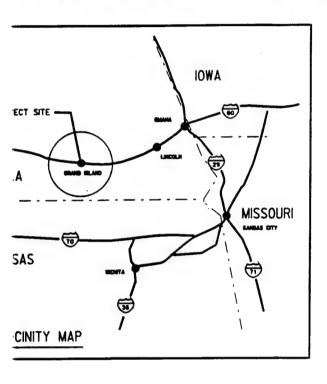
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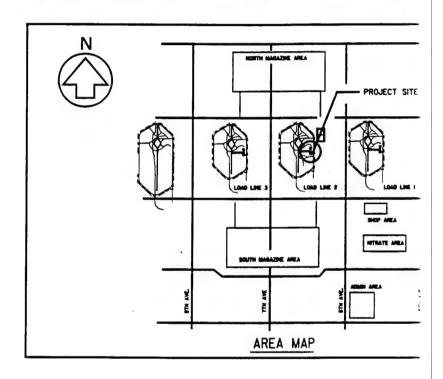
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TRW SPACE & TECHNOLOGY DIVISION REDONDO BEACH, CALIFORNIA

PROXONE DEMONSTRATION PROGRAM PROHUSKER ARMY AMMUNITION PLANT





AREA MAPS

ND SYMBOLS

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ULE AND DETAILS

. SECTIONS AND DETAILS

JMENTATION DIAGRAM

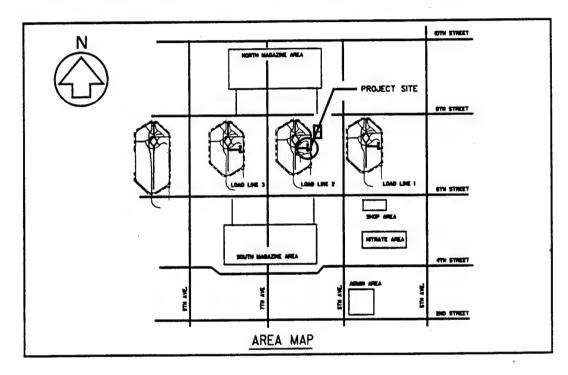
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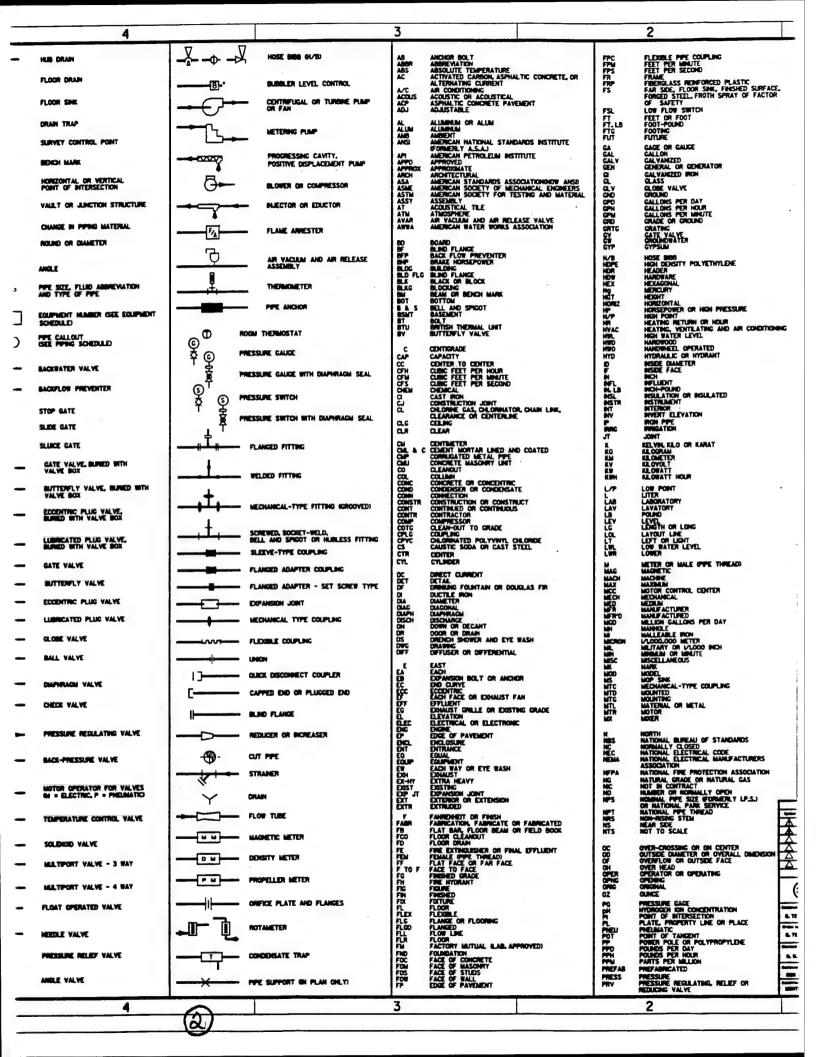
CHNOLOGY DIVISION CH, CALIFORNIA TRATION PROGRAM AMMUNITION PLANT



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A M 000.7 - 10 10	/10/17			-	RAJ
AS BOTH FROM .	B 10/30/17			-	441
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MONTE	OMERY WATSON	100 274	CE & TROPE	1,007 (F) 1,700 (F)	(MICH
& TEY	PEROXONE D CORNALISKER	EMONSTRA ARMY AM	TION PRO	GRAM PLANT	
-					
6. TET	TITLE SHEE	T AND	ADEA	MAD	•
-	IIILE SHEE	. I AND	AREA	mAr.	9
6. S. DW46					
	-	-	_		
AND SOLUTION					
	-	6-4			
AMPROVED THE PARTY BATTERS		7 0-1			
SERVICE OF SATISFIES		G-1			

(3)

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ſ	9.10	MASONRY	<u> </u>	HUB ORAIN	HOSE BEER
		CAST STON		FLOOR DRAIN	[E].* BUBBLER L:
		STEEL.	×	FLOOR SHK	CENTRIFUGA OR FAM
	recent.	BROKZE	<u>~</u>	DRAIN TRAP	METERNIC F
	*******	INSULATION	•	SURVEY CONTROL POINT	L5-
D	THERMAN	GRAVEL	Δ	BENCH MARK	PROGRESSI POSITIVE D
	40 PASA	CONCRETE	Δ	HORIZONTAL OR VERTICAL POINT OF BITERSECTION	BLOWER OF
		EARTH		VALET OR JUNCTION STRUCTURE	BLECTOR C
	170,000,000	SAID	*	CHANGE IN PIPING MATERIAL	FLAME ARE
	(50000000	ALLMANUM OR METAL DECKING	Ø	ROUND OR DIAMETER	AR VACUU
	3000000	CHECKERED PLATE	_	ANGLE	ASSEMBLY
	2M	GRATEN	24' RW-RCP	PIPE SIZE, FILID ABBREVIATION AND TYPE OF PIPE	THEPMOLET
	1/////	PLASTIC, RUBBER OR HEOPREJE	EV-3-SF-7	EQUIPMENT NUMBER (SEE EQUIPMENT SCHEDULE)	77, 300
	, '` >>≥	WOOD STRESHO	(F-90'-(B)	PPE CALLOUT CSEE PPING SCHEDULED	© ROOM THERMOSTAT
		WOOD GROUCH FRAMING OR OPENING OR DEPRESSION IN SLAB OR WALL	→ (∑)—	BACKBATER VALVE	© PRESSURE GALGE 1
	πO	FIRE EXTINGUISHER	+022-	BACKFLOW PREVENTER	S PRESSURE GAUGE T
С	<u>رم ح</u>	· UNIT NEATER	()	STOP GATE	(S) PRESSURE SWITCH T
		CONTENUE	(——)	SLIDE GATE	PRESSURE SWITCH T
		PROPERTY LINE	<u></u>	SLUKCE GATE	FLANGED FITTIN
		HEW STRUCTURE OR FACILITY		GATE VALVE, BLINED WITH VALVE BOX	WELDED FITTHC
		EXISTING STRUCTURE OR FACILITY		BUTTENFLY VALVE, BURED WITH VALVE BOX	.+.
		FUTURE STRUCTURE OR FACILITY NEW FENCE	─	ECCENTRIC PLUG VALVE.	MECHANICAL-TYF
	-0-0-	TEMPORARY FENCE		LUBRICATED PLUG VALVE. BURBED WITH VALVE BOX	SCHEWED, SOCKE
		DOSTING FENCE		CATE VALVE	SLEEVE-TYPE C
		NEW PIPELNE (CIVIL SHEETS) 36' DIA AND LANGER		BUTTENFLY VALVE	FLANGED ADAPT
	E(OH0	ELECTRICAL, GVETHEAD		ECCENTRIC PLUG VALVE	ENPARSON JON
		EXISTING PIPELINE (SCREENED) DITCH CENTENLINE WITH FLOW DIRECTION		LUBRICATED PLUG VALVE	MECHANICAL TY
В		SLOPE	— —	GLOBE VALVE	FLEXIBLE COUP.
	100-	EXISTING GRADE CONTOUR (SCREEDED)		BALL VALVE	UNION UNION
	⊗ 1230.2	FINISHED ELEVATION	<u>-</u> ∞-	DIAPHRAGN VALVE	CAPTED DID OF
	X 1230.2	EXISTING ELEVATION		CHECK ANTAE	BLNO FLANCE
		TO BE CONSTRUCTED	-A	PRESSURE REGULATING VALVE	
	heliberte	NEW A.C. PAYING	- 2	BACK-PRESSURE VALVE	- CUT PIPE
	المداد الأطر	EXISTING A.C. PAYING (SCREENED)	₩ ₽	MOTOR OPERATOR FOR VALVES	STRANER
		RALBIG		ON . ELECTRIC, P . PHELMATIC	DRAIN
	-	FIRE HYDRANT	 	TEMPERATURE CONTROL VALVE	FLOW TUBE
	 	MANNOLE	——————————————————————————————————————	SOLENOID VALVE	M M MAGNETIC METE
A	PCOTG O-	- MESSURE CLEANOUT TO GRADE	<u></u>	MALTPORT VALVE - 3 BAY	O M DENSITY METER
alo c	٠,	ICO	─────	MILTPORT VALVE - 4 BAY	PROPELLER MET
	<u></u>	WALL CLEANOUT		FLOAT OPERATED VALVE	OMFICE PLATE
et s	F60 - QE	FLOOR CLEANOUT	₹—	NEEDLE VALVE	ROTAMETER
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	00TG _00	CLEANOUT TO GRADE	<u>\$</u>	PRESSURE RELET VALVE	CONDENSATE TR
4-0CT-1997 14:19 4-0CT-1997 14:19 4-0CT-1997 14:19		at the last account to		ANGLE VALVE	PPE SUPPORT
7.2		SLOW OFF ASSEMBLY	T	4	
	(1)	3			



FLEDBLE PPE COUPLING
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OF SAFETY
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FEET OR FROT
FOOT-POUND
FOOT-POUND
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FUTURE PRESSURE SWITCH
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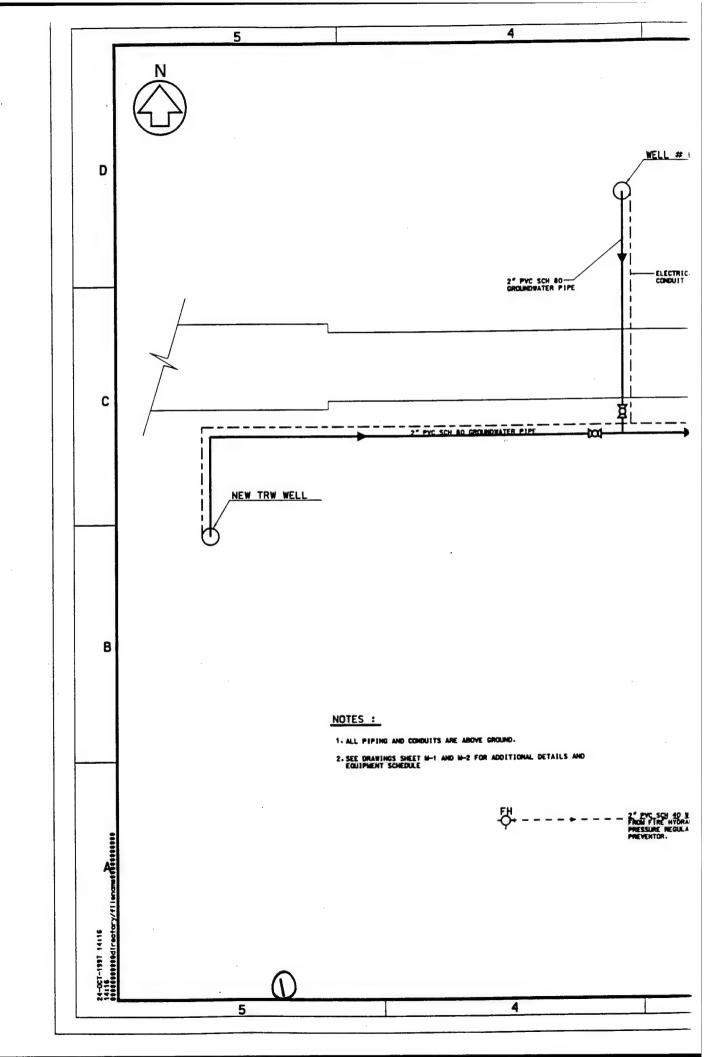
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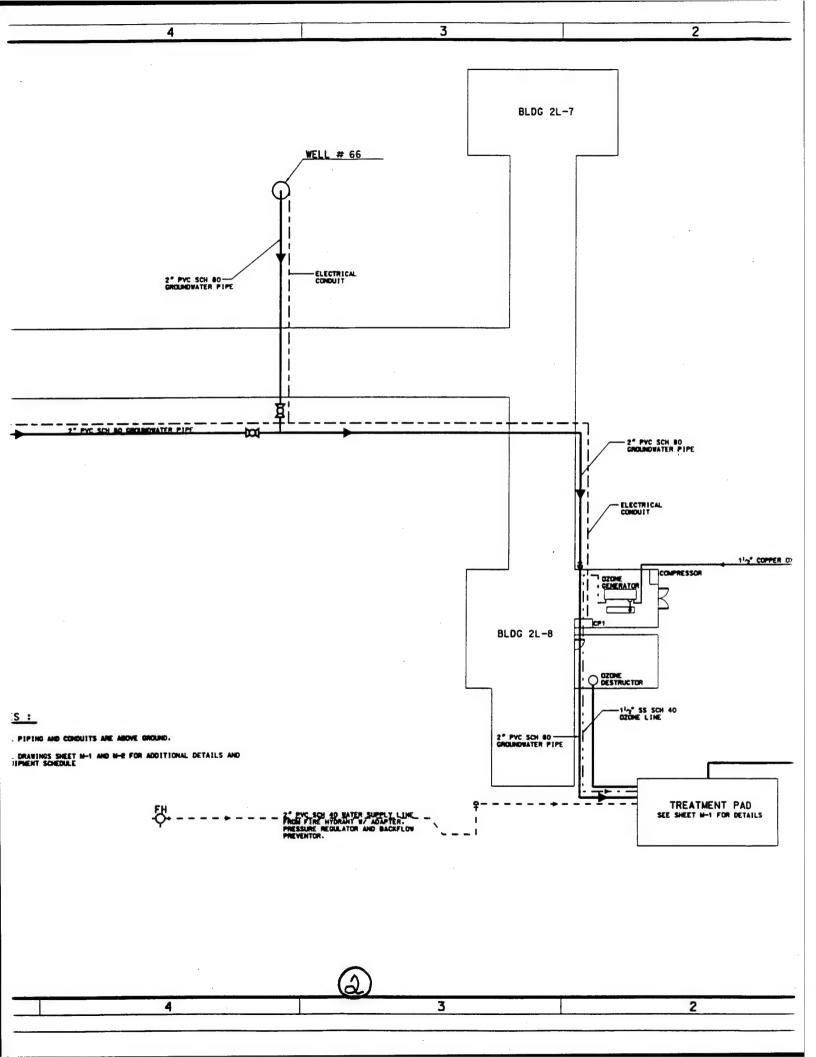
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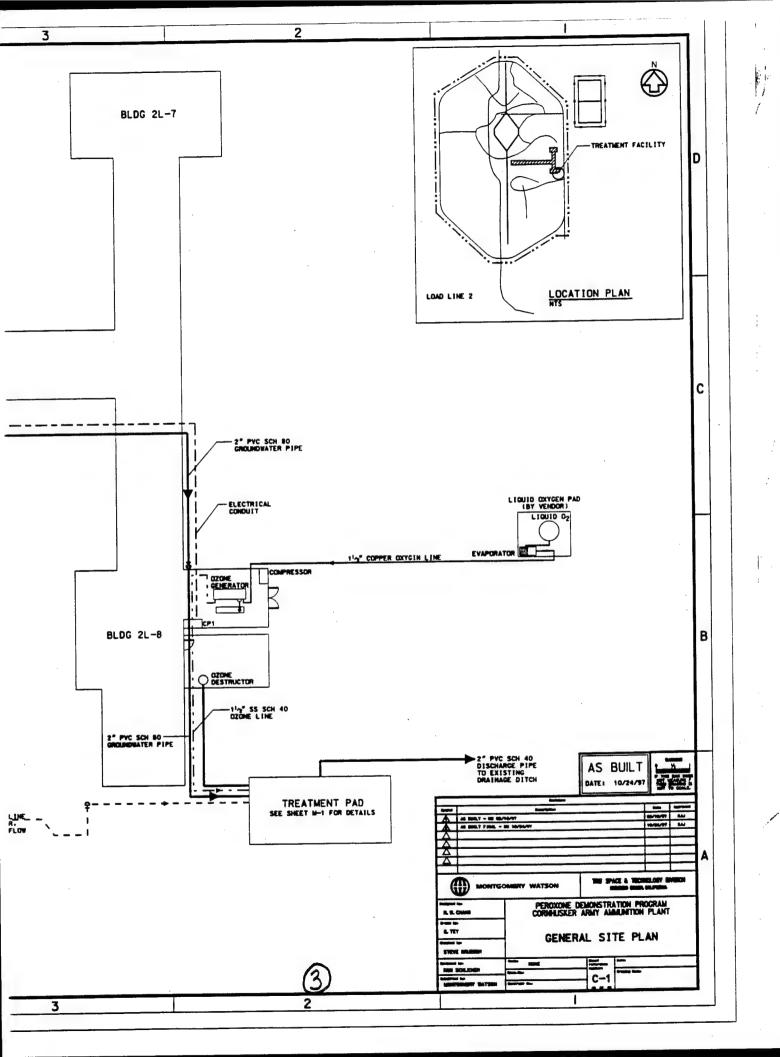
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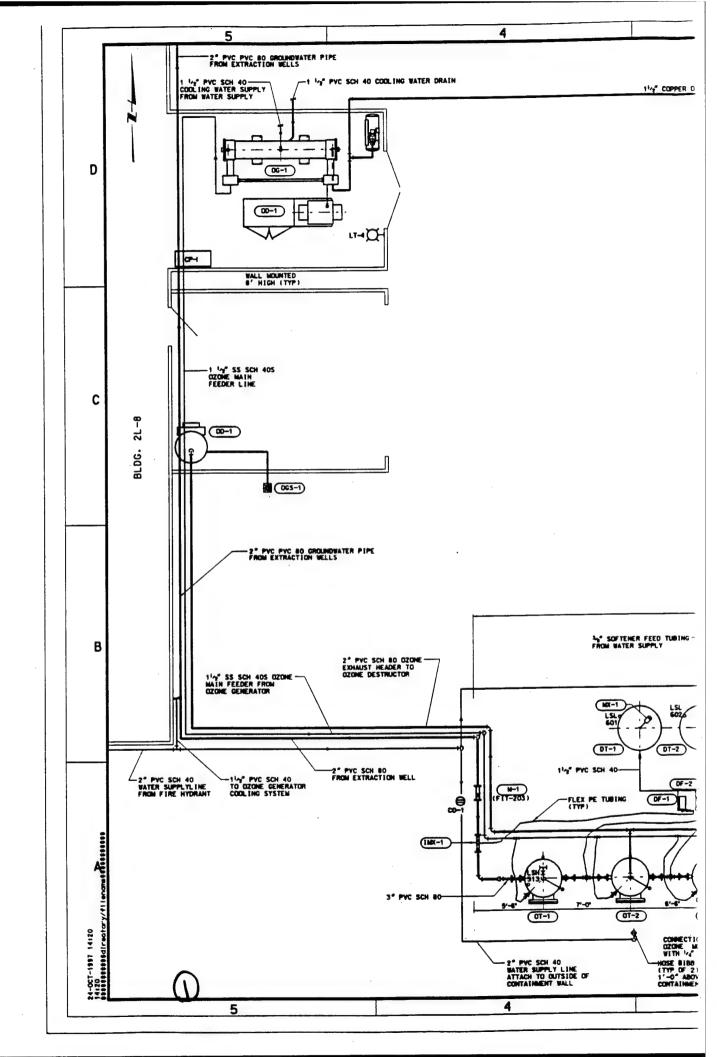
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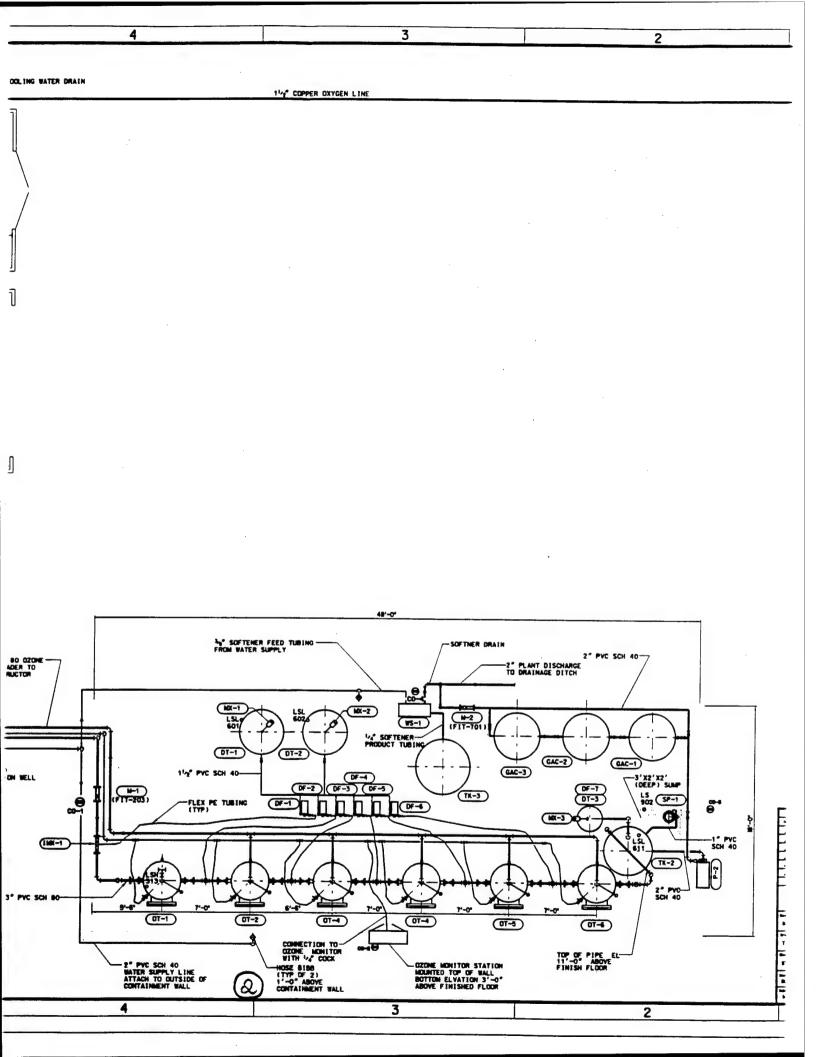
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OUTSIDE DAMETER OR OVERALL DIMENSION
OVERFLOW OR OUTSIDE FACE
OVER HEAD
OFFINIS
OFFINIS
OURCE TO F THE PAGE & TREMELOW COM MONTGOMERY WATSON OUNCE
PRESSURE CAGE
HYDROGEN ION CONSENTRATION
POINT OF INTERSECTION
PLATE, PROPERTY LINE OR PLACE
PRESIDENT
POINT OF TANCENT
POWER POLE OR POLYPROPYLENE
POUNDS PER DAY
POLINGS FER HALLON
PREFABRICATED
PRESSURE POR AL PROPERTY AND PROPERTY AN PERCICONE DEMONSTRATION PROGRAM CORNHLISKER ARMY AMAILINITION PLANT FLOOR
FLEXIBLE
FLANGE OR FLOORING
FLANGED
FLOOR LINE & MY EX ... & TET ABBREVIATIONS AND SYMBOLS FLOOR FACTORY MUTUAL (LAS. APPROVED) FOUNDATION
FACE OF CONCRETE
FACE OF MASONRY
FACE OF STLDS
FACE OF WALL
EDGE OF PAVEMENT ... 1654 AMI 104.DA PRESSURE PRESSURE REGULATING, RELIEF OR REDUCING VALVE G-2

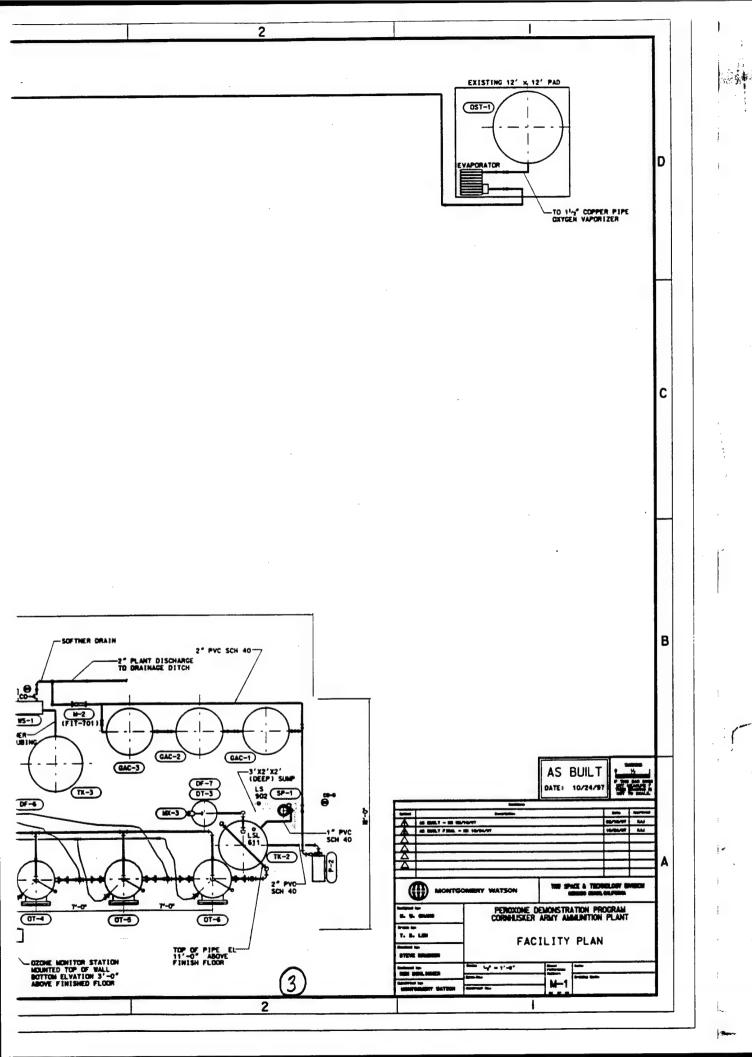


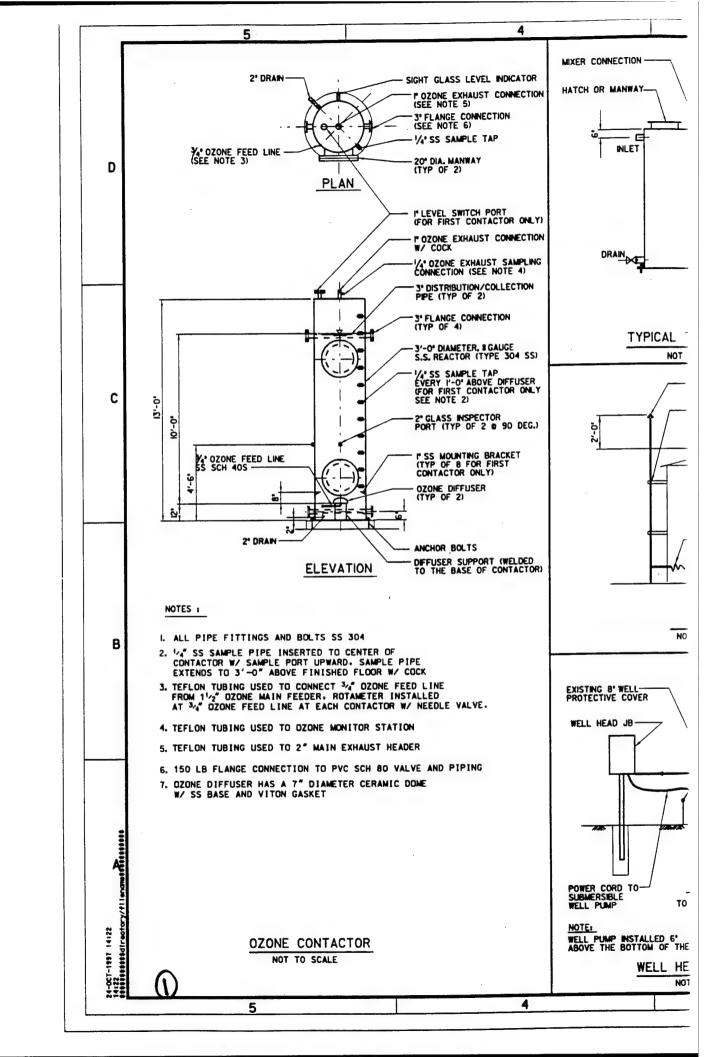














- 3° FLANGE CONNECTION (SEE NOTE 6) - 1/4° SS SAMPLE TAP - 20° DIA, MANWAY (TYP OF 2)

(SEE NOTE 5)

SIGHT GLASS LEVEL INDICATOR

POZONE EXHAUST CONNECTION

4

- PLEVEL SWITCH PORT (FOR FIRST CONTACTOR ONLY)

- POZONE EXHAUST CONNECTION
W/ COCK

-1/4° OZONE EXHAUST SAMPLING CONNECTION (SEE NOTE 4) - 3° DISTRIBUTION/COLLECTION

PIPE (TYP OF 2)

3' FLANGE CONNECTION
(TYP OF 4)

-3'-0" DIAMETER, II GAUGE S.S. REACTOR (TYPE 304 SS)

1/4°SS SAMPLE TAP EVERY 1'-0' ABOVE DIFFUSER (FOR FIRST CONTACTOR ONLY SEE NOTE 2)

-2° GLASS INSPECTOR PORT (TYP OF 2 @ 90 DEG.)

"SS MOUNTING BRACKET (TYP OF 8 FOR FIRST CONTACTOR ONLY)

OZONE DIFFUSER (TYP OF 2)

- ANCHOR BOLTS

OFFUSER SUPPORT (WELDED TO THE BASE OF CONTACTOR)

R OF

#PLE PIPE
JR W/ COCK

ONE FEED LINE
ER INSTALLED
CTOR W/ NEEDLE VALVE.

STATION

T HEADER

80 VALVE AND PIPING RAMIC DOME

HATCH OR MANWAY

LEVEL SWITCH PORT

SPIGOT OR PIPE CONNECTION
OUTLET

POSITION ANGLE PLATE
(TYP OF 4)

TYPICAL TANK SECTION

NOT TO SCALE

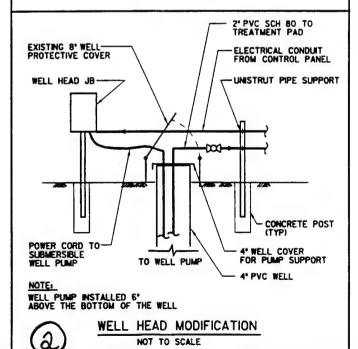
2' STL STACK W/ RAIN CAP

UNISTRUT SADDLE PIPE SUPPORT
ATTACHED TO EXISTING BUILDING
COLUMN (TYP)

EXIST BUILDING

FLEXIBLE 2" METAL DUCT FROM OZONE DESTRUCTOR

STACK NOT TO SCALE



MECHANICAL EQUIPME EQUIP. MALE DESCRIPTION EXISTING 4' OR LARGER CASING NEW TRW WELL EXTRACTION WELL NO.66 EXISTING 4" OR LARGER CASING SUBMERSIBLE WELL PUMP EP-I EXTRACTION PUMP NO. I SURMERSIBLE WELL PLANS EP-2 EXTRACTION PUMP NO. 2 PADOLE WEEL TYPE METER MELLIENT FLOW METER M-I H₂Q₂ IN-LINE MIXER

UNPACKED COLUMN W/CO-CURR

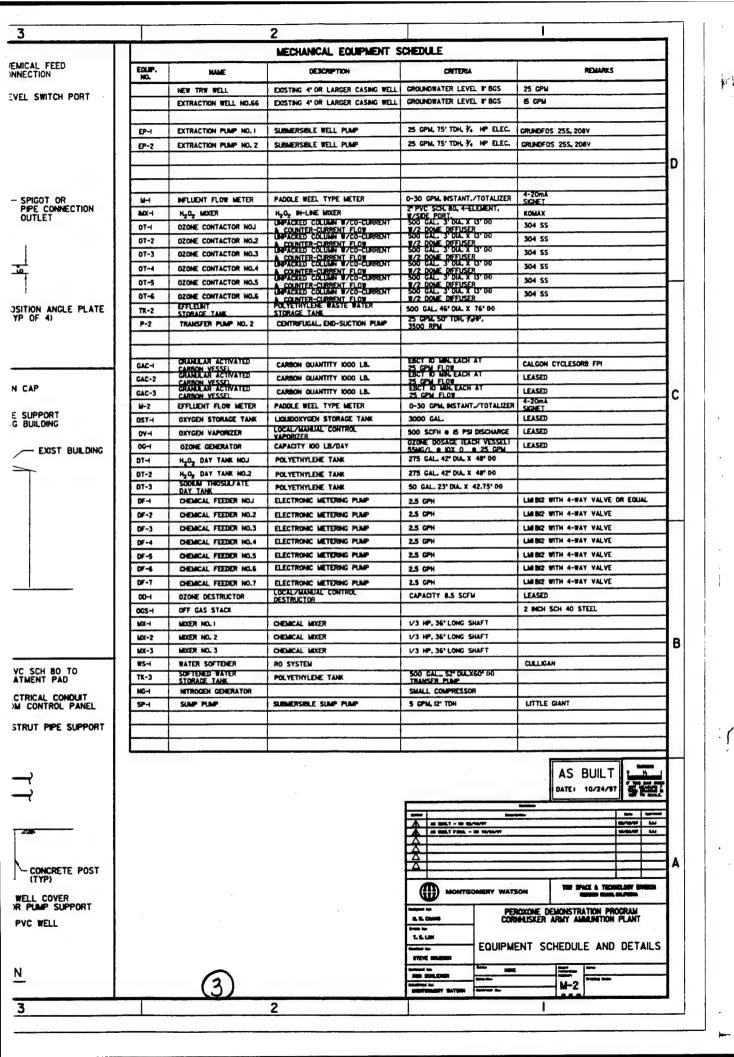
A COUNTER-CURRENT FLOW

UNPACKED COLUMN W/CO-CURR H_O_ MODER OZONE CONTACTOR NOJ OT-2 DZONE CONTACTOR NO.2 A COUNTER-CURRENT FLOW OT-3 DZONE CONTACTOR NO.3 A COUNTER-CURRENT FLOW OT-4 DZONE CONTACTOR NO.4 A COUNTER-CURRENT FLOW
UNPACKED COLUMN W/CO-CURR OT-5 DZONE CONTACTOR NO.5 A COUNTER-CURRENT FLOW DZONE CONTACTOR NO.S A COUNTER-CURRENT FLOW POLYETHYLENE WASTE WATER STORAGE TANK TK-2 STORAGE TANK CENTRIFUGAL, END-SUCTION PL TRANSFER PUMP NO. 2 P-2 CHANLLAR ACTIVATED CARBON QUANTITY 1000 LB. GAC-I CARBON VESSEL CARRON QUANTITY 1000 LB. GAC-2 CARBON VESSEL CARBON QUANTITY 1000 LB. GAC-3 M-2 EFFLUENT FLOW METER PADOLE WEEL TYPE METER LICARDOXYGEN STORAGE TANK OST-OXYGEN STORAGE TANK LOCAL/MANUAL CONTROL OXYGEN VAPORIZER 0V-1 VAPORIZER. CAPACITY 100 LB/DAY 0G-I OZONE GENERATOR H₂O₂ DAY TANK NOJ DTH POLYETHYLENE TANK H₂O₂ DAY TANK NO.2 SODUM THOSULFATE DAY TANK POLYETHYLENE TANK DT-2 DT-3 POLYETHYLENE TANK DF-I CHEMICAL FEEDER NO.I ELECTRONIC METERING PUMP ELECTRONIC METERING PLANS DF-2 CHEMICAL FEEDER NO.2 CHEMICAL FEEDER NO.3 FLECTRONIC METERING PLMP DF-3 ELECTRONIC METERING PLANP CHEMICAL FEEDER NO.4 DF-4 ELECTRONIC METERING PUMP DF-5 CHEMICAL FEEDER NO.5 DF-6 CHEMICAL FEEDER NO.6 ELECTRONIC METERING PUMP ELECTRONIC METERING PUMP LOCAL/MANUAL CONTROL DF-7 CHEMICAL FEEDER NO.7 00-1 DZONE DESTRUCTOR DESTRUCTOR OGS-I OFF GAS STACK CHEMICAL MIXER MX-I MOXER NO. I MIXER NO. 2 CHEMICAL MIXED MX-2 MX-3 MEXER NO. 3 CHEMICAL MIXER 85-WATER SOFTENER RO SYSTEM SOFTENED WATER TK-3 POLYETHYLENE TANK NITROGEN GENERATOR SUMP PUMP SUBMERSIBLE SUMP PUMP SP-I

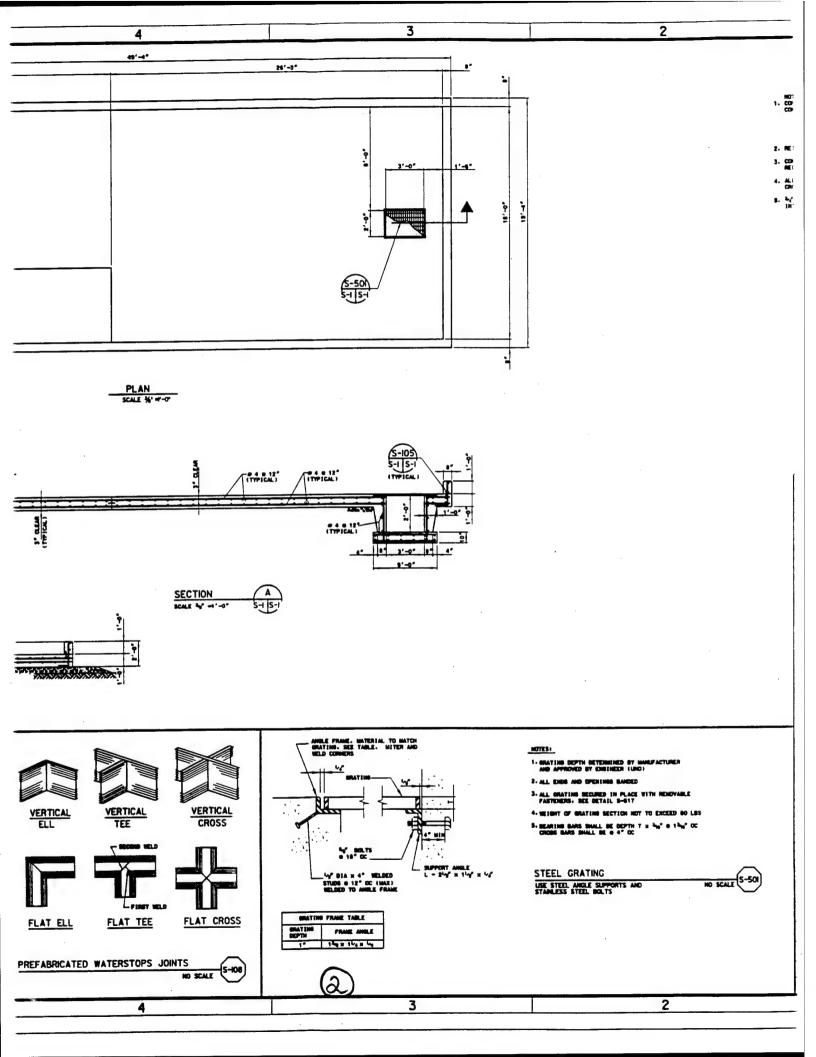
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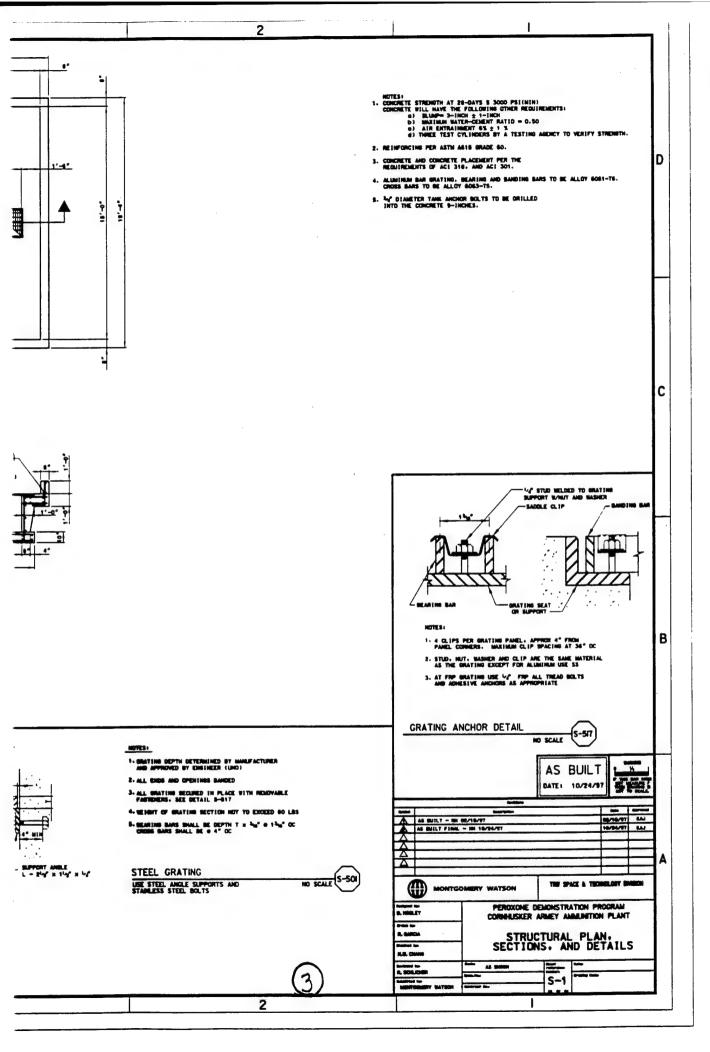
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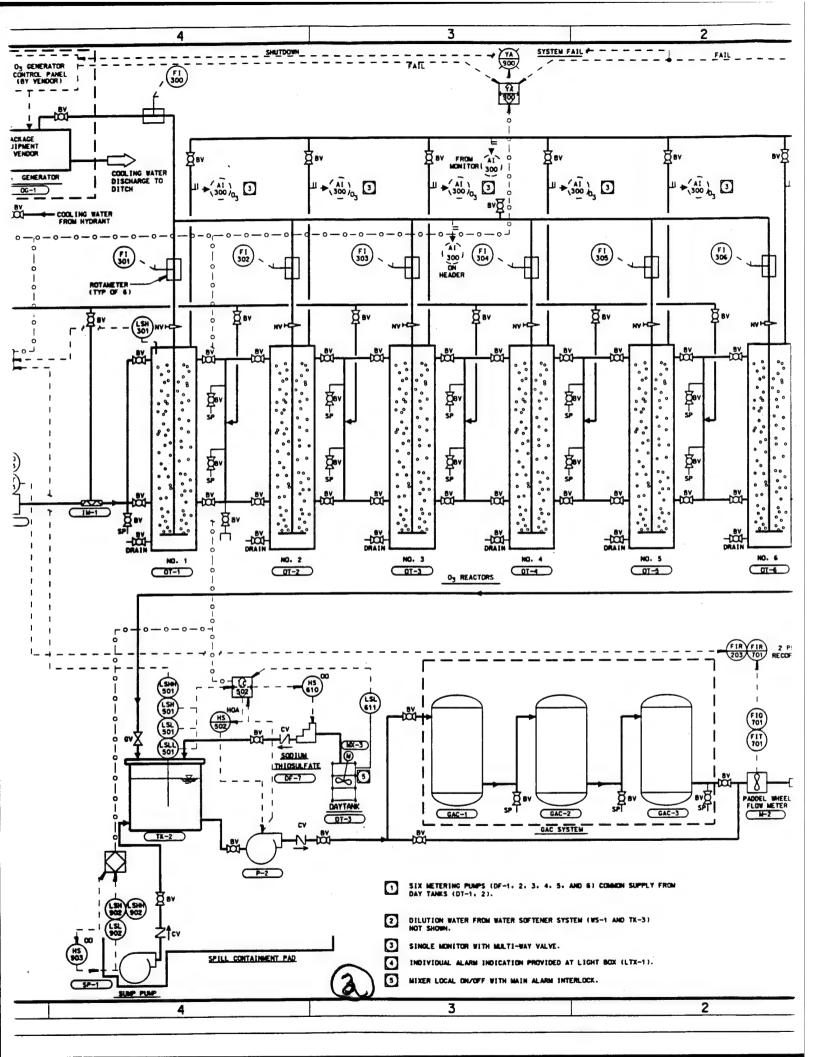


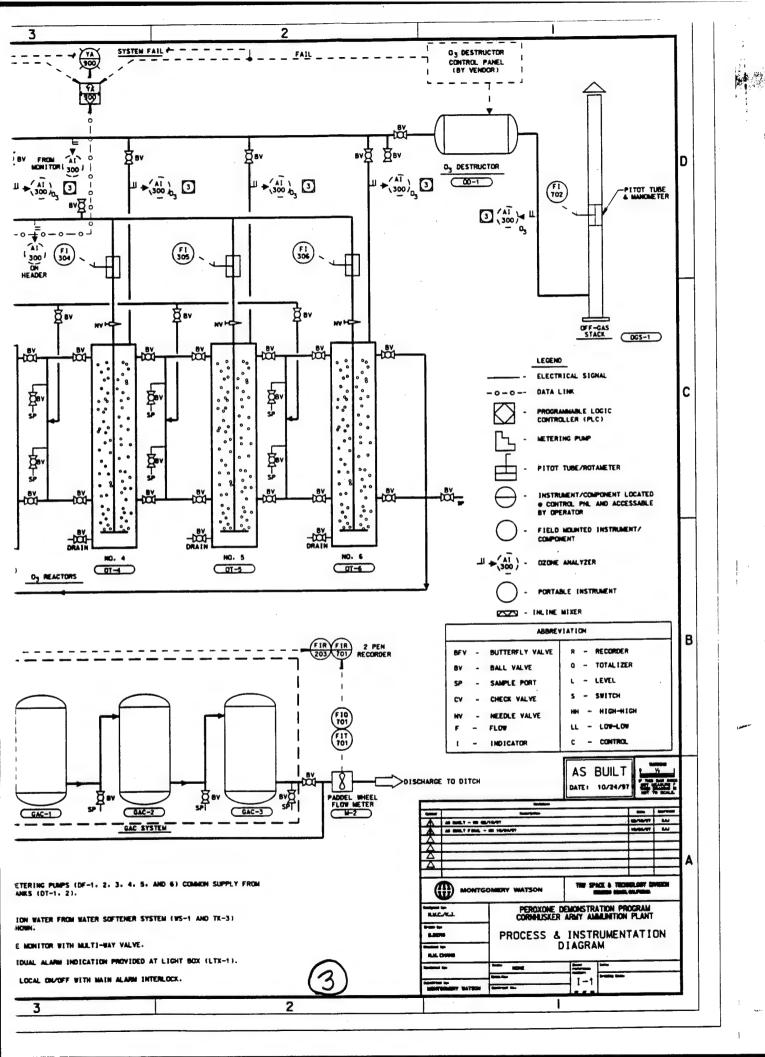
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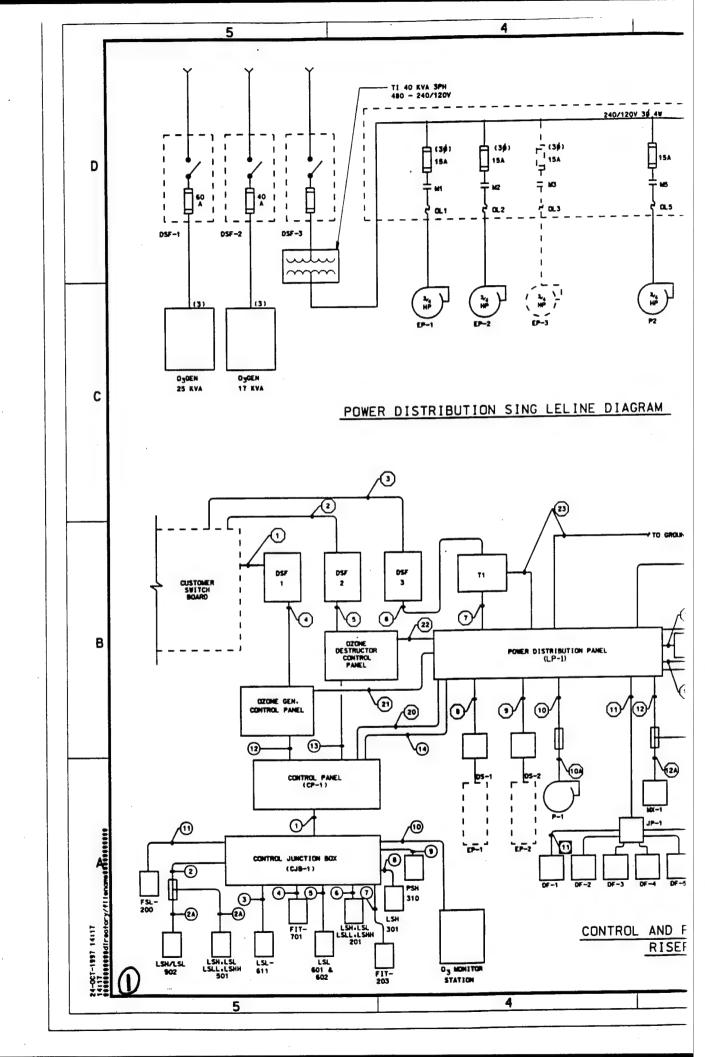


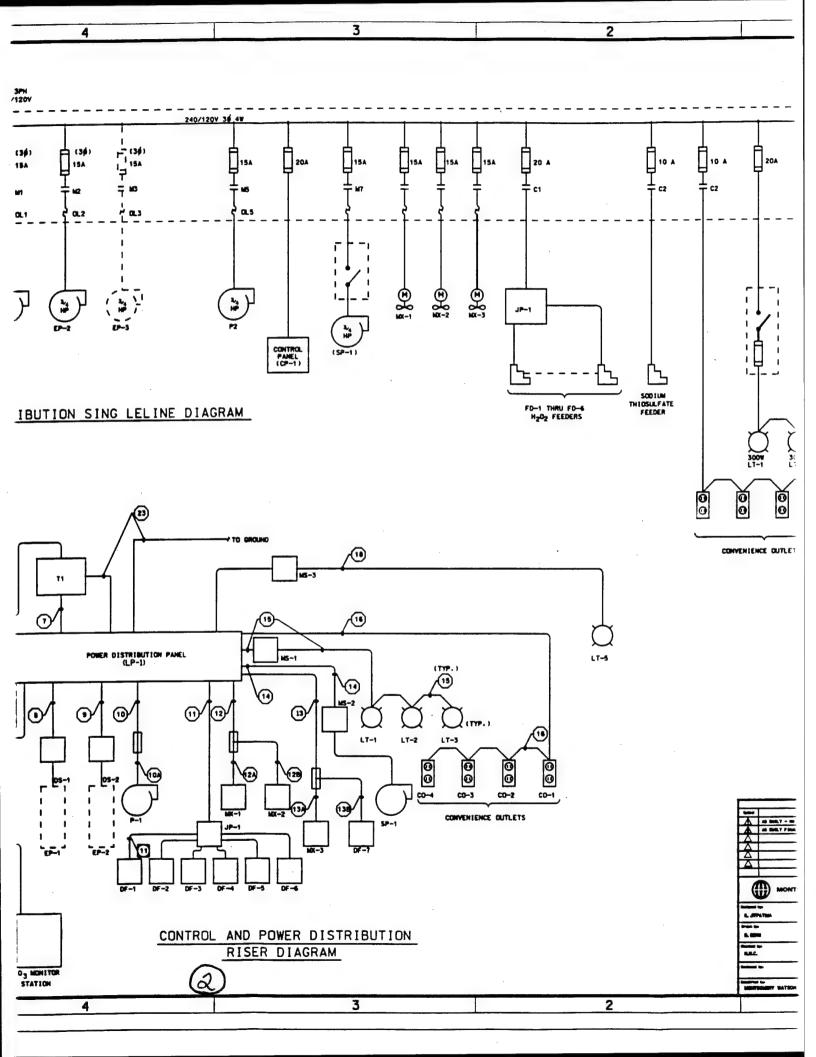


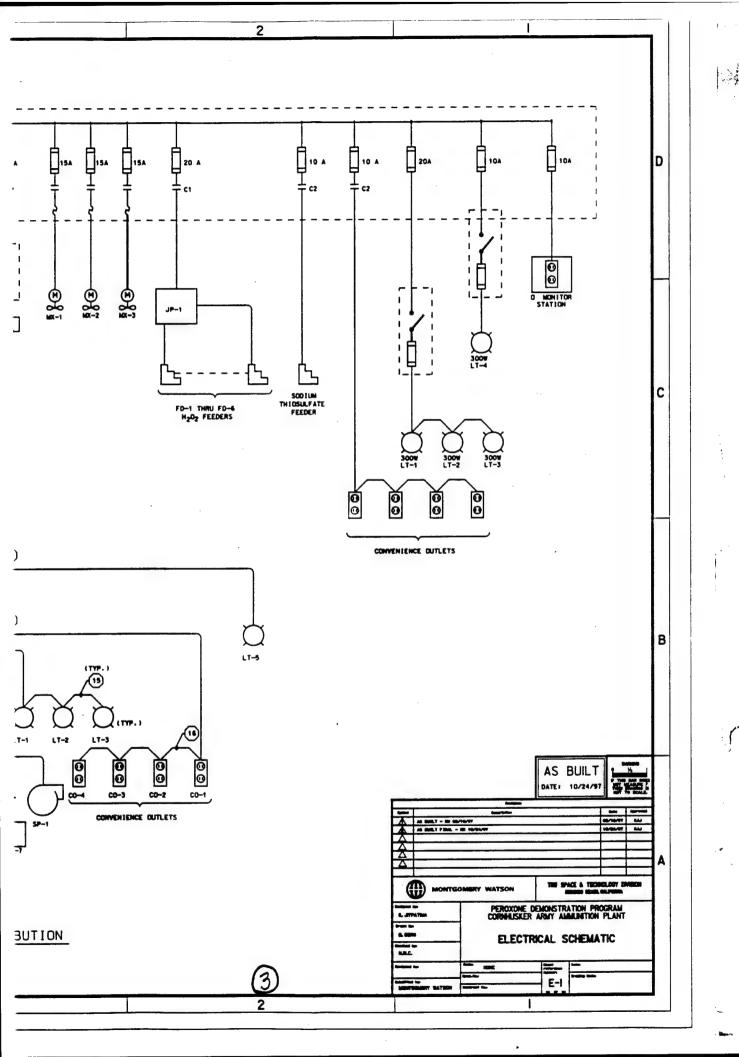
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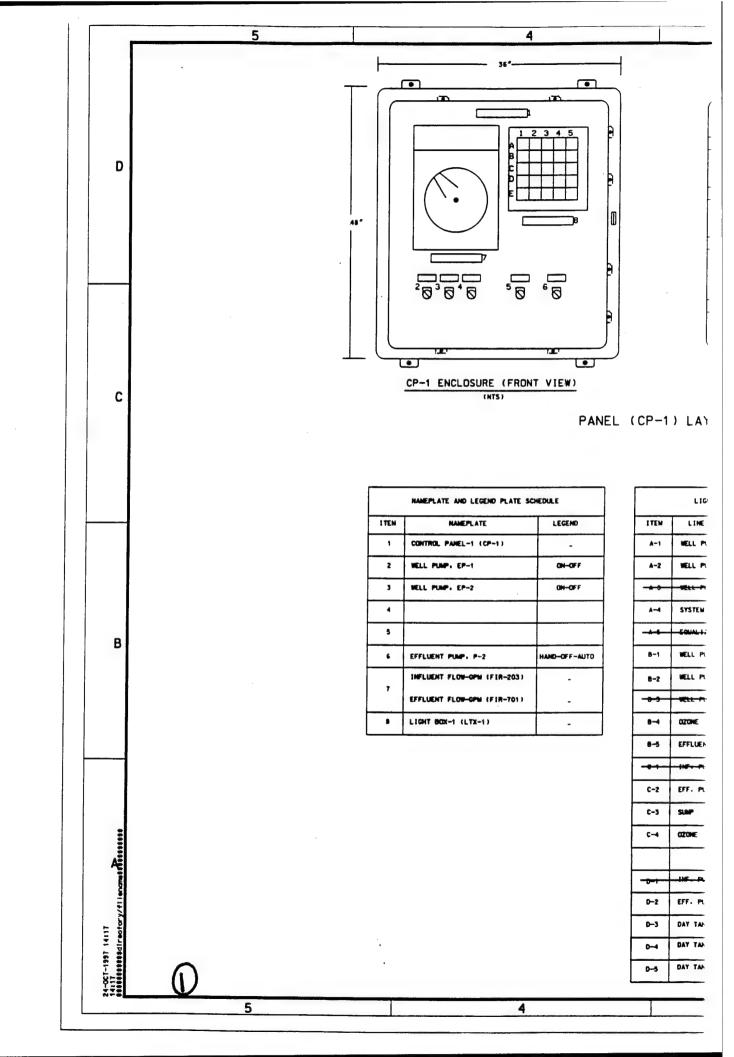


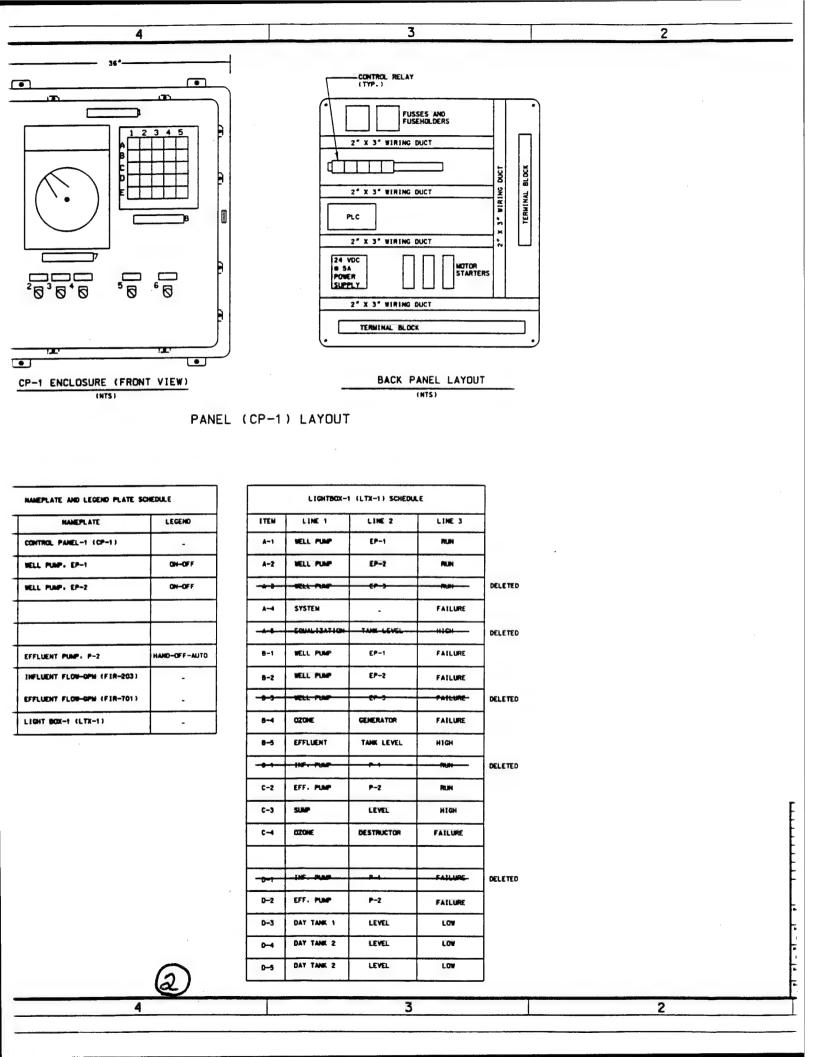


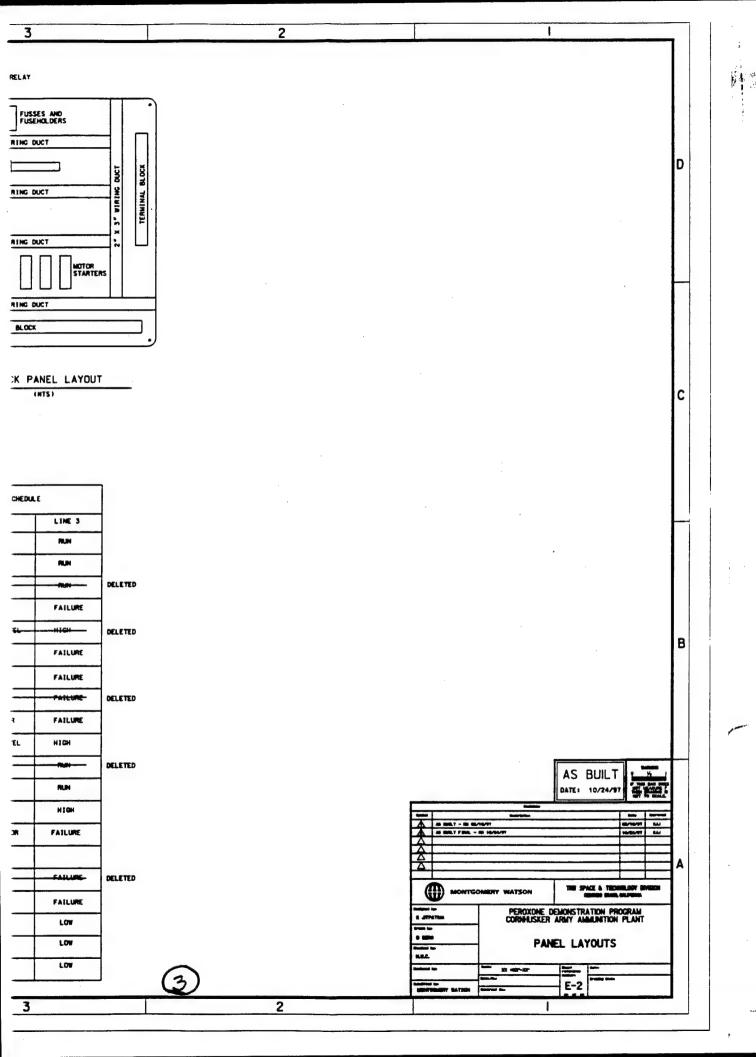












Appendix E

Project Experimental Plan

CORNHUSKER ARMY AMMUNITION PLANT PEROXONE GROUNDWATER TREATMENT PROJECT

EXPERIMENTAL PLAN

submitted to:

TRW Space & Technolgoy

Prepared by:

MONTGOMERY WATSON

250 North Madison Avenue Pasadena, CA 91101

JULY 5, 1996

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SECTION 1 INTRODUCTION & GENERAL APPROACH

1.1: INTRODUCTION

TRW Space and Technology retained the services of Montgomery Watson to design, construct, and operate a 25 gpm Peroxone groundwater treatment demonstration plant. The groundwater at the Cornhusker Army Ammunition Plant (CAAP) in Grand Island, Nebraska is contaminated with various energetic compounds, including TNT, RDX, TNB, and other nitrobodies. The objective of the project is to determine the ozone dose, hydrogen peroxide dose, and hydraulic retention time needed to reliably achieve the required removals of these contaminants to acceptable levels. The anticipated levels of contaminant concentrations in the groundwater, and their respective treated water goals are listed in Table 1.

Table 1

Anticipated Contaminant Levels and Corresponding Treated Water Goals

Contaminant	Anticipated Groundwater Concentration (mg/L)	Target Concentration After Peroxone Treatment (mg/L)
TNT	0.5	0.002
RDX	0.2	0.002
TNB	0.1	0.002
Total Nitrobodies	1.0	0.030

Figure 1 shows a schematic of the groundwater treatment demonstration plant. The design of the demonstration plant includes six (6) ozone contactors in series with ozone and hydrogen peroxide fed independently to each contactor. A GAC contactor is provided at the effluent side of the plant with an EBCT of 30 minutes at 25 gpm to ensure that no contaminants are discharged with the plant water during testing. The maximum design applied ozone dose to each contactor is 60 mg/L for a total applied ozone dose of 360 mg/L. The hydrogen peroxide system is designed to deliver a maximum of 18 mg/L to each contactor for a total of 108 mg/L at 25 gpm (this provides a $H_2O_2/Ozone$ Ratio of 0.3 mg/mg). At the design flow rate of 25 gpm, the average hydraulic retention time (HRT) in each contactor is 20 minutes for a total system HRT of 120 minutes. The system will be tested on waters from two (2) groundwater wells. The notations used in this document for the two wells are "Well A" and "Well B".

This document details the experimental plan to be implemented at the demonstration plant to achieve the project objectives.

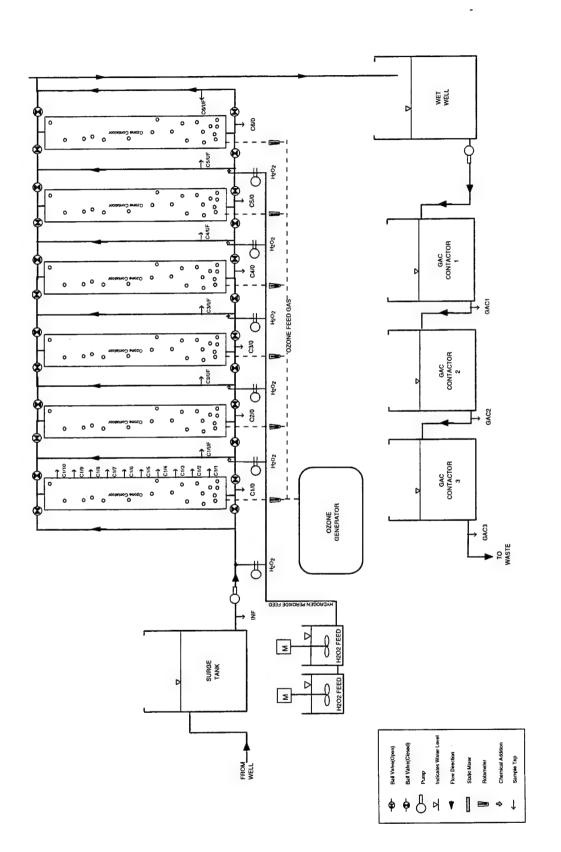


FIGURE 1 PEROXONE DEMONSTRATION PLANT SCHEMATIC

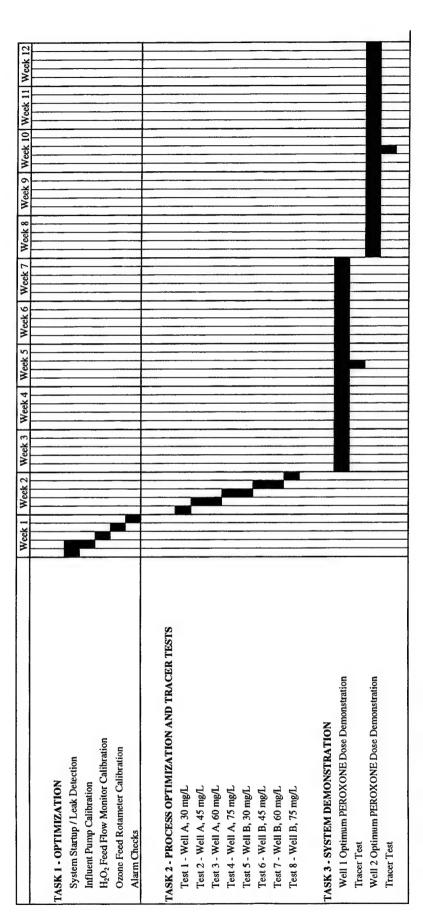


FIGURE 2 PROJECT SCHEDULE

1.2: GENERAL APPROACH

The overall testing schedule, which is planned for a total of 12 weeks, is outlined in Figure 2. After the plant is constructed and all the equipment are installed, the demonstration plant operators will conduct three primary tasks.

1.2.1: Task 1 (1 week):—Conduct System Debugging

During the first week, the plant pumps and chemical feed systems will be started up at a low flow rate (approximately 10 gpm) using tap water, and checked for any water or chemical leaks. The system will also be checked for malfunctions of chemical feed equipment and shut-down alarms. After the leaks and malfunctions, if any, are repaired, the flow rate through the plant will be continuously increased until the design flow of 25 gpm is reached. The plant will then be operated at that flow rate for a period of two (2) hours. During this period all water and chemical feed equipment will be checked for operational stability.

1.2.2: Task 2 (1 week):—Conduct Process Optimization

During the 2nd week, process optimization testing will be conducted using water from each of the two wells. Process optimization will involve operating the system at various ozone doses, collecting water samples from the effluent of each of the six contactors, as well as from the wall taps along the water depth of the first contactor, and analyzing them for ozone residual and target contaminants' concentrations. The applied ozone dose tested will range from 30 mg/L to 75 mg/L. The flowrate will be held constant at the design value of 25 gpm. Since the ozone generator is capable of producing a maximum dose of 60 mg/L at 25 gpm water flowrate, the flowrate will be reduced to 15 gpm when evaluating the ozone dose ranging from 60 mg/L to 75 mg/L. These tests will allow for the determination of the optimum operating conditions that will result in the reduction of the contaminants to their respective target concentrations.

1.2.3: Task 3 (10 weeks):—Conduct System Demonstration

During the period extending from the 3rd week through the 12th week, the system will be operated under constant conditions. This period will serve to demonstrate that the system can achieve the anticipated performance on a long-term basis. The demonstration period will be divided into two (2) segments of 5 weeks each, with water from one of each of the two wells used as the raw water source in each segment. In addition, two tracer tests will be conducted on the system to determine the hydraulic characteristics of the contactor design at the selected water flow rate.

SECTION 2 TESTING PLAN

The following paragraphs detail the tests to be conducted during each task. In order to facilitate the implementation of this experimental plan, Figure 1 includes a schematic of the demonstration plant showing the plant components and all the sampling taps installed. Each tap is letter-coded for ease of identification and sampling tracking.

2.1: TASK 1—CONDUCT SYSTEM DEBUGGING

The objectives of this 1-week task are as follows:

- 1. start up the demonstration plant,
- 2. ensure that all its components are fully operational,
- 3. calibrate all chemical feed systems,
- 4. test all alarms and emergency shut-down systems, and
- 5. check for leaks and malfunctions.

The following is a description of the tests to be conducted in this task:

2.1.1: System Startup & Leak Detection

Fill the surge tank with tap water and pump water into the system at an indicated flowrate of 25 gpm to fill up the six contactors with water. When the six contactors and the GAC contactor are full with water, turn off the water flow rate, look for any major leaks, and then wait for 30 minutes and check for any minor leaks throughout the system including, but not limited to, the following:

- 1. the sides of the contactors
- 2. sampling taps
- 3. pipes and pipe connections
- 4. pumps and chemical injection ports

If any leaks are detected, the leaking component of the system will be isolated, drained from water, and repaired. The system will be refilled with water and re-checked for leaks. This process should be repeated until no leaks are detected.

Once the system is void of leaks, the water will be started at a flow rate of 10 gpm. The ozone system will be turned on, and ozone will be fed to the six contactors at 40 percent of capacity (which should translate into a total dose of 360 mg/L to a flow of 10 gpm). The Soap-Bubble test will be conducted on all gas-phase pipe connections outside the ozone generator, monitor, and destruct unit. While ozone is being fed to the system, the hydrogen peroxide feed system to the six contactors will be turned on. The system will be checked for any hydrogen peroxide leaks. If any leaks in the ozone system or the hydrogen peroxide

system are detected, the system will be shut down, and the leaks repaired. This test will be repeated until both feed systems are void of detectable leaks.

After all system components are checked for leaks, the water flow rate will be increased gradually to 25 gpm, accompanied by a corresponding increase in ozone generator setting and hydrogen peroxide feed rate to deliver the design doses of 360 mg/L ozone and 108 mg/L hydrogen peroxide. The system will be operated under these conditions for a period of 30 minutes during which a final leak check will be conducted on all system components.

2.1.2: Equipment Calibration

The following instruments and monitoring equipment will be calibrated during this task:

- 1. influent water flowmeter
- 2. hydrogen peroxide metering pumps

2.1.2.1: Influent Water Flowmeter

The influent water flowmeter will be calibrated using a polyethylene 55-gallon drum. Tap water will be used in this test. A total of three (3) indicated flow rates will be evaluated: 10, 18, and 25 gpm. A constant flowrate will be allowed to flow through the flowmeter. The water will be diverted from the effluent of the first contactor through a flexible hose to the drain. After 10 minutes of steady flow, the water will be diverted into the 55-gallon calibration drum. Time will be kept using a stopwatch until the 50 gallon mark is reached. During the test, one operator will watch the flowmeter to ensure that the reading is stable at the test flowrate. The ratio of 50 gallons by the fill time (in minutes) will constitute the actual flowrate value in gpm. This test will be repeated in triplicates for each of the three test flowrates. It is important that the temperature of the water be measured and recorded during each test run. The datasheet to be used in this test is shown in Figure 3. Once the calibration curve is developed, the "actual" flowrate, instead of the "indicated" flowrate, should be used in all subsequent testing.

2.1.2.2: Hydrogen Peroxide Metering Pumps

The column calibration method will be used to calibrate the metering pumps. No water will be flowing through the contactors during this test. However, the contactors should be full. A total of three (3) pump settings will be calibrated for each of the pumps installed. A 50-mL graduated burette will be filled with water and connected to the suction side of the pump being calibrated. The pump is then turned on at one of the three settings being tested. After the first 30 seconds, the timing will begin and the water level in the burette will be read and recorded. Once the water level reaches the 5-mL mark the timer will be stopped. The ratio of the volume drawn (in mL) divided by the draw time (min) will constitute the flow rate in mL/min. This test will be repeated in duplicates for each of the three test settings. The temperature of the test water should be measured and recorded in each test. The datasheet to

DATA LOGSHEET

TASK 1 - RAW WATER FLOWMETER CALIBRATION

	Comment										
Calculated Flowrate	gpm		,								
Time to	Fill, min										
Fill Volume	gallons										
Indicated Flowrate	gpm										
Target Flowrate	gpm	10	10	10	18	18	18	25	25	25	
	Time										
	Date										

Note: Tests to be conducted in triplicate at indicated flowrates of 10, 18 and 25 gpm.

FIGURE 3

be used in this test is shown in Figure 4. Once the calibration curve is developed, the "actual" flow rate should be used in all subsequent testing.

2.1.2.3: Ozone Feed-Gas Flowmeters

The ozone gas flowmeters to the six contactors need to be calibrated during this task. A wet gas flowmeter will be leased to the project. The following procedure will be used to calibrate the feed-gas flow meter to each contactor at each of three gas flow settings (0.5, 1, and 2 scfm indicated flow).

- 1. turn on the gas to the test contactor at one of the three test flowrate settings,
- 2. connect the wet gas flowmeter to the off-gas line from the test contactor,
- 3. measure the gas flowrate using the manufacturer's directions.
- 4. take a duplicate gas flowrate reading,
- 5. repeat the duplicate measurements at the other two indicated flowrates,
- 6. repeat the above 5 steps on each of the remaining five contactors.

Note that the ozone generator setting should be at "zero" and the feed-gas should contain no ozone. Also, no water flow through the contactor is necessary. However, it is important that each contactor be full of water to the operating water level. In addition, a pressure gauge and a temperature gauge will be installed downstream of the flowmeter to the first contactor to measure the actual gas temperature and pressure. This information is necessary to correct the gas flow for temperature and pressure. Figure 5 shows the data logsheet to be used to record the data collected from this calibration test.

2.1.3: Alarm Checks

The following alarms will be checked by the operator during this task:

- 1. Overflow alarm on the 1st contactor,
- 2. Overflow alarm on the wet well between the 6th contactor and the GAC contactors,
- 3. Spill alarm in the containment pad, and
- 4. Overflow alarm on the surge feed tank.

All alarm checks will be conducted using tap water.

2.1.3.1: Overflow Alarm on 1st Contactor

To check whether the overflow alarm on the 1st contactor is operating properly, the valve between the 1st and 2nd contactor will be closed off and tap water will be turned on at 25 gpm flowrate into the 1st contactor. The contactor will fillup until the water reaches the level sensor. At that time, the alarm should shut down the entire power system. This includes the extraction well pumps, influent water pump, transfer pump in the wet well, the ozone generator, and the hydrogen peroxide pumps.

DATA LOGSHEET

TASK 1 - CHEMICAL FEED PUMP CALIBRATION

				Burrette	Burrette	Draw	Draw	Measure	đ
				builette	Dullette	We leave a	Time	Flormote	
			rump	cart Mar	nd Mar	volume	1111111	Flowrate	
Date	Time	ump No	Setting	mL	mL	mL	min.	nL/min	Comment
	-								
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		-				-			
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					ļ				

DATA LOGSHEET

TASK 1 - OZONE GAS FLOWMETER CALIBRATION

		Ozone	Gas Flow	Measured	Gas	Gas	
		Contactor				1	
Date	Time	Number	(scfm)	(scfm)	(atm)	(°C)	Comment
Date	Time	1	0.5	(00222)	(4.0444)	()	
		1	0.5				
		1	1				
		1	1		-		
		1	2				
		1	2				
			 				
		2	0.5				
		2	0.5				
		2	1				
		2	1				
	,	2	2				
		2	2				
		3	0.5				
		3	0.5				
		3	1				
		3	1				
		3	2				
		3	2				
		4	0.5				
		4	0.5				
		4	1				
		4	11				
		4	2				
		4	2				
		5	0.5				
		5	0.5				
		5	1		-		-
		5	1				
		5	2				
		5	2				
			0.5				
		6	0.5				
		6	0.5				
		6	1				
		6	1				
		6	2				
		6	2				
						L	

2.1.3.2: Overflow Alarm on the Wet Well

The same type of test will be conducted on the wet well. The transfer pump between the wet well and the GAC contactor will be turned off. The wet well will be filled with tap water. As the wet well fills up and the water reaches the level sensor, the alarm system should shut down the entire power system. This should include the extraction well pumps, raw water pump, the ozone generator, and the hydrogen peroxide feed pumps.

2.1.3.3: Spill Alarm on Containment Pad

The spill alarm in the containment pad will be checked. Using a flexible hose, tap water will be diverted into the sump of the containment pad. As the pad fills up and the water reaches the level sensor, the alarm should shut down the power system. This should include the extraction pump wells, raw water pump, the transfer pump in the wet well, the ozone generator, and the hydrogen peroxide feed pumps.

2.1.3.4: Overflow Alarm on Surge Tank

Testing of the overflow alarm on the surge tank is similar to that of the alarm on the 1st contactor or the wet well. The pump from the surge tank to the first contactor will be turned off, and the tank will be filled with tap water. As the surge tank fills up and the water reaches the level sensor, the alarm system should shut down the entire power system. This should include the extraction well pumps, raw water pump, the ozone generator, and the hydrogen peroxide feed pumps.

2.2: TASK 2—CONDUCT PROCESS OPTIMIZATION

After completing system startup and instrument calibration, the 1-week process optimization task will begin. The objective of this task is to evaluate contaminant removal under a wide range of ozone dose in order to select the optimum set of conditions for system demonstration. Figure 6 shows the datalogsheet that will be used during each process optimization test. It is noted that all testing will be conducted in a countercurrent flow mode.

This task includes a total of 8 tests to be conducted over a one-week period. The concept behind this task is to run the PEROXONE plant at four applied ozone doses ranging from 30 mg/L to 75 mg/L, which extend well below and above the anticipated required dose of 60 mg/L. The applied doses to be evaluated are 30 mg/L, 45 mg/L, 60 mg/L, and 75 mg/L. All doses, except the 75 mg/L dose, will be evaluated at a hydraulic flowrate of 25 gpm. Due to the limitation of the ozone generator capacity, the flowrate will have to be reduced to 15 gpm in order to evaluate the system performance at the applied dose of 75 mg/L. These tests will be conducted on each of the two wells to be evaluated. All applied and transferred ozone doses will be accurately measured by monitoring the ozone concentration in the feed gas and the off gas streams to and from each of the six contactors. In addition, the hydrogen peroxide stock solution will be prepared at 2% strenght (20,000 mg/L) by diluting the neat 35%

PEROXONE GROUNDWATER TREATMENT PROJECT DATA LOGSHEETS

TASK 2 - PROCESS OPTIMIZATION TASK

CONTACTOR OFF-

GAS OZONE CONC TIME CONTACTOR (mg/L) TEST CONDITIONS: Cl DATE: C2 TEST NO .: СЗ WELL CODE: C4 WATER FLOW RATE (gpm): C5 FEED-GAS OZONE CONC (mg/L): C6 H2O2 FEED TANK CONC (mg/L): CONTAMINANT OZONE OZONE SAMPLE SAMPLE RESIDUAL RESIDUAL 2 SAMPLE TIME LOCATION CODE CODE (MG/L) OPTIONAL (24HR) (MG/L) 02--INF-INF -INF-02-INF -INF-INF 02--INF-INF 02--C1/0-C1/0 02--C1/UF-C1/UF 02--C1/2-C1/2 02--C1/4-C1/4 02--C1/6-C1/6 02--C1/8-C1/8 02--C2/0-C2/0 02--C2/UF-C2/UF 02--C3/0-C3/0 02--C3/UF-C3/UF 02--C4/0-C4/0 -C4/UF-C4/UF 02-02--C5/0-C5/0 02--C5/UF-C5/UF 02--C6/0-C6/0 02--C6/UF-C6/UF C6/0UQ 02--C6/0UQ-02--C6/UFUQ C6/UFUQ GAC3 02--GAC3-02--GAC1-GAC1 -GAC2-GAC2

Note: "UQ" following sample location designator indicates explosives sample not quenched with Thiosulfate

FIGURE 6

solution with DI water. All throughout these eight tests, the gas flowrate to each contactor will be set at 0.8 scfm.

The following is a detailed description of how each of the eight tests will be conducted.

2.2.1: Test #1 (Week #2; Monday): Conditions: Well A; Flow Rate = 25 gpm; Ozone Dose = 30 mg/L per contactor; H₂O₂ Dose = 9 mg/L per contactor.

At 8:00 AM on Monday morning, set the flow rate through the system at 25 gpm with the ozone generator set to produce an applied dose of 30 mg/L, and the hydrogen peroxide flowrate to each contactor set at 42.3 mL/min which translates into a hydrogen peroxide dose of 9 mg/L to each contactor. At 8:30 AM the concentration of ozone in the feed gas will be measured. If the applied ozone dose is not 30 mg/L ± 3 mg/L, then the ozone generator will be adjusted and rechecked after period of 15 minutes. This process will be repeated until the dose is within this acceptable range. For a water flowrate of 25 gpm, and a gas flowrate of 0.8 scfm, a ozone gas-phase concentration of 125 mg/L will result in the target applied ozone dose of 30 mg/L to each contactor.

Assuming that each contactor is completely mixed, then the six contactors in series can be simulated by six completely stirred tank reactors (CSTRs) in series. Therefore, steady-state conditions are expected to be reached in 4 hours of operating time. While waiting for steady-state conditions to be reached, 4 influent samples will be collected from the surge tank and analyzed for explosives concentrations.

After steady-state conditions are reached, samples will be collected from the effluent of each of the six contactors, as well as from five taps along the water depth of the first contactor. The samples will then be analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors will then be analyzed for ozone gasphase concentration. It is anticipated that the off-gas ozone concentrations will be different between the six contactors, and therefore it is important that they be measured individually.

2.2.2: Test #2 (Week #2; Monday): Conditions: Well A; Flow Rate = 25 gpm; Ozone Dose = 45 mg/L per contactor; H₂O₂ Dose = 13.5 mg/L per contactor

After all the samples from test #1 are taken (approximately 5:00 PM), the hydrogen peroxide flowrate to each contactor will be set at 64 mg/L, which translates into a hydrogen peroxide dose of 13.5 mg/L. The ozone generator will be set to produce 188 mg/L ozone in the gasphase, which translates into an applied ozone dose of 45 mg/L to each contactor (water flowrate = 25 gpm, and gas flowrate = 0.8 scfm to each contactor). After 30 minutes, the ozone concentration in the feed-gas will be analyzed to confirm that the target dose is achieved. Adjustments, if necessary, will be made to the ozone generator setting, and the ozone gas-phase concentration will be rechecked until the target ozone dose is reached. The system will then be left to run overnight.

At 8:00 AM on Tuesday morning, the ozone generator setting, the hydrogen peroxide feed rate, and the water flow rate will be checked and recorded to ensure that they have not changed overnight. Water samples will then be collected from the effluent of each contactor, and from five taps along the water depth of the first contactor, and analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors should analyzed for ozone gas-phase concentration to record the applied and transferred ozone dose.

The samples should be collected before 10:00 AM on Tuesday, and the system will then be setup for Test #3.

2.2.3: Test #3 (Week #2; Tuesday): Conditions: Well A; Flow Rate = 25 gpm; Ozone Dose = 60 mg/L per contactor; H₂O₂ Dose = 18 mg/L per contactor

After the samples from Test #2 are taken (10:00 AM), the ozone gas-phase concentration will be set at 250 mg/L \pm 25 mg/L (which translates into an applied ozone dose of 60 mg/L \pm 6 mg/L), and the hydrogen peroxide flowrate set at 85 mL/min, which translates into a hydorgen peroxide dose of 18 mg/L. At 10:30 AM, the concentration of ozone in the feed gas will be measured. If the applied ozone dose is not 60 mg/L \pm 6 mg/L, then the ozone generator will be adjusted and rechecked after period of 15 minutes. This process will be repeated until the dose is within this acceptable range. The system is then allowed to reach steady-state conditions. This should be reached within four hours. While waiting for steady-state conditions to be reached, 4 influent samples will be collected from the surge tank and analyzed for explosives concentrations.

After steady-state conditions are reached, samples will be collected from the effluent of each of the six contactors, as well as from five taps along the water depth of the first contactor. The samples will then be analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors will then be analyzed for ozone gas-phase concentration. It is anticipated that the off-gas ozone concentrations will be different between the six contactors, and therefore it is important that they be measured individually.

2.2.4: Test #4 (Week #2; Tuesday): Conditions: Well A; Flow Rate = 15 gpm; Ozone Dose = 75 mg/L per contactor; H₂O₂ Dose = 22.5 mg/L per contactor

After all the samples from test #3 are taken (approximately 5:00 PM), the hydrogen peroxide flowrate to each contactor will be set at 64 mg/L, which translates into a hydrogen peroxide dose of 22.5 mg/L. The ozone generator will be set to produce 188 mg/L ozone in the gasphase, which translates into an applied ozone dose of 75 mg/L to each contactor (water flowrate = 15 gpm, and gas flowrate = 0.8 scfm to each contactor). After 30 minutes, the ozone concentration in the feed-gas will be analyzed to confirm that the target dose is achieved. Adjustments, if necessary, will be made to the ozone generator setting, and the ozone gas-phase concentration will be rechecked until the target ozone dose is reached. The system will then be left to run overnight. It is noted that a total of 7 hours of operation time is required before steady-state conditions are reached for a flowrate of 15 gpm.

At 8:00 AM on Wednesday morning, the ozone generator setting, the hydrogen peroxide feed rate, and the water flow rate will be checked and recorded to ensure that they have not changed overnight. Water samples will then be collected from the effluent of each contactor, and from five taps along the water depth of the first contactor, and analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors should analyzed for ozone gas-phase concentration to record the applied and transferred ozone dose.

The samples should be collected before 10:00 AM on Wednesday, and the system will then be setup for Test #5.

2.2.5: Test #5 (Week #2; Wednesday): Conditions: Well B; Flow Rate = 25 gpm; Ozone Dose = 30 mg/L per contactor; H₂O₂ Dose = 9 mg/L per contactor

After Test #4 samples were taken (before 10:00 AM), the feed water will be switched to well B, and the flow rate through the system will be set at 25 gpm. The ozone generator will then be set to produce an applied dose of 30 mg/L, and the hydrogen peroxide flowrate to each contactor set at 42.3 mL/min which translates into a hydrogen peroxide dose of 9 mg/L to each contactor. At 10:30 AM the concentration of ozone in the feed gas will be measured. If the applied ozone dose is not 30 mg/L ± 3 mg/L, then the ozone generator will be adjusted and rechecked after period of 15 minutes. This process will be repeated until the dose is within this acceptable range. For a water flowrate of 25 gpm, and a gas flowrate of 0.8 scfm, a ozone gas-phase concentration of 125 mg/L will result in the target applied ozone dose of 30 mg/L to each contactor.

Steady-state conditions are expected to be reached in 4 hours of operating time. While waiting for steady-state conditions to be reached, 4 influent samples will be collected from the surge tank and analyzed for explosives concentrations.

After steady-state conditions are reached, samples will be collected from the effluent of each of the six contactors, as well as from five taps along the water depth of the first contactor. The samples will then be analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors will then be analyzed for ozone gasphase concentration. It is anticipated that the off-gas ozone concentrations will be different between the six contactors, and therefore it is important that they be measured individually.

2.2.6: Test #6 (Week #2; Wednesday): Conditions: Well B; Flow Rate = 25 gpm; Ozone Dose = 45 mg/L per contactor; H,O, Dose = 13.5 mg/L per contactor

After all the samples from test #5 are taken (approximately 5:00 PM), the hydrogen peroxide flowrate to each contactor will be set at 64 mg/L, which translates into a hydrogen peroxide dose of 13.5 mg/L. The ozone generator will be set to produce 188 mg/L ozone in the gasphase, which translates into an applied ozone dose of 45 mg/L to each contactor (water flowrate = 25 gpm, and gas flowrate = 0.8 scfm to each contactor). After 30 minutes, the

ozone concentration in the feed-gas will be analyzed to confirm that the target dose is achieved. Adjustments, if necessary, will be made to the ozone generator setting, and the ozone gas-phase concentration will be rechecked until the target ozone dose is reached. The system will then be left to run overnight.

At 8:00 AM on Tuesday morning, the ozone generator setting, the hydrogen peroxide feed rate, and the water flow rate will be checked and recorded to ensure that they have not changed overnight. Water samples will then be collected from the effluent of each contactor, and from five taps along the water depth of the first contactor, and analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors should analyzed for ozone gas-phase concentration to record the applied and transferred ozone dose.

The samples should be collected before 10:00 AM on Thursday, and the system will then be setup for Test #7.

2.2.7: Test #7 (Week #2; Thursday): Conditions: Well B; Flow Rate = 25 gpm; Ozone Dose = 60 mg/L per contactor; H,O, Dose = 18 mg/L per contactor

After the samples from Test #6 are taken (10:00 AM), the ozone gas-phase concentration will be set at 250 mg/L \pm 25 mg/L (which translates into an applied ozone dose of 60 mg/L \pm 6 mg/L), and the hydrogen peroxide flowrate set at 85 mL/min, which translates into a hydorgen peroxide dose of 18 mg/L. At 10:30 AM, the concentration of ozone in the feed gas will be measured. If the applied ozone dose is not 60 mg/L \pm 6 mg/L, then the ozone generator will be adjusted and rechecked after period of 15 minutes. This process will be repeated until the dose is within this acceptable range. The system is then allowed to reach steady-state conditions. This should be reached within four hours. While waiting for steady-state conditions to be reached, 4 influent samples will be collected from the surge tank and analyzed for explosives concentrations.

After steady-state conditions are reached, samples will be collected from the effluent of each of the six contactors, as well as from five taps along the water depth of the first contactor. The samples will then be analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors will then be analyzed for ozone gasphase concentration. It is anticipated that the off-gas ozone concentrations will be different between the six contactors, and therefore it is important that they be measured individually.

2.2.8: Test #8 (Week #2; Thursday): Conditions: Well B; Flow Rate = 15 gpm; Ozone Dose = 75 mg/L per contactor; H,O, Dose = 22.5 mg/L per contactor

After all the samples from test #7 are taken (approximately 5:00 PM), the hydrogen peroxide flowrate to each contactor will be set at 64 mg/L, which translates into a hydrogen peroxide dose of 22.5 mg/L. The ozone generator will be set to produce 188 mg/L ozone in the gasphase, which translates into an applied ozone dose of 75 mg/L to each contactor (water flowrate = 15 gpm, and gas flowrate = 0.8 scfm to each contactor). After 30 minutes, the

ozone concentration in the feed-gas will be analyzed to confirm that the target dose is achieved. Adjustments, if necessary, will be made to the ozone generator setting, and the ozone gas-phase concentration will be rechecked until the target ozone dose is reached. The system will then be left to run overnight. It is noted that a total of 7 hours of operation time is required before steady-state conditions are reached for a flowrate of 15 gpm.

At 8:00 AM on Friday morning, the ozone generator setting, the hydrogen peroxide feed rate, and the water flow rate will be checked and recorded to ensure that they have not changed overnight. Water samples will then be collected from the effluent of each contactor, and from five taps along the water depth of the first contactor, and analyzed for ozone residual and explosives concentrations. The feed-gas and the off-gas from each of the six contactors should analyzed for ozone gas-phase concentration to record the applied and transferred ozone dose. After all samples are taken, the system will be shut down.

Based on the above discussion, a total of 104 explosives samples will be collected during the Process Optimization task. Table 2 shows a breakdown of the explosives sampling schedule during this task.

Table 2

Explosives Sampling Schedule During Process Optimization Testing

Test #	Day	Influent	Contactor Effluent	Taps Along 1st Contactor Depth	Total
1	Mon.	4	6	5	15
2	Mon./Tues.		6	5	11
3	Tues.	4	6	5	15
4	Tues./Wed.		6	5	11
5	Wed.	4	6	5	15
6	Wed./Thurs.	_	6	5	11
7	Thurs.	4	6	5	15
8	Thurs./Fri.		6	5	11
TOTAL		16	48	40	104

Important Note: The schedule for conducting Task 2 (Process Optimization) is very compact, and does not allow for any interruptions to the plant operation. Unfortunately, it is Montgomery Watson's experience with similar projects that interruptions can occur. In order to keep the project on schedule, if interruptions do take place, optimization will be conducted on one well only. However, both wells will still be evaluated in Task 3 (System Demonstration).

2.3: TASK 3—CONDUCT SYSTEM DEMONSTRATION

During this 10-week task, the plant will be operated under constant ozone and hydrogen peroxide doses and water flow rate for a period of three weeks with each of the three wells. The exact operating conditions, such as ozone dose, hydrogen peroxide dose, and water flowrate, will be determined based on the results of Task 2 (Process Optimization Task). The selected operating conditions will be those that result in the removal of the contaminants to their corresponding target finished water levels at the lowest possible treatment cost. The plant will be operated five days a week, 24 hours per day, but will be attended for only 8 hours/day.

In addition, two tracer tests will be conducted at different days during this task. The objective of the tracer tests is to characterize the hydraulic behavior of the system and assess the degree of mixing taking place in the contactors. This information will be used to determine whether packing material will be necessary in the 1000-gpm full-scale plant.

2.3.1: Plant Demonstration

During the plant operation, the sampling schedule detailed in Table 3 will be implemented during each 5-week period for each of the two wells to be tested. It is noted that ORP stands for Oxidation Reduction Potential which will be measured online using ORP probes. The analytical results obtained from this task will be logged into the data logsheet shown in Figures 7.a and 7.b for countercurrent (downflow) and co-current (upflow) operation, respectively. However, it is noted that the plant will be operated in the countercurrent mode, unless otherwise decided during the first project progress meeting.

DATA LOGSHEETS

TASK 3 - SYSTEM DEMONSTRATION

PRED GAS OZONE CONO. (mg/L)	TEST CONDITIONS:	TARGET	TARGET APPLIED OZONE (mg/L):		Γ	TARGE	HYDROGEN PEROXIDE:				
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Part Part	DOWNFLO		GAS FLOW RATE (scfm):			Н2О2 ГЕБО	TANK CONC (mg/L):			Verification Time	
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DATA LOGSHEETS

TASK 3 - SYSTEM DEMONSTRATION

TEST CONDITIONS:			OZONE:			HYDR	HYDROGEN PEROXIDE:	•••			
DATE	L	ARGET APPLI	TARGET APPLIED OZONE (mg/L):			TARGET	TARGET H2O2 DOSE (mg/L):	L			
WELL CODE:	ír.	EED-GAS OZ	FEED-GAS OZONE CONC (mg/L):		H,0	FEED FLO	H2O2 FEED FLOW RATE (mL/min):	Ļ			
TER FLOW RATE (gpm):		FEED-GAS	FEED-GAS FLOW RATE (scfm):		H	O2 FEED T	H2O2 FEED TANK CONC (mg/L):			Verification Time	
CONTACTOR FLOW (UP/I	UPFLOW										
SAMPLE	SAMPLE	OZONE	OZONE RESIDUAL 2	Hd	ORP	OZONE OFF-GAS	TRANSFERRED	H ₂ O ₂	H ₂ O ₂	CONTAMINANTS	MISC.
CODE		(1/ pus)	(optional)			CONC	DOSE		(1/)	CODE	CODE
		(a /9)	(2 (3)	2/D	2/D	(m/2) m	(2 /Sm)	(min)	(mg/ r/	D (2/W INF, W	
		4/D		(D INF) (D INF)	D INF)	Z/D	Z/D	Q	۵	GAC1,2)	A
INF						Section Con-				03-00-INF-	03-00-INF-
C1/UF										03-00-C1/UF-	
C1/UF										03-00-C1/UF-	
C1/UF										03-00-C1/UF-	
C1/UF										03-00-C1/UF-	Supplied of the Contra
C2/UF										03-00-C2/UF-	A Company of
C2/UF										03-00-C2/UF-	
C2/UF										03-00-C2/UF-	
C2/UF										03-00-C2/UF-	Squitte Company
C3/UF										03-00-C3/UF-	
C3/UF										03-00-C3/UF-	
C3/UF										03-00-C3/UF-	
C3/UF										03-00-C3/UF-	
C4/UF										03-00-C4/UF-	18 8/38 A
C4/UF										03-00-C4/UF-	A 188 188 188 188 188 188 188 188 188 18
C4/UF										03-00-C4/UF-	A CONTRACTOR
C4/UF										03-00-C4/UF-	
C5/UF										03-00-C5/UF-	AND 2018
C5/UF										03-00-C5/UF-	3 1822 9 2 2
C5/UF										03-00-C5/UF-	Section of the second
C5/UF										03-00-C5/UF-	
C6/UF										03-00-C6/UF-	03-00-C6/UF-
C6/UF										03-00-C6/UF-	03-00-C6/UF
C6/UF											03-00-C6/UF
C6/UF										03-00-C6/UF-	03-00-C6/UF
C6/UFUG										03-00-C6/UFUG-	
C6/UFUQ										03-00-C6/UFUQ-	

GAC2

Note: Form of Sample Location ID for Task 3 is location code followed by "W" and well identifier, e.g. C1/0W1.

03-00-GAC3-03-00-GAC3-03-00-GAC3-03-00-GAC3-

03-00-GAC3-03-00-GAC3-03-00-GAC3-03-00-GAC3-03-00-GAC1-03-00-GAC2-

03-00-C6/UFUG-

CG/UFUG CG/UFUG

GAC3 GAC3 GAC3 GAC3

03-00-CG/UFUG-

Table 3
Sampling Schedule During Task 3 (System Demonstration)

Sampl	ing Location			Analyte		
ID	Description	Explosives	O, Res.	pH/ORP	Doses	Misc.
INF— <well#></well#>	Influent to Treatment System	2/W		D	_	\mathbf{W}
	Effluent of 1st Contactor	D	4/D	2/D	2/D	
C2— <well#></well#>	Effluent of 2nd Contactor	D	4/D	2/D	2/D	
C3— <well#></well#>	Effluent of 3rd Contactor	D	4/D	2/D	2/D	
C4— <well#></well#>	Effluent of 4th Contactor	D	4/D	2/D	2/D	
C5- <well#></well#>	Effluent of 5th Contactor	D	4/D	2/D	2/D	
C6— <well#></well#>	Effluent of 6th Contactor	D	4/D	2/D	2/D	\mathbf{W}
GAC1— <well#></well#>	Effluent of 1st GAC Contactor	W	D	D		
GAC2- <well#></well#>	Effluent of 2nd GAC Contactor	W	D	D		
GAC3— <well#></well#>	Effluent of 3rd GAC Contactor	D	D	D		W

D = Sample collected once per day

W = Sample collected once per week

#/D = Sample collected # times per day

#/W = Sample collected # times per week

Misc.: General Mineral which includes TOC, turbidity, alkalinity, hardness, TDS, calcium, magnesium, iron, & manganese.

Doses: Includes transferred ozone dose and hydrogen peroxide dose to each contactor.

Note: The well number notation on each sample identifies the groundwater source that was being tested when

the sample was taken.

2.3.2: Tracer Testing

Both tracer tests will be conducted on the first contactor only. Water should be running for at least three (3) detention times at the test flow rate BEFORE the tracer is injected into the influent line. Tracer tests will be conducted using sodium fluoride, and will use the "pulse" or "slug" feed method. In other words, a pre-calculated mass of the tracer will be injected at time "zero" into the influent line to the first contactor. The injection period should be less than 10 seconds. Samples will then be taken at 5-minute intervals from the effluent of the first contactor, as well as from five taps along the depth of the contactor. The samples are then analyzed for fluoride concentration using a fluoride-selective probe. During this period, influent water samples should be collected from the raw water surge tank every 15 minutes and analyzed for background fluoride concentration. The analytical results obtained from the tracer tests will be logged into the data logsheet shown in Figure 8. The test will be conducted at two different days during the 10-week System Demonstration task.

It is anticipated that the fluoride concentration at the effluent of the 1st contactor will reach background levels within the 3-HRT sampling period. If the effluent tracer concentration is still higher than background after the 3-HRT sampling period, sampling should be continued for an additional hour.

DATA LOGSHEETS

TRACER TESTING

TEST CONDITIONS:		_				
DATE:					CALIBRATION DA	
WELL CODE:]			Fluoride Conc. (m	Probe Output (m\
TER FLOW RATE (gpm):		1				
9		•				
CK VOLUME INJECTED:		ī				
OCK CONCENTRATION:		1				
TRACER DOSE (mg):		†				
		1				
ACER INJECTION TIME:		J			Calibration Result	e
						NTERCEPT
					SLUPE	NIERCEFI
					r² =	
		FLUORIDE	FLUORIDE	INF. SAMPLE	FLUORIDE	FLUORIDE
				(Sample Every		
Complete Village	Ti	DEADOITE (m)	(met/L)	15 Minutes)	READOUT (mV)	(mg/L)
Sample ID/Location	Time	READOUT (mV)	(mg/L)	15 Minutes)	TG/IDOOT (IIIV)	(11.5/ 2)
					 	
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1		1		1		化二氯甲基二酚 经基础

2.4: ANALYTICAL METHODS

Five analyses will be conducted by the Montgomery Watson operators: ozone residual, pH, ORP, hydrogen peroxide, and fluoride tracer concentration measurements. All other analyses will be conducted by the independent evaluator using an on-site laboratory. Those analyses include all target contaminants, and general mineral constituents of the water including TOC, turbidity, alkalinity, hardness, calcium, magnesium, TDS, iron, and manganese.

A HACH CEL/700 portable analyzer will be used to measure the ozone residual concentration, the pH of the water, and fluoride concentration. A bench-top ORION meter with an ORP probe will be used to measure the ORP of the water samples collected during plant operation. The hydrogen peroxide concentration will be measured using the Cobalt method.

2.5: H,O, FEED STOCK SOLUTION PREPARATION

A day tank will be used as the feed reservoir for hydrogen peroxide. The target concentration of hydrogen peroxide in the day tank is 2%, or 20,000 mg/L. This solution will be prepared from a commercial hydrogen peroxide stock solution, which has an approximate concentration of 35%, or 350,000 mg/L. The hydrogen peroxide will be diluted from 35% to 2% using deionized water. A deionized water system will be provided with the demonstration plant. The DI system should use tap water as its influent water source. To determine the exact dilution ratio to get the target 2% concentration, the exact stock solution concentration should first be measured. This is done by creating several dilutions of the stock solution and measuring their concentrations using the Cobalt method. After the day tank solution is prepared, its exact H_2O_2 concentration should be measured and recorded. The H_2O_2 flowrate to each contactor will then be adjusted to provide the target dose based on the feed tank concentration.

The stability of the H_2O_2 stock solution and the day tank feed solution is of concern. Therefore, the following measures should be implemented at all times:

- 1. The H₂O₂ stock solution should be stored in the dark at 4°C when not in use.
- 2. The day tank should be covered and protected from any sunlight.
- 3. The H₂O₂ concentrations in the stock solution and in the day tank should be checked evertime the day tank feed solution is prepared.
- 4. The stability of the H₂O₂ feed solution strenght in the feed day tank should be checked during a 24 hour period to ensure that the concentration of H₂O₂ does not change between preparation times. This test is done by measuring the feed solution strength at different times during the day.

2.6: SAMPLE IDENTIFICATION

A standard notation should be used for all water quality samples provided to the independent evaluator for analysis. The notation should include the following parts:

- 1. Task Number
- 2. Test Number (if applicable)
- 3. Tap Code
- 4. Time & Date

For example, a sample collected from the effluent of the second contactor (tap C2/0) during Test #3 of Task #2 (Process Optimization task) at 2:15 PM on August 17 will have the notation:

If no test number exists (e.g., during the 10-week demonstration period [Task #3]), the test number should be substituted with "00". Labels for all samples collected for immediate analysis by the operators should, at a minimum, include the tap letter-code.

To satisfy a 10% QA/QC requirement, one sample from the effluent of the 6th contactor will be collected in duplicates every week. The ID for these samples should have a notation at the end to show that they are duplicates. For example, the following identifiers are for duplicate samples collected from the effluent of the 6th contactor at 3:10 PM on September 12:

Appendix F
Project Team Contact List

USAEC Peroxone Groundwater Treatment Demonstration Study Points of Contact

US Army Environmental Center

Project Manager:

Mr. James Heffinger

U.S. Army Environmental Center

Building E4430

Aberdeen Proving Gr. MD 21010-5401

Phone: 410.612.6846 Fax: 410.612.6836

Email: jgheffin@usaec1.apgea.army.mil

Defense Evaluation Support Activity

Independent Evaluator and Project

Manager:

Ms. Lynda Liptak

DESA/TRI

2251 Wyoming Blvd. SE Kirtland AFB NM 87117

Phone: 505.262.4583

Fax: 505.260.2759

Email: liptakl@desa.osd.mil

TRW

Project Manager:

Dr. John Lamb

TRW

One Space Park O1/2030 Redondo Beach CA 90278

Phone: 310.813.9354

Fax: 310.813.9349

Email: john.lamb@trw.com

Design Support:

Mr. Loren McClanathan

TRW

O1/2080 One Space Park Redondo Beach CA 90278

Phone: 310.813.9378 Fax: 310.812.1717

Email: loren.mcclanathan@trw.com

Subcontract Manager:

Ms. Rory Brooks

TRW

One Space Park O1/2261 Redondo Beach CA 90278

Phone: 310.814.4344 Fax: 310.814.7466

Email: rory.brooks@trw.com

Montgomery Watson

Project Manager:

Mr. Thomas Metzger Montgomery Watson 1300 Walnut Street

Suite 200

Boulder CO 80302 Phone: 303.546.6671 Fax: 303.546.6726

Email: tom.metzger@us.mw.com

Technical Manager

Dr. Issam Najm Montogomery Watson 250 N. Madison Avenue Pasadena, CA 91101

Phone: 818.568.6744 Fax: 818.568.6323

Email: issam.najm@us.mw.com

System Design, Integration and Demonstration:

Mr. Vivek Agrawal Montgomery Watson 4525 S. Wasatch Blvd., 200 Salt Lake City, Utah

Phone: 801.273.2466 Fax: 801.272.0430

Email: vivek.agrawal@us.mw.com

Mr. Ning-Wu Chang Montgomery Watson 250 No. Madison Ave. Pasadena, CA 91101 Phone: 818.568.6504 Fax: 818.568.6515

Email: ning_wu_chang@smtpgw.mw.com

Mr. Keith Johnson Montgomery Watson 4525 South Wasatch Blvd., Suite 200 Salt Lake City, Utah 84124-4799

Phone: 801.273.2408 Fax: 801.272.0430

Email: keith.johnson@us.mw.com

Mr. Karl Gramith Montgomery Watson 250 N. Madison Avenue Pasadena, CA 91101 Phone: 818.568.6740

Fax: 818.568.6323

Email: karl.gramith@us.mw.com

Mr. Joseph Marcinko Montgomery Watson 250 N. Madison Avenue Pasadena, CA 91101 Phone: 818.303.5845

Fax: 818.392.3923

Email: joseph.marcinko@us.mw.com

Advisory Board

Ms. Elizabeth Fleming US Army Corps of Engineers 3909 Halls Ferry Rd. Vicksburg, MS 39180 Phone: 601.634.3943

Fax: 601.634.3833

Email: flemine@ex1.wes.army.mil

Dr. William Glaze University of North Carolina Dept. of Envr. Sci & Engr. Chapel Hill NC 27599 Phone: 919.966.1024

Fax: 919.966.2583

Email: bill_glaze@unc.edu

Mr. Lindsey Lien US Army Corps of Engineers 12565 West Center Road Omaha NE 68144-3869 Phone: 402,697,2580

Fax: 402.697.2595

Email:

Lindsey.K.Lien@mrd01.usace.army.mil

Dr. Stephen Maloney US Army Corps of Engineers

P.O. Box 9005

Champaign IL 61826-9005 Phone: 217.373.3482

Fax: 217.373.3490

Email: s-maloney@cecer.army.mil

Mr. Jerry Miller

US Army Corps of Engineers 3909 Halls Ferry Rd. Vicksburg, MS 39180

Phone: 601.634.3931 Fax: 601.634.4263

Email: millerj@ex1.wes.army.mil

Mr. Kerwin Rakness Process Applications Inc. 2627 Redwing Road Fort Collins, CO 80526

Phone: 970.223.5787 Fax: 970.223.5786

Email: 75017.725@compuserve.com

Mr. Ted Streckfuss

US Army Corps of Engineers

Attn: CEMRO-HX-T 12565 West Center Road Omaha NE 68144-3869 Phone: 402.697.2560

Fax: Email:

ted.h.streckfuss@mrO01.usace.army.mil